

Supporting Information

Influence of the Cation in Lithium and Magnesium Polysulphide Solutions in Dependence of the Solvent Chemistry

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Preparation of the Mg and Li polysulphide solutions

The Mg polysulphide solutions were prepared as 1.0 mM stock solutions by stirring $[\text{Mg}(\text{N-Melm})_6]\text{S}_8$ or $[\text{Mg}(\text{DMSO})_6]\text{S}_8$ at 40°C for 7 d in the investigated solvents.

In order to increase accuracy upon weighing the sulphur and Li_2S powders, the “ LiS_8 ” solutions were prepared at a higher concentration of 10 mM. Stoichiometric amounts of Li_2S and sulphur were stirred in the investigated solvent at 40 °C for 4 d, after which the solutions were diluted to 1 mM and stirred for another 3 d at 40°C. The total preparation time of 7 d is comparable to the preparation of the Mg polysulphide solutions. However, due to their slower dissolution rate, the 10 mM “ LiS_8 ” samples in ACN and $\text{Pyr}_{14}\text{TFSI}$ were stirred for 9 d at 40°C in order to achieve a complete dissolution. After a complete dissolution, also these solutions were diluted to a concentration of 1.0 mM and stirred for another 3 d at 40°C.

The Mg and Li 1.0 mM stock solutions were diluted to 0.5 mM, 0.2 mM, and 0.1 mM samples directly before measuring the UV/Vis absorption spectra.

UV/Vis studies

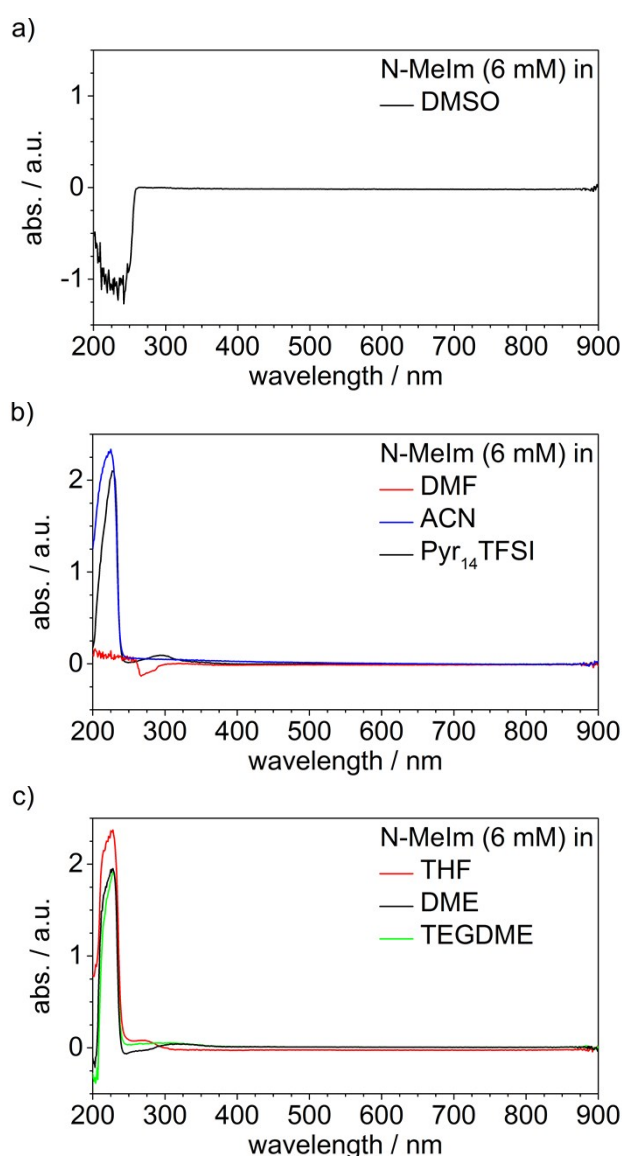


Figure S1: UV/Vis spectra of N-Melm (6 mM) in a) DMSO, b) DMF, ACN, and $\text{Pyr}_{14}\text{TFSI}$, and c) THF, DME, and TEGDME.

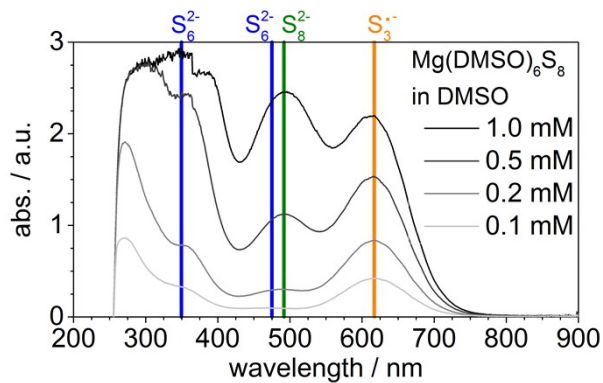


Figure S2: UV/Vis spectra of $\text{Mg}(\text{DMSO})_6\text{S}_8$ solutions in DMSO.

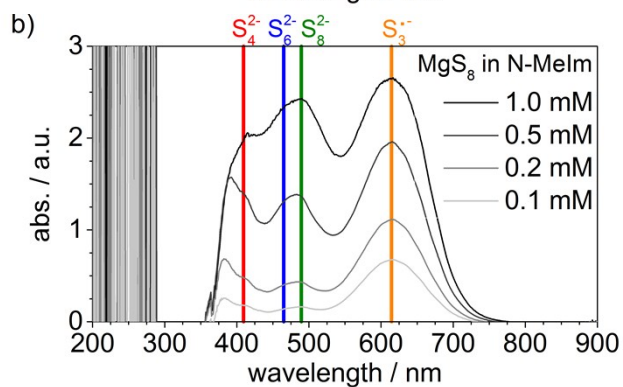
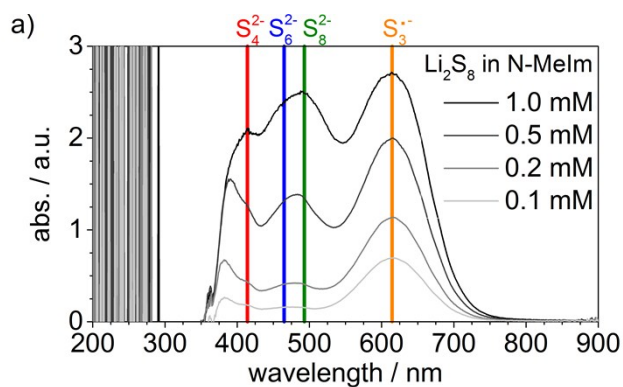


Figure S3: UV/Vis spectra of a) Li_2S_8 and b) MgS_8 solutions in N-Melm.

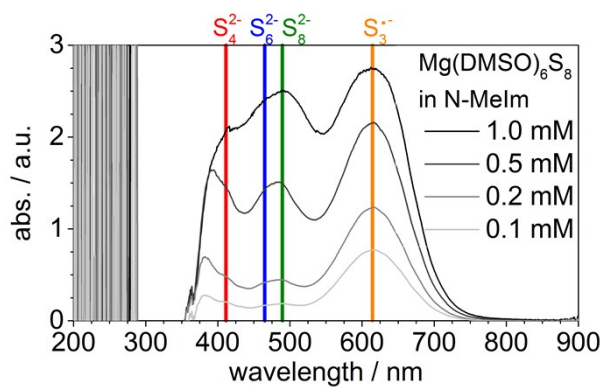


Figure S4: UV/Vis spectra of $\text{Mg}(\text{DMSO})_6\text{S}_8$ solutions in N-Melm.