Determinants of the host-guest interactions between

α-, β- and γ-cyclodextrins and group IA, IIA and IIIA metal cations: a DFT/PCM study

S. E. Angelova,^a V. K. Nikolova^b and T. M. Dudev^{c,*}

^a Institute of Organic Chemistry with Centre of Phytochemistry, Bulgarian Academy of Sciences, 1113 Sofia,

Bulgaria.

^b Faculty of Chemistry and Pharmacy, Sofia University "St. Kl. Ohridski", 1164 Sofia, Bulgaria. Email:t.dudev@chem.uni-sofia.bg

Supplementary Information

Experimental structural data from $[Mg(H_2O)_2(15\text{-}crown-5)]^{2+}$ cation of $[Mg(H_2O)_2(15\text{-}crown-5)]_3[NO_3]_6$ complex have been used to calibrate the theoretical method employed. An averaged experimental distance of 2.190 Å was calculated from the experimental Mg-O_{crown} distances of the X-ray structure determined by Junk and Steed^[S1] (Figure S1):



Figure S1.

Table S1. Experimental and calculated mean Mg^+-O_{crown} distances in $[Mg(H_2O)_2(15\text{-}crown-5)]^{2+}$ cation of $[Mg(H_2O)_2(15\text{-}crown-5)]_3[NO_3]_6$ complex.

	Mg-O _{crown} distance, Å	Exp. – calc., Å
Exp. ^[S1]	2.190	
M062X/6-31G(d,p)	2.179	0.011
M062X/6-31+G(d,p)	2.178	0.012
M062X/6-31+G(2d,p)	2.176	0.014
M062X/6-311G(d,p)	2.174	0.016
M062X/6-311+G(d,p)	2.173	0.017
B3LYP/6-31G(d,p)	2.207	-0.017
B3LYP/6-31+G(d,p)	2.205	-0.015
B3LYP/6-31+G(2d,p)	2.204	-0.014
B3LYP/6-311G(d,p)	2.203	-0.013
B3LYP/6-311+G(d,p)	2.202	-0.012

^[S1] P. C. Junk, J. W. Steed, J. Chem. Soc., Dalton Trans. 1999, 407-414.