

Electronic Supplementary Information

## Tunable the electronic structure and magnetic moment of the C<sub>2</sub>N nanoribbons with different edge functionalization atoms

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**Table S1.** The lattice constants (Å), the C–C–N bond angle ( $\alpha$ ), the C–N–C bond angle( $\beta$ ), the C–C–C bond angle( $\gamma$ ), bond length (Å)in pristine and saturated systems

system	Z-C <sub>2</sub> NNR						A1-C <sub>2</sub> NNR	A2-C <sub>2</sub> NNR
	b	$\alpha$ (deg)	$\beta$ (deg)	$\gamma$ (deg)	d <sub>C-N</sub>	d <sub>C-C</sub>	a <sub>1</sub>	a <sub>2</sub>
sheet	14.428	118.836	117.620	119.980	1.336	1.469	8.330	8.330
pristine	14.470	137.20	163.110	123.100	1.238	1.382	8.359	8.364
H	14.428	120.67	120.980	120.410	1.321	1.443	8.332	8.333
O	14.420	113.400	111.763	114.466	1.359	1.527	8.340	8.343
F	14.437	124.24	126.780	123.620	1.304	1.449	8.334	8.334
OH	14.429	115.055	112.376	118.630	1.346	1.530	8.338	8.336
OF	14.438	116.087	109.822	122.888	1.344	1.517	8.339	8.334

**Table S2.** The different direction diameter ( $d_1$ ,  $d_2$ ,  $d_3$ ) of the inscribed circle formed by the nitride atoms in pristine and saturated systems for a series of ribbon widths

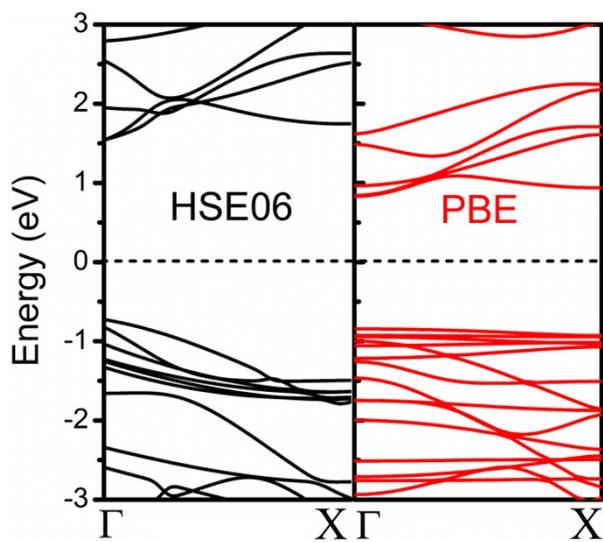
Width	C <sub>2</sub> NNR	Z-C <sub>2</sub> NNR			A1-C <sub>2</sub> NNR			A2-C <sub>2</sub> NNR		
		$d_1(\text{\AA})$	$d_2(\text{\AA})$	$d_3(\text{\AA})$	$d_1(\text{\AA})$	$d_2(\text{\AA})$	$d_3(\text{\AA})$	$d_1(\text{\AA})$	$d_2(\text{\AA})$	$d_3(\text{\AA})$
1	Pristine	6.636	5.576	5.689	5.568	5.540	5.542	5.585	5.615	5.611
	H	5.545	5.491	5.496	5.519	5.497	5.494	5.519	5.516	5.515
	O	5.400	5.601	5.579	5.533	5.528	5.522	5.541	5.532	5.525
	F	5.698	5.579	5.578	5.522	5.508	5.549	5.519	5.517	5.514
	OH	5.507	5.547	5.545	5.523	5.504	5.498	5.525	5.518	5.514
	OF	5.328	5.600	5.596	5.529	5.512	5.502	5.519	5.517	5.519
2	Pristine	6.116	5.591	5.599	5.543	5.535	5.521	5.553	5.599	5.600
	H	5.541	5.513	5.509	5.520	5.499	5.511	5.518	5.526	5.516
	O	5.511	5.572	5.562	5.526	5.525	5.517	5.529	5.518	5.526
	F	5.630	5.549	5.551	5.521	5.502	5.514	5.519	5.525	5.513
	OH	5.520	5.533	5.536	5.524	5.504	5.514	5.521	5.515	5.526
	OF	5.441	5.554	5.561	5.524	5.508	5.517	5.519	5.512	5.530
3	Pristine	6.123	5.583	5.576	5.534	5.532	5.521	5.541	5.598	5.601
	H	5.526	5.516	5.509	5.519	5.504	5.509	5.519	5.521	5.530
	O	5.512	5.575	5.568	5.523	5.511	5.528	5.526	5.525	5.520
	F	5.621	5.549	5.544	5.520	5.506	5.511	5.519	5.524	5.514
	OH	5.503	5.530	5.533	5.520	5.506	5.510	5.521	5.525	5.523
	OF	5.543	5.550	5.551	5.521	5.515	5.508	5.520	5.526	5.525
4	Pristine	6.125	5.578	5.559	5.531	5.525	5.532	5.519	5.542	5.552
	H	5.536	5.517	5.500	5.519	5.509	5.506	5.518	5.528	5.532
	O	5.507	5.578	5.565	5.523	5.528	5.530	5.524	5.525	5.526
	F	5.621	5.550	5.541	5.519	5.507	5.511	-----	-----	-----
	OH	5.511	5.537	5.523	5.520	5.503	5.513	5.521	5.527	5.520
	OF	5.444	5.554	5.541	5.520	5.510	5.515	-----	-----	-----

**Table S3.** The energy differences of FM and AFM1, AFM2, and AFM3 with respect to NM phase for Z-C<sub>2</sub>NNR, taking the energy of NM phase as zero

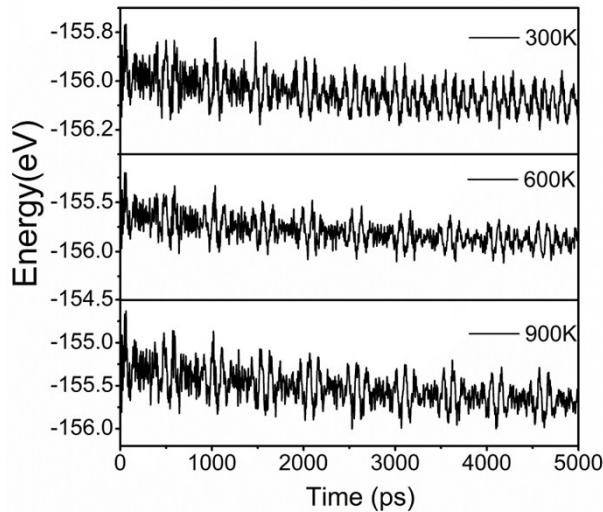
Magnetic phase	Magnetic ordering		Energy (meV)
	Edge 1	Edge 2	
NM			0
FM	↑ ↑	↑ ↑	-523.77
AFM1	↑ ↑	↓ ↓	-497.5
AFM2	↑ ↓	↑ ↓	-517.34
AFM3	↑ ↓	↓ ↑	-498.91

**Table S4.** The calculated binding energy E<sub>b</sub>(eV) of C<sub>2</sub>NNRs with H, O and F saturated for a series of ribbon widths

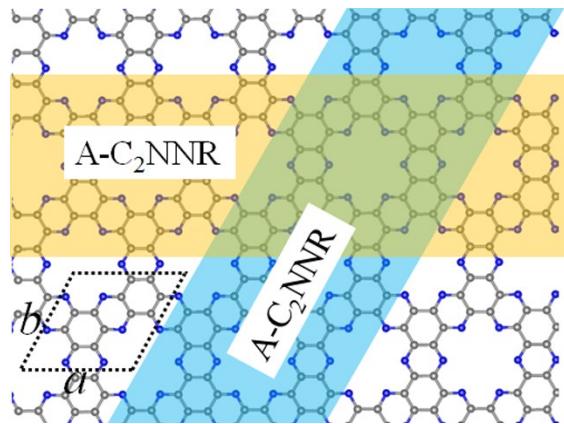
Width \ C <sub>2</sub> NNR	Z-C <sub>2</sub> NNR			A1-C <sub>2</sub> NNR			A2-C <sub>2</sub> NNR		
	H	O	F	H	O	F	H	O	F
1	4.142	7.402	5.146	5.371	7.431	5.332	5.065	6.952	4.321
2	4.154	7.421	5.161	5.376	7.435	5.336	5.068	6.954	4.348
3	4.158	7.421	5.162	5.376	7.435	5.337	5.071	6.957	4.351
4	4.158	7.422	5.162	5.378	7.438	5.337	4.796	6.681	1.962



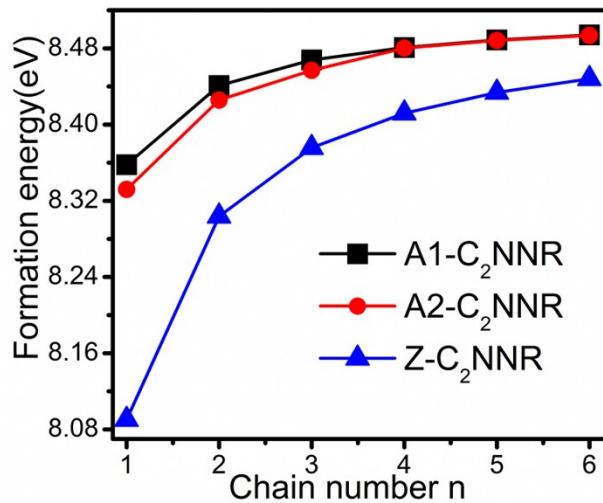
**Figure S1.** The band structure of H saturated Al-C<sub>2</sub>NNR calculated by HSE06 and PBE method respectively.



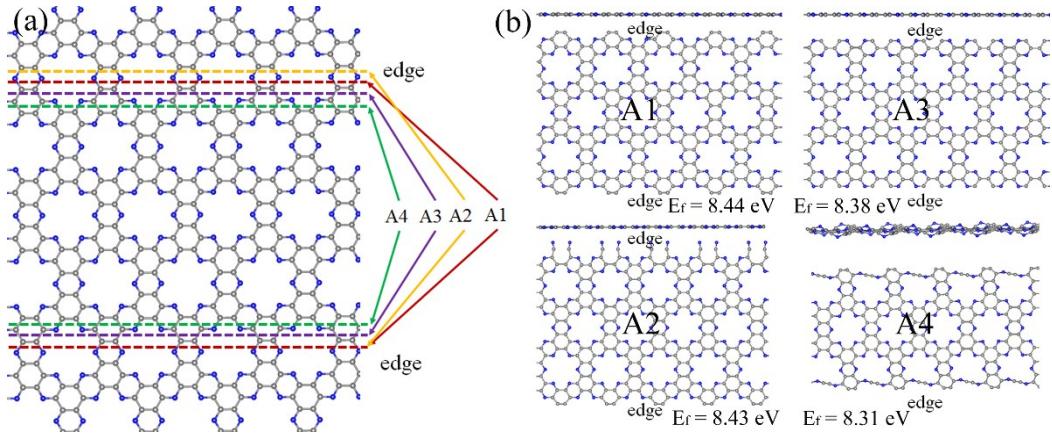
**Figure S2.** Relationships of total energy and time during 300K, 600K and 900K MD simulations of monolayer C<sub>2</sub>N.



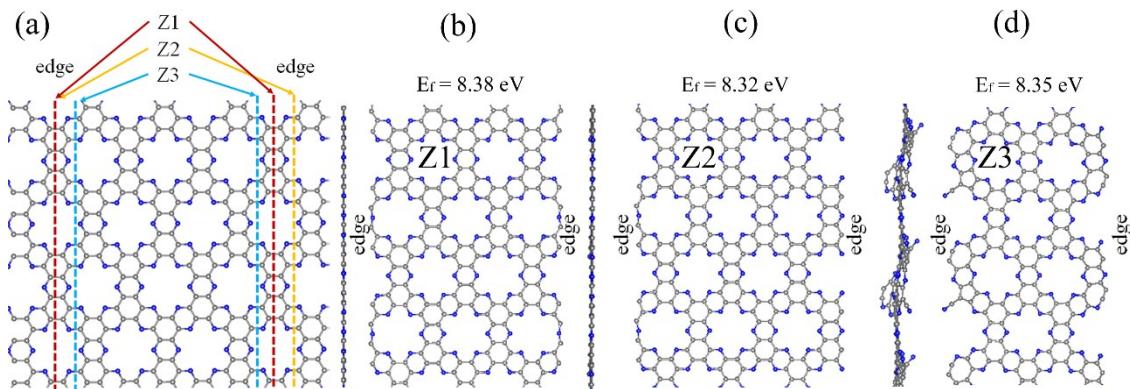
**Figure S3.** The atomic structure of armchair nanoribbons slicing along a and b direction of unit cell.



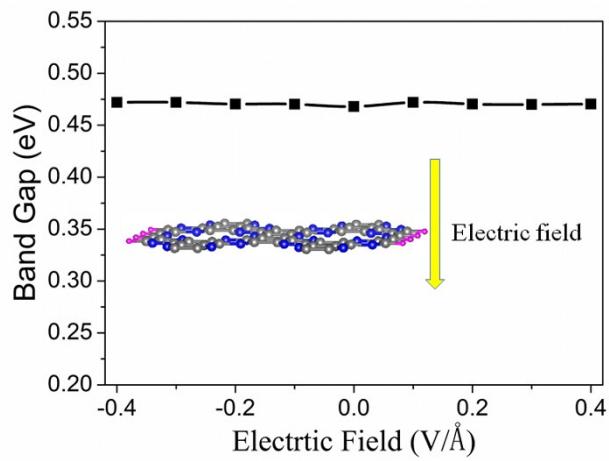
**Figure S4.** The formation energy of different C<sub>2</sub>NNRs depending on the chain number n.



**Figure S5.** (a) The atomic structure of different armchair nanoribbons, named as A1, A2, A3 and A4 C<sub>2</sub>NNRs. (b) The optimized atomic structures of armchair nanoribbons.



**Figure S6.** (a) The atomic structure of different zigzag nanoribbons, named as Z1-C<sub>2</sub>NNRs. The optimized atomic structures of (b) Z1-C<sub>2</sub>NNR, (c) Z2-C<sub>2</sub>NNR and (d) Z3-C<sub>2</sub>NNR.



**Figure S7.** The band gap of H-Z-C<sub>2</sub>NNR depending on the vertical electric field.