Supporting Information

More Accurately Depicting Adsorption Energy on Transition Metal using

Work Function as One Additional Descriptor

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Figure S1. (A) Different adsorption sites on transition metal (111) surface. (B) Side view of the transition metal surface model.



Figure S2. Top view of adsorption model of (A) O, (B) OH, (C) OOH species on transition metal (111) surface. (•: oxygen, •: hydrogen)



Figure S3. $\Delta q - W$ plots and linear correlation for O, OH and OOH adsorption to transition metal (111) surface.



Figure S4. A proportional correlation between Δq and ΔW for O adsorption to transition metals.



Figure S5. A proportional correlation between Δq and ΔW for OH adsorption to transition metals.



Figure S6. A proportional correlation between Δq and ΔW for OOH adsorption to transition metals.



Figure S7. The E_{ionic} – ΔW correlation for O, OH and OOH adsorption to transition metals.

	ε _d /	W/eV	$E_{ads}(O) / eV$	E _{ads} (OH) / eV	E _{ads} (OOH) / eV
Au(111)	-3.55	4.96	-0.62	-2.47	-1.00
Cu(111)	-2.60	4.62	-2.25	-3.58	-2.05
Pd(111)	-2.17	5.20	-1.88	-3.04	-1.72
Pt(111)	-2.78	5.24	-1.58	-2.49	-1.56
Rh(111)	-2.47	4.91	-2.45	-3.59	-2.25
Ag(111)	-4.27	4.26	-1.05	-3.28	-1.72
Ni(111)	-1.66	4.98	-3.08	-3.75	-2.20

Table S1. DFT calculated $\varepsilon_{\rm d}$, *W* and adsorption energies.

		0	ОН	ООН
E _{ionic} –W	k	-0.13	-0.11	-0.29
	W ₀	8.97	8.10	5.99
	R_{adj}^2	0.8524	0.9647	0.9497
$E_{\rm covalent}$ — $\varepsilon_{\rm d}$	slope intercept	-1.23	-0.48	-0.43
		-3.08	-3.36	-2.59
	R_{adj}^2	0.9357	0.5893	0.6747

Table S2. Fitting results of the E_{ionic} -W and the E_{covalent} - ε_{d} relation.

	0	ОН	ООН
Z ₀	-2.29	-2.36	-2.83
а	-1.19	-0.70	-0.60
k	-0.17	-0.26	-0.47
W ₀	8.97	8.10	5.99
R_{adj}^2	0.9059	0.9053	0.7142
$R_{adj}^2 (E_{ads} - \varepsilon_d)^*$	0.6865	0.0116	0.2096

Table S3. Fitting results of the E_{ads} –(ε_d , W) model and E_{ads} – ε_d model.

* $R_{adj}^2(E_{ads}-\varepsilon_d)$ values are for the $E_{ads}-\varepsilon_d$ linear fitting correlation.