## ELECTRONIC SUPPLEMENTARY INFORMATION

for

## Can Kohn-Sham Density Functional Theory Predict Accurate Charge Distributions for Both Single-Reference and Multi-Reference Molecules?

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molecule	BLYP	GAM	HCTH/407	HLE16	HSE06	M06-L	M11-L	MN15-L	PBE	<b>PBE0</b>	PBEsol	TPSS	HLE17	Expt.
def2-TZVP														
CrO	3.79	3.71	3.94	4.52	4.23	4.12	4.40	4.40	3.68	4.26	3.59	3.85	4.50	3.88
CuF	4.67	4.92	4.97	5.76	5.33	4.90	4.98	5.04	4.69	5.37	4.62	4.81	5.57	5.77
CuO	4.19	4.25	4.42	5.07	4.75	4.50	4.55	4.51	4.26	4.79	4.25	4.40	4.98	4.45
$H_2CO$	2.21	2.15	2.15	2.27	2.32	2.08	1.96	2.16	2.16	2.31	2.16	2.23	2.41	2.33
$H_2O$	2.03	2.01	2.01	2.00	2.07	2.03	2.01	2.03	2.04	2.07	2.07	2.04	2.01	1.85
$H_2S$	1.07	1.15	1.13	1.13	1.16	1.13	1.07	1.18	1.13	1.16	1.17	1.13	1.07	0.98
HCN	2.95	2.92	2.92	2.95	3.04	2.94	2.92	3.01	2.93	3.03	2.96	2.96	2.98	2.99
HF	1.90	1.88	1.89	1.87	1.93	1.89	1.85	1.90	1.90	1.93	1.92	1.90	1.89	1.83
$HfO_2$	7.44	7.40	7.45	7.61	8.09	8.01	7.90	8.06	7.44	8.10	7.42	7.71	8.01	7.92
HfO	3.39	3.24	3.49	4.15	3.46	3.16	3.29	3.55	3.24	3.43	3.20	3.47	4.49	3.43
LaO	4.33	3.78	3.70	5.32	4.10	3.87	4.39	3.96	3.99	4.00	3.95	3.94	5.32	3.21
$N_2O$	0.09	0.22	0.18	0.19	0.02	0.17	0.18	0.06	0.16	0.02	0.20	0.10	0.02	0.16
$NH_3$	1.67	1.66	1.67	1.57	1.69	1.70	1.72	1.71	1.70	1.69	1.71	1.70	1.58	1.47
PbO	4.41	4.44	4.39	4.79	4.67	4.55	4.22	4.49	4.32	4.66	4.24	4.40	4.95	4.64
PbS	3.88	3.99	4.01	4.42	4.11	4.25	3.82	4.27	3.89	4.15	3.75	4.04	4.51	3.59
$PH_3$	0.57	0.70	0.67	0.75	0.69	0.65	0.55	0.66	0.65	0.70	0.70	0.68	0.69	0.57
$SO_2$	1.69	1.54	1.57	1.50	1.70	1.56	1.49	1.53	1.60	1.69	1.58	1.60	1.64	1.63
TiO	3.71	3.43	3.39	4.52	3.86	3.80	3.80	3.76	3.54	3.81	3.47	3.67	3.88	2.96
YO	4.72	4.35	4.10	4.08	4.65	4.64	4.00	4.59	4.41	4.55	4.39	4.44	4.34	4.52
$ZrO_2$	7.49	7.41	7.38	7.25	7.93	7.94	7.77	8.01	7.45	7.92	7.42	7.68	7.72	7.8
ZrO	3.19	2.90	3.17	3.98	3.37	3.19	3.14	3.12	3.12	3.39	3.04	3.40	4.54	2.55
						def2	-QZVP							
CrO	3.66	3.66	3.93	4.56	4.09	3.99	4.08	4.25	3.54	4.13	3.45	3.73	4.48	3.88
CuF	4.55	4.84	4.83	5.59	5.21	4.81	4.75	4.94	4.56	5.25	4.49	4.66	5.41	5.77
CuO	4.16	4.23	4.35	4.99	4.69	4.47	4.42	4.47	4.22	4.74	4.21	4.35	4.90	4.45
$H_2CO$	2.25	2.19	2.20	2.29	2.35	2.11	2.00	2.22	2.21	2.34	2.20	2.27	2.42	2.33
$H_2O$	1.87	1.84	1.87	1.88	1.91	1.85	1.85	1.87	1.87	1.91	1.89	1.87	1.87	1.85
$H_2S$	0.98	1.07	1.04	1.05	1.05	1.03	0.99	1.10	1.02	1.06	1.05	1.02	0.98	0.98
HCN	2.96	2.94	2.95	2.96	3.04	2.92	2.93	3.03	2.95	3.04	2.97	2.96	2.98	2.99

**Table S1.** Comparison of def2-TZVP and def2-QZVP basis sets for dipole moments (in D) of 21 molecules calculated with 13 density functionals.

	BLYP	GAM	HCTH/407	HLE16	HSE06	M06-L	M11-L	MN15-L	PBE	<b>PBE0</b>	PBEsol	TPSS	HLE17	Expt.
MUE <sup><i>a,c</i></sup>	0.27	0.24	0.22	0.44	0.21	0.20	0.25	0.22	0.26	0.21	0.27	0.23	0.42	0.00
MUE $^{a,b}$	0.30	0.27	0.26	0.49	0.28	0.27	0.31	0.28	0.29	0.27	0.30	0.26	0.46	0.00
ZrO	3.15	2.83	3.10	3.74	3.27	2.90	2.87	3.05	3.07	3.30	2.97	3.33	4.36	2.55
$ZrO_2$	7.45	7.37	7.34	7.15	7.85	7.83	7.64	8.00	7.39	7.85	7.36	7.60	7.61	7.8
YO	4.71	4.25	4.04	4.02	4.58	4.43	3.82	4.58	4.38	4.48	4.34	4.36	4.23	4.52
TiÔ	3.48	3.15	3.15	4.37	3.52	3.39	3.15	3.39	3.26	3.45	3.19	3.37	3.71	2.96
SO <sub>2</sub>	1.66	1.50	1.55	1.49	1.65	1.49	1.44	1.50	1.57	1.65	1.55	1.57	1.62	1.63
$PH_3$	0.53	0.67	0.63	0.72	0.65	0.63	0.56	0.65	0.60	0.66	0.64	0.63	0.65	0.57
PbS	3.92	4.08	4.06	4.38	4.11	4.25	3.79	4.35	3.90	4.15	3.75	4.02	4.45	3.59
PbO	4.51	4.56	4.50	4.83	4.73	4.64	4.26	4.62	4.40	4.73	4.30	4.45	4.97	4.64
$\tilde{NH_3}$	1.52	1.51	1.53	1.48	1.55	1.54	1.55	1.56	1.54	1.55	1.55	1.53	1.47	1.47
$N_2O$	0.07	0.20	0.16	0.18	0.03	0.16	0.17	0.03	0.14	0.03	0.18	0.09	0.01	0.16
LaO	4.25	3.77	3.59	4.81	3.94	3.79	4.19	3.79	3.88	3.84	3.82	3.81	5.03	3.21
HfO	3.38	3.21	3.45	4.03	3.42	3.13	3.20	3.45	3.24	3.39	3.17	3.47	4.41	3.43
$HfO_2$	7.45	7.42	7.47	7.56	8.06	7.92	7.79	8.10	7.44	8.07	7.41	7.68	7.95	7.92
HF	1.81	1.78	1.80	1.80	1.84	1.79	1.76	1.82	1.80	1.83	1.82	1.81	1.81	1.83

 $a^{a}$ MUE = mean unsigned error  $b^{b}$ def2-TZVP  $c^{c}$ def2-QZVP

molecule	spin state	$\Delta E$	< <u>S</u> <sup>2</sup> >	dipole moment
CrN	quartet	0.0	4.07	3.06
	doublet	12.9	1.93	2.38
CrO	quintet	0.0	6.16	3.99
	triplet	14.8	2.77	2.29
CuF	singlet	0.0	_b	4.81
	triplet	42.5	2.00	1.95
CuO	doublet	0.0	0.77	4.47
	quartet	35.4	3.77	0.11
$HfO_2$	singlet	0.0	_b	7.92
	triplet	34.0	2.03	4.45
HfO	singlet	0.0	_b	3.13
	triplet	27.8	2.00	4.71
LaO	doublet	0.0	0.75	3.79
	quartet	77.9	3.76	3.23
TiO	triplet	0.0	2.02	3.39
	singlet	-0.8	0.94	2.84
YO	doublet	0.0	0.75	4.43
	quartet	79.7	3.76	3.14
$ZrO_2$	singlet	0.0	_b	7.83
	triplet	49.0	2.03	4.15
ZrO	singlet	0.0	0.80	2.90
	triplet	9.0	2.00	3.77
FeO	quintet	0.0	6.11	4.31
	triplet	24.5	3.03	4.21
NiH	doublet	0.0	0.77	2.48
	quartet	33.8	3.75	0.41
ScF	singlet	0.0	0.74	2.29
	triplet	11.9	2.00	2.71
TiH	quartet	0.0	3.76	3.64
	doublet	4.4	1.59	2.33
VN	triplet	0.0	2.07	4.12
	singlet	5.8	1.03	2.82
VO	quartet	0.0	3.81	3.32
	doublet	10.6	1.75	2.42

**Table S2.** Spin-splitting energies ( $\Delta E$  in kcal/mol),<sup>*a*</sup> <*S*<sup>2</sup>> values, and dipole moments (in D) using M06-L/def2-QZVP.

<sup>*a*</sup>For each molecule,  $\Delta E$  is calculated with respect to the energy of the experimental ground spin state. <sup>b</sup>These singlet calculations were restricted Kohn-Sham calculations.

E	C-F		Man	DLO	DLC
runctional			MgU		2.02
BLYP	4.55	4.16	6.95	4.51	3.92
PBE	4.56	4.22	7.04	4.40	3.90
HCTH/407	4.83	4.35	7.23	4.50	4.06
OLYP	4.77	4.34	7.14	4.47	4.09
PBEsol	4.49	4.21	7.10	4.30	3.75
OreLYP	4.72	4.30	7.10	4.44	4.05
SOGGA11	4.71	4.59	7.76	4.43	4.38
HLE16	5.59	4.99	8.16	4.83	4.38
N12	4.79	4.22	7.12	4.55	3.98
GAM	4.84	4.23	7.02	4.56	4.08
τ-HCTH	4.92	4.48	7.38	4.44	3.95
TPSS	4.66	4.35	7.21	4.45	4.02
M06-L	4.81	4.47	7.09	4.64	4.25
M11-L	4.75	4.42	7.20	4.26	3.79
MGGA_MS2	4.50	4.21	6.88	4.24	3.85
HLE17	5.41	4.90	7.91	4.97	4.45
MN12-L	4.74	4.44	7.08	4.25	3.88
MN15-L	4.94	4.47	7.27	4.62	4.35
BHandHLYP	5.78	5.28	8.11	5.13	4.40
B3LYP	5.12	4.59	7.45	4.78	4.12
B1LYP	5.24	4.68	7.52	4.85	4.18
mPW1PW	5.26	4.76	7.71	4.74	4.15
B97-1	5.20	4.70	7.53	4.75	4.21
PBE0	5.25	4.74	7.67	4.73	4.15
MPW1K	5.66	5.18	8.19	4.95	4.31
B3LYP*	4.97	4.48	7.34	4.70	4.05
CAM-B3LYP	5.31	4.87	7.93	4.92	4.24
MPW3LYP	5.14	4.61	7.48	4.79	4.12
B97-3	5.38	4.85	7.57	4.89	4.36
LC-wPBE	5.41	5.14	8.63	4.93	4.42
HSE06	5.21	4.69	7.64	4.73	4.11
SOGGA11-X	5.49	4.95	7.95	4.98	4.33
N12-SX	5.20	4.75	7.79	4.82	4.21
$\tau_{-}$ HCTHhyb	5.05	4 58	7 53	4.63	4 04
TPSSh	1 94	4.50	7.55	4.58	4.04
MPWR1K	5.64	5.14	8.04	4.96	4.26
M05	5 37	5.14 4 77	7.82	5.01	4.20
M05 2Y	5.87	4.77 5.37	8 32	5.01	4.38
DW6R05	5.31	5.57 A 77	0.52 7.61	J.21 4.82	4.58
	622	4.// 6.03	0.00	4.02 5.25	4.13
M06	0.33 5 15	0.05	7.09 7.69	J.JJ 1 02	4.20
MOG 2V	5.15	4.30	7.02	4.73	4.24
MOQ UV	5.00	J.33 5 52	/.74	5.20	4.37
ΝΙΟδ-ΗΧ	3.98	5.55	ð.41	5.50	4./1

**Table S3.** Dipole moments (in D) of CuF, CuO, MgO, PbO and PbS calculated with 47 density functionals and the def2-QZVP basis set.

M08-SO	5.94	5.40	7.92	5.15	4.24
M11	6.04	5.67	9.17	5.10	4.31
MN12-SX	5.39	4.91	7.60	4.65	3.97
MN15	5.45	5.00	7.67	4.82	4.13
Expt.	5.77	4.45	6.2	4.64	3.59