

ELECTRONIC SUPPLEMENTARY INFORMATION

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Reaction of SO₂ with OH in the Atmosphere

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Basis set abbreviations

Basis set	Reference	Abbrev.
cc-pVDZ-F12	1	DZF
cc-pVTZ-F12	1	TZF
aug-cc-pV(T+d)Z	2	aTZ
maug-cc-pV(T+d)Z	3	maTZ
6-311++G(2df,2p)	4	STZ
modified G3Large semidiffuse	5	MG3S

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- 1 Peterson, K. A.; Adler, T. B.; Werner, H.-J. Systematically convergent basis sets for explicitly correlated wavefunctions: The atoms H, He, B--Ne, and Al--Ar. *J. Chem. Phys.* **2008**, *128*, 084102.
 - 2 Dunning, T. H. Jr.; Peterson, K. A.; Wilson, A. K. Gaussian basis sets for use in correlated molecular calculations. X. The atoms aluminum through argon revisited. *J. Chem. Phys.* **2001**, *114*, 9244-9253.
 - 3 Papajak, E.; Leverentz, H. R.; Zheng, J.; Truhlar, D. G. Efficient Diffuse Basis Sets: cc-pVxZ+ and maug-cc-pVxZ. *J. Chem. Theory. Comput.* **2009**, *5*, 1197-1202.
 - 4 Frisch, M. J.; Pople, J. A.; Binkley, J. S. Self-consistent molecular orbital methods 25. Supplementary functions for Gaussian basis sets. *J. Chem. Phys.* **1984**, *80*, 3265-3269.
 - 5 Lynch, B. J.; Zhao, Y.; Truhlar, D. G. The Effectiveness of Diffuse Basis Functions for Calculating Relative Energies by Density Functional Theory. *J. Phys. Chem. A* **2003**, *107*, 1384-1388.

Table S1. The optimized bond lengths (in Å) of the transition structure for the OH + SO₂ reaction by spin-orbit-free theoretical methods with mean unsigned error (MUE) and maximum absolute derivation (MAD)

Method	HO1	SO1	SO2	SO3	MUE	MAD
CCSD(T)-F12a/TZF	0.973	2.472	1.431	1.434	0.000	0.000
CCSD(T)-F12a/DZF	0.973	2.472	1.431	1.435	0.000	0.000
M06/MG3S	0.973	2.464	1.425	1.432	0.004	0.008
M06/STZ	0.973	2.463	1.431	1.439	0.004	0.009
PW6B95/STZ	0.971	2.462	1.430	1.438	0.004	0.010
MPW1B95/STZ	0.971	2.462	1.430	1.438	0.004	0.010
BMK/STZ	0.972	2.464	1.425	1.431	0.005	0.008
mPW1PW91/STZ	0.972	2.493	1.433	1.440	0.008	0.021
SOGGA11-X/STZ	0.970	2.493	1.427	1.432	0.008	0.021
wB97XD/STZ	0.971	2.507	1.429	1.435	0.010	0.035
CCSD(T)/aTZ	0.975	2.438	1.440	1.443	0.014	0.034
MPW3LYP/STZ	0.975	2.510	1.439	1.447	0.015	0.038
BB1K/STZ	0.966	2.437	1.418	1.423	0.017	0.035
TPSSIKCIS/STZ	0.977	2.507	1.445	1.454	0.018	0.035
MPW1K/STZ	0.964	2.436	1.417	1.422	0.018	0.036
B97-1/STZ	0.974	2.524	1.440	1.448	0.019	0.052
B98/STZ	0.974	2.539	1.437	1.444	0.021	0.067
MN12-L/STZ	0.971	2.402	1.421	1.427	0.022	0.070
M11-L/STZ	0.965	2.433	1.404	1.408	0.025	0.039
B1LYP/STZ	0.974	2.564	1.438	1.443	0.027	0.092
B3LYP/STZ	0.976	2.552	1.441	1.448	0.027	0.080
M05/STZ	0.970	2.574	1.434	1.439	0.028	0.102
M05-2X/STZ	0.972	2.365	1.428	1.434	0.028	0.107
MN12-SX/STZ	0.975	2.371	1.423	1.429	0.029	0.101
M11/STZ	0.975	2.349	1.431	1.438	0.032	0.123
B97-3/STZ	0.971	2.594	1.434	1.439	0.033	0.122
B97-2/STZ	0.970	2.594	1.434	1.440	0.034	0.122
M06-2X/STZ	0.973	2.334	1.426	1.430	0.037	0.138
CCSD/aTZ	0.973	2.334	1.427	1.430	0.037	0.138
M08-SO/STZ	0.978	2.320	1.429	1.435	0.040	0.152
M08-HX/maTZ	0.972	2.303	1.422	1.426	0.047	0.169
M08-HX/aTZ	0.972	2.307	1.422	1.426	0.046	0.165
BHandHLYP/STZ	0.963	2.320	1.415	1.419	0.048	0.152
M08-HX/STZ	0.972	2.289	1.426	1.431	0.048	0.183
M08-HX/MG3S	0.972	2.291	1.421	1.425	0.050	0.181
MP2(Full)/aTZ	0.971	2.219	1.436	1.441	0.067	0.253
BHandH/STZ	0.968	2.245	1.408	1.421	0.067	0.227
MP2(Full)/STZ	0.970	2.197	1.440	1.445	0.075	0.275
M06-L/STZ	0.973	2.657	1.441	1.449	0.053	0.185
O3LYP/STZ	0.973	2.779	1.443	1.446	0.083	0.307
τ -HCTHhyb/STZ	0.976	2.843	1.444	1.444	0.099	0.371
TPSSh/STZ	0.978	2.838	1.447	1.452	0.101	0.366
BLYP/STZ	0.984	2.790	1.468	1.477	0.102	0.318

Table S2. The optimized bond angles (in degrees) of the transition structure for the OH + SO₂ reaction by spin-orbit-free theoretical methods

Method	Bend HO1S	Bend O1SO2	Bend O1SO3	Torsion HO1SO2	Torsion HO1SO3	MUE	MAD
CCSD(T)-F12a/TZF	100.2	107.4	104.5	169.9	41.9	0.0	0.0
CCSD(T)-F12a/DZF	100.1	107.4	104.5	169.9	41.8	0.0	0.1
M08-HX/MG3S	99.9	106.7	104.8	167.3	39.8	1.2	2.6
M08-HX/maTZ	100.0	106.9	104.7	167.1	39.6	1.2	2.8
M08-HX/STZ	100.0	106.9	104.4	165.4	38.3	1.8	4.5
M08-HX/aTZ	99.8	106.8	104.3	164.7	37.5	2.2	5.2
BHandHLYP/STZ	101.5	106.7	107.0	174.8	46.2	2.7	4.9
CCSD(T)/aTZ	99.1	108.4	108.1	176.4	45.3	3.1	6.5
CCSD/aTZ	99.6	106.7	107.7	176.5	46.8	3.2	6.6
MP2(Full)/STZ	99.3	107.0	107.9	180.0	50.0	4.6	10.1
MP2(Full)/aTZ	99.9	106.2	108.4	176.6	53.5	4.7	11.6
M06-2X/STZ	98.9	108.3	103.3	159.3	32.1	4.8	10.6
M08-SO/STZ	98.8	108.9	101.9	154.4	27.8	7.0	15.5
MPW1K/STZ	100.5	108.9	100.9	151.5	25.7	8.0	18.4
M11-L/STZ	98.9	109.6	98.4	142.4	17.4	12.3	27.5
O3LYP/STZ	105.4	111.1	101.5	143.6	16.4	12.7	26.3
MN12-SX/STZ	99.8	109.4	97.8	140.3	16.0	12.9	29.6
M05-2X/STZ	97.3	110.2	97.8	140.8	16.2	13.4	29.1
B97-3/STZ	101.4	112.3	98.6	140.7	14.7	13.7	29.2
M11/STZ	97.6	110.2	96.7	138.5	14.3	14.4	31.4
BB1K/STZ	98.0	111.5	96.1	137.0	12.8	15.3	32.9
SOGGA11-X/STZ	99.2	112.3	95.6	134.6	10.4	16.3	35.3
BMK/STZ	98.1	113.4	95.3	134.9	10.4	16.8	35.0
TPSSh/STZ	98.6	100.1	82.0	145.1	27.9	17.6	24.8
B1LYP/STZ	101.5	112.6	94.7	128.9	5.2	18.8	41.0
M05/STZ	100.7	112.7	93.4	125.9	3.0	20.0	44.0
wB97XD/STZ	99.4	112.2	93.9	128.8	5.6	23.4	-41.1
MN12-L/STZ	97.9	109.7	96.5	139.1	115.3	23.4	73.4
M06/STZ	99.7	113.1	87.7	117.3	-2.6	24.0	52.6
M06/MG3S	99.0	113.1	88.5	117.2	-3.5	24.2	52.7
mPW1PW91/STZ	100.7	110.7	89.3	109.4	-10.8	26.4	60.5
MPW1B95/STZ	99.7	112.4	86.9	110.4	-9.1	26.7	59.5
PW6B95/STZ	99.7	112.4	86.9	110.4	-9.1	26.7	59.5
B97-2/STZ	101.4	111.9	89.0	109.1	-11.3	27.0	60.8
B3LYP/STZ	102.1	112.3	87.4	105.8	-13.8	28.7	64.1
B98/STZ	101.4	111.8	87.2	105.3	-14.2	28.7	64.6
MPW3LYP/STZ	102.0	111.7	86.5	104.2	-14.9	29.3	-65.7
B97-1/STZ	101.5	111.1	85.1	97.7	-20.6	31.8	72.2
TPSSIKCIS/STZ	100.9	109.2	82.8	83.9	-33.3	37.1	86.0
BHandH/STZ	101.7	108.3	83.8	100.0	-17.3	40.7	69.9
M06-L/STZ	97.9	106.2	75.6	70.7	-44.6	43.6	99.2
BLYP/STZ	109.2	149.7	67.2	87.4	-17.5	46.1	82.5
τ -HCTHhyb/STZ	99.6	91.4	91.4	56.0	-59.0	61.2	113.9

Table S3. The enthalpies of activation ($\Delta H_0^{\ddagger,0}$, 0 K) and reaction (ΔH_0^0 , 0 K) of the OH + SO₂ reaction by spin-orbit-free theoretical methods (in kcal/mol)

Method ^{a,b}	$\Delta H_0^{\ddagger,0}$	ΔH_0^0	MUE ^b	Scale factor ^c
MW3.2//C	0.07	-26.57	0.00	0.984 ^d
W3X-L//C	0.01	-26.64	0.07	0.984 ^d
M08-HX/MG3S	0.15	-26.65	0.08	0.973 ^e
W3X//C	-0.02	-26.32	0.17	0.984 ^d
CCSD(T)-F12a/TZF	0.33	-26.66	0.18	0.984 ^d
W3X-L//QCISD/STZ	-0.39	-26.64	0.27	0.973 ^d
W2X//C	0.35	-26.86	0.29	0.984 ^d
CCSD(T)-F12a/DZF	0.53	-26.27	0.38	0.983 ^d
CCSD(T)-F12b/VQZ-F12//C	0.36	-27.11	0.42	0.984 ^d
CCSD(T)-F12b/VQZ-F12//CCSD(T)-F12a/DZF	0.43	-27.12	0.46	0.983 ^d
M08-HX/maTZ	0.65	-26.96	0.49	0.975 ^e
M08-HX/aTZ	0.22	-27.49	0.54	0.976 ^e
M06-2X/maTZ	-0.31	-28.23	1.02	0.971 ^e
M06-2X/aTZ	-0.60	-28.60	1.35	0.971 ^e
CCSD(T)/aTZ	0.51	-24.03	1.49	0.987 ^d
M06/MG3S	-2.41	-27.65	1.78	0.981 ^e

^a//C denotes “at the geometry calculated by CCSD(T)-F12a/TZF”.

^bMean unsigned error averaged over the two previous columns, rrelative to the first row.

^cUnitless factor by which vibrational frequencies were scaled.

^dThe scale factor is calculated in the present work.

^eThe scale factor is obtained from reference 6.

⁶ Alecu, I. M.; Zheng, J.; Zhao, Y.; Truhlar, D. G. Computational Thermochemistry: Scale Factor Databases and Scale Factors for Vibrational Frequencies Obtained from Electronic Model Chemistries, *J. Chem. Theory Comput.* **2010**, *6*, 2872-2887.

Table S4. The barrier height (ΔE^\ddagger , 0 K) and reaction energy (ΔE , 0 K) of the OH + SO₂ reaction by spin-orbit-free theoretical methods (in kcal/mol)

Method ^a	V^\ddagger	ΔV	MUE ^b
MW3.2//C	-1.20	-30.93	0.00
W3X-L//C	-1.26	-31.00	0.06
M08-HX/MG3S	-1.24	-30.74	0.12
W3X//C	-1.29	-30.68	0.17
CCSD(T)-F12a/TZF	-0.93	-31.01	0.17
W2X//C	-0.92	-31.21	0.28
W3X-L//QCISD/STZ	-1.77	-30.94	0.3
M08-HX/maTZ	-0.71	-31.05	0.31
CCSD(T)-F12a/DZF	-0.81	-30.61	0.35
M08-HX/aTZ	-1.12	-31.59	0.37
CCSD(T)-F12b/VQZ-F12//C	-0.91	-31.47	0.41
CCSD(T)-F12b/VQZ-F12//CCSD(T)-F12a/DZF	-0.91	-31.47	0.41
M06-2X/MG3S	-1.91	-31.93	0.86
M06-2X/maTZ	-1.59	-32.35	0.91
M06/STZ	-3.41	-30.79	1.18
M06-2X/aTZ	-1.92	-32.72	1.26
CCSD(T)/aTZ	-0.76	-28.34	1.51
M06/MG3S	-3.65	-31.90	1.71
MP2(full)/aTZ	1.97	-33.27	2.76

CCSD(T)-F12a/TZF^{''}

^bMean unsigned error averaged over the two previous columns

Table S5. The rate constant of the OH + SO₂ reaction at the different temperatures and pressures.(a) bath gas: N₂

T/K	P/bar	k (cm ³ molecule ⁻¹ s ⁻¹)	T/K	P/bar	k (cm ³ molecule ⁻¹ s ⁻¹)
200	1.30 × 10 ⁻³	9.25 × 10 ⁻¹⁵	250	1.30 × 10 ⁻³	1.03 × 10 ⁻¹⁵
200	1.00 × 10 ⁻²	6.68 × 10 ⁻¹⁴	250	1.00 × 10 ⁻²	7.91 × 10 ⁻¹⁵
200	5.00 × 10 ⁻²	2.61 × 10 ⁻¹³	250	5.00 × 10 ⁻²	3.84 × 10 ⁻¹⁴
200	1.00 × 10 ⁻¹	4.10 × 10 ⁻¹³	250	1.00 × 10 ⁻¹	7.40 × 10 ⁻¹⁴
200	5.00 × 10 ⁻¹	7.59 × 10 ⁻¹³	250	5.00 × 10 ⁻¹	2.89 × 10 ⁻¹³
200	1.00 × 10 ⁰	8.51 × 10 ⁻¹³	250	1.00 × 10 ⁰	4.53 × 10 ⁻¹³
200	5.00 × 10 ⁰	9.46 × 10 ⁻¹³	250	5.00 × 10 ⁰	8.43 × 10 ⁻¹³
200	1.00 × 10 ¹	9.60 × 10 ⁻¹³	250	1.00 × 10 ¹	9.51 × 10 ⁻¹³
200	5.00 × 10 ¹	9.71 × 10 ⁻¹³	250	5.00 × 10 ¹	1.07 × 10 ⁻¹²
200	1.00 × 10 ²	9.72 × 10 ⁻¹³	250	1.00 × 10 ²	1.09 × 10 ⁻¹²
200	1.00 × 10 ³	9.74 × 10 ⁻¹³	250	1.30 × 10 ⁻³	1.11 × 10 ⁻¹²
T/K	P/bar	k (cm ³ molecule ⁻¹ s ⁻¹)	T/K	P/bar	k (cm ³ molecule ⁻¹ s ⁻¹)
298	1.30 × 10 ⁻³	1.07 × 10 ⁻¹⁵	350	1.30 × 10 ⁻³	1.03 × 10 ⁻¹⁵
298	1.00 × 10 ⁻²	8.21 × 10 ⁻¹⁵	350	1.00 × 10 ⁻²	7.89 × 10 ⁻¹⁵
298	5.00 × 10 ⁻²	3.98 × 10 ⁻¹⁴	350	5.00 × 10 ⁻²	3.84 × 10 ⁻¹⁴
298	1.00 × 10 ⁻¹	7.69 × 10 ⁻¹⁴	350	1.00 × 10 ⁻¹	7.42 × 10 ⁻¹⁴
298	5.00 × 10 ⁻¹	3.01 × 10 ⁻¹³	350	5.00 × 10 ⁻¹	2.95 × 10 ⁻¹³
298	1.00 × 10 ⁰	4.75 × 10 ⁻¹³	350	1.00 × 10 ⁰	4.71 × 10 ⁻¹³
298	5.00 × 10 ⁰	9.05 × 10 ⁻¹³	350	5.00 × 10 ⁰	9.40 × 10 ⁻¹³
298	1.00 × 10 ¹	1.04 × 10 ⁻¹²	350	1.00 × 10 ¹	1.10 × 10 ⁻¹²
298	5.00 × 10 ¹	1.20 × 10 ⁻¹²	350	5.00 × 10 ¹	1.32 × 10 ⁻¹²
298	1.00 × 10 ²	1.22 × 10 ⁻¹²	350	1.00 × 10 ²	1.37 × 10 ⁻¹²
298	1.00 × 10 ³	1.25 × 10 ⁻¹²	350	1.00 × 10 ³	1.41 × 10 ⁻¹²
T/K	P/bar	k (cm ³ molecule ⁻¹ s ⁻¹)	T/K	P/bar	k (cm ³ molecule ⁻¹ s ⁻¹)
400	1.30 × 10 ⁻³	4.59 × 10 ⁻¹⁶	450	1.30 × 10 ⁻³	9.83 × 10 ⁻¹⁷
400	1.00 × 10 ⁻²	3.52 × 10 ⁻¹⁵	450	1.00 × 10 ⁻²	7.56 × 10 ⁻¹⁶
400	5.00 × 10 ⁻²	1.74 × 10 ⁻¹⁴	450	5.00 × 10 ⁻²	3.77 × 10 ⁻¹⁵
400	1.00 × 10 ⁻¹	3.42 × 10 ⁻¹⁴	450	1.00 × 10 ⁻¹	7.51 × 10 ⁻¹⁵
400	5.00 × 10 ⁻¹	1.53 × 10 ⁻¹³	450	5.00 × 10 ⁻¹	3.65 × 10 ⁻¹⁴
400	1.00 × 10 ⁰	2.70 × 10 ⁻¹³	450	1.00 × 10 ⁰	7.06 × 10 ⁻¹⁴
400	5.00 × 10 ⁰	7.18 × 10 ⁻¹³	450	5.00 × 10 ⁰	2.81 × 10 ⁻¹³
400	1.00 × 10 ¹	9.33 × 10 ⁻¹³	450	1.00 × 10 ¹	4.51 × 10 ⁻¹³
400	5.00 × 10 ¹	1.33 × 10 ⁻¹²	450	5.00 × 10 ¹	9.51 × 10 ⁻¹³
400	1.00 × 10 ²	1.43 × 10 ⁻¹²	450	1.00 × 10 ²	1.16 × 10 ⁻¹²
400	1.00 × 10 ³	1.57 × 10 ⁻¹²	450	1.00 × 10 ³	1.60 × 10 ⁻¹²

(b) bath gas: He

T/K	P/bar	k (cm ³ molecule ⁻¹ s ⁻¹)	T/K	P/bar	k (cm ³ molecule ⁻¹ s ⁻¹)
200	1.30×10^{-3}	2.88×10^{-15}	250	1.30×10^{-3}	3.07×10^{-16}
200	1.00×10^{-2}	2.17×10^{-14}	250	1.00×10^{-2}	2.36×10^{-15}
200	5.00×10^{-2}	9.95×10^{-14}	250	5.00×10^{-2}	1.17×10^{-14}
200	1.00×10^{-1}	1.80×10^{-13}	250	1.00×10^{-1}	2.31×10^{-14}
200	5.00×10^{-1}	5.14×10^{-13}	250	5.00×10^{-1}	1.06×10^{-13}
200	1.00×10^0	6.71×10^{-13}	250	1.00×10^0	1.93×10^{-13}
200	5.00×10^0	8.90×10^{-13}	250	5.00×10^0	5.57×10^{-13}
200	1.00×10^1	9.30×10^{-13}	250	1.00×10^1	7.32×10^{-13}
200	5.00×10^1	9.65×10^{-13}	250	5.00×10^1	9.94×10^{-13}
200	1.00×10^2	9.69×10^{-13}	250	1.00×10^2	1.05×10^{-12}
200	1.00×10^3	9.73×10^{-13}	250	1.30×10^{-3}	1.10×10^{-12}
T/K	P/bar	k (cm ³ molecule ⁻¹ s ⁻¹)	T/K	P/bar	k (cm ³ molecule ⁻¹ s ⁻¹)
298	1.30×10^{-3}	3.09×10^{-16}	350	1.30×10^{-3}	2.89×10^{-16}
298	1.00×10^{-2}	2.37×10^{-15}	350	1.00×10^{-2}	2.22×10^{-15}
298	5.00×10^{-2}	1.17×10^{-14}	350	5.00×10^{-2}	1.10×10^{-14}
298	1.00×10^{-1}	2.32×10^{-14}	350	1.00×10^{-1}	2.18×10^{-14}
298	5.00×10^{-1}	1.07×10^{-13}	350	5.00×10^{-1}	1.01×10^{-13}
298	1.00×10^0	1.95×10^{-13}	350	1.00×10^0	1.86×10^{-13}
298	5.00×10^0	5.77×10^{-13}	350	5.00×10^0	5.69×10^{-13}
298	1.00×10^1	7.71×10^{-13}	350	1.00×10^1	7.78×10^{-13}
298	5.00×10^1	1.09×10^{-12}	350	5.00×10^1	1.16×10^{-12}
298	1.00×10^2	1.16×10^{-12}	350	1.00×10^2	1.27×10^{-12}
298	1.00×10^3	1.24×10^{-12}	350	1.00×10^3	1.40×10^{-12}
T/K	P/bar	k (cm ³ molecule ⁻¹ s ⁻¹)	T/K	P/bar	k (cm ³ molecule ⁻¹ s ⁻¹)
400	1.30×10^{-3}	1.26×10^{-16}	450	1.30×10^{-3}	2.65×10^{-17}
400	1.00×10^{-2}	9.66×10^{-16}	450	1.00×10^{-2}	2.04×10^{-16}
400	5.00×10^{-2}	4.81×10^{-15}	450	5.00×10^{-2}	1.02×10^{-15}
400	1.00×10^{-1}	9.58×10^{-15}	450	1.00×10^{-1}	2.03×10^{-15}
400	5.00×10^{-1}	4.63×10^{-14}	450	5.00×10^{-1}	1.01×10^{-14}
400	1.00×10^0	8.90×10^{-14}	450	1.00×10^0	2.00×10^{-14}
400	5.00×10^0	3.40×10^{-13}	450	5.00×10^0	9.29×10^{-14}
400	1.00×10^1	5.30×10^{-13}	450	1.00×10^1	1.71×10^{-13}
400	5.00×10^1	1.02×10^{-12}	450	5.00×10^1	5.37×10^{-13}
400	1.00×10^2	1.20×10^{-12}	450	1.00×10^2	7.54×10^{-13}
400	1.00×10^3	1.52×10^{-12}	450	1.00×10^3	1.40×10^{-12}

(c) bath gas: Ar

T/K	P/bar	k (cm ³ molecule ⁻¹ s ⁻¹)	T/K	P/bar	k (cm ³ molecule ⁻¹ s ⁻¹)
200	1.30×10^{-3}	4.40×10^{-15}	250	1.30×10^{-3}	4.78×10^{-16}
200	1.00×10^{-2}	3.28×10^{-14}	250	1.00×10^{-2}	3.67×10^{-15}
200	5.00×10^{-2}	1.44×10^{-13}	250	5.00×10^{-2}	1.81×10^{-14}
200	1.00×10^{-1}	2.51×10^{-13}	250	1.00×10^{-1}	3.55×10^{-14}
200	5.00×10^{-1}	6.13×10^{-13}	250	5.00×10^{-1}	1.56×10^{-13}
200	1.00×10^0	7.50×10^{-13}	250	1.00×10^0	2.72×10^{-13}
200	5.00×10^0	9.17×10^{-13}	250	5.00×10^0	6.72×10^{-13}
200	1.00×10^1	9.44×10^{-13}	250	1.00×10^1	8.27×10^{-13}
200	5.00×10^1	9.68×10^{-13}	250	5.00×10^1	1.03×10^{-12}
200	1.00×10^2	9.71×10^{-13}	250	1.00×10^2	1.07×10^{-12}
200	1.00×10^3	9.74×10^{-13}	250	1.30×10^{-3}	1.10×10^{-12}
T/K	P/bar	k (cm ³ molecule ⁻¹ s ⁻¹)	T/K	P/bar	k (cm ³ molecule ⁻¹ s ⁻¹)
298	1.30×10^{-3}	4.87×10^{-16}	350	1.30×10^{-3}	4.61×10^{-16}
298	1.00×10^{-2}	3.74×10^{-15}	350	1.00×10^{-2}	3.54×10^{-15}
298	5.00×10^{-2}	1.84×10^{-14}	350	5.00×10^{-2}	1.75×10^{-14}
298	1.00×10^{-1}	3.62×10^{-14}	350	1.00×10^{-1}	3.44×10^{-14}
298	5.00×10^{-1}	1.60×10^{-13}	350	5.00×10^{-1}	1.53×10^{-13}
298	1.00×10^0	2.80×10^{-13}	350	1.00×10^0	2.71×10^{-13}
298	5.00×10^0	7.07×10^{-13}	350	5.00×10^0	7.11×10^{-13}
298	1.00×10^1	8.84×10^{-13}	350	1.00×10^1	9.10×10^{-13}
298	5.00×10^1	1.14×10^{-12}	350	5.00×10^1	1.24×10^{-12}
298	1.00×10^2	1.19×10^{-12}	350	1.00×10^2	1.31×10^{-12}
298	1.00×10^3	1.25×10^{-12}	350	1.00×10^3	1.41×10^{-12}
T/K	P/bar	k (cm ³ molecule ⁻¹ s ⁻¹)	T/K	P/bar	k (cm ³ molecule ⁻¹ s ⁻¹)
400	1.30×10^{-3}	2.02×10^{-16}	450	1.30×10^{-3}	4.29×10^{-17}
400	1.00×10^{-2}	1.55×10^{-15}	450	1.00×10^{-2}	3.30×10^{-16}
400	5.00×10^{-2}	7.73×10^{-15}	450	5.00×10^{-2}	1.65×10^{-15}
400	1.00×10^{-1}	1.54×10^{-14}	450	1.00×10^{-1}	3.29×10^{-15}
400	5.00×10^{-1}	7.28×10^{-14}	450	5.00×10^{-1}	1.63×10^{-14}
400	1.00×10^0	1.37×10^{-13}	450	1.00×10^0	3.20×10^{-14}
400	5.00×10^0	4.67×10^{-13}	450	5.00×10^0	1.43×10^{-13}
400	1.00×10^1	6.78×10^{-13}	450	1.00×10^1	2.53×10^{-13}
400	5.00×10^1	1.15×10^{-12}	450	5.00×10^1	6.87×10^{-13}
400	1.00×10^2	1.30×10^{-12}	450	1.00×10^2	9.09×10^{-13}
400	1.00×10^3	1.55×10^{-12}	450	1.00×10^3	1.49×10^{-12}

Table S6. Computed $p_{1/2}$ (in the unit of bar) for $\text{SO}_2 + \text{OH}$ reaction with N_2 as bath gas at various temperatures.

T/K	$p_{1/2} / \text{bar}$
190.0	0.04
200.0	0.13
210.0	0.32
220.0	0.62
230.0	0.95
240.0	1.25
250.0	1.47
260.0	1.63
270.0	1.68
298.0	1.70
300.0	1.70
310.0	1.71
320.0	1.75
330.0	1.85
340.0	1.99
350.0	2.19
360.0	2.52
370.0	3.04
380.0	3.75
390.0	4.85
400.0	6.43
410.0	8.64
420.0	10.48
430.0	15.07
440.0	22.16
450.0	33.01

Table S7. Computed $k(T, p)$ (in the unit of $\text{cm}^3 \text{ molecule}^{-1} \text{ s}^{-1}$) for $\text{SO}_2 + \text{OH}$ reaction with N_2 as bath gas, and the fitted $k(T, p)$ at various pressures (bar) and at 190 K, 298 K and 450 K.

T/K	p/bar	B/bar^{-1}	d / bar^2	F^a	Fitted $k(T, p)$	Computed $k(T, p)$	Error%
298	1.0E-04	6.22E-01	-2.60E+00	6.21E-05	7.80E-17	8.27E-17	-5.7
298	5.0E-04	6.22E-01	-2.60E+00	3.11E-04	3.90E-16	4.13E-16	-5.7
298	1.0E-03	6.22E-01	-2.60E+00	6.21E-04	7.79E-16	8.26E-16	-5.7
298	5.0E-03	6.22E-01	-2.60E+00	3.10E-03	3.89E-15	4.12E-15	-5.6
298	1.0E-02	6.22E-01	-2.60E+00	6.18E-03	7.75E-15	8.21E-15	-5.6
298	5.0E-02	6.22E-01	-2.60E+00	3.01E-02	3.78E-14	3.98E-14	-5.1
298	1.0E-01	6.22E-01	-2.60E+00	5.85E-02	7.34E-14	7.69E-14	-4.5
298	5.0E-01	6.22E-01	-2.60E+00	2.37E-01	2.98E-13	3.01E-13	-1.1
298	8.0E-01	6.22E-01	-2.60E+00	3.33E-01	4.18E-13	4.15E-13	0.6
298	1.0E+00	6.22E-01	-2.60E+00	3.85E-01	4.83E-13	4.75E-13	1.6
298	1.5E+00	6.22E-01	-2.60E+00	4.96E-01	6.22E-13	5.90E-13	5.4
298	5.0E+00	6.22E-01	-2.60E+00	7.55E-01	9.48E-13	9.05E-13	4.7
298	1.0E+01	6.22E-01	-2.60E+00	8.61E-01	1.08E-12	1.04E-12	4.2
450	1.0E-04	3.84E-02	-6.95E+02	3.84E-06	6.72E-18	7.57E-18	-11.2
450	5.0E-04	3.84E-02	-6.95E+02	1.92E-05	3.36E-17	3.78E-17	-11.2
450	1.0E-03	3.84E-02	-6.95E+02	3.84E-05	6.72E-17	7.57E-17	-11.2
450	5.0E-03	3.84E-02	-6.95E+02	1.92E-04	3.36E-16	3.78E-16	-11.2
450	1.0E-02	3.84E-02	-6.95E+02	3.84E-04	6.72E-16	7.56E-16	-11.2
450	5.0E-02	3.84E-02	-6.95E+02	1.92E-03	3.35E-15	3.77E-15	-11.0
450	1.0E-01	3.84E-02	-6.95E+02	3.83E-03	6.69E-15	7.51E-15	-10.9
450	5.0E-01	3.84E-02	-6.95E+02	1.89E-02	3.30E-14	3.65E-14	-9.7
450	8.0E-01	3.84E-02	-6.95E+02	2.98E-02	5.22E-14	5.73E-14	-8.9
450	1.0E+00	3.84E-02	-6.95E+02	3.70E-02	6.48E-14	7.06E-14	-8.3
450	1.5E+00	3.84E-02	-6.95E+02	5.46E-02	9.54E-14	1.03E-13	-7.0
450	5.0E+00	3.84E-02	-6.95E+02	1.62E-01	2.83E-13	2.81E-13	0.9
450	1.0E+01	3.84E-02	-6.95E+02	2.81E-01	4.91E-13	4.51E-13	8.8
190	1.0E-04	2.14E+01	-2.23E-03	2.14E-03	2.03E-15	2.33E-15	-12.6
190	5.0E-04	2.14E+01	-2.23E-03	1.06E-02	1.01E-14	1.15E-14	-12.4
190	1.0E-03	2.14E+01	-2.23E-03	2.10E-02	2.00E-14	2.27E-14	-12.3
190	5.0E-03	2.14E+01	-2.23E-03	9.70E-02	9.22E-14	1.04E-13	-11.0
190	1.0E-02	2.14E+01	-2.23E-03	1.77E-01	1.69E-13	1.87E-13	-9.6
190	5.0E-02	2.14E+01	-2.23E-03	4.15E-01	3.95E-13	5.20E-13	-24.1
190	1.0E-01	2.14E+01	-2.23E-03	6.72E-01	6.40E-13	6.70E-13	-4.5
190	5.0E-01	2.14E+01	-2.23E-03	9.13E-01	8.68E-13	8.74E-13	-0.7
190	8.0E-01	2.14E+01	-2.23E-03	9.44E-01	8.98E-13	9.01E-13	-0.3
190	1.0E+00	2.14E+01	-2.23E-03	9.54E-01	9.08E-13	9.10E-13	-0.2
190	1.5E+00	2.14E+01	-2.23E-03	9.69E-01	9.22E-13	9.23E-13	-0.1
190	5.0E+00	2.14E+01	-2.23E-03	9.91E-01	9.42E-13	9.41E-13	0.1
190	1.0E+01	2.14E+01	-2.23E-03	9.95E-01	9.47E-13	9.45E-13	0.2

^a The ratio of fitted $k(T, p)$ to fitted $k^\infty(T)$.

Table S8. The Cartesian coordinates (Å) of the optimized geometries by different theoretical methods.

Species	Methods	Cartesian Coordinates			
OH	CCSD(T)-F12a/TZF	O	0.00000000	0.00000000	0.1088362627
		H	0.00000000	0.00000000	-0.8617186527
	CCSD(T)-F12a/DZF	O	0.00000000	0.00000000	0.1089007776
		H	0.00000000	0.00000000	-0.8617831676
	CCSD(T)/aTZ	O	0.00000000	0.00000000	-0.1133432910
		H	0.00000000	0.00000000	0.8600099510
	M08-HX/MG3S	O	0.00000000	0.00000000	0.10779300
		H	0.00000000	0.00000000	-0.86234500
	QCISD/STZ	O	0.00000000	0.00000000	0.10764100
		H	0.00000000	0.00000000	-0.86112600
M08-HX/maTZ	O	0.00000000	0.00000000	0.10781800	
	H	0.00000000	0.00000000	-0.86254300	
M08-HX/aTZ	O	0.00000000	0.00000000	0.10780400	
	H	0.00000000	0.00000000	-0.86242900	
M06-2X/MG3S	O	0.00000000	0.00000000	0.10787900	
	H	0.00000000	0.00000000	-0.86303400	
M06-2X/maTZ	O	0.00000000	0.00000000	0.10799200	
	H	0.00000000	0.00000000	-0.86393500	
M06-2X/aTZ	O	0.00000000	0.00000000	0.10799200	
	H	0.00000000	0.00000000	-0.86393700	
SO ₂	CCSD(T)-F12a/TZF	S	0.0000000000	0.0000000000	0.3604699381
		O	0.0000000000	-1.2370798113	-0.3634167190
		O	0.0000000000	1.2370798113	-0.3634167190
	CCSD(T)-F12a/DZF	S	0.00000000	0.00000000	0.3602770192
		O	0.0000000000	-1.2379025591	-0.3633202596
		O	0.0000000000	1.2379025591	-0.3633202596
	CCSD(T)/aTZ	S	0.0000000000	0.0000000000	0.3643273502
		O	0.0000000000	-1.2449818111	-0.3653454251
		O	0.0000000000	1.2449818111	-0.3653454251
	M08-HX/MG3S	S	0.00000000	0.00000000	0.36300600
O		0.00000000	1.22444100	-0.36300600	
O		0.00000000	-1.22444100	-0.36300600	
QCISD/STZ	S	0.00000000	0.00000000	0.36504300	
	O	0.00000000	1.23596100	-0.36504300	
	O	0.00000000	-1.23596100	-0.36504300	
M08-HX/maTZ	S	0.00000000	0.00000000	0.36287100	
	O	0.00000000	1.22567500	-0.36287100	
	O	0.00000000	-1.22567500	-0.36287100	
M08-HX/aTZ	S	0.00000000	0.00000000	0.36326900	
	O	0.00000000	1.22555900	-0.36326900	
	O	0.00000000	-1.22555900	-0.36326900	
M06-2X/MG3S	S	0.00000000	0.00000000	0.36221200	
	O	0.00000000	1.22496800	-0.36221200	
	O	0.00000000	-1.22496800	-0.36221200	
M06-2X/maTZ	S	0.00000000	0.00000000	0.36194300	

		O	0.00000000	1.22685200	-0.36194300
		O	0.00000000	-1.22685200	-0.36194300
	M06-2X/aTZ	S	0.00000000	0.00000000	0.36238100
		O	0.00000000	1.22678800	-0.36238100
		O	0.00000000	-1.22678800	-0.36238100
HOSO ₂	CCSD(T)-F12a/TZF	S	0.0004710960	-0.0014666539	0.0015885695
		O	0.0002101956	0.0007530638	1.4422455393
		O	1.1968183345	-0.0006376395	-0.7865867996
		O	-0.8511032141	-1.2700578522	-0.4973010186
		H	-1.4913831920	-1.4854263382	0.1964792294
	CCSD(T)-F12a/DZF	S	0.0012651531	-0.0013833590	0.0017895656
		O	-0.0003331832	0.0009567940	1.4428915547
		O	1.1976173132	-0.0001470897	-0.7871547314
		O	-0.8522594347	-1.2707420587	-0.4981627953
		H	-1.4912766284	-1.4855197066	0.1970619264
	CCSD(T)/aTZ	S	0.0011695704	0.0022079264	0.0021858715
		O	0.0007104686	0.0033270529	1.4524480086
		O	1.2039036026	0.0020390917	-0.7935195470
		O	-0.8538066653	-1.2775654825	-0.5005076481
		H	-1.4969637563	-1.4868440085	0.1958188350
	M08-HX/MG3S	S	-0.12494700	0.07800900	0.25138200
		O	0.27097800	1.39135600	-0.18348800
		O	-1.32515600	-0.55986700	-0.19562700
		O	1.06678700	-0.93220300	-0.10585900
		H	1.89828600	-0.44242800	-0.14230600
	QCISD/STZ	S	-0.12475000	0.07925100	0.24876400
		O	0.27386300	1.39952100	-0.18252900
		O	-1.33287900	-0.56189000	-0.19215100
		O	1.07178400	-0.94164200	-0.10777200
		H	1.89385500	-0.43593100	-0.12060700
	M08-HX/maTZ	S	-0.12530200	0.07596700	0.25148800
		O	0.25038000	1.39615600	-0.18389700
		O	-1.31863300	-0.57741000	-0.19473800
		O	1.08071400	-0.91865000	-0.10815800
		H	1.90514500	-0.41623900	-0.12946500
M08-HX/aTZ	S	-0.12504900	0.07624000	0.25171200	
	O	0.25579100	1.39528200	-0.18365400	
	O	-1.32088000	-0.57255900	-0.19547300	
	O	1.07746600	-0.92259200	-0.10690200	
	H	1.90176700	-0.42089000	-0.13916400	
M06-2X/MG3S	S	-0.12495400	0.07623700	0.25011600	
	O	0.24623200	1.39625700	-0.18296800	
	O	-1.31538500	-0.58079900	-0.19372700	
	O	1.08067400	-0.91625400	-0.10688400	
	H	1.90710500	-0.41341900	-0.13322600	
M06-2X/maTZ	S	-0.12493200	0.07566500	0.25013000	
	O	0.24544100	1.39756600	-0.18292600	
	O	-1.31704800	-0.58147000	-0.19349000	

		O 1.08289600 -0.91617400 -0.10769900
		H 1.90860100 -0.41000800 -0.12915300
	M06-2X/aTZ	S -0.12487000 0.07563400 0.25028000
		O 0.24625700 1.39770400 -0.18271800
		O -1.31759800 -0.58073500 -0.19403600
		O 1.08270000 -0.91707700 -0.10668000
		H 1.90704900 -0.40927200 -0.13700700
TS1	CCSD(T)-F12a/TZF	S -0.3763806742 -0.0183999348 0.2982645747
		O -0.7508807861 1.2605473056 -0.2315844983
		O -0.9652475821 -1.2062308201 -0.2390247388
		O 2.0604501591 -0.1614920281 -0.0914524962
		H 2.3262598835 0.7422364775 0.1510281585
	CCSD(T)-F12a/DZF	S -0.3762115117 -0.0184393107 0.2977858498
		O -0.7506934102 1.2614193328 -0.2313860377
		O -0.9653292912 -1.2070494410 -0.2388196549
		O 2.0604374538 -0.1615725757 -0.0913770181
		H 2.3259977594 0.7423029946 0.1510278608
	CCSD(T)/aTZ	S -0.3417128143 -0.0037698818 0.2654856243
		O -0.7913746012 1.2738394689 -0.2328771924
		O -0.9595600494 -1.2129528057 -0.2135348379
		O 2.0605124288 -0.1634727430 -0.1200822067
		H 2.3263360361 0.7230169616 0.1882396126
	M08-HX/MG3S	S -0.31127100 0.00584300 0.28998200
		O -0.64411000 1.28608800 -0.24005600
		O -0.94366400 -1.15237400 -0.23626800
		O 1.93124000 -0.22956600 -0.11392100
		H 2.23261600 0.67333600 0.08225000
	QCISD/STZ	S -0.31582500 0.00455900 0.28813600
		O -0.65702800 1.29720200 -0.23729800
		O -0.96278200 -1.16095200 -0.23580000
		O 1.97022800 -0.23064200 -0.11155100
		H 2.24985700 0.68218500 0.06701500
	M08-HX/maTZ	S -0.31373300 0.00565700 0.29011300
		O -0.64201400 1.28827600 -0.23987600
		O -0.95162300 -1.15043600 -0.23702100
		O 1.94072900 -0.23295200 -0.11328700
		H 2.24297700 0.67038000 0.07967200
	M08-HX/aTZ	S -0.31650600 0.00564300 0.29302300
		O -0.63318500 1.28993000 -0.24110600
		O -0.95594600 -1.14648100 -0.24184100
		O 1.94233100 -0.23884400 -0.10947400
		H 2.23849200 0.67287200 0.05099500
	M06-2X/MG3S	S -0.31965800 0.00469900 0.28840400
		O -0.64146500 1.28920800 -0.23722100
		O -0.96550700 -1.14566600 -0.23801400
		O 1.96478600 -0.23764200 -0.10844800
		H 2.25201900 0.67762500 0.05500900
	M06-2X/maTZ	S -0.32275200 0.00444800 0.28897600

		O	-0.62114100	1.29684500	-0.23559000
		O	-0.98784900	-1.13490000	-0.24130900
		O	1.97177000	-0.25410100	-0.10476200
		H	2.26177700	0.66607600	0.02967600
	M06-2X/aTZ	S	-0.32498200	0.00434800	0.29126900
		O	-0.61569800	1.29771400	-0.23650500
		O	-0.99078800	-1.13234100	-0.24505500
		O	1.97435900	-0.25761000	-0.10193300
		H	2.25672100	0.66833900	0.00763400

Table S9. Absolute energies in hartrees.

Methods	Species			
	OH	SO ₂	TS1	HOSO ₂
MW3.2//CCSD(T)-F12a/TZF	-75.79421151	-549.8021775	-625.5982935	-625.6456752
W3X-L//CCSD(T)-F12a/TZF	-75.79399865	-549.8014140	-625.5974131	-625.6448092
M08-HX/MG3S	-75.72715708	-548.6021208	-624.3312600	-624.3782680
W3X//CCSD(T)-F12a/TZF	-75.77673679	-549.5832442	-625.3620352	-625.4088677
CCSD(T)-F12a/TZF	-75.67409900	-548.0938420	-623.7694280	-623.8173670
W3X-L//QCISD/STZ	-75.79399936	-549.8013890	-625.5982086	-625.6447014
W2X//CCSD(T)-F12a/TZF	-75.79353781	-549.7991471	-625.5941439	-625.6424224
CCSD(T)-F12a/DZF	-75.66380000	-548.0639980	-623.7290890	-623.7765840
CCSD(T)-F12b/VQZ-F12//CCSD(T)-F12a/TZF	-75.67392067	-548.0932588	-623.7686299	-623.8173276
CCSD(T)-F12b/VQZ-F12//CCSD(T)-F12a/DZF	-75.67392054	-548.0932550	-623.7686191	-623.8173195
M08-HX/maTZ	-75.73346734	-548.6182246	-624.3528246	-624.4011812
M08-HX/aTZ	-75.73398003	-548.6197711	-624.3555288	-624.4040975
M06-2X/maTZ	-75.73338632	-548.6340144	-624.3699276	-624.4189573
M06-2X/aTZ	-75.73381017	-548.6351094	-624.3719801	-624.4210654
CCSD(T)/aTZ	-75.64554700	-548.0062710	-623.6530330	-623.6969750
M06-2X/MG3S	-75.72913496	-548.6206904	-624.3528654	-624.4007039

Table S10. Parameters used to fit $k(T,p)$.^a

	$k_{\infty}(T)$	$k_0(T)$		$p_{1/2}$
A	3.275×10^{-12}	3.8735×10^{107}	a_1	-2.81111
n	0.1966	219.980	a_2	4.88071
E	0.6022	103.4219	a_3	0.61832
T_0	259.802	243.8243	l_1	191.8051
			l_2	425.2390
			T_1	15.1164
			T_2	31.0877

^aUnits are given in text.

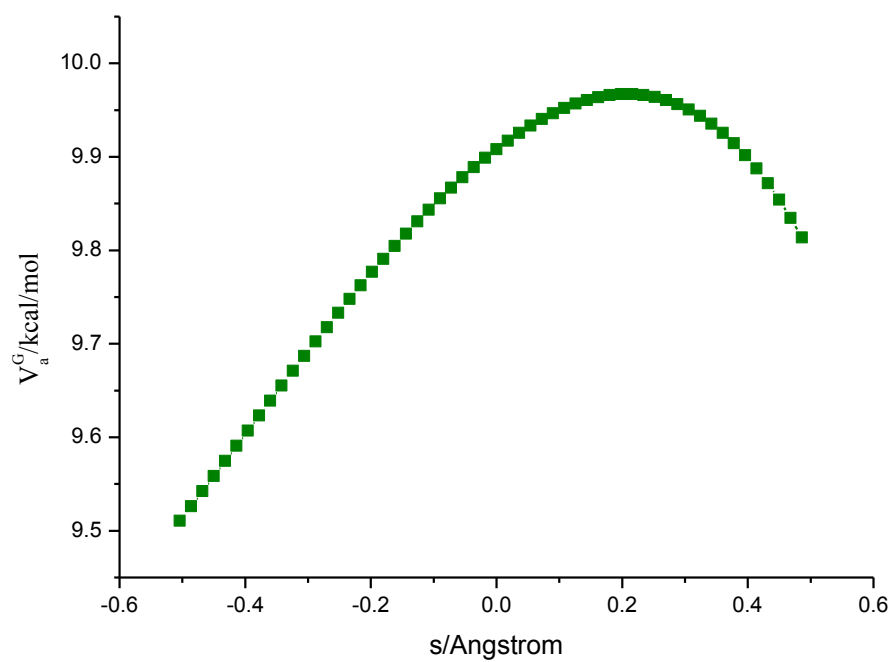


Figure S1. Vibrationally adiabatic potential energy along the minimum energy path for the OH + SO₂ reaction.

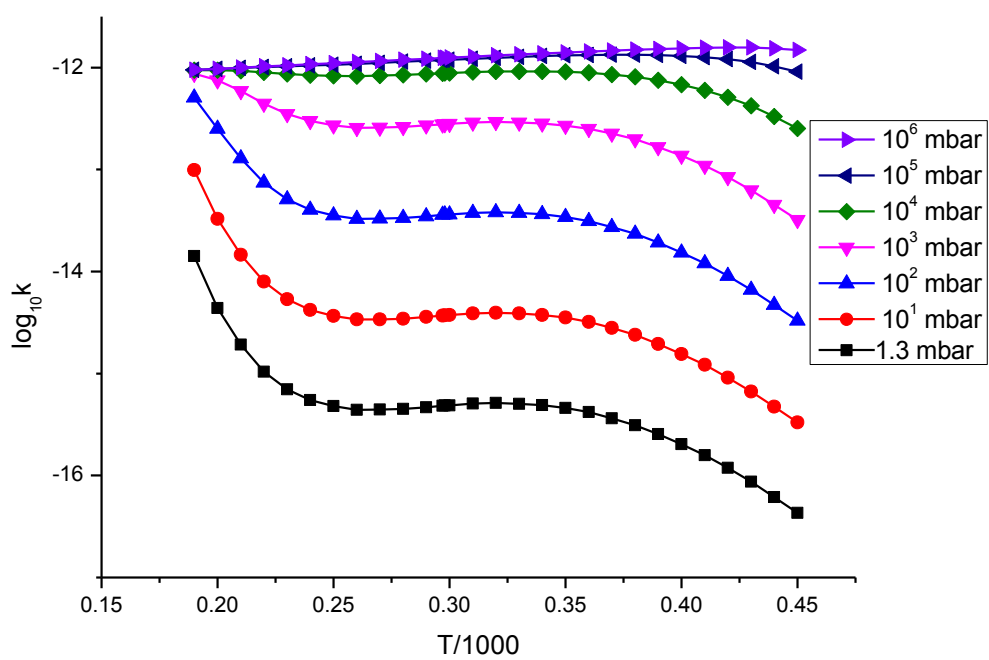


Figure S2. The rate constant of the OH + SO₂ reaction at the different temperatures and pressures (bath gas: Ar).

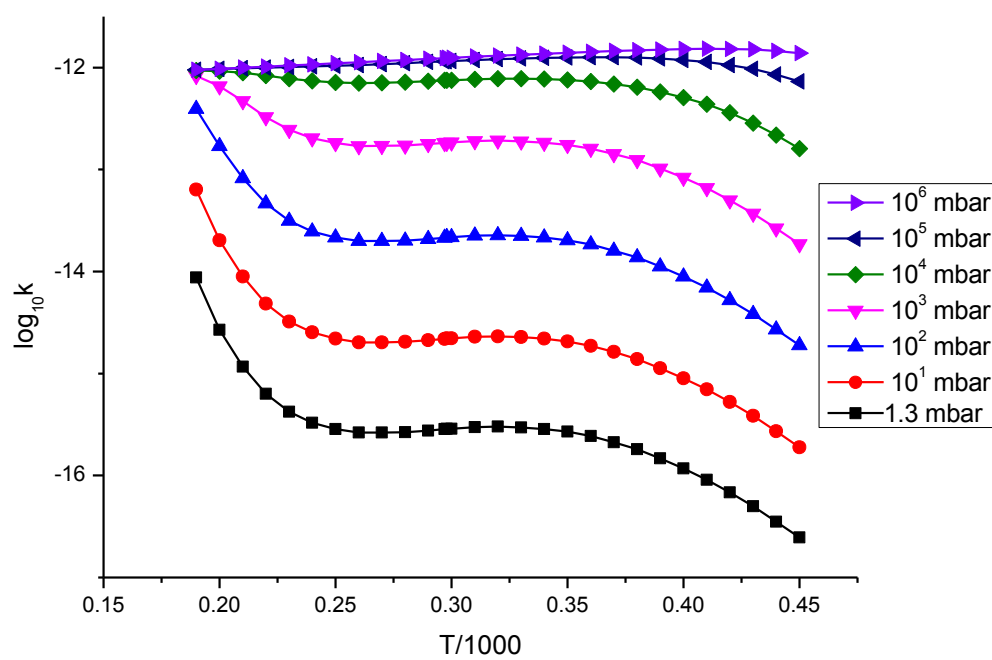


Figure S3. The rate constant of the OH + SO₂ reaction at the different temperatures and pressures (bath gas: He).