

## SUPPLEMENTARY INFORMATIONS

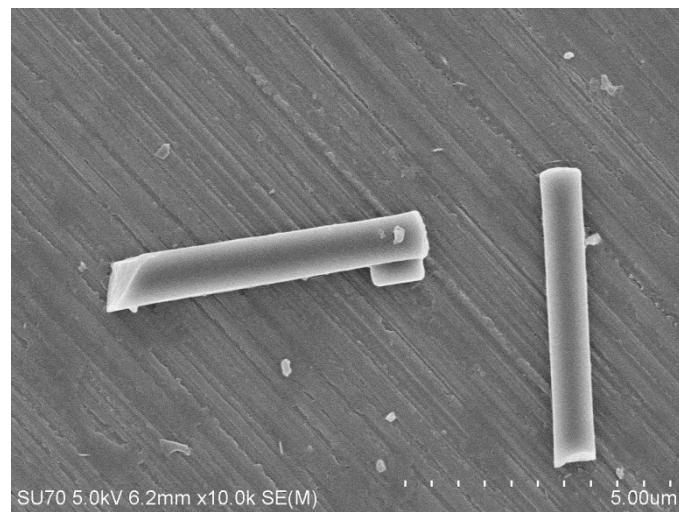
### A lanthanide functionalized MOF hybrid for ratiometric luminescent detection of an anthrax biomarker

Denan Zhang,<sup>a</sup> You Zhou,<sup>\*a</sup> Jing Cuan,<sup>b</sup> Ning Gan

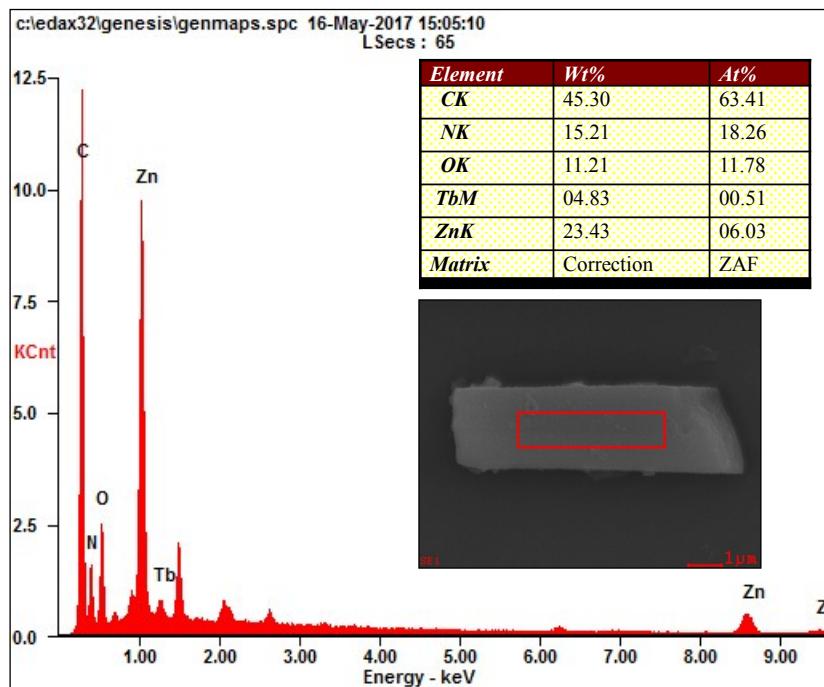
<sup>a</sup> Faculty of Materials Science and Chemical Engineering, Ningbo University, Ningbo 315211, Zhejiang, China. E-mail: [zhouyou@nbu.edu.cn](mailto:zhouyou@nbu.edu.cn), [ganning@nbu.edu.cn](mailto:ganning@nbu.edu.cn).

<sup>b</sup> Institute for Superconducting & Electronic Materials, School of Mechanical, Materials and Mechatronics Engineering, University of Wollongong, Wollongong, NSW 2522, Australia.

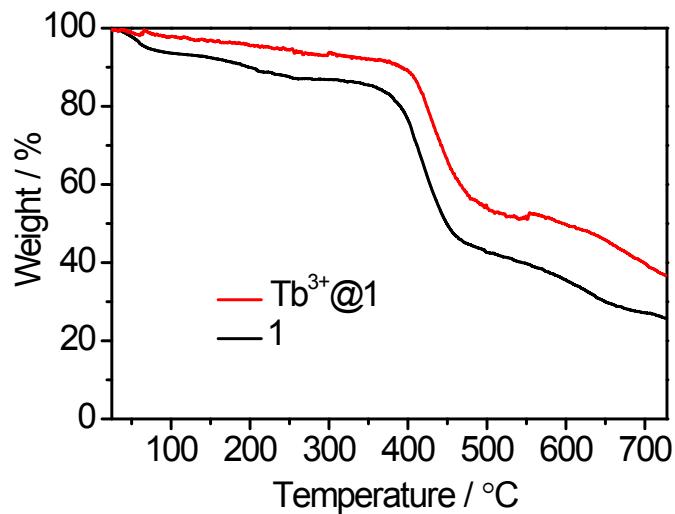
\* Corresponding Author, E-mail: [zhouyou@nbu.edu.cn](mailto:zhouyou@nbu.edu.cn)



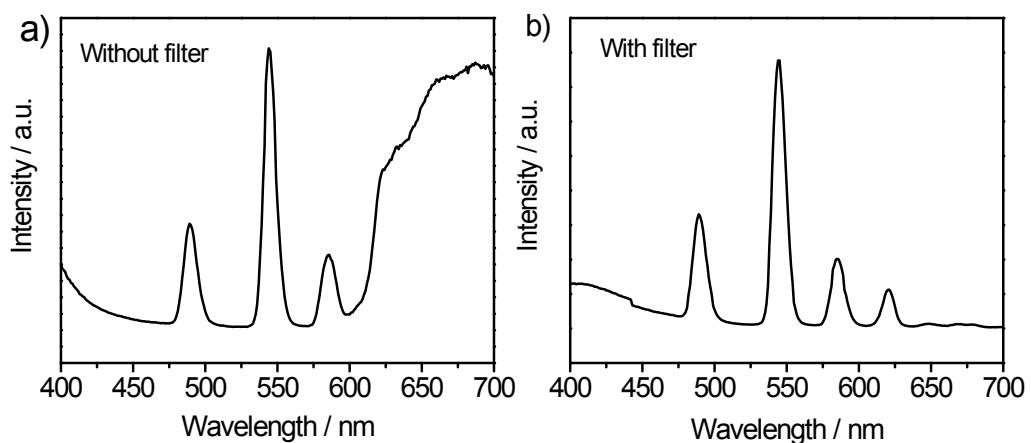
**Figure S1.** SEM images of  $\text{Tb}^{3+}@\mathbf{1}$  microcrystalline rods.



**Figure S2.** EDS spectrum of  $\text{Tb}^{3+}@\mathbf{1}$ . The spectrum was recorded in the region marking with a red square on the microcrystal of  $\text{Tb}^{3+}@\mathbf{1}$ , which is shown in the inset. The Table in the inset is results of mass ratios (Wt%) and atomic ratios (At%) of the elements.



**Figure S3.** TGA curves of pristine 1 (black) and  $\text{Tb}^{3+}@\mathbf{1}$  (red).



**Figure S4.** Emission spectra of  $\text{Tb}^{3+}@\mathbf{1}$  recorded without (a) and with (b) using a high-pass filter (400 nm). Both of the emission spectra are collected at the range of 400-700 nm.

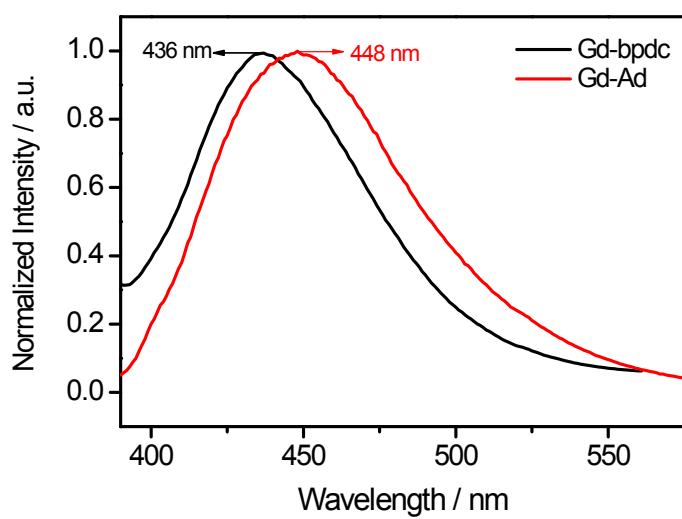
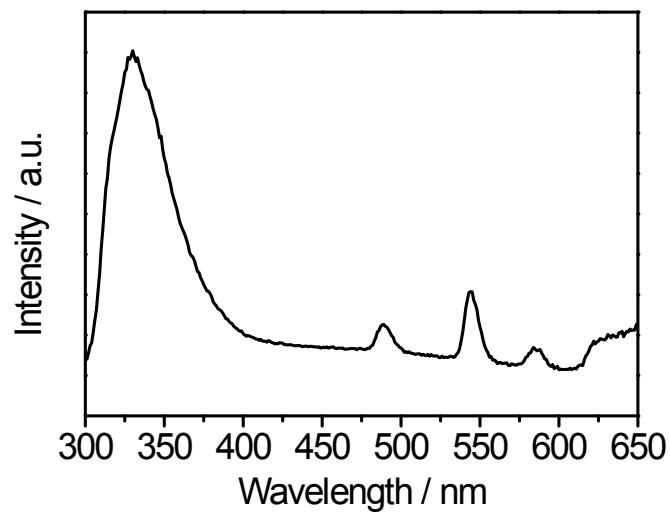
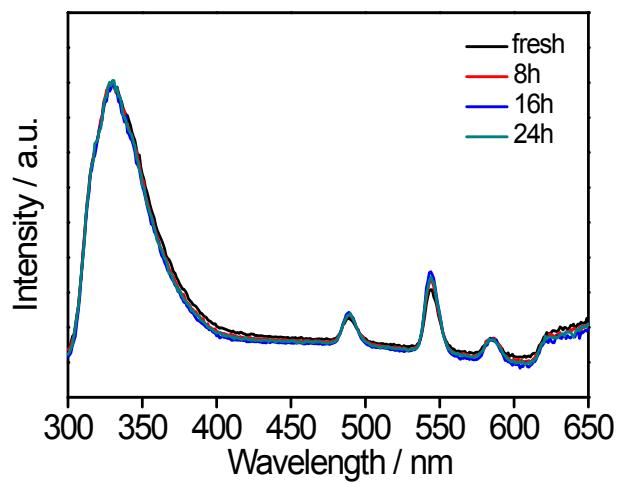


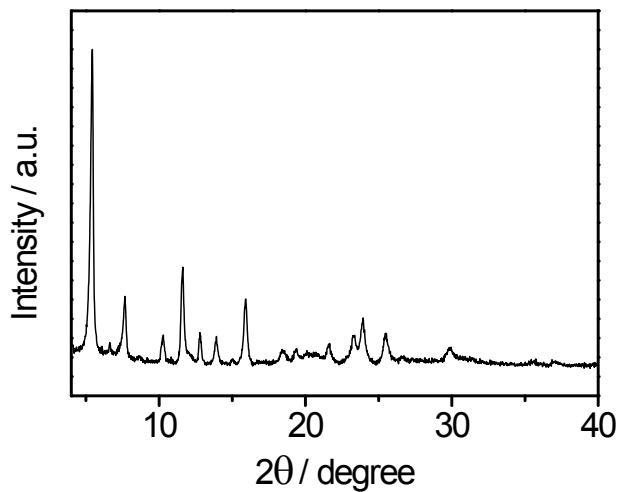
Figure S5. The phosphorescent spectra Gd-Ad (Ad, Adenine) and Gd-bpdc (bpdc, 4,4'-biphenyldicarboxylic) complexes recorded at 77 K.



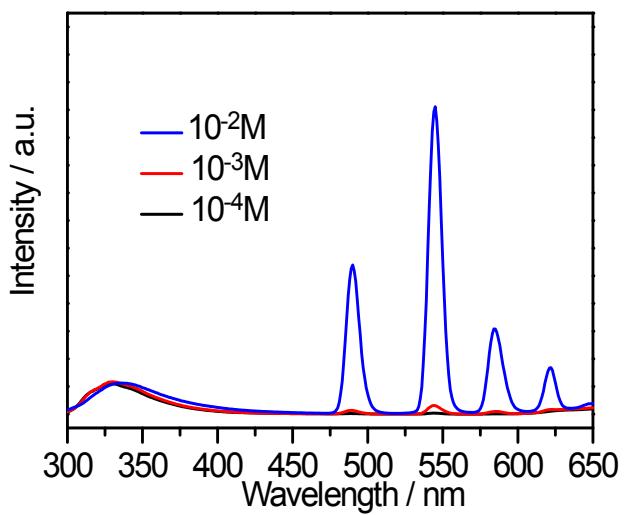
**Figure S6.** The photoluminescence emission spectrum of  $\text{Tb}^{3+}@\mathbf{1}$  aqueous suspension.



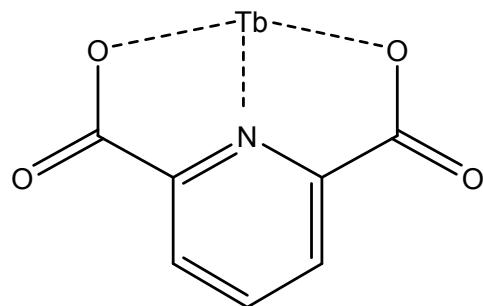
**Figure S7.** Time-dependent emission spectra of  $\text{Tb}^{3+}@\mathbf{1}$  aqueous suspension.



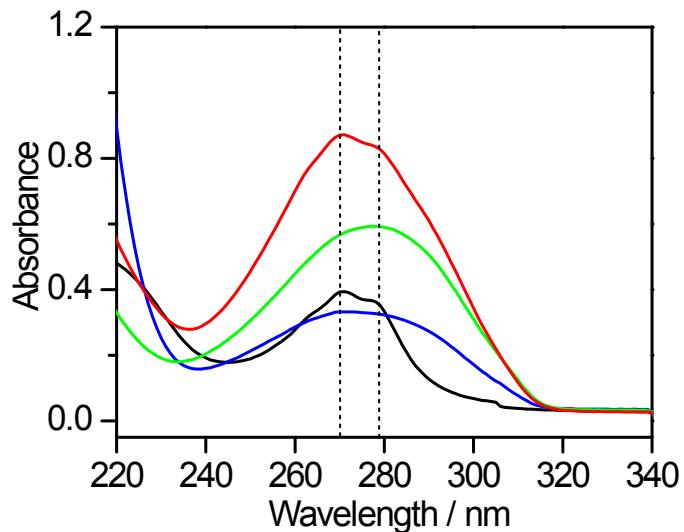
**Figure S8.** PXRD pattern of the  $\text{Tb}^{3+}@\mathbf{1}$  sample after immersing in water for 24 h.



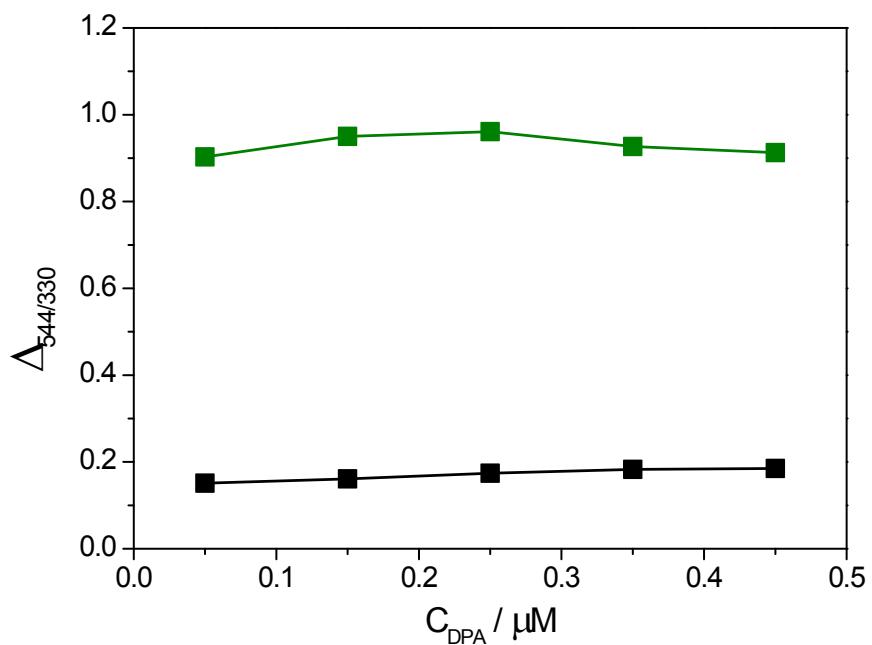
**Figure S9.** Emission spectra ( $\lambda_{\text{ex}} = 280$  nm) of the  $\text{Tb}^{3+}@\mathbf{1}$  products obtained with different feeding concentration ( $10^{-4}$ ,  $10^{-3}$ ,  $10^{-2}$  M) of  $\text{TbCl}_3$ .



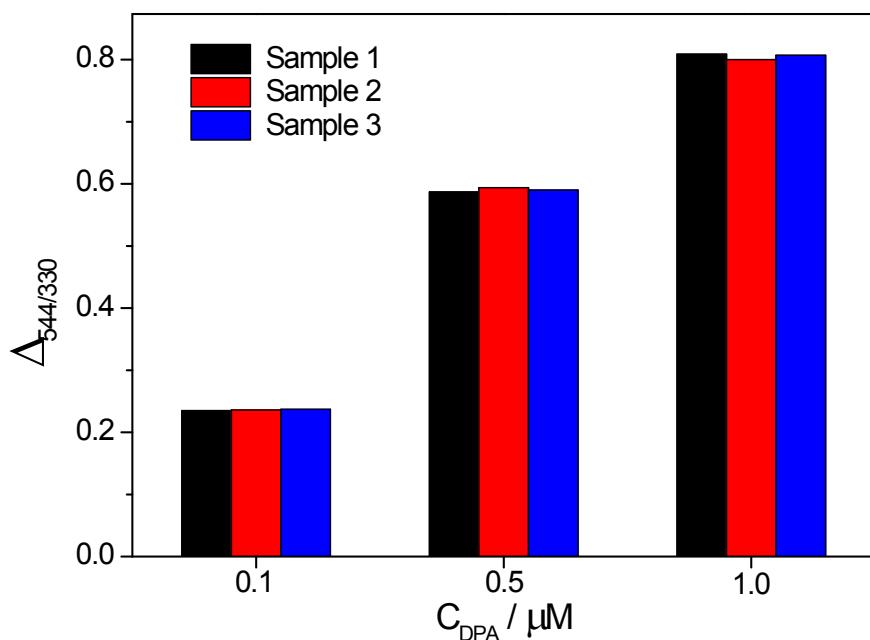
**Figure S10.** Schematic illustration of the chelation between DPA molecule and  $\text{Tb}^{3+}$ .



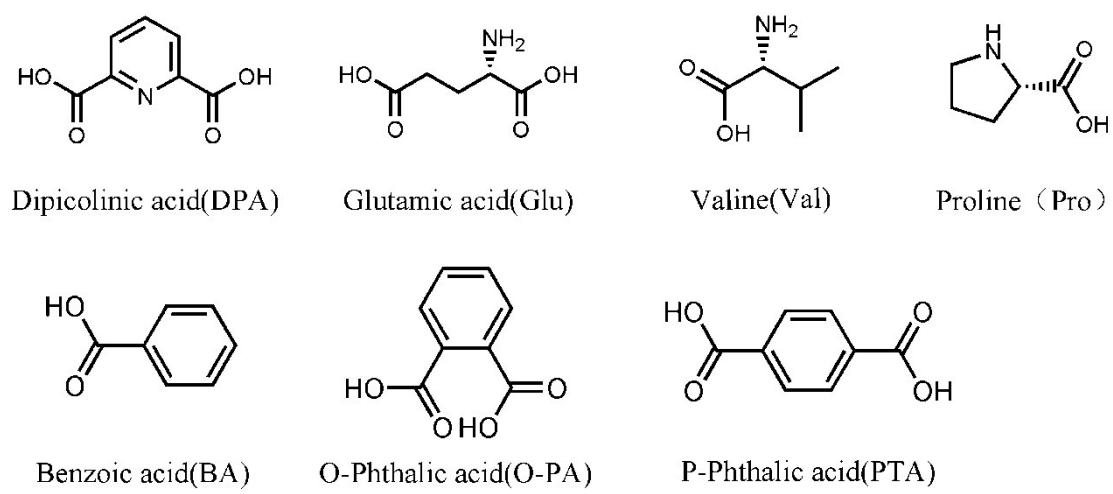
**Figure S11.** UV-vis spectra of DPA (black), pristine **1** (blue), Tb<sup>3+</sup>@**1** (green) and the thoroughly washed Tb<sup>3+</sup>@**1** (red) upon contacting with DPA.



**Figure S12.** Emission intensity ratios ( $\Delta_{544/330}$ ) of various concentrations of Tb<sup>3+</sup>@**1** in the presence (green) and absence (black) of DPA (1  $\mu\text{M}$ ).



**Figure S13.** Emission intensity ratios ( $\Delta_{544/330}$ ) of the  $\text{Tb}^{3+}@\mathbf{1}$  prepared from different batches in different concentration of DPA.



**Figure S14.** The molecular structures of DPA and interfering species used in this study.

**Table S1.** Detection of DPA in serum samples.

| Added (nM) | Founded (nM) | Recovery (%) | RSD (n = 3) (%) |
|------------|--------------|--------------|-----------------|
| 20         | 20.71 ± 0.26 | 103.55       | 1.25            |
| 100        | 109.4 ± 0.45 | 109.40       | 0.41            |
| 200        | 191.2 ± 0.35 | 95.60        | 0.18            |