Synthesis, Structure and Optoelectronic Properties of Rigid 3D Tetraindeno-Fused Spirofluorene with Thioxanthene or dioxothioxanthene Substitutes

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Table of Contents

General experimental methods	2
Experimental Section	2
TGA curves	4
Absorption and emission spectra	4
TD-DFT calculations	5
HR-Mass spectra	
¹ H–NMR and ¹³ C–NMR Spectra	11
Single crystal structure	14

General experimental methods

¹H NMR and ¹³C NMR spectra were recorded in C₂D₂Cl₄ or CDCl₃ using a Bruker DPX 400 or 500 spectrometer. High-resolution MALDI mass spectrometry measurements were performed on a Solarix ESI-/MALDI-ICR (9.4 T) system. UVvis absorption spectra were recorded using a Shimadu and SPECORD S600 spectrophotometer. Photoluminescence performance was investigated by a steadystate fluorescence spectrometer FLS920 at room temperature with the scanning range between 200 nm to 800 nm. CV measurements were carried out on a computercontrolled GSTAT12 in a three-electrode cell in a DCM solution of Bu₄NPF₆ (0.1 M), using glassy carbon discs as the working electrode, Pt wire as the counter electrode, Ag/AgCl electrode as the reference electrode. Thermogravimetric analysis (TGA) was carried out on Shanghai Tian Mei Balance (Ltd ZRY-2P) at a heating rate of 10 °C /min under nitrogen flow. Density functional theory (DFT) calculations were performed using the Gaussian 09 program, with the B3LYP hybrid functional and basis set 6-31G(d) for the ground-state geometry optimization. All reagents and starting materials were obtained from commercial suppliers and used without further purification. All reported yields are isolated yields.

Experimental Section

Spiro-4S: (2-Bromophenyl)(phenyl)sulfane (2.64 g, 10 mmol) was dissolved in 40 mL dry THF, which was dropwised to a flask equipped with iodine (10 mg, 0.04 mmol) and magnesium (0.29 g, 12 mmol) at the beginning to initiate the Grignard reaction at argon atmosphere. Then, the mixture was heated to 80 °C for 5 h. Thereafter, **Spiro-4O** (1.9 g, 2 mmol) was added to the mixture all at once and refluxed 12 h. After cooling to room temperature, the reaction was quenched by 1M HCl (50 mL) and then extracted with DCM (3×50 mL). The combined organic solvent was removed by rotary evaporator to obtain tertiary alcohol intermediate

Spiro-4OH, which was used directly in the next step without purification. The resulting **Spiro-4OH** was dissolved in 100 mL acetic acid and refluxed overnight. Then the acetic acid was removed by distillation under reduced pressure. The residue was subjected to column chromatography to obtain desired pure compound as white solid (2.6 g, 80%). ¹H NMR (500 MHz, $C_2D_2Cl_4$) δ 8.17 (s, 4H), 7.49 (d, *J* = 7.3 Hz, 8H), 7.44 (s, 4H), 7.22 (d, *J* = 7.3 Hz, 4H), 7.19 (t, *J* = 8.0 Hz, 8H), 7.12 (d, *J* = 8.1 Hz, 4H), 6.99 (s, 4H), 6.90 (t, *J* = 8.0 Hz, 8H), 6.58 (d, *J* = 7.6 Hz, 8H), 1.22 (s, 36H). ¹³C NMR (126 MHz, $C_2D_2Cl_4$): δ 155.91, 151.47, 150.17, 149.36, 141.53, 140.70, 138.68, 138.09, 131.35, 128.73, 127.25, 126.21, 126.50, 126.07, 125.85, 116.85, 116.44, 115.60, 59.85, 34.88, 31.68.

HR-MS (MALDI) m/z calculated for $C_{117}H_{89}S_4$

[M⁺] 1621.5847. Found 1621.5859. Crystallographic data (CCDC deposition number): CCDC 1564493.

Spiro-4SO2:

Spiro-4S (1.62 g, 1.0 mmol) was dissolved in CH₂Cl₂ (250 mL) and metachloroperbenzoic acid (1.73 g, 10 mmol) was added slowly. The mixture was stirred overnight. Then, the organic layer was washed with saturated NaHCO₃ solution (3×75 mL) and concentrated under reduced pressure. The crude product was purified by column chromatography to obtain target compound **Spiro-4SO2** with 30% yield. ¹H NMR (400 MHz, CDCl₃) δ 8.31 (d, *J* = 7.8 Hz, 8H), 7.89 (s, 2H), 7.57 – 7.50 (m, 12H), 7.36 – 7.26 (m, 10H), 7.25 – 7.19 (m, 8H), 7.17 (d, *J* = 8.2 Hz, 4H), 6.70 (dd, *J* = 8.0, 3.2 Hz, 8H), 1.28 (s, 36H). ¹³C NMR (100 MHz, C₂D₂Cl₄): δ 152.64, 152.21, 150.13, 149.70, 142.21, 141.14, 140.53, 140.24, 136.78, 133.12, 129.69, 128.49, 127.09, 125.26, 123.55, 117.78, 117.04, 116.60, 57.71, 35.02, 31.68.

HR-MS (MALDI) m/z calculated for $C_{117}H_{88}O_8NaS_4$

[M⁺] 1771.5260. Found 1771.5273. Crystallographic data (CCDC deposition number): CCDC 1564494.

TGA curves



Figure S1. TGA curve for Spiro-4S and Spiro-4SO2 measured under a nitrogen atmosphere at a heating rate of 10 °C/min.

Absorption and emission spectra



Figure S2. Absorption spectra of Spiro-4S film (black line) and Spiro-4SO2 film (red line).



Figure S3. Photoluminescence spectra of Spiro-4S film (black line) and Spiro-4SO2 film (red line) ($\lambda_{Exc} = 337$ nm).

TD-DFT calculations

	E_{g} (eV)	E _g (nm)	f	transition	coefficient	transition	coefficient
S_1	3.3105	374.52	0.3177	HOMO-1 -> LUMO+1	-0.12349	HOMO -> LUMO	0.69126
S_2	3.3119	374.36	0.3166	HOMO-1 -> LUMO	0.12800	HOMO -> LUMO+1	0.69044
S_3	3.4712	357.18	0.7812	HOMO-1 -> LUMO	0.68838	HOMO -> LUMO+1	-0.11914
S_4	3.4720	357.10	0.7640	HOMO-1 -> LUMO+1	0.68917	HOMO -> LUMO	-0.11456
				HOMO-5 -> LUMO	-0.10653	HOMO-5 -> LUMO+1	-0.15771
S_5	3.5449	349.76	0.0031	HOMO-4 -> LUMO	0.47567	HOMO-4 -> LUMO+1	-0.38504
				HOMO-3 -> LUMO	0.20193	HOMO-3 -> LUMO+1	-0.16611
				HOMO-5 -> LUMO	0.37411	HOMO-5 -> LUMO+1	-0.15771
S_6	3.5449	349.76	0.0041	HOMO-4 -> LUMO	0.47567	HOMO-4 -> LUMO+1	0.38504
				HOMO-2 -> LUMO	0.21190	HOMO-2 -> LUMO+1	0.25693
ç	3 5462	240.62	.62 0.0007	HOMO-5 -> LUMO	-0.22454	HOMO-5 -> LUMO+1	-0.26926
3 7	S ₇ 3.5462 34	349.02		HOMO-2 -> LUMO	0.38863	HOMO-2 -> LUMO+1	0.46660
S 25460 240.55	240 55	55 0.0014	HOMO-4 -> LUMO	-0.21301	HOMO-4 -> LUMO+1	-0.17509	
38	5.5409	549.55 0.0014	0.0014	HOMO-3 -> LUMO	-0.49957	HOMO-3 -> LUMO+1	-0.40775
S ₉	3.7180	333.47	0.0000	HOMO-2 -> LUMO	0.54324	HOMO-2 -> LUMO+1	-0.45241
S ₁₀	3.7234	332.99	0.0000	HOMO-3 -> LUMO	0.44693	HOMO-3 -> LUMO+1	0.54505
т	T ₁ 2.4435 507.41 (0.0000	HOMO-1 -> LUMO	0.17199	HOMO-1 -> LUMO+1	-0.39133	
11		307.41	0.0000	HOMO -> LUMO	0.47352	HOMO-1 -> LUMO+1	-0.10830
т	2 4 4 4 1	2.4441 507.28 0.4	507.28 0.0000	HOMO-1 -> LUMO	-0.39260	HOMO-1 -> LUMO+1	-0.17229
12	1 ₂ 2.4441			HOMO -> LUMO	0.10823	HOMO-1 -> LUMO+1	0.47232
				HOMO-1 -> LUMO+12	0.12500	HOMO -> LUMO+11	-0.15695
				HOMO -> LUMO+7	-0.25890	HOMO-1 -> LUMO+6	0.25358
T_3	3.0797	402.58	0.0000	HOMO-5 -> LUMO+1	0.11660	HOMO+7 -> LUMO+1	-0.11030
				HOMO-9 -> LUMO+1	-0.22035	HOMO-4 -> LUMO	0.19700
				HOMO-6 -> LUMO	-0.12317	HOMO-8 -> LUMO	0.20252

Table S1. Calculated energy levels, oscillator strengths (f), and orbital transition analyses of Spiro-4S.

				HOMO -> LUMO+12	-0.12501	HOMO -> LUMO+7	-0.13573
T 2.09(0 401.()			HOMO-1 -> LUMO+11	0.15804	HOMO-1 -> LUMO+7	0.23518	
	401.66		HOMO -> LUMO+6	-0.23913	HOMO-1 -> LUMO+6	-0.14456	
14	5.0808	401.00	0.0000	HOMO-4-> LUMO+1	0.17575	HOMO-6-> LUMO+1	-0.12211
				HOMO-8-> LUMO+1	0.19920	HOMO-5-> LUMO	0.12549
				HOMO-7-> LUMO	-0.11133	HOMO-9-> LUMO	-0.22235
т	T ₅ 3.3469 370.45	370.45	0.0000	HOMO-> LUMO+1	0.48402	HOMO $1 > U I MO+1$	0 12003
15		0.0000	HOMO-> LUMO	0.47658		0.12995	
Т	2 2 4 7 2 2 7 0 4 2	370 42	0.0000	HOMO-> LUMO+1	0.48402	HOMO-1-> LUMO+1	0.12993
16	5.5472	570.42	0.0000	HOMO-1-> LUMO	0.47996		
				HOMO-22-> LUMO+22	-0.12611	UOMO > UUMO + 2	0 13338
				HOMO-16-> LUMO+16	0.12541	HOMO 1 > $UIMO+3$	0.10925
Т.	3 3794	366.88	0.0000	HOMO-2-> LUMO+3	0.10145	$HOMO_{-1} > LUMO_{+3}$	0.18698
17	5.5774	500.88	0.0000	HOMO-5-> LUMO+3	0.22265	$HOMO_{-6-} > I UMO_{+3}$	0.18134
				HOMO-7-> LUMO+3	0.31151	$HOMO_20>UIMO+3$	0.12045
			HOMO-22-> LUMO+3	-0.15345	Homo 20 × Lomo (5	0.12045	
				HOMO-13-> LUMO+26	0.10209	HOMO-22-> LUMO+23	-0.11022
T ₈ 3.3799			HOMO-23-> LUMO+23	0.14539	HOMO-17-> LUMO+17	-0.13034	
		366.82	0.0000	HOMO-> LUMO+2	0.11145	HOMO-1-> LUMO+2	0.13461
	3.3799			HOMO-3-> LUMO+2	0.10226	HOMO-4-> LUMO+2	0.17166
			HOMO-5-> LUMO+2	-0.22416	HOMO-6-> LUMO+2	0.33338	
				HOMO-7-> LUMO+2	-0.16043	HOMO-14-> LUMO+2	0.12196
			HOMO-15-> LUMO+2	-0.10138	HOMO-17-> LUMO+2	-0.14614	
				HOMO-21-> LUMO+23	0.16543	HOMO-17-> LUMO+18	0 10731
			HOMO-18-> LUMO+18	0.10271	$HOMO_{-1} > I UMO+5$	0.10279	
		366.52	0.0000	HOMO-4-> LUMO+5	-0.20299	HOMO-5-> LUMO+5 HOMO-7-> LUMO+5 HOMO-9-> LUMO+5 HOMO-17-> LUMO+5	0.24414
T9	3.3828			HOMO-6-> LUMO+5	0.17506		-0.12076
				HOMO-8-> LUMO+5	0.24816		0.16816
		366.36		HOMO-14-> LUMO+5	-0.11288		-0.11834
				HOMO-18-> LUMO+5	-0.10023		
				HOMO-20-> LUMO+23	-0.11720	HOMO-20-> LUMO+22	-0.1370
				HOMO-> LUMO+5	0.10338	HOMO-4-> LUMO+5	0.19693
T ₁₀	3.3842			HOMO-5-> LUMO+5	0.23384	HOMO-7-> LUMO+5	-0.21037
				HOMO-8-> LUMO+5	-0.22916	HOMO-9-> LUMO+5	0.19038
				HOMO-14-> LUMO+5	0.11312	HOMO-16-> LUMO+5	0.10850





Excited State 1 Excited State 2 Excited State 3 Excited State 4 Energy: 3.31 Energy: 3.31 Energy: 3.47 Energy: 3.47 Wavelength: 374.52 nm Wavelength: 374.36 nm Wavelength: 357.18 nm Wavelength: 357.10 nm Oscillator strength: 0.32 Oscillator strength: 0.32 Oscillator strength: 0.78 Oscillator strength: 0.76 Configurations: Configurations: Configurations: Configurations: HOMO-1→LUMO+1 HOMO-1→LUMO HOMO-1→LUMO HOMO-1→LUMO+1 HOMO→LUMO HOMO→LUMO+1 HOMO→LUMO+1 HOMO→LUMO

Figure S4. Calculated absorption spectrum of **Spiro-4S** (up); molecular orbitals and their energy levels (in eV) of **Spiro-4S**, as well as some major excitations calculated by TD-DFT at the B3LYP6-31G(d) level.

	Eg (eV)	E _g (nm)	f	transition	coefficie nt	transition	coefficient
S_1	3.2731	378.80	0.2110	HOMO-1 -> LUMO	-0.10058	HOMO -> LUMO	0.69620
S_2	3.3260	372.77	0.4215	HOMO-1 -> LUMO+1	0.15199	HOMO -> LUMO+1	0.68587
S_3	3.4537	358.99	0.8813	HOMO-1 -> LUMO	0.69492		
S_4	3.4753	356.76	0.6292	HOMO-1 -> LUMO+1	0.68740	HOMO -> LUMO+1	-0.14767
S_5	3.7653	329.28	0.0012	HOMO-1 -> LUMO+2	-0.23373	HOMO -> LUMO+2	0.65792
S_6	3.8453	322.43	0.0010	HOMO-1 -> LUMO+3 HOMO -> LUMO+3	0.31705 0.52692	HOMO-> LUMO+6	0.28315
S ₇	3.8503	322.01	0.0026	HOMO-1 -> LUMO+3 HOMO-1 -> LUMO+4 HOMO -> LUMO+6	-0.17119 0.28926 0.29175	HOMO -> LUMO+3 HOMO -> LUMO+4	-0.17360 0.48481

Table S2. Calculated energy levels, oscillator strengths (f), and orbital transition analyses of Spiro-SO2.

				HOMO-1 -> LUMO+3	-0.15760	$UOMO \rightarrow UUMO \downarrow 4$	0 22104
				HOMO-1 -> LUMO+6	-0.13143	HOMO-1 \rightarrow LUMO+4	-0.22194
S_8	3.8593	321.26	0.0098	HOMO-> LUMO+3	-0.12863	$HOMO-1 \rightarrow LUMO+7$	0.12850
				HOMO -> LUMO+5	0.10577	HOMO \rightarrow LUMO+4	-0.27049
				HOMO -> LUMO+7	0.25743	HOMO -> LUMO+6	0.41/13
				HOMO-7 -> LUMO	-0.10595	$UOMO \rightarrow UUMO \downarrow 4$	0 12014
c	2 8720	220.12	0.0062	HOMO-1 -> LUMO+7	0.24276	HOMO-1 -> LUMO+4	0.12814
39	5.8729	520.15	0.0005	HOMO -> LUMO+5	0.21974	$HOMO \rightarrow LUMO + 4$	0.10084
				HOMO -> LUMO+7	0.49135	HOMO -> LUMO+0	-0.19393
G	2 0741	220.04	0.0010	HOMO-1-> LUMO+5	-0.30653	UOMO > UUMO + 7	0 10010
S ₁₀	3.8/41	320.04	0.0010	HOMO-1 -> LUMO+5	0.59463	HOMO->LUMO+/	-0.18818
T1	2.4341	509.37	0.0000	HOMO-1 -> LUMO+1	-0.34805	HOMO -> LUMO+1	0.54171
T_2	2.4386	508.43	0.0000	HOMO-1 -> LUMO	0.49452	HOMO -> LUMO	0.41717
				HOMO-6 -> LUMO	0.18189		0.12(12
				HOMO-1 -> LUMO+9	-0.13963	HOMO-3 -> LUMO+1	-0.13613
T_3	3.0739	403.34	0.0000	HOMO-1 -> LUMO+15	0.18113	$HOMO-2 \rightarrow LUMO$	0.37266
				HOMO-1 -> LUMO+19	-0.10340	HOMO-1 -> LUMO+16	0.11506
				HOMO -> LUMO+15	0.14998	HOMO -> LUMO+9	-0.19080
				HOMO-8 -> LUMO+1	0.13579	HOMO-5 -> LUMO+1	0.16244
				HOMO-3 -> LUMO+1	0.34547	HOMO-2 -> LUMO	0.15266
T_4	3.0851	401.88	0.0000	HOMO-1 -> LUMO+9	-0.15643	HOMO-1 -> LUMO+12	0.14227
				HOMO-1 -> LUMO+14	-0.10733	HOMO -> LUMO+12	-0.20387
				HOMO -> LUMO+14	0.16766	HOMO-> LUMO+19	-0.11279
T_5	3.2989	375.83	0.0000	HOMO-1 -> LUMO	-0.42943	HOMO -> LUMO	0.54762
T_6	3.3744	367.43	0.0000	HOMO-1 -> LUMO+1	0.56690	HOMO-> LUMO+1	0.38652
				HOMO-1 -> LUMO+5	0.11827		0.12426
T_7	3.4291	361.56	0.0000	HOMO-1 -> LUMO+7	0.41830	HOMO 1 \rightarrow LUMO+7	0.13436
				HOMO-> LUMO	0.10204	HOMO-1 -> LUMO+/	0.45851
	2 4465	250 74	0.0000	HOMO-1 -> LUMO+6	-0.25969	HOMO-> LUMO+1	0.15028
18	3.4465	359.74	0.0000	HOMO-1 -> LUMO+6	0.52518	HOMO -> LUMO+14	0.13957
				HOMO-19 -> LUMO+10	-0.12617		0.100/2
				HOMO-22 -> LUMO+21	0.14117	HOMO-22 -> LUMO+15	-0.10062
				HOMO-18 -> LUMO+3	-0.19923	HOMO-22 -> LUMO+22	0.11423
T ₉	3.5862	345.73	0.0000	HOMO-16 -> LUMO+3	-0.18421	HOMO-1/ \rightarrow LUMO+3	-0.23961
				HOMO-14 -> LUMO+21	0.12858	HOMO-14 -> $LUMO+9$	0.13427
				HOMO-4 -> LUMO+10	0.11135	HOMO-14 -> LUMO+22	0.11135
				HOMO-12 -> LUMO+10	0.12729	HOMO-14 -> LUMO+10	0.10062
				HOMO-28 -> LUMO+8	-0.15926		0 12712
				HOMO-23 -> LUMO+20	0.16556	$HOMO-23 \rightarrow LUMO+14$	-0.12/12
				HOMO-19 -> LUMO+6	-0.12538	HOMO $10 > LUMO + 12$	-0.31544
T_{10}	3.5902	345.34	0.0000	HOMO-19 -> LUMO+20	0.18020	$HOMO-19 \rightarrow LUMO+12$	0.10589
				HOMO-8 -> LUMO+8	-0.16665	HOMO-18 -> LUMO+2	-0.18641
				HOMO-3 -> LUMO+8	0.12214	HUMU-5 -> LUMU+8	0.11/5/
				HOMO -> LUMO+8	-0.11749	HOMO -> LUMO+2	-0.11399



Figure S5. Calculated absorption spectrum of **Spiro-4SO2** (up).Molecular orbitals and their energy levels (in eV) of **Spiro-4SO2**, as well as some major excitations calculated by TD-DFT at the B3LYP6-31G(d) level.

HR-Mass spectra

Elemental Composition Report

Single Mass Analysis

Tolerance = 30.0 PPM / DBE: min = -1.5, max = 200.0 Isotope cluster parameters: Separation = 1.0 Abundance = 1.0%

Monoisotopic Mass, Odd and Even Electron lons 5 formula(e) evaluated with 1 results within limits (up to 50 closest results for each mass)





Elemental Composition Report

Single Mass Analysis

Tolerance = 30.0 PPM / DBE: min = -1.5, max = 200.0 Isotope cluster parameters: Separation = 1.0 Abundance = 1.0%

Monoisotopic Mass, Odd and Even Electron lons 18 formula(e) evaluated with 4 results within limits (up to 50 closest results for each mass)

Xia, Spiro-SO2 ESI20453 34 (0.872) AM (Top,4, Ar,8000.0,1971.61,1.00,LS 5); Sm (Mn, 2x1.00); Sb (1,40.00); Cm (31:68) 1: TOF MS ES+ 125 1772,4360 100-1771.5273 1773.4375 % 1774.4391 1774,9167 1775,5179 1770.2803 1776.3505 1772.1434 1773.0830 1771.2194 1773.9921 _____ m/z 11 . I. 0 J.J. 1776.00 1771.00 1772.00 1773.00 1774.00 1775.00 Minimum: -1.5 200.0 30.0 200.0 Maximum: Calc. Mass mDa PPM DBE Score Formula Mass C117 H88 23Na 0.17 73.5 73.5 08 S4 1771.5273 1771.5260 1.3 3 C117 C117 010 23Na S3 06 23Na S5 H88 -16.4 -9.3 1771.5437 1771.5082 1771.4905 19.1 36.8 2 H88 10.8 73.5 20.8 73.5 1 C117 H88 04 23Na S6

Figure S7. High-resolution MALDI mass spectrum of Spiro-4SO2.

¹H-NMR and ¹³C-NMR Spectra



¹³C-NMR spectrum of **Spiro-4S** (298K, C₂D₂Cl₄, 126 MHz).



Aromatic region of H-H COSY spectrum of Spiro-4S (298K, C₂D₂Cl₄, 500 MHz).



Aromatic region of H-H TOCSY spectrum of Spiro-4S (298K, C₂D₂Cl₄, 500 MHz).



Aromatic region of H-H NOESY spectrum of Spiro-4S (298K, C₂D₂Cl₄, 500 MHz).



¹H-NMR spectrum of **Spiro-4SO2** (298K, CDCl₃, 400 MHz).



Single crystal structure

Crystal structure of Spiro-4S

Cambridge Crystallographic Data Centre deposition number: CCDC 1564493



 Table S3. Crystallographic table.

Compound	Spiro-4S
Molecular formula	$C_{157}H_{168}O_{10}S_4$
Formula weight	2343.14 gmol ⁻¹
Absorption coefficient	$\mu = 0.135 \text{ mm}^{-1}$
Crystal size	
Space group	P-1 (triclinic)
Lattice parameters	a = 15.9701(10) Å alpha = 86.261(4) deg.
	b = 17.8188(8) Å beta = $87.234(4) deg.$
	c = 25.3977(11) Å gamma = 63.893(5) deg.
Volume	6474.6(6) Å ³
Z value	2
F (000)	2508.0
Calculated density	$d_{xray} = 1.202 \text{mg/m}^3$
Temperature	173.0 K
Theta range for data collection	$2.9^\circ < \theta < 25.3^\circ$
Limiting indices	$-19 \le h \le 19, -21 \le k \le 21, -30 \le l \le 30$
Total number of reflections	63701
Unique number of reflections	23660 ($R_{int} = 0.0571$)
Refinement method	Full-matrix least-squares on F^2
Final R indices [I>2sigma(I)]	R1 = 0.1217, $wR2 = 0.3276$
R indices (all data)	R1 = 0.1912, wR2 = 0.3935
Goodness of fit	S = 1.206
Largest diff. peak and hole	2.135 and -0.551 eÅ ⁻³

Crystal structure of Spiro-4SO2

Cambridge Crystallographic Data Centre deposition number: CCDC 1564494



 Table S4. Crystallographic table.

Compound	Spiro-4SO2
Molecular formula	$C_{133}H_{104}Cl_{32}O_8S_4$
Formula weight	3092.80 gmol ⁻¹
Absorption coefficient	$\mu = 0.514 \text{ mm}^{-1}$
Space group	C 1 2/ c 1 (monoclinic)
Lattice parameters	a = 31.7737(8) Å alpha = 90 deg.
	b = 18.4299(6) Å beta = 101.028(2) deg.
	c = 34.7940(6) Å gamma = 90 deg.
Volume	19998.6(9) Å ³
Z value	4
F (000)	6296
Calculated density	$d_{xray} = 1.027 mg/m^3$
Temperature	173.0 K
Theta range for data collection	$3.1^\circ < \theta < 25.0^\circ$
Limiting indices	$-37 \le h \le 34, -21 \le k \le 19, -41 \le l \le 41$
Total number of reflections	36447
Unique number of reflections	17507 ($R_{int} = 0.0365$)
Refinement method	Full-matrix least-squares on F^2
Final R indices [I>2sigma(I)]	R1 = 0.1438, $wR2 = 0.4203$
R indices (all data) Goodness of fit	R1 = 0.1836, $wR2 = 0.4561S = 1.612$
Largest diff. peak and hole	1.585 and -0.874 eÅ ⁻³