

## Electronic supplementary information (ESI)

### Alkali Earth MO<sub>x</sub> (x=6, 7, 9, 12) PolyhedraTuned Cadmium Selenites with Different Dimensions and Diverse SeO<sub>3</sub><sup>2-</sup> Coordinations

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Table S1. Atomic coordinates ( $\text{\AA}\times 10^4$ ) and equivalent isotropic displacements parameters ( $\text{\AA}^2\times 10^3$ ) for **1-4**.  $U_{(eq)}$  is defined as one third of the trace of the orthogonalized  $U_{ij}$  tensor.

Table S2. Selected bond distances ( $\text{\AA}$ ) for **1-4**.

Table S3. Energy-Dispersive Spectrometry (EDS) for 1-4.

Fig. S1. The morphology for 1(a), 2(b), 3(c) and 4(d).

Fig. S2. Experimental and simulated PXRD powder patterns of 1(a), 2(b), 3(c) and 4(d).

Fig. S3. Energy-Dispersive Spectrometry (EDS) plot of 1(a), 2(b), 3(c) and 4(d).

Fig. S4. The coordinate environments of the SeO<sub>3</sub><sup>2-</sup> in **1**.

Fig. S5. The coordinate environments of the SeO<sub>3</sub><sup>2-</sup> in **2**.

Fig. S6. The coordinate environments of the SeO<sub>3</sub><sup>2-</sup> in **3**.

Fig. S7. The coordinate environments of the SeO<sub>3</sub><sup>2-</sup> in **4**.

Fig. S8. The TGA and DSC curves of 1(a), 2(b), 3(c) and 4(d).

Fig. S9. PXRD powder patterns for the residual of 1(a), 2(b), 3(c) and 4(d).

Fig.S10. The IR spectrum of 1-4.

Table S1. Atomic coordinates ( $\text{\AA}\times 10^4$ ) and equivalent isotropic displacements parameters ( $\text{\AA}^2\times 10^3$ ) for **1-4**.  $U_{(\text{eq})}$  is defined as one third of the trace of the orthogonalized  $U_{ij}$  tensor.

<b>(1)</b>				
atoms	x	y	z	U(eq)
Cd(1)	8659(1)	2100(1)	4084(1)	35(1)
Cd(2)	3335(1)	-311(1)	1846(1)	37(1)
Cd(3)	2434(1)	4045(1)	3575(1)	34(1)
Se(1)	665(1)	2166(1)	6653(1)	32(1)
Se(2)	7067(1)	1754(1)	1236(1)	30(1)
Se(3)	5174(1)	6669(1)	3092(1)	31(1)
Se(4)	1299(1)	3480(1)	902(1)	33(1)
Mg(1)	5000	0	5000	18(1)
Mg(2)	5000	5000	0	21(1)
O(1)	-261(12)	563(10)	6552(12)	52(2)
O(2)	-229(11)	3510(9)	5624(9)	39(2)
O(3)	2822(9)	1806(9)	5197(8)	35(2)
O(4)	6205(11)	349(9)	671(8)	37(2)
O(5)	5347(10)	3389(9)	1662(8)	35(2)
O(6)	6857(10)	1101(9)	3017(8)	36(2)
O(7)	4567(11)	6885(9)	1567(8)	36(2)
O(8)	3577(10)	5733(9)	4473(8)	36(2)
O(9)	-927(10)	4029(10)	1970(9)	42(2)
O(10)	2226(10)	4994(9)	1189(8)	36(2)
O(11)	1720(11)	2178(9)	2261(8)	37(2)
O(12)	5893(10)	1357(8)	6186(8)	33(1)

**(2)**

atoms	x	y	z	U(eq)
Cd(1)	4199(1)	4602(1)	1562(1)	11(1)
Cd(2)	476(1)	109(1)	3023(1)	13(1)
Se(2)	4144(1)	1844(1)	3778(1)	10(1)
Se(1)	623(1)	2978(1)	620(1)	11(1)
Se(3)	7512(1)	4715(1)	4561(1)	10(1)
Cd(1')	7496(1)	2472(1)	2331(1)	12(1)
O(6)	4917(3)	3007(2)	634(3)	15(1)
O(7)	8793(3)	3334(2)	829(3)	18(1)
O(5)	5821(4)	3938(2)	3705(3)	25(1)
O(8)	5780(3)	1417(2)	3252(3)	15(1)
O(4)	1586(3)	4204(2)	1253(3)	20(1)
O(9)	3305(3)	552(2)	3774(3)	13(1)
O(1)	8141(3)	4166(2)	6278(3)	18(1)
O(3)	235(3)	3138(2)	-1220(3)	15(1)
O(2)	8867(4)	4134(2)	3779(3)	21(1)
Ca(1)	7496(1)	2472(1)	2331(1)	12(1)

**(3)**

atoms	x	y	z	U(eq)
Cd(1)	7746(2)	3529(1)	3065(1)	21(1)
Se(1)	12207(2)	2959(1)	5843(1)	21(1)
Se(2)	2296(2)	5076(1)	2833(1)	20(1)
Sr(1)	12603(2)	3840(1)	9706(1)	16(1)
O(1)	11800(20)	2946(5)	7582(8)	35(2)
O(2)	12902(17)	1852(5)	5668(8)	27(2)
O(3)	8618(16)	2993(5)	5194(8)	27(2)
O(4)	2926(16)	3941(5)	2789(8)	26(2)
O(5)	-1423(17)	5025(5)	2906(10)	33(2)
O(6)	2642(18)	5335(5)	1130(8)	29(2)

**(4)**

atom	x	y	z	U(eq)
Cd(1)	4488(1)	2500	6221(1)	12(1)
Se(1)	3671(1)	7500	8693(1)	13(1)
Se(2)	4279(1)	2500	1782(1)	13(1)
Ba(1)	2831(1)	7500	3797(1)	12(1)
O(1)	3528(2)	5111(7)	7192(6)	18(1)
O(2)	4456(3)	152(8)	3320(6)	25(1)
O(3)	3292(4)	2500	1658(10)	34(2)
O(4)	2818(4)	7500	9922(9)	28(1)

Table S2. Selected bond distances (Å) for **1-4**.

**(1)**

Cd(1)-O(1)#3	2.340(9)	Cd(3)-O(2)	2.342(8)	Mg(2)-O(7)	2.125(7)
Cd(1)-O(8)#6	2.341(8)	Cd(3)-O(11)	2.379(7)	Mg(2)-O(7)#8	2.125(7)
Cd(1)-O(12)	2.477(7)	Cd(3)-O(2)#1	2.386(8)	Se(1)-O(2)	1.708(7)
Cd(1)-O(9)#10	2.446(8)	Cd(3)-O(3)	2.412(7)	Se(1)-O(1)	1.708(8)
Cd(1)-O(2)#10	2.497(8)	Cd(3)-O(10)	2.418(7)	Se(1)-O(3)	1.714(7)
Cd(1)-O(6)	2.426(7)	Mg(1)-O(6)	2.154(7)	Se(2)-O(6)	1.692(6)
Cd(1)-O(8)#6	2.341(8)	Mg(1)-O(12)	2.087(7)	Se(2)-O(5)	1.701(7)
Cd(2)-O(11)	2.210(8)	Mg(1)-O(3)	2.065(7)	Se(2)-O(4)	1.719(7)
Cd(2)-O(4)#2	2.259(7)	Mg(1)-O(3)#3	2.065(7)	Se(3)-O(8)	1.700(7)
Cd(2)-O(12)#3	2.294(7)	Mg(1)-O(12)#3	2.087(7)	Se(3)-O(7)	1.714(7)
Cd(2)-O(4)	2.323(8)	Mg(1)-O(6)#3	2.154(7)	Se(3)-O(12)#6	1.753(7)
Cd(2)-O(7)#4	2.363(7)	Mg(2)-O(10)#8	2.087(7)	Se(4)-O(9)	1.667(7)
Cd(2)-O(1)#5	2.392(8)	Mg(2)-O(10)	2.087(7)	Se(4)-O(10)	1.706(7)
Cd(3)-O(8)	2.245(7)	Mg(2)-O(5)#8	2.088(7)	Se(4)-O(11)	1.741(7)
Cd(3)-O(5)	2.305(7)	Mg(2)-O(5)	2.088(7)		

**(2)**

Cd(1)-O(4)	2.191(3)	Cd(2)-O(4)#6	2.339(3)	Se(1)-O(3)	1.685(3)
Cd(1)-O(8)#1	2.208(8)	Cd(2)-O(7)#4	2.405(3)	Se(1)-O(4)	1.712(3)
Cd(1)-O(5)	2.219(3)	Cd(1')-O(8)	2.304(2)	Se(2)-O(8)	1.693(2)
Cd(1)-O(6)	2.286(3)	Cd(1')-O(7)	2.304(3)	Se(2)-O(6)#3	1.687(3)
Cd(1)-O(9)#1	2.513(2)	Cd(1')-O(6)	2.362(3)	Se(2)-O(9)	1.719(3)
Cd(1)-O(9)#2	2.522(3)	Cd(1')-O(1)#2	2.363(3)	Se(3)-O(1)	1.685(3)
Cd(2)-O(3)#3	2.274(3)	Cd(1')-O(3)#9	2.411(3)	Se(3)-O(5)	1.693(3)
Cd(2)-O(2)#4	2.287(3)	Cd(1')-O(2)	2.510(3)	Se(3)-O(2)	1.701(3)
Cd(2)-O(1)#5	2.318(3)	O(1)-Ca(1)#3	2.363(3)		
Cd(2)-O(9)	2.336(2)	Se(1)-O(7)#7	1.678(3)		

**(3)**

Cd(1)-O(3)	2.211(7)	Sr(1)-O(2)#5	2.569(7)	Sr(1)-O(5)#9	3.057(9)
Cd(1)-O(4)	2.258(7)	Sr(1)-O(6)#6	2.633(8)	Se(1)-O(1)	1.684(8)
Cd(1)-O(5)#1	2.298(7)	Sr(1)-O(6)#7	2.640(8)	Se(1)-O(3)	1.702(7)
Cd(1)-O(1)#2	2.313(8)	Sr(1)-O(2)#8	2.722(8)	Se(1)-O(2)	1.712(7)
Cd(1)-O(2)#2	2.371(8)	Sr(1)-O(6)#9	2.752(8)	Se(2)-O(5)	1.679(8)
Cd(1)-O(4)#1	2.438(7)	Sr(1)-O(3)#8	2.843(8)	Se(2)-O(6)	1.691(7)
Sr(1)-O(1)	2.454(7)	Sr(1)-O(4)#7	2.953(8)	Se(2)-O(4)	1.740(7)

## (4)

Cd(1)-O(1)#8	2.245(4)	Ba(1)-O(1)#3	2.902(4)	Ba(1)-O(3)	3.214(4)
Cd(1)-O(1)	2.245(4)	Ba(1)-O(1)#4	2.902(4)	Ba(1)-O(3)#7	3.214(4)
Cd(1)-O(2)#9	2.296(4)	Ba(1)-O(1)#5	2.978(4)	Se(1)-O(4)	1.663(6)
Cd(1)-O(2)#10	2.296(4)	Ba(1)-O(1)	2.978(4)	Se(1)-O(1)	1.701(4)
Cd(1)-O(2)	2.430(5)	Ba(1)-O(4)#6	3.039(3)	Se(1)-O(1)#5	1.701(4)
Cd(1)-O(2)#8	2.430(5)	Ba(1)-O(4)#3	3.039(3)	Se(2)-O(3)	1.640(6)
Ba(1)-O(4)#1	2.758(6)	Ba(1)-O(2)#7	3.078(4)	Se(2)-O(2)#8	1.709(4)
Ba(1)-O(3)#2	2.761(6)	Ba(1)-O(2)#8	3.078(4)	Se(2)-O(2)	1.709(4)

Symmetry codes: for: **1** :#1 -x, -y+1, -z+1; #2 -x+1, -y, -z; #3 -x+1, -y, -z+1; #4 x, y-1, z; #5 -x, -y, -z+1; #6 -x+1, -y+1, -z+1; #7 x, y+1, z; #8 -x+1, -y+1, -z; #9 x-1, y, z; #10 x+1, y, z. For **2** : #1 -x+1,y+1/2,-z+1/2; #2 x,-y+1/2,z-1/2; #3 x,-y+1/2,z+1/2; #4 -x+1,y-1/2,-z+1/2; #5 x-1,-y+1/2,z-1/2; #6 -x,y-1/2,-z+1/2; #7 x-1,y,z; #8 -x,y+1/2,-z+1/2; #9 x+1,-y+1/2,z+1/2; #10 x+1,y,z. For **3** : #1 x+1, y, z; #2 x-1/2, -y+1/2, z-1/2; #3 x-1, y, z-1; #4 x, y, z-1; #5 x-1/2, -y+1/2, z+1/2; #6 -x+2, -y+1, -z+1; #7 x+1, y, z+1; #8 x+1/2, -y+1/2, z+1/2; #9 -x+1, -y+1, -z+1; #10 x-1, y, z; #11 x+1/2, -y+1/2, z-1/2. For **4** : #1 x, y, z-1; #2 -x+1/2, -y+1, z+1/2; #3 -x+1/2, -y+1, z-1/2; #4 -x+1/2, y+1/2, z-1/2; #5 x, -y+3/2, z; #6 -x+1/2, -y+2, z-1/2; #7 x, y+1, z; #8 x, -y+1/2, z; #9 -x+1, y+1/2, -z+1; #10 -x+1, -y, -z+1; #11 -x+1/2, -y+2, z+1/2; #12 x, y-1, z; #13 x, y, z+1.

Table S3. Energy-Dispersive Spectrometry (EDS) for 1-4.

## (1)

Point 1				Point 2			
Element	Weight %	Atomic %	Formula	Element	Weight %	Atomic %	Formula
O K	17.15	55.61		O K	12.61	56.06	
Mg K	3.11	5.80	1.0	Mg K	3.10	5.59	1.0
Se L	39.25	22.41	3.86	Se L	41.33	22.36	4.0
Cd L	40.49	16.18	2.79	Cd L	42.96	16.60	2.97
Totals	100			Totals	100		
Point 3				Average ratio: Mg <sub>1.0</sub> Cd <sub>2.9</sub> Se <sub>3.9</sub>			
Element	Weight %	Atomic %	Formula				
O K	4.79	21.21					
Mg K	3.44	10.06	1.0				
Se L	44.50	39.08	3.88				
Cd L	47.27	29.65	2.95				
Totals	100						

## (2)

Point 1				Point 2			
Element	Weight %	Atomic %	Formula	Element	Weight %	Atomic %	Formula
O K	28.12	52.58		O K	37.95	64.46	
Ca K	5.73	5.39	0.7	Ca K	4.74	4.13	0.7
Se L	32.96	23.06	2.99	Se L	29.45	18.43	3.11
Cd L	33.19	18.97	2.46	Cd L	27.86	12.98	2.20
Totals	100			Totals	100		
Point 3							
Element	Weight %	Atomic %	Formula	Average ratio: $\text{Ca}_{0.7}\text{Cd}_{2.3}\text{Se}_{3.1}$			
O K	28.51	68.77					
Ca K	5.29	4.89	0.7				
Se L	32.99	22.07	3.16				
Cd L	33.21	15.58	2.23				
Totals	100						

(3)

Point 1				Point 2			
Element	Weight %	Atomic %	Formula	Element	Weight %	Atomic %	Formula
O K	25.34	65.40		O K	22.9	62.39	
Se L	32.67	17.07	2.15	Se L	32.84	18.13	2.01
Sr L	20.36	9.58	1.21	Sr L	21.00	10.44	1.16
Cd L	21.63	7.95	1.0	Cd L	23.26	9.03	1.0
Totals	100			Totals	100		
Point 3							
Element	Weight %	Atomic %	Formula	Average ratio: $\text{Sr}_{1.1}\text{Cd}_{1.0}\text{Se}_{2.0}$			
O K	26.98	67.43					
Se L	31.93	16.18	1.96				
Sr L	17.83	8.13	0.98				
Cd L	23.25	8.26	1.0				
Totals	100						

(4)

Point 1				Point 2			
Element	Weight %	Atomic %	Formula	Element	Weight %	Atomic %	Formula
O K	27.83	70.98		O K	12.46	47.87	
Se L	28.77	14.89	2.20	Se L	33.33	25.86	2.22
Cd L	18.59	6.76	1.0	Cd L	21.36	11.64	1.0
Ba L	24.78	7.37	1.09	Ba L	32.85	14.64	1.26
Totals	100			Totals	100		
Point 3							
Element	Weight %	Atomic %	Formula	Average ratio: $Ba_{1.2}Cd_{1.0}Se_{2.1}$			
O K	13.35	50.02					
Se L	30.79	23.37	1.85				
Cd L	23.71	12.61	1.0				
Ba L	32.15	14.00	1.11				
Totals	100						

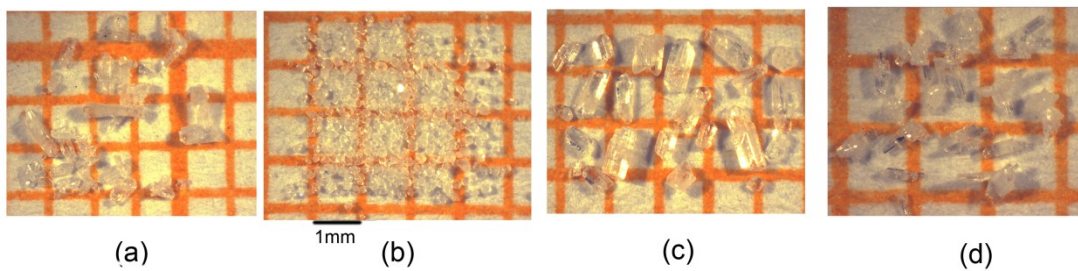


Fig. S1. The morphology for 1(a), 2(b), 3(c) and 4(d).

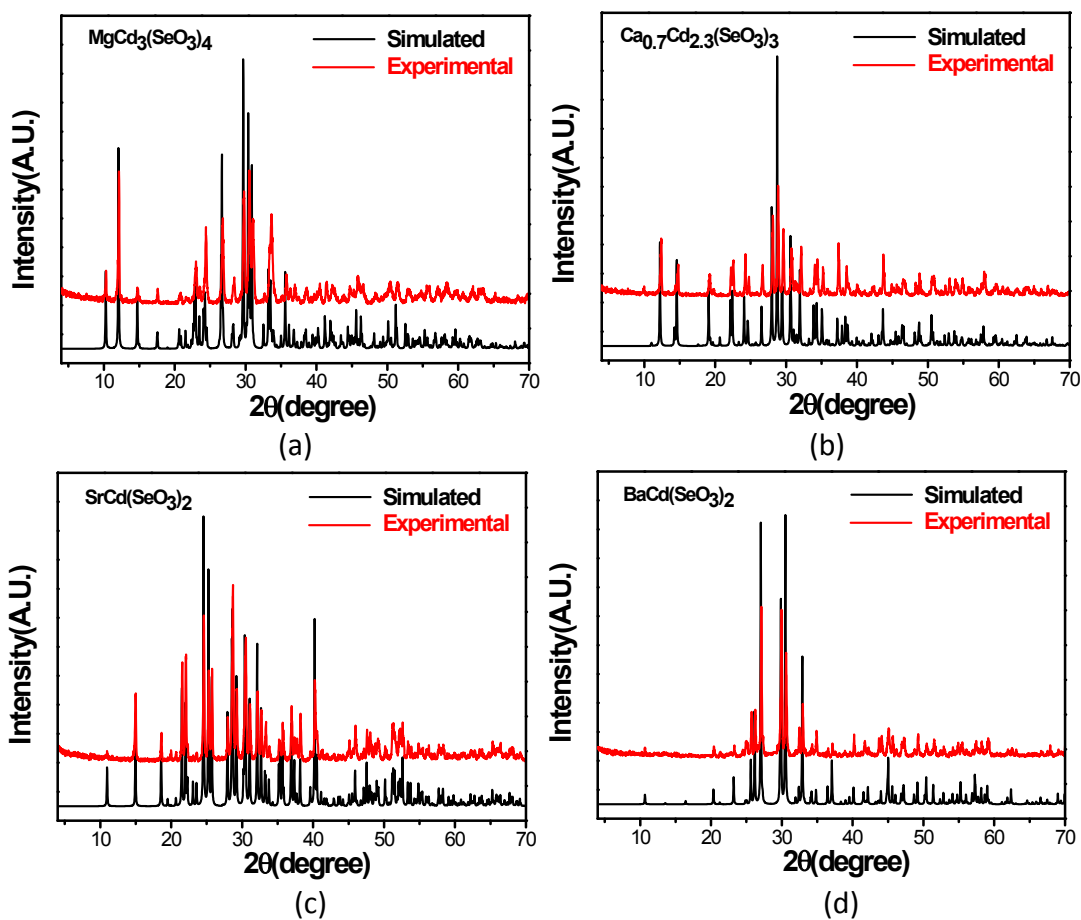
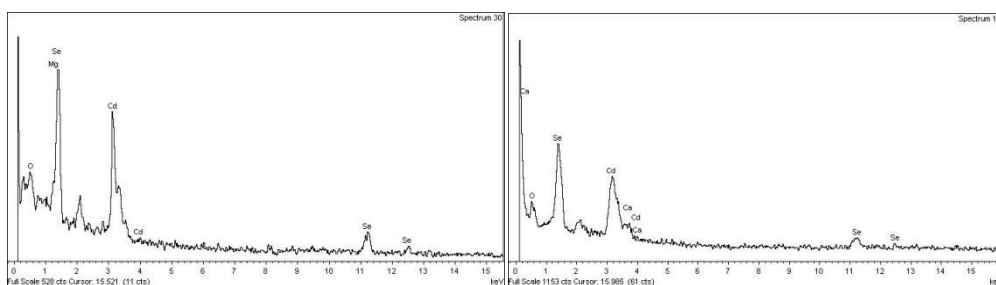


Fig. S2. Experimental and simulated PXRD powder patterns of 1(a), 2(b), 3(c) and 4(d).





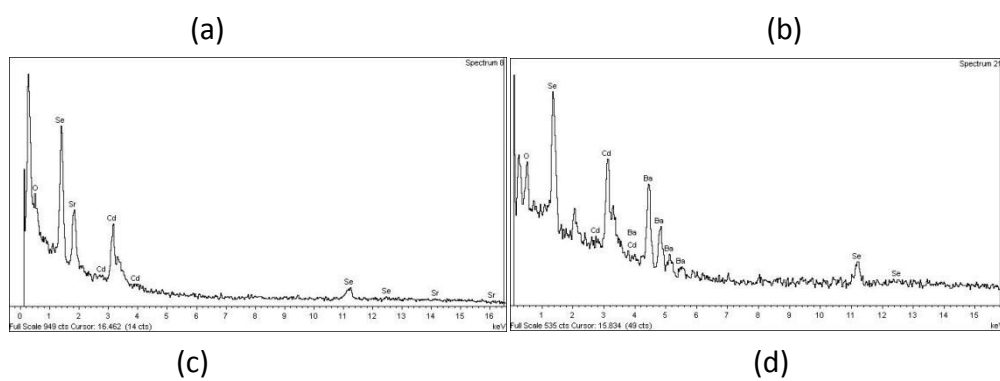


Fig. S3. Energy-Dispersive Spectrometry (EDS) plot of 1(a), 2(b), 3(c) and 4(d).

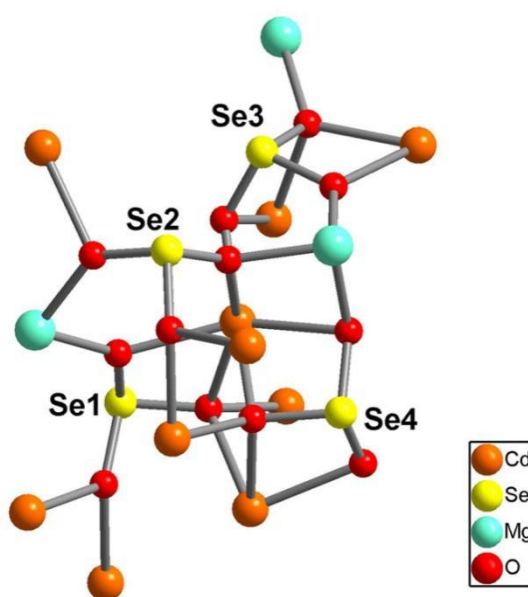


Fig. S4. The coordinate environments of the  $\text{SeO}_3^{2-}$  in **1**.

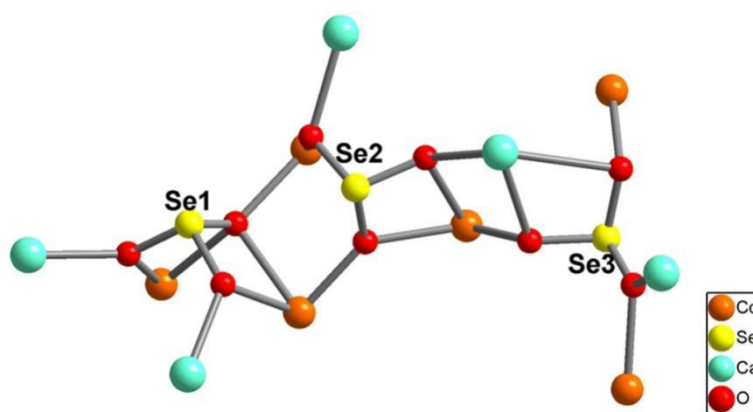


Fig. S5. The coordinate environments of the  $\text{SeO}_3^{2-}$  in **2**.

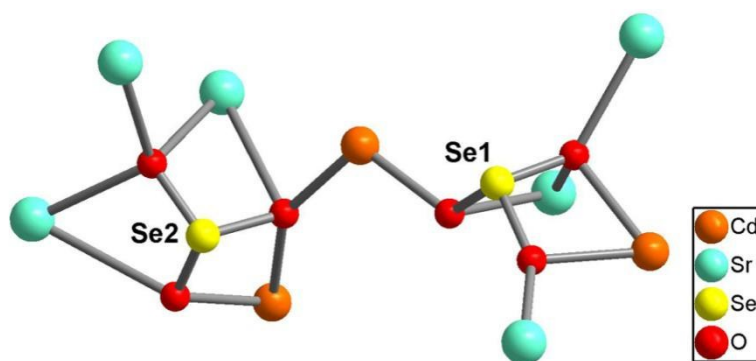


Fig. S6. The coordinate environments of the  $\text{SeO}_3^{2-}$  in **3**.

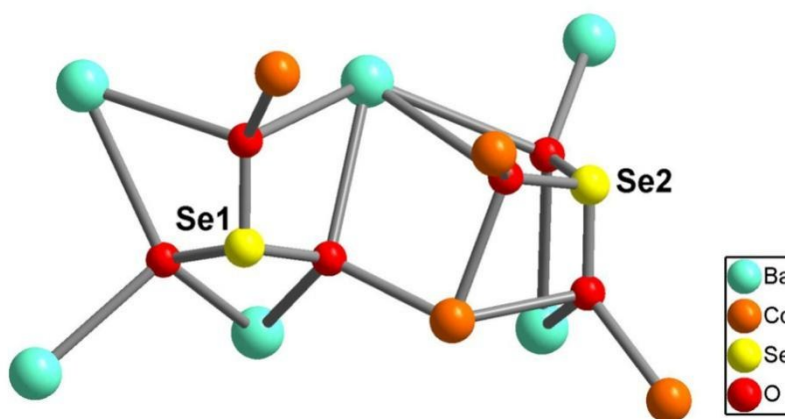
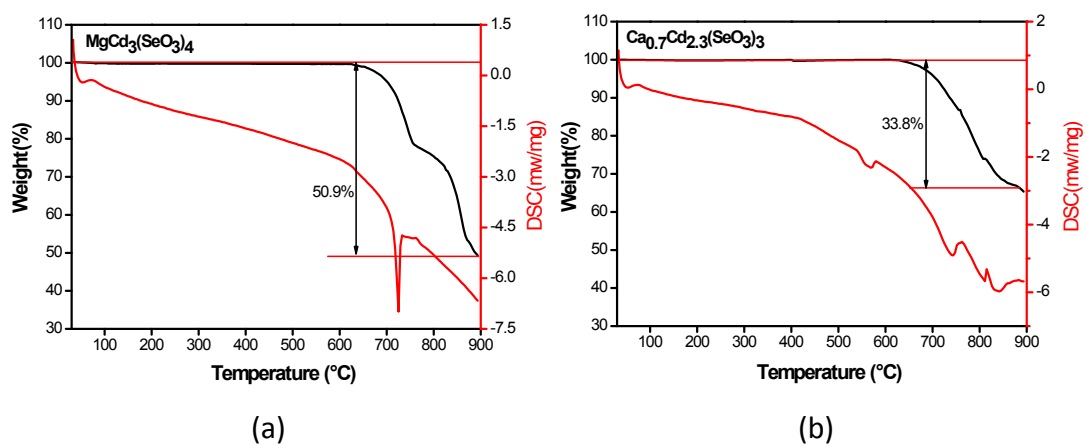


Fig. S7. The coordinate environments of the  $\text{SeO}_3^{2-}$  in **4**.



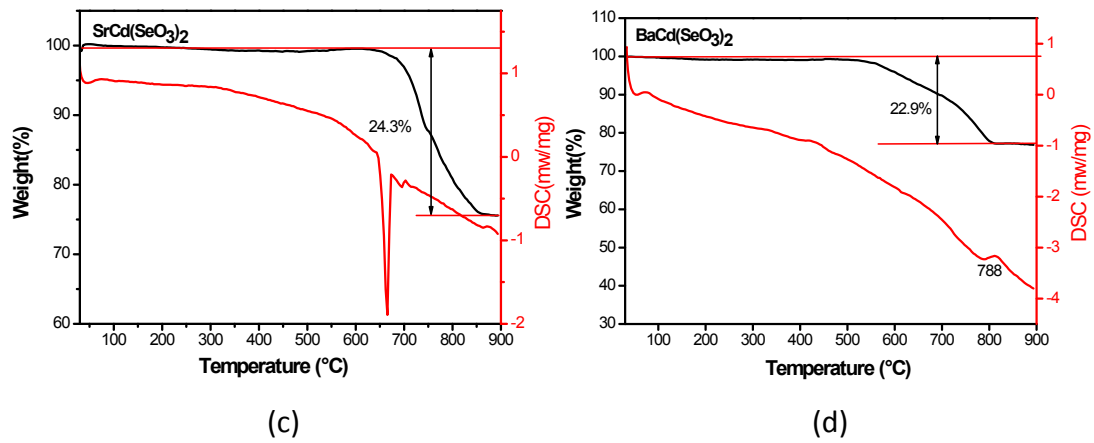
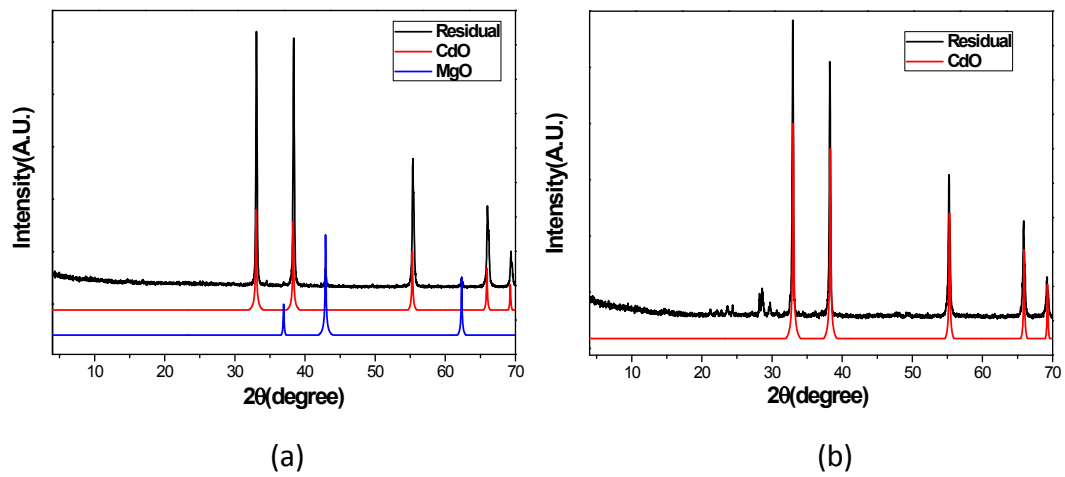


Fig. S8. TGA and DSC curves of 1(a), 2(b), 3(c) and 4(d).



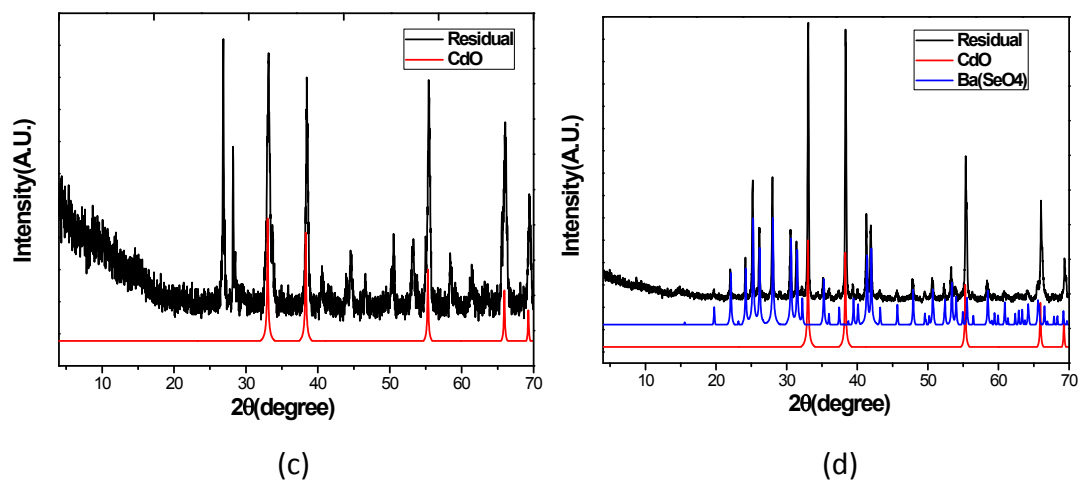


Fig. S9. PXRD powder patterns for the residual of 1(a), 2(b), 3(c) and 4(d).

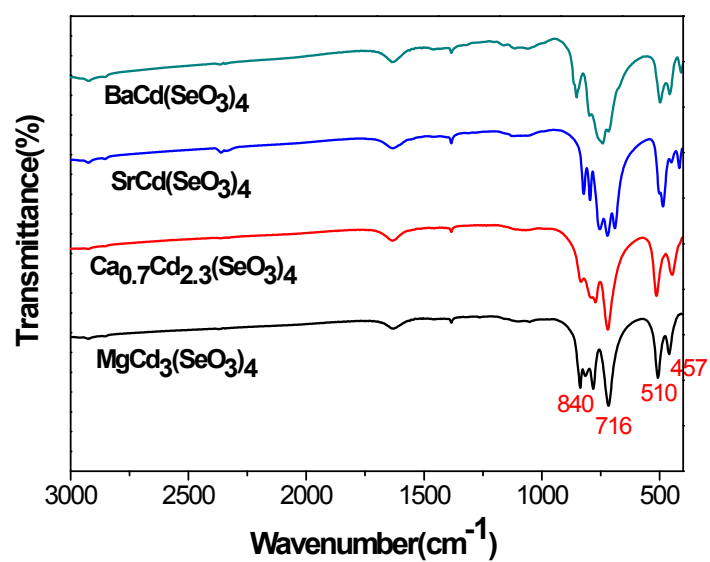


Fig. S10. IR spectrum of 1-4.