Electronic Supplementary Information

Silicon doped boron clusters: how to make stable ribbons?

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Figure S1. a) 16 localized CMOs and b) the ELF(σ _loca) of $B_{12}Si_2^{2-}$.



a) $B_{14}H_2^{2-}$



b) ELF(π) and ELF(σ _delo) for the B₁₄H₂²⁻



c) $B_{12}C_2H_2$



d) ELF(π) and ELF(σ _delo) for the B₁₂C₂H₂



e) $B_{14}Li_2H_2$



f) ELF(π) and ELF(σ _delo) for the B₁₄Li₂H₂

Figure S2. Ribbon structures of (a) $B_{14}H_2^{2-}$, (c) $B_{12}C_2H_2$, and (e) $B_{14}Li_2H_2$. The ELF(π) (red) and ELF(σ _delo) (green) are plotted simultaneously for (b) the $B_{14}H_2^{2-}$, (d) $B_{12}C_2H_2$ and (f) $B_{14}Li_2H_2$. The bond length (Å) is given in Figures (a), (c), and (e). Yellow areas indicate places affected by π delocalized electrons while green bonds are affected by σ delocalized electrons.



a) ELF(π) and ELF(σ _delo) for the α spin electrons of B₁₀Si₂²⁻ triplet



b) ELF(π) and ELF(σ _delo) for the β spin electrons of $B_{10}Si_2^{2-}$ triplet

Figure S3. The ELF(π) (red) and ELF(σ _delo) (green) are plotted simultaneously for (a) α spin electrons, and (b) β spin electrons of triplet B₁₀Si₂²⁻.



Figure S4. Ribbon structure of $B_{12}H_2^{2-}$ in (a) singlet and (c) triplet states. ELF(π) (red) and ELF(σ _delo) (green) are plotted simultaneously for (b) singlet $B_{10}H_2^{2-}$ and (d) triplet $B_{10}H_2^{2-}$. Bond lengths (Å) are given in Figure (a) and (c). Yellow areas indicate the places affected by π delocalized electrons while green bonds are effected by σ delocalized electrons.



Figure S5. Ribbon structure of $B_{12}Li_2H_2$ in (a) singlet and (c) triplet states. ELF(π) (red) and ELF(σ _delo) (green) are plotted simultaneously for (b) singlet $B_{12}Li_2H_2$ and (d) triplet $B_{12}Li_2H_2$. Bond lengths (Å) are given in Figure (a) and (c). Yellow areas indicate the places affected by π delocalized electrons while green bonds are effected by σ delocalized electrons.



Figure S6. Ribbon structure of B_{14} . ELF(π) (red) and ELF(σ _delo) (green) are plotted simultaneously for (b) singlet $B_{12}Li_2H_2$ and (d) triplet $B_{12}Li_2H_2$. Bond lengths (Å) are given in Figure (a) and (c). Yellow areas indicate the places affected by π delocalized electrons while green bonds are effected by σ delocalized electrons.





Figure S7. The ELF(σ _loca) of (a) B₁₄, (b) B₁₄H₂²⁻, (c) B₁₄Li₂H₂²⁻, and (d) B₁₂Si₂²⁻.



Figure S8. Optimized structures of (a) C_4H_6 , (b) C_6H_8 , (c) C_8H_{10} , (d) $C_{10}H_{12}$, and (e) $C_{12}H_{14}$ using DFT with the PBE1 functional and the 6-311+G(d,p) basis set.