Supporting Information for:

Water effect on effective Goldschmidt tolerance factor and photoelectric conversion efficiency of organic-inorganic perovskite: insight from first-principles calculations

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S1. The geometrical configuration of interstitial H₂O molecule infiltrated into the MAPbI₃ supercells with different interstitial H₂O ratios

Figure S1 (a) shows the geometrical configuration of pure $2 \times 2 \times 1$ MAPbI₃ supercell. When a single interstitial H₂O molecule infiltrate into the $2 \times 2 \times 1$ MAPbI₃ supercell, the two hydrogen atoms of interstitial H₂O molecule will bonding with iodine anion by weak hydrogen-bonds, as shown in the Figure S1 (b). However, there are kinds of geometric constructions for two interstitial H₂O molecules in the 2×2×1 MAPbI₃ supercell. Considered the crystal structure and symmetry, 13 different configurations of two interstitial H₂O molecules in the MAPbI₃ supercell were chosen as the possible structures in our calculations, as shown in Figure S1 (c-o). Besides, the corresponding total energies and lattice parameters are listed on the Table S1. We note that the 2H₂O-1 structure has the lowest total energy of 0.004-0.328 eV among other considered structures. In the 2H₂O-1 structure, two H₂O molecules are located on the two sides of two nearby Pb-I6 octahedron. Symmetrical water molecule position results in the smallest total additional dipole moment and crystal lattice distortion. Thus, 2H₂O-1 structure is the most stable configure with the lowest total energy. Besides, we note that the 2H₂O-2 structure also has small crystal lattice distortion and total energy. Base on the 2H₂O-1 and 2H₂O-2 configure in the MAPbI₃ structure, we built the related configures of the 3, 4, 8, 12, and 16 interstitial H₂O molecule in the MAPbI₃ structure, shown in the Figure S1 (p-t), respectively.

Structure	E (eV)	a (Å)	b (Å)	c (Å)	α(°)	β(°)	γ(°)	V (Å ³)
2H ₂ O-1	0	17.472	17.438	12.888	88.744	90.484	89.897	3926.234
2H ₂ O-2	0.004	17.458	17.452	12.893	88.877	90.468	89.824	3927.460
2H ₂ O-3	0.236	17.448	17.450	12.911	89.273	90.251	89.942	3931.096
2H ₂ O-4	0.207	17.435	17.473	12.907	88.970	90.268	89.899	3931.387
2H ₂ O-5	0.309	17.501	17.437	12.890	89.252	89.888	89.848	3933.535
2H ₂ O-6	0.241	17.474	17.419	12.905	88.908	90.610	89.824	3927.636
2H ₂ O-7	0.164	17.482	17.419	12.890	88.714	90.668	89.813	3924.269
2H ₂ O-8	0.194	17.484	17.432	12.901	89.247	90.251	89.860	3931.886
2H ₂ O-9	0.154	17.484	17.426	12.896	88.887	90.501	89.826	3928.858
2H ₂ O-10	0.328	17.481	17.441	12.904	88.910	90.146	89.924	3934.129
2H ₂ O-11	0.293	17.422	17.494	12.908	89.051	90.133	89.932	3933.693
2H ₂ O-12	0.304	17.404	17.514	12.898	88.971	90.258	89.813	3931.202
2H ₂ O-13	0.256	17.474	17.453	12.892	89.133	90.521	89.744	3931.453

Table S1 The formation energies and lattice parameters of two interstitial H_2O molecules in the MAPbI₃ supercell.



Figure S1 Snapshots of Li-ion transfer in the LiBOB-DEC (a-c), LiBOB-MPN (d-f) and LiBOB-GLN (g-i) system. The yellow, red, pink, light green, blue, and white balls represent Li, O, B, C, N, and H atoms, respectively.

S2. The ground state electronic structure of interstitial H₂O molecule infiltrated MAPbI₃ supercells with different ratios of MAPbI₃ and H₂O

It is well known that primitive MAPbI₃ structure is a direct band gap semiconductor with a value of 1.50 eV, as shown in the Figure S2 (a). Thus, it is necessary to know whether the MAPbI₃ structures with interstitial H₂O molecules own unique electronic structures. In order to study the detail electronic structures of the interstitial H₂O molecule infiltrated MAPbI₃ structures, the ground state band structures for the MA(H₂O)_xPbI₃ with seven different interstitial H₂O ratios are investigated, and the calculated results are shown in Figure S2 (b-h). The band structures show that the MA(H₂O)_xPbI₃ structures are semiconductors with direct band gaps at the Gamma point. Besides, the band structures of MA(H₂O)_xPbI₃ structures are similar with that of the pure MAPbI₃ structure, which are also semiconductors with direct band gaps. However, the band gaps of MA(H₂O)_xPbI₃ structures will increase with the increasing interstitial H_2O ratio x. That is to say, all $MA(H_2O)_xPbI_3$ structures exhibit different kinds of optical absorption properties.



Figure S2 The calculated band structures of the H_2O molecular insert MAPbI₃ with different $H_2O:MAPbI_3$ ratios. The (a-h) represent the band structure of the H_2O molecular insert MAPbI₃ with $H_2O:MAPbI_3$ ratio of 0, 1/16, 1/8, 3/16, 1/4, 1/2, 3/4, and 1, respectively.