

Support information

Supplementary Figure 1: XRD patterns of NaEuF₄ products obtained at different dosage of oxalic acid and the mole ratio of nEu³⁺/nF⁻.

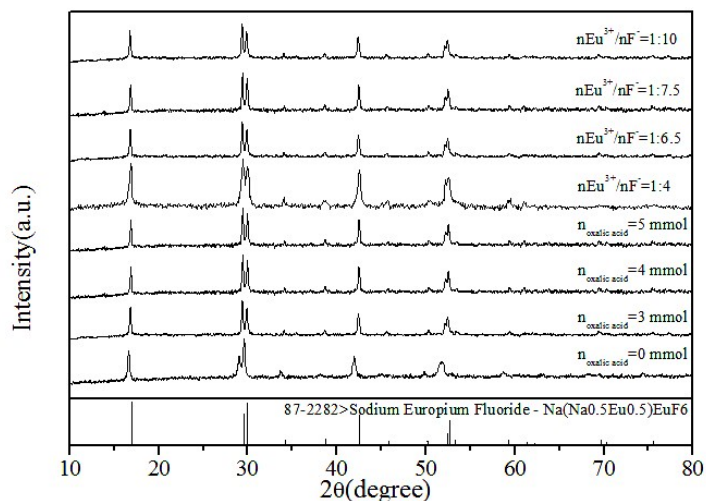
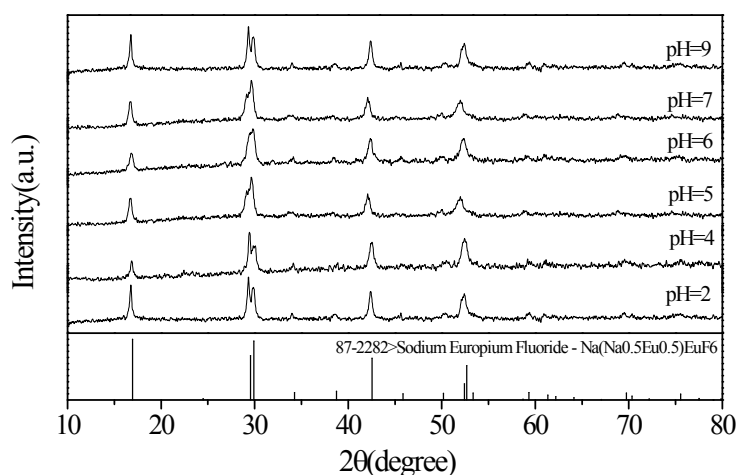


Fig.S1 shows the XRD patterns of the NaEuF₄ product synthesized at different dosage of oxalic acid and the mole ratio of nEu³⁺/nF⁻ (no changed other condition). The samples exhibit the peaks of the pure hexagonal phase Na(Na_{0.5}Eu_{0.5})EuF₆ (JCPDS Card No.87-2282). No peaks of other impurity crystalline phases were detected.



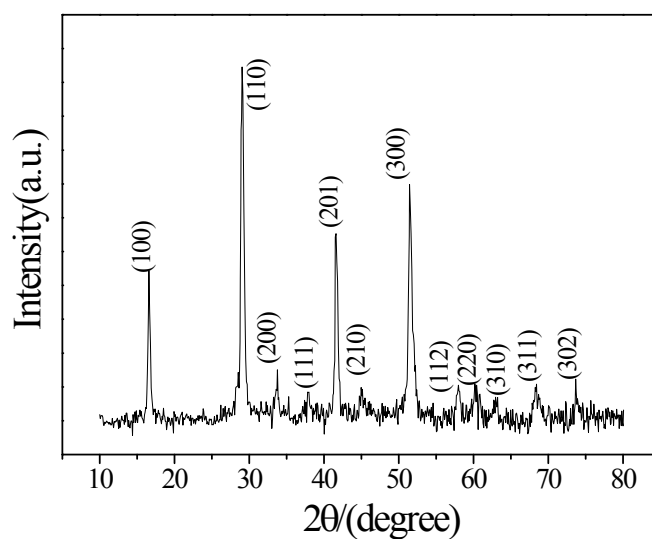
Supplementary Figure 2: XRD patterns of NaEuF₄ products obtained at different pH value.

Fig.S2 shows the XRD patterns of the NaEuF₄ product synthesized at different pH value. The samples exhibit the peaks of the pure hexagonal phase Na(Na_{0.5}Eu_{0.5})EuF₆ (JCPDS Card No.87-2282). No peaks of other impurity crystalline phases were detected.

Synthesis of NaSmF₄ flowerlike structure

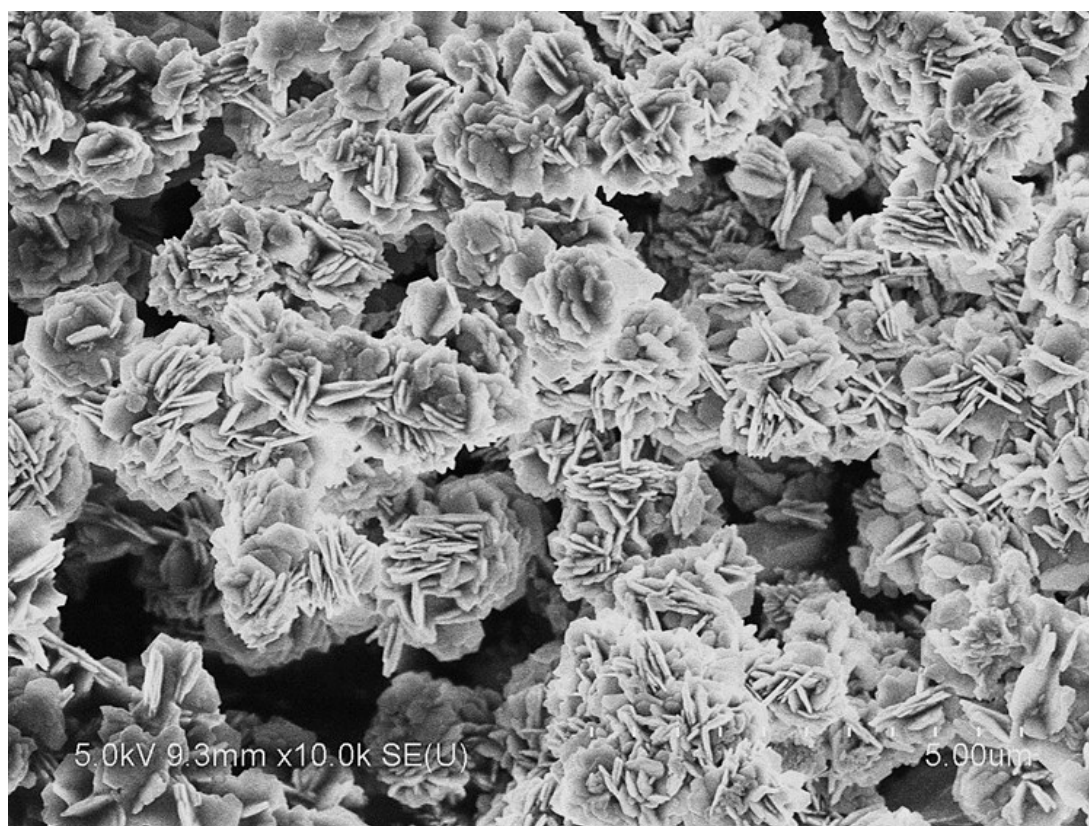
0.5 mmol Sm₂O₃ were dissolved in a dilute HNO₃ (3 mol L⁻¹, 40 mL), and then added 4 mmol oxalic acids (10 mL) to form the metal-oxalic complex with magnetic stirring. After that, 4 mmol NaF was added to the above solution under vigorous stirring and adjusted the pH = 6 using a dilute NaOH solution. The mixture was transferred into a 30 mL Teflon-lined autoclave. The autoclave was tightly sealed and heated at 180 °C for 12h. Finally, it was cooled to room temperature naturally, and the products were collected, washed several times with distilled water and absolute ethanol in turn, and then dried in air at 60 °C for 12h.

Supplementary Figure 3: XRD patterns of NaSmF₄ samples



XRD patterns show well-resolved diffraction peaks consistent with high crystallinity, which can be indexed to hexagonal phase NaSmF₄ (JCPDS No.27-0779). The XRD result indicates that the phase of the as-obtained flowerlike structure is very pure.

Supplementary Figure 4: FESEM of NaSmF₄ flowerlike structure



Supplementary Figure 5: FESEM of NaSmF₄ oblate structure

