# Testing the limits of halogen bonding in coordination chemistry

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# Supplementary information

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### 1. Single crystal X-ray crystallography

Compound	<b>1</b> a	1b	2a	2b
Formula	$C_{20}H_{22}CoCl_2N_2O_4$	C <sub>20</sub> H <sub>22</sub> NiCl <sub>2</sub> N <sub>2</sub> O <sub>4</sub>	$C_{20}H_{22}CoBr_2N_2O_4$	$C_{20}H_{22}NiBr_2N_2O_4$
$M_{ m r}$	484.22	484.00	573.14	572.92
Colour and habit	dark orange,	Clear blue,	dark orange,	dark blue,
Crystal system, space	block Triclinic,	block Triclinic,	block Triclinic,	prism Monoclinic,
group	<i>P</i> -1	<i>P</i> -1	<i>P</i> -1	$P 2_1/n$
Crystal dimensions (mm <sup>3</sup> )	0.32x0.24x0.20	0.25x0.25x0.18	0.27x0.25x0.18	0.32x0.16x0.13
a (Å)	9.0807(5)	9.1894(2)	6.4978(5)	7.9966(5)
b (Å)	10.2369(6)	9.9870(3)	8.8500(7)	9.5422(7)
c (Å)	13.0194(8)	13.4819(5)	10.3960(5)	15.2152(8)
α (°)	74.910(5)	73.422(3)	100.254(5)	90
$\beta(°)$	74.027(5)	71.424(3)	101.706(5)	97.623(6)
γ (°)	86.362(5)	85.191(2)	99.895(7)	90
$V(Å^3)$	1123.37(12)	1124.05(6)	562.47(7)	1150.74(13)
Ζ	2	2	1	2
$D_{\text{calc}}$ (g cm <sup>-3</sup> )	1.432	1.430	1.692	1.653
$\mu$ (mm <sup>-1</sup> )	1.029	1.128	4.342	4.342
F(000)	498	500	285	572
$\theta$ range for data collection (°)	4.37 - 26.00	4.26 - 28.00	4.27 - 26.00	4.48 - 25.99
<i>h,k,l</i> range	-9:11, -12:12, - 16:16	-12:12, -13:13, - 17:17	-8:6, -10:10, -11:12	-9:9, -11:10, -18:8
Scan type	Ø	Ø	Ø	Ø
No. measured reflections	9149	19684	4189	4420
No. independent reflections ( $R_{int}$ )	4395 (0.0189)	4487 (0.0211)	1820 (0.0468)	1505 (0.0306)
No. observed reflections, $I \ge 2\sigma(I)$	3549	5409	2194	2243
No. refined parameters	269	269	135	135
$R, wR[I \ge 2\sigma(I)]$	0.0341, 0.0829	0.0336, 0.0830	0.0358, 0.0794	0.0463, 0.1076
R, wR [all data]	0.0459, 0.0890	0.0438, 0.0882	0.0468, 0.0851	0.0812, 0.1204
Goodness of fit on $F^2$ , S	1.027	1.041	1.079	1.027
Max., min. electron density (e $Å^{-3}$ )	0.273, -0.304	0.274, -0.279	0.566, -0.660	0.824, -0.573
CCDC number	1582547	1582539	1582550	1582538

Table S1. Crystal data and details of the structure determination for 1a–7b.

Compound	<b>3</b> a	3b	<b>4</b> a	<b>4</b> b
Formula	$C_{20}H_{22}CoI_2N_2O_4$	$C_{20}H_{22}NiI_2N_2O_4$	$C_{18}H_{20}CoCl_2N_4O_4\\$	$C_{18}H_{20}NiCl_2N_4O_4$
$M_{ m r}$	667.12	666.90	486.21	485.99
Colour and habit	Orange,	Blue,	Orange,	Light blue,
	prism	prism	block	plate
Crystal system, space	Monoclinic,	Monoclinic,	Triclinic,	Triclinic,
group	$P 2_1/n$	$P \ 2_1/n$	<i>P</i> -1	<i>P</i> -1
Crystal dimensions (mm <sup>3</sup> )	0.18x0.08x0.07	0.27x0.11x0.1	0.23x0.21x0.1	0.17x0.09x0.03
a (Å)	7.9717(5)	8.0225(2)	6.3162(4)	6.3212(5)
b (Å)	9.7058(4)	9.6265(2)	8.8584(8)	8.9502(7)
c (Å)	15.4552(5)	15.4896(3)	10.1891(5)	10.0197(8)
α (°)	90	90	99.706(6)	100.977(7)
$\beta(°)$	96.710(4)	96.406(2)	101.645(5)	100.919(7)
y (°)	90	90	100.278(7)	99.073(6)
$V(Å^3)$	1187.60(10)	1188.77(5)	536.88(7)	535.27(8)
Ζ	2	2	1	1
$D_{\text{calc}}$ (g cm <sup>-3</sup> )	1.866	1.863	1.504	1.508
$u (\mathrm{mm}^{-1})$	3.348	3.439	1.079	1.187
F(000)	642	644	249	250
$\theta$ range for data collection (°)	4.41 - 27.00	4.42 - 27.00	4.35 - 27.99	4.15 - 26.00
h,k,l range	-9:10, -7:12, - 18:19	-10:10, -12:12, - 19:19	-8:8, -11:9, -13:13	-7:7, -9:11, -12:12
Scan type	ω	ω	ω	ω
No. measured reflections	5140	28298	4382	4103
No. independent reflections $(R_{int})$	2583 (0.0232)	2594 (0.0333)	2587(0.0228)	2100(0.0221)
No. observed reflections, $l \ge 2\sigma(l)$	1993	2291	2288	1803
No. refined parameters	135	135	135	135
$R, WR[I \ge 2\sigma(I)]$	0.0423, 0.1073	0.0352, 0.0881	0.0364, 0.0926	0.0388, 0.0813
R, w $R$ [all data]	0.0576, 0.1168	0.0407, 0.0921	0.0428. 0.0977	0.0498. 0.0859
Goodness of fit on $F^2$ ,	1.066	1.067	1.075	1.061
Max., min. electron density (e Å <sup>-3</sup> )	0.551, -0.1024	0.669, -1.319	0.335, -0.475	0.306, -0.258
CCDC number	1582542	1582540	1582548	1582545

Compound	5a	5b	6a	6b	7a
Formula	C <sub>18</sub> H <sub>20</sub> CoBr <sub>2</sub> N <sub>4</sub> O <sub>4</sub>	$C_{18}H_{20}NiBr_2N_4O_4$	C <sub>18</sub> H <sub>20</sub> CoI <sub>2</sub> N <sub>4</sub> O <sub>4</sub>	C <sub>18</sub> H <sub>20</sub> NiI <sub>2</sub> N <sub>4</sub> O <sub>4</sub>	$C_{18}H_{20}CoBr_2N_4O_4$
$M_{ m r}$	575.13	574.91	669.11	668.89	575.13
Colour and habit	Dark orange,	Light blue,	Dark orange,	Light blue,	Light orange,
	block	prism	block	block	plate
Crystal system,	Triclinic,	Triclinic,	Triclinic,	Triclinic,	Triclinic,
space group	<i>P-1</i>	<i>P-1</i>	<i>P-1</i>	P-1	P-1
Crystal dimensions (mm <sup>3</sup> )	0.35x0.23x0.19	0.2x0.06x0.04	0.22x0.21x0.19	0.21x0.18x0.15	0.21x0.15x0.07
a (Å)	6.3290(5)	6.3344(4)	6.3513(9)	6.3490(3)	6.1501(4)
<i>b</i> (Å)	8.9473(6)	9.0280(8)	9.2560(11)	9.2589(5)	8.7162(7)
<i>c</i> (Å)	10.2742(7)	10.1145(7)	10.3631(8)	10.2457(4)	10.7960(8)
α (°)	100.396(6)	101.295(7)	102.241(8)	102.659(4)	100.733(6)
$\beta(\degree)$	101.149(7)	100.541(6)	100.737(9)	100.162(4)	97.813(6)
γ (°)	100.044(6)	99.102(7)	98.708(11)	98.163(4)	101.437(6)
$V(Å^3)$	548.09(7)	546.23(7)	573.27(12)	568.11(5)	548.26(7)
Ζ	1	1	1	1	1
$D_{\rm calc}~({ m g~cm^{-3}})$	1.742	1.748	1.938	1.955	1.742
$\mu(\mathrm{mm}^{-1})$	4.459	4.576	3.471	3.601	4.457
<i>F</i> (000)	285	286	321	322	285
$\theta$ range for data collection (°)	4.34 - 27.00	4.14 - 27.00	4.27 - 26.00	4.15 - 27.00	4.25 - 26.00
<i>h,k,l</i> range	-7:8, -11:11,	-8:5, -11:11,	-7:7, -11:11,	-8:8, -9:11,	-7:7, -10:10,
	-13:13	-12:12	-11:12	-13:13	-13:13
Scan type	ω	ω	ω	ω	ω
No. measured reflections	4340	4386	4009	4563	4740
No. independent reflections ( $R_{int}$ )	2373(0.0179)	2375 (0.0148)	2226(0.0244)	2476(0.0177)	2153 (0.0223)
No. observed reflections, $I \ge 2\sigma(I)$	1893	1970	1748	2139	1782
No. refined parameters	135	135	135	135	135
$R, wR[I \ge 2\sigma(I)]$	0.0354, 0.0807	0.0312, 0.0632	0.0329, 0.0765	0.0287, 0.0623	0.0330, 0.0770
R, wR [all data]	0.0490, 0.0864	0.0415, 0.0675	0.0467. 0.0848	0.0349, 0.0662	0.0439, 0.0818
Goodness of fit on		,	,	,	,
$F^2, S$	1.032	1.037	1.046	1.065	1.052
Max., min.					
electron density	0.503, -0.558	0.622, -0.590	0.564, -0.954	0.580, -0.911	0.265, -0.439
(e Å <sup>-3</sup> )					
CCDC number	1582546	1582541	1582549	1582543	1582544

	1a	1b
M1-01	2.030(1)	2.012(1)
M1–02	2.036(2)	2.018(1)
M2-03	2.030(1)	2.009(1)
M2-04	2.038(2)	2.017(1)
M1-N1	2.192(2)	2.126(1)
M2-N2	2.205(2)	2.134(1)
01-M1-02	89.53(6)	91.57(5)
01–M1–02 <sup>i</sup>	90.47(6)	88.43(5)
01-M1-N1	91.59(6)	88.72(5)
02-M1-N1	91.99(7)	92.03(6)
01–M1–N1 <sup>i</sup>	88.41(6)	91.28(5)
02–M1–N1 <sup>i</sup>	88.01(7)	87.97(6)
03–M2–04	89.06(6)	91.33(6)
03–M2–04 <sup>ii</sup>	90.94(6)	88.67(6)
O3-M2-N2	86.67(6)	87.33(5)
O4-M2-N2	89.83(6)	90.55(6)
03–M2–N2 <sup>ii</sup>	93.34(6)	92.67(5)
04–M2–N2 <sup>ii</sup>	90.17(6)	89.45(6)

Table S2. Selected bond distances (Å) and angles (°) for 1a and 1b.

Symmetry codes (i): -x, -y+1, -z+2; (ii): -x+1, -y+1, -z+1. M = Co(II) (1a), Ni(II) (1b).

	2a	2b	<b>3</b> a	3b
M1-01	2.014(2)	2.007(3)	2.019(3)	2.011(2)
M1–02	2.043(2)	2.018(3)	2.042(3)	2.027(2)
M1N1	2.233(2)	2.120(4)	2.188(4)	2.121(3)
01-M1-02	89.67(9)	91.69(12)	90.29(12)	91.94(9)
01–M1–02 <sup>i</sup>	90.33(9)	88.30(12)	89.71(12)	88.06(9)
01-M1-N1	89.78(9)	89.30(13)	89.29(13)	89.41(10)
02-M1-N1	91.79(9)	91.61(12)	92.06(12)	91.57(10)
O1–M1–N1 <sup>i</sup>	90.22(9)	90.70(13)	90.71(13)	90.59(10)
O2–M1–N1 <sup>i</sup>	88.21(9)	88.38(12)	87.94(12)	88.43(10)
	4a	4b	5a	5b
M1-01	2.011(1)	1.998(2)	2.015(2)	2.005(2)
M1–02	2.029(1)	2.007(2)	2.030(2)	2.012(2)
M1-N1	2.246(2)	2.171(2)	2.245(2)	2.172(2)
01-M1-02	89.85(6)	91.77(7)	89.90(8)	91.64(7)
01–M1–02 <sup>i</sup>	90.15(6)	88.23(7)	90.10(8)	88.36(7)
01-M1-N1	90.55(6)	89.50(7)	89.67(8)	90.47(7)
02-M1-N1	87.22(6)	92.26(7)	92.54(8)	87.97(7)
O1–M1–N1 <sup>i</sup>	89.45(6)	90.50(7)	90.33(8)	89.53(7)
O2–M1–N1 <sup>i</sup>	92.78(6)	87.74(7)	87.46(8)	92.03(7)
	6a	6b	7a	
M1–01	2.018(2)	1.998(2)	2.016(2)	
M1–02	2.026(3)	2.014(2)	2.034(2)	
M1-N1	2.233(3)	2.169(2)	2.236(2)	
01-M1-02	89.91(11)	91.76(9)	89.51(8)	
01–M1–02 <sup>i</sup>	90.08(11)	88.24(9)	90.49(8)	
01-M1-N1	90.34(11)	89.65(9)	91.82(8)	
02-M1-N1	88.64(12)	91.30(9)	85.87(8)	
O1–M1–N1 <sup>i</sup>	89.66(11)	90.35(9)	88.18(8)	
O2–M1–N1 <sup>i</sup>	91.36(12)	88.70(9)	94.13(8)	

 Table S3. Selected bond distances (Å) and angles (°) for 2a-7a.

Symmetry codes (i): -x+1, -y+1, -z+1. M = Co(II) (**2a–7a**), Ni(II) (**2b–6b**).



**Figure S1.** ORTEP-style plot of (a) [Co(acac)<sub>2</sub>(3-Clpy)<sub>2</sub>] **(1a)** and (b) [Ni(acac)<sub>2</sub>(3-Clpy)<sub>2</sub>] **(1b)** with partial labeling scheme. Thermal ellipsoids are drawn at 30% probability level at 296(2) K.



**Figure S2.** ORTEP-style plot of [Ni(acac)<sub>2</sub>(3-Brpy)<sub>2</sub>] **(2b)** with partial labeling scheme. Thermal ellipsoids are drawn at 30% probability level at 296(2) K.



**Figure S3.** ORTEP-style plot of (a) [Co(acac)<sub>2</sub>(3-Ipy)<sub>2</sub>] **(3a)** and (b) [Ni(acac)<sub>2</sub>(3-Ipy)<sub>2</sub>] **(3b)** with partial labeling scheme. Thermal ellipsoids are drawn at 30% probability level at 296(2) K.



**Figure S4.** ORTEP-style plot of (a) [Co(acac)<sub>2</sub>(2-Clpz)<sub>2</sub>] **(4a)** and (b) [Ni(acac)<sub>2</sub>(2-Clpz)<sub>2</sub>] **(4b)** with partial labeling scheme. Thermal ellipsoids are drawn at 30% probability level at 296(2) K.



**Figure S5.** ORTEP-style plot of [Ni(acac)<sub>2</sub>(2-Brpz)<sub>2</sub>] **(5b)** with partial labeling scheme. Thermal ellipsoids are drawn at 30% probability level at 296(2) K



**Figure S6.** ORTEP-style plot of (a) [Co(acac)<sub>2</sub>(2-Ipz)<sub>2</sub>] **(6a)** and (b) [Ni(acac)<sub>2</sub>(2-Ipz)<sub>2</sub>] **(6b)** with partial labeling scheme. Thermal ellipsoids are drawn at 30% probability level at 296(2) K

#### Secondary interactions in the crystal packing of 1a-7a.

In all crystal structures (**1a**–**7a**) stacking interactions were observed but they do delicately adopt to different types of primary interactions that were identified.



**Figure S7.** The 1-D chains, formed via type I Cl···Cl contacts in **1a**, are further extended into 2-D layers through stacking interactions. The corresponding nickel(II) complex,  $[Ni(acac)_2(3-Clpy)_2]$  (**1b**), is isostructural with **1a**. Hydrogen atoms on the acac ligands are omitted for clarity.



**Figure S8.** (a) Stacking interactions in  $[Co(acac)_2(3-Brpy)_2]$  (**2a**). 1-D strands formed via primary C–H···O hydrogen-bond interactions are further extended into 2-D layers through stacking interactions. The corresponding pyrazine complexes (**4a–6b**) are isostructural with **2a**. (b) Stacking interactions in  $[Co(acac)_2(5-Brpm)_2]$  (**7a**), 1-D strands formed via primary C–H···O and C–H···N hydrogen-bond interactions are further extended into 2-D layers through stacking. Hydrogen atoms on the acac ligands are omitted for clarity.



**Figure S9**,  $\pi$ -stacking between the neighbouring molecules of [Ni(acac)<sub>2</sub>(3-Brpy)<sub>2</sub>] (**2b**), isostructural with **3a** and **3b**, extends in 1-D chains.

	Ring	α	β	Cg···Cg	CgI_Perp	Slippage
1a	$Cg(1)\cdots Cg(1)^{i}$	0.02(11)	37.9	4.367(1)	3.446(1)	2.682
1b	$Cg(1)$ ···· $Cg(1)^{i}$	0.02(10)	35.7	4.322(1)	3.510(1)	2.522
2a	$Cg(2)$ ···· $Cg(2)^{ii}$	0.02(17)	41.3	4.999(2)	3.756(1)	3.299
<b>2b</b>	$Cg(2)$ ···· $Cg(2)^{iii}$	0.0(2)	31.7	4.127(3)	3.510(1)	2.171
3a	$Cg(2)$ ···· $Cg(2)^{iii}$	0.0(2)	31.4	4.074(3)	3.476(2)	2.124
<b>3</b> b	$Cg(2)$ ···· $Cg(2)^{iii}$	0.00(18)	32.6	4.133(2)	3.481(1)	2.228
<b>4</b> a	$Cg(3)\cdots Cg(3)^{iv}$	0	39.8	4.658(1)	3.580(1)	2.980
<b>4</b> b	$Cg(3)$ ···· $Cg(3)^{ii}$	0.00(14)	39.8	4.653(2)	3.576(1)	2.978
5a	$Cg(3)$ ···· $Cg(3)^{ii}$	0.03(17)	40.3	4.790(2)	3.653(1)	3.098
5b	$Cg(3)$ ···· $Cg(3)^{iv}$	0.04(14)	40.3	4.786(2)	3.650(1)	3.096
6a	$Cg(3)$ ···· $Cg(3)^{ii}$	0.0(2)	40.8	4.974(2)	3.768(1)	3.248
6b	$Cg(3)$ ···· $Cg(3)^{iv}$	0.02(16)	41.0	4.995(1)	3.769(1)	3.278
7a	$Cg(3)$ ···· $Cg(3)^{iv}$	0.03(16)	45.9	5.321(1)	3.705(1)	3.819

**Table S4** Details on  $\pi \cdots \pi$  geometry for **1a–7a**.

Symmetry codes: (i): 1–x, 2–y, 1–z; (ii): 1–x, 1–y, 2–z; (iii): 1–x, –y, 1–z; (iv): 1–x, 1–y, –z. Cg(1): N2, C11-C15; Cg(2): N1, C1-C5; Cg(3): N1, N2, C1-C4.

#### 2. Powder X-ray crystallography



Figure S10. Experimental (black) and calculated (blue) PXRD traces of [Co(acac)<sub>2</sub>(3-Clpy)<sub>2</sub>] (1a)



Figure S11. Experimental (black) and calculated (blue) PXRD traces of [Ni(acac)<sub>2</sub>(3-Clpy)<sub>2</sub>] (1b)



Figure S12. Experimental (black) and calculated (blue) PXRD traces of [Co(acac)<sub>2</sub>(3-Brpy)<sub>2</sub>] (2a)



Figure S13. Experimental (black) and calculated (blue) PXRD traces of [Ni(acac)<sub>2</sub>(3-Brpy)<sub>2</sub>] (2b)



Figure S14. Experimental (black) and calculated (blue) PXRD traces of [Co(acac)<sub>2</sub>(3-Ipy)<sub>2</sub>] (3a)



Figure S15. Experimental (black) and calculated (blue) PXRD traces of [Ni(acac)<sub>2</sub>(3-Ipy)<sub>2</sub>] (3b)



Figure S16. Experimental (black) and calculated (blue) PXRD traces of [Co(acac)<sub>2</sub>(2-Clpz)<sub>2</sub>] (4a)



Figure S17. Experimental (black) and calculated (blue) PXRD traces of [Ni(acac)<sub>2</sub>(2-Clpz)<sub>2</sub>] (4b)



Figure S18. Experimental (black) and calculated (blue) PXRD traces of [Co(acac)<sub>2</sub>(2-Brpz)<sub>2</sub>] (5a)



Figure S19. Experimental (black) and calculated (blue) PXRD traces of [NI(acac)<sub>2</sub>(2-Brpz)<sub>2</sub>] (5b)



Figure S20. Experimental (black) and calculated (blue) PXRD traces of  $[Co(acac)_2(2-Ipz)_2]$  (6a)



Figure S21. Experimental (black) and calculated (blue) PXRD traces of [Ni(acac)<sub>2</sub>(2-Ipz)<sub>2</sub>] (6b)



Figure S22. Experimental (black) and calculated (blue) PXRD traces of [Co(acac)<sub>2</sub>(5-Brpm)<sub>2</sub>] (7a)