

SUPPORTING INFORMATION

Heme isomers substantially affect heme electronic structure and function

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Table S1. Electronic energies (E_{el}) and zero-point energies (ZPE) from numerically computed frequencies of geometry-optimized states (BP/def2-SVP COSMO, $\epsilon = 10$); in atomic units. Q = total charge; M = multiplicity; and thermodynamic corrections to the energy ΔE_{therm} , entropy ΔS , and free energy ΔG .

Model	State	Q	M	E_{el} (a.u.)	ZPE (a.u.)	ΔE_{therm} (kJ/mol)	ΔS (kJ/molK)	ΔG (kJ/mol)
EEO	Fe(II) LS	-2	1	-3096.5373	0.5529	1546.0	1.015	1245.9
	Fe(II) HS	-2	5	-3096.5313	0.5647	1566.2	0.885	1304.8
	Fe(III) LS	-1	2	-3096.4030	0.5525	1545.6	1.019	1244.2
	Fe(III) HS	-1	6	-3096.3864	0.5648	1566.9	0.890	1304.2
EEO2	Fe(II) LS	-2	1	-3512.2062	0.6631	1851.1	1.110	1522.6
	Fe(II) HS	-2	5	-3512.1731	0.6510	1826.9	1.229	1463.0
	Fe(III) LS	-1	2	-3512.0657	0.6601	1847.2	1.164	1502.7
	Fe(III) HS	-1	6	-3512.0248	0.6571	1842.0	1.191	1489.5
EEcpl	doublet	-1	2	-3436.9559	0.6550	1829.8	1.128	1495.9
	quartet	-1	4	-3436.9565	0.6571	1835.5	1.123	1503.3
EZO	Fe(II) LS	-2	1	-3096.5396	0.5511	1540.2	1.021	1238.2
	Fe(II) HS	-2	5	-3096.5326	0.5495	1537.4	1.023	1234.8
	Fe(III) LS	-1	2	-3096.4051	0.5518	1542.2	1.004	1245.3
	Fe(III) HS	-1	6	-3096.3884	0.5501	1540.9	1.036	1234.4
EZO2	Fe(II) LS	-2	1	-3512.2079	0.6612	1848.1	1.133	1512.7
	Fe(II) HS	-2	5	-3512.1748	0.6509	1829.1	1.258	1456.6
	Fe(III) LS	-1	2	-3512.0672	0.6603	1847.1	1.148	1507.2
	Fe(III) HS	-1	6	-3512.0266	0.6573	1839.4	1.143	1501.2
EZcpl	doublet	-1	2	-3436.9575	0.6563	1834.2	1.137	1497.8
	quartet	-1	4	-3436.9578	0.6525	1827.8	1.188	1476.1

ZEO	Fe(II) LS	-2	1	-3096.5383	0.5507	1540.1	1.042	1232.0
	Fe(II) HS	-2	5	-3096.5332	0.5493	1539.5	1.054	1227.8
	Fe(III) LS	-1	2	-3096.4037	0.5516	1541.6	1.011	1242.7
	Fe(III) HS	-1	6	-3096.3859	0.5493	1537.3	1.036	1230.8
ZEO2	Fe(II) LS	-2	1	-3512.2079	0.6608	1847.4	1.139	1510.2
	Fe(II) HS	-2	5	-3512.1750	0.6592	1844.0	1.138	1507.3
	Fe(III) LS	-1	2	-3512.0673	0.6603	1847.1	1.151	1506.5
	Fe(III) HS	-1	6	-3512.0266	0.6577	1842.3	1.161	1498.7
ZEcpl	doublet	-1	2	-3436.9576	0.6569	1835.2	1.126	1502.0
	quartet	-1	4	-3436.9578	0.6582	1837.1	1.107	1509.5
ZZO	Fe(II) LS	-2	1	-3096.5396	0.5523	1545.0	1.043	1236.4
	Fe(II) HS	-2	5	-3096.5343	0.5608	1558.6	0.910	1289.6
	Fe(III) LS	-1	2	-3096.4055	0.5523	1543.1	1.004	1246.1
	Fe(III) HS	-1	6	-3096.3899	0.5503	1541.3	1.035	1235.1
ZZO2	Fe(II) LS	-2	1	-3512.2133	0.6549	1838.3	1.243	1470.1
	Fe(II) HS	-2	5	-3512.1760	0.6513	1829.6	1.249	1459.6
	Fe(III) LS	-1	2	-3512.0737	0.6551	1836.4	1.196	1482.4
	Fe(III) HS	-1	6	-3512.0280	0.6578	1842.4	1.162	1498.5
ZZcpl	doublet	-1	2	-3436.9568	0.6582	1837.0	1.108	1509.0
	quartet	-1	4	-3436.9584	0.6580	1837.0	1.114	1507.4

Table S2. Electronic energies (E_{el}) in atomic units of models and states computed by TPSSh, B3LYP, and PBE with and without dispersion D3 corrections, using the def2-TZVPP basis set. Q = total charge; M = multiplicity.

Model	State	Q	M	TPSSh-D3	D3	B3LYP-D3	D3	PBE-D3	D3
EEO	Fe(II) LS	-2	1	-3098.5947	-0.0750	-3097.3200	-0.0918	-3096.0069	-0.0550
	Fe(II) HS	-2	5	-3098.5900	-0.0745	-3097.3265	-0.0912	-3095.9955	-0.0547
	Fe(III) LS	-1	2	-3098.4427	-0.0753	-3097.1687	-0.0922	-3095.8645	-0.0552
	Fe(III) HS	-1	6	-3098.4269	-0.0748	-3097.1589	-0.0916	-3095.8476	-0.0549
EEO2	Fe(II) LS	-2	1	-3514.7051	-0.1001	-3513.1753	-0.1210	-3511.6622	-0.0748
	Fe(II) HS	-2	5	-3514.6693	-0.0983	-3513.1587	-0.1189	-3511.6099	-0.0734
	Fe(III) LS	-1	2	-3514.5585	-0.1003	-3513.0336	-0.1213	-3511.5155	-0.0749
	Fe(III) HS	-1	6	-3514.5222	-0.0982	-3513.0112	-0.1187	-3511.4667	-0.0734
EEcpdl	doublet	-1	2	-3439.3659	-0.0977	-3437.8742	-0.1183	-3436.3901	-0.0728
	quartet	-1	4	-3439.3660	-0.0977	-3437.8737	-0.1183	-3436.3918	-0.0728
EZO	Fe(II) LS	-2	1	-3098.5938	-0.0757	-3097.3199	-0.0927	-3096.0074	-0.0555
	Fe(II) HS	-2	5	-3098.5901	-0.0743	-3097.3264	-0.0910	-3095.9958	-0.0545
	Fe(III) LS	-1	2	-3098.4416	-0.0750	-3097.1687	-0.0918	-3095.8660	-0.0550
	Fe(III) HS	-1	6	-3098.4287	-0.0757	-3097.1610	-0.0928	-3095.8493	-0.0555
EZO2	Fe(II) LS	-2	1	-3514.7078	-0.0999	-3513.1784	-0.1209	-3511.6647	-0.0745
	Fe(II) HS	-2	5	-3514.6735	-0.0979	-3513.1635	-0.1184	-3511.6136	-0.0731
	Fe(III) LS	-1	2	-3514.5597	-0.1000	-3513.0348	-0.1210	-3511.5168	-0.0746
	Fe(III) HS	-1	6	-3514.5239	-0.0979	-3513.0129	-0.1184	-3511.4684	-0.0731
EZcpdl	doublet	-1	2	-3439.3673	-0.0975	-3437.8755	-0.1180	-3436.3911	-0.0726
	quartet	-1	4	-3439.3668	-0.0976	-3437.8745	-0.1181	-3436.3928	-0.0727

ZEO	Fe(II) LS	-2	1	-3098.5951	-0.0746	-3097.3226	-0.0914	-3096.0091	-0.0548
	Fe(II) HS	-2	5	-3098.5919	-0.0743	-3097.3284	-0.0910	-3095.9973	-0.0545
	Fe(III) LS	-1	2	-3098.4427	-0.0750	-3097.1693	-0.0919	-3095.8660	-0.0550
	Fe(III) HS	-1	6	-3098.4297	-0.0747	-3097.1610	-0.0915	-3095.8497	-0.0548
ZEO2	Fe(II) LS	-2	1	-3514.7072	-0.0998	-3513.1775	-0.1207	-3511.6642	-0.0745
	Fe(II) HS	-2	5	-3514.6712	-0.0982	-3513.1605	-0.1188	-3511.6117	-0.0733
	Fe(III) LS	-1	2	-3514.5595	-0.0998	-3513.0345	-0.1208	-3511.5166	-0.0745
	Fe(III) HS	-1	6	-3514.5233	-0.0978	-3513.0116	-0.1183	-3511.4709	-0.0731
ZEcpl	doublet	-1	2	-3439.3675	-0.0974	-3437.8759	-0.1180	-3436.3917	-0.0725
	quartet	-1	4	-3439.3672	-0.0975	-3437.8750	-0.1181	-3436.3928	-0.0726
ZZO	Fe(II) LS	-2	1	-3098.5965	-0.0744	-3097.3216	-0.0911	-3096.0089	-0.0545
	Fe(II) HS	-2	5	-3098.5922	-0.0739	-3097.3285	-0.0906	-3095.9977	-0.0542
	Fe(III) LS	-1	2	-3098.4416	-0.0746	-3097.1691	-0.0914	-3095.8660	-0.0547
	Fe(III) HS	-1	6	-3098.4296	-0.0766	-3097.1612	-0.0937	-3095.8499	-0.0562
ZZO2	Fe(II) LS	-2	1	-3514.7080	-0.0995	-3513.1787	-0.1205	-3511.6651	-0.0742
	Fe(II) HS	-2	5	-3514.6791	-0.0977	-3513.1712	-0.1182	-3511.6130	-0.0729
	Fe(III) LS	-1	2	-3514.5691	-0.1057	-3513.0438	-0.1277	-3511.5239	-0.0790
	Fe(III) HS	-1	6	-3514.5253	-0.1036	-3513.0132	-0.1251	-3511.4726	-0.0729
ZZcpl	doublet	-1	2	-3439.3689	-0.0973	-3437.8765	-0.1179	-3436.3914	-0.0724
	quartet	-1	4	-3439.3674	-0.0972	-3437.8752	-0.1177	-3436.3928	-0.0724

Table S3. Relative Energies (kJ/mol) of Isomers in Different Spin and Redox States (relative to the corresponding EE isomers = 0 kJ/mol).

		TPSSh-D3	TPSSh	B3LYP-D3	B3LYP	PBE-D3	PBE
EE0	Fe(II) LS	0.0	0.0	0.0	0.0	0.0	0.0
	Fe(II) HS	0.0	0.0	0.0	0.0	0.0	0.0
	Fe(III) LS	0.0	0.0	0.0	0.0	0.0	0.0
	Fe(III) HS	0.0	0.0	0.0	0.0	0.0	0.0
EE O2	Fe(II) LS	0.0	0.0	0.0	0.0	0.0	0.0
	Fe(II) HS	0.0	0.0	0.0	0.0	0.0	0.0
	Fe(III) LS	0.0	0.0	0.0	0.0	0.0	0.0
	Fe(III) HS	0.0	0.0	0.0	0.0	0.0	0.0
EEcpdl	doublet	0.0	0.0	0.0	0.0	0.0	0.0
	quartet	0.0	0.0	0.0	0.0	0.0	0.0
EZ0	Fe(II) LS	2.2	4.1	0.5	2.8	-1.3	-0.1
	Fe(II) HS	-0.3	-0.7	0.4	-0.1	-0.9	-1.3
	Fe(III) LS	2.9	2.2	-0.1	-1.0	-4.0	-4.6
	Fe(III) HS	-4.9	-2.5	-5.6	-2.5	-4.5	-2.9
EZ O2	Fe(II) LS	-7.1	-7.7	-8.0	-8.5	-6.4	-7.0
	Fe(II) HS	-10.9	-12.0	-12.6	-13.8	-9.7	-10.6
	Fe(III) LS	-3.3	-4.0	-3.2	-3.9	-3.6	-4.3
	Fe(III) HS	-4.6	-5.3	-4.5	-5.2	-4.5	-5.1
EZcpdl	doublet	-3.7	-4.2	-3.5	-4.1	-2.6	-3.1
	quartet	-2.3	-2.6	-2.1	-2.6	-2.6	-2.9
ZE0	Fe(II) LS	-1.1	-1.9	-6.8	-7.7	-5.9	-6.6
	Fe(II) HS	-5.0	-5.7	-4.9	-5.6	-4.9	-5.5
	Fe(III) LS	0.0	-0.6	-1.6	-2.4	-3.9	-4.5
	Fe(III) HS	-7.4	-7.7	-5.6	-5.9	-5.7	-6.0
ZE O2	Fe(II) LS	-5.5	-6.3	-5.8	-6.6	-5.2	-6.1
	Fe(II) HS	-5.0	-5.3	-4.8	-5.1	-4.8	-5.1
	Fe(III) LS	-2.6	-3.8	-2.2	-3.6	-2.9	-4.0
	Fe(III) HS	-3.0	-4.0	-1.1	-2.1	-11.1	-12.0
ZEcpdl	doublet	-4.0	-4.8	-4.6	-5.3	-4.2	-4.9
	quartet	-3.1	-3.6	-3.5	-4.1	-2.6	-3.1

ZZ0	Fe(II) LS	-4.9	-6.4	-4.1	-5.9	-5.3	-6.6
	Fe(II) HS	-5.8	-7.3	-5.2	-6.8	-6.0	-7.3
	Fe(III) LS	2.9	1.3	-1.2	-3.1	-3.9	-5.2
	Fe(III) HS	-7.2	-2.7	-6.2	-0.6	-6.1	-2.7
ZZ O2	Fe(II) LS	-7.5	-9.0	-8.8	-10.1	-7.5	-9.1
	Fe(II) HS	-25.6	-27.2	-32.8	-34.5	-8.0	-9.4
	Fe(III) LS	-28.0	-13.9	-26.8	-10.0	-22.1	-11.4
	Fe(III) HS	-8.2	6.1	-5.3	11.6	-15.5	-16.7
ZZcpdl	doublet	-7.8	-8.8	-6.1	-7.1	-3.5	-4.5
	quartet	-3.8	-5.3	-4.0	-5.6	-2.6	-3.9

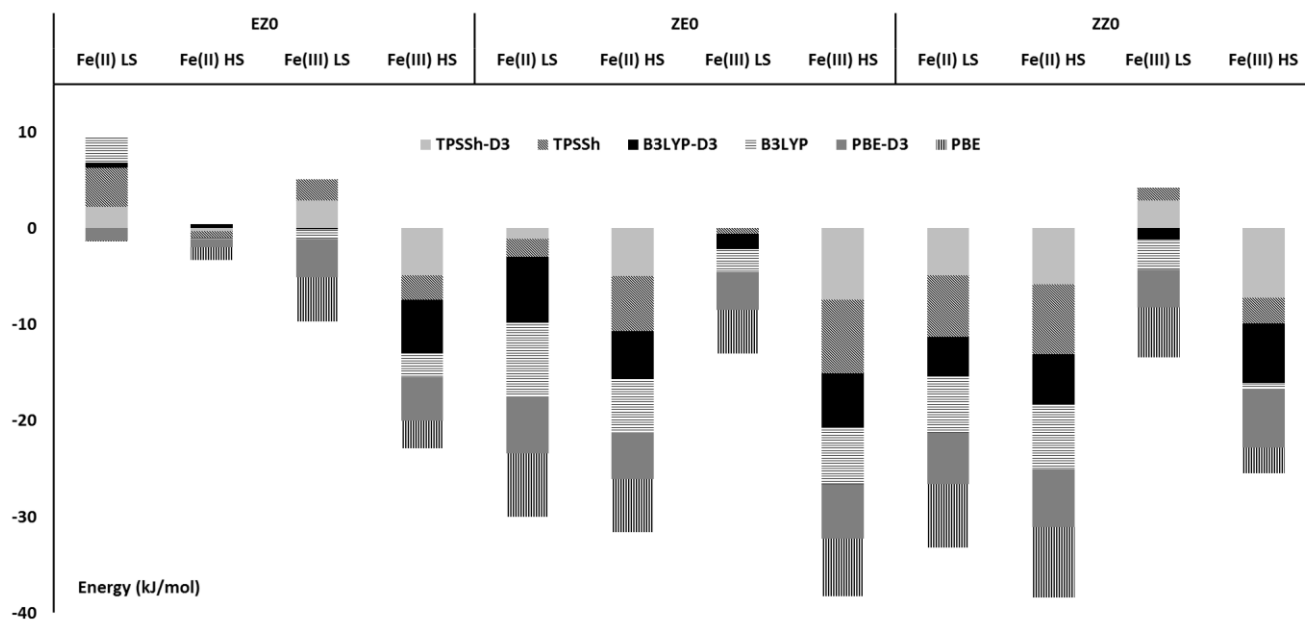


Figure S1. Stability (in kJ/mol) relative to EE0 isomer for the four electronic states of EZ0, ZE0, and ZZ0, with various bars representing TPSSh-D3, TPSSh, B3LYP-D3, B3LYP, PBE-D3, and PBE; most methods give energies within 5 kJ/mol and of similar sign.

Table S4. Computed Half Reduction Potentials vs. SHE (V) for Fe(III) + e⁻ → Fe(II), Corrected for Vibrational and Thermal Free energy Contributions.

System	State	TPSSh-D3	B3LYP-D3	PBE-D3
EE0	LS	0.32	0.34	0.59
	HS	0.01	-0.12	0.42
EE O2	LS	0.66	0.79	0.65
	HS	0.16	0.15	0.27
EZ0	LS	0.22	0.25	0.52
	HS	0.05	-0.05	0.46
EZ O2	LS	0.47	0.59	0.47
	HS	-0.09	-0.12	0.03
ZE0	LS	0.18	0.16	0.44
	HS	0.00	-0.15	0.39
ZE O2	LS	0.46	0.59	0.46
	HS	0.51	0.48	0.70
ZZ0	LS	0.12	0.19	0.45
	HS	0.58	0.45	0.98
ZZ O2	LS	0.54	0.64	0.47
	HS	-0.15	-0.26	0.22

Table S5. Fe–O, Fe–N_{ax}, and O–O bond lengths (Å) of optimized geometries.

System	State	Q	M	Fe–O	Fe–N _{ax}	O–O
EE O2	Fe(II) LS	-2	1	1.757	2.074	1.282
	Fe(II) HS	-2	5	1.895	2.130	1.282
	Fe(III) LS	-1	2	1.859	2.062	1.272
	Fe(III) HS	-1	6	1.983	2.109	1.265
EE cpdl	doublet	-1	2	1.631	2.132	
	quartet	-1	4	1.645	2.124	
EZ O2	Fe(II) LS	-2	1	1.757	2.076	1.282
	Fe(II) HS	-2	5	1.898	2.148	1.281
	Fe(III) LS	-1	2	1.857	2.063	1.272
	Fe(III) HS	-1	6	1.984	2.112	1.265
EZ cpdl	doublet	-1	2	1.630	2.134	
	quartet	-1	4	1.645	2.130	
ZE O2	Fe(II) LS	-2	1	1.757	2.074	1.282
	Fe(II) HS	-2	5	1.895	2.133	1.282
	Fe(III) LS	-1	2	1.859	2.062	1.272
	Fe(III) HS	-1	6	1.983	2.110	1.265
ZE cpdl	doublet	-1	2	1.630	2.133	
	quartet	-1	4	1.645	2.128	
ZZ O2	Fe(II) LS	-2	1	1.825	2.057	1.285
	Fe(II) HS	-2	5	1.899	2.151	1.281
	Fe(III) LS	-1	2	1.898	2.069	1.266
	Fe(III) HS	-1	6	1.981	2.124	1.265
ZZ cpdl	doublet	-1	2	1.631	2.138	
	quartet	-1	4	1.644	2.132	

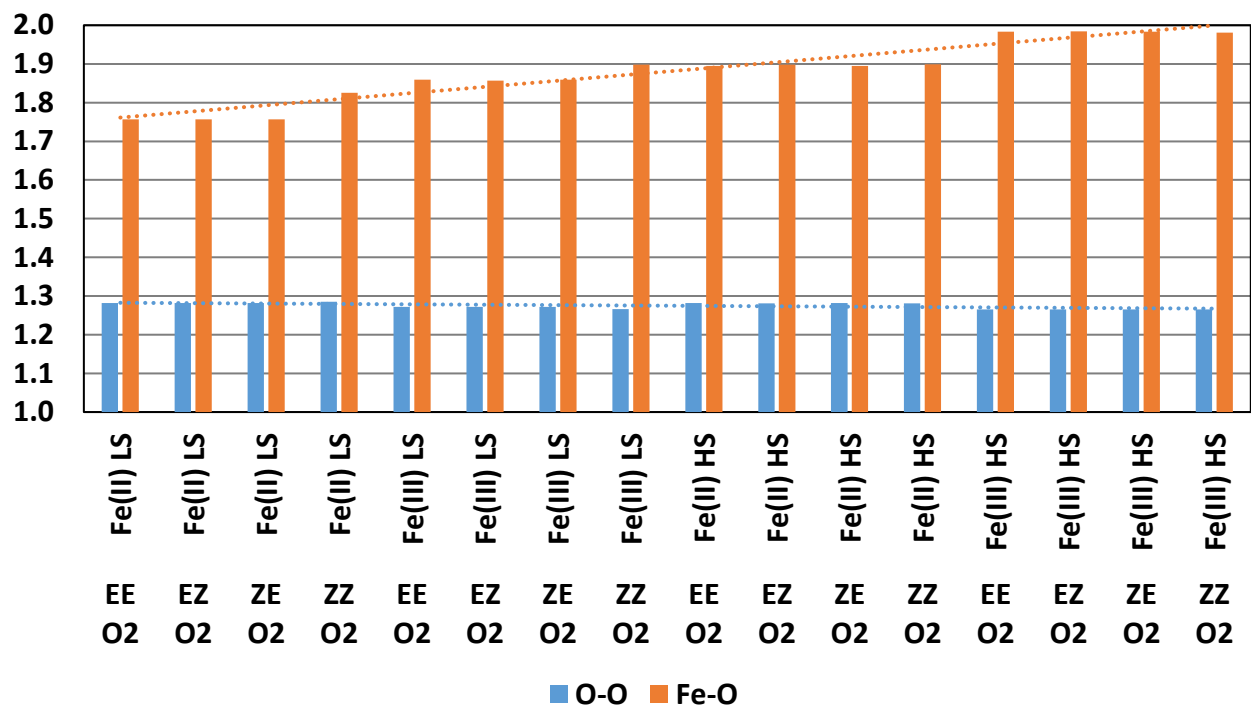


Figure S2. Differences in equilibrium Fe-O and O-O bond lengths of O₂-adduct isomer states (Å). Notice the high-spin effect on Fe-O bonds and the inverse relationship of Fe-O and O-O due to back-bonding.

Table S6. Energy Separations (def2-TZVPP, in kJ/mol) between High-Spin and Low-Spin States, Corrected for Zero-Point Energies, for Planar 4-Coordinate Models.

Model	State	Q	TPSSh-D3	B3LYP-D3	PBE-D3
EEO	Fe(II)	-2	-43.3	-13.9	-60.9
	Fe(III)	-1	-73.9	-58.2	-76.9
EZO	Fe(II)	-2	-5.6	21.3	-26.2
	Fe(III)	-1	-29.1	-15.7	-39.4
ZEO	Fe(II)	-2	-4.7	18.9	-27.4
	Fe(III)	-1	-28.0	-15.8	-36.8
ZZO	Fe(II)	-2	-33.8	-4.2	-51.6
	Fe(III)	-1	-26.1	-15.5	-37.1

Table S7. Energy Separations (def2-TZVPP, in kJ/mol) between High-Spin and Low-Spin States, Corrected for Zero-Point Energies, for O₂-Adducts (all favor low-spin).

Model	State	Q	TPSSh-D3	B3LYP-D3	PBE-D3
EE O2	Fe(II)	-2	-62.4	-12.0	-105.8
	Fe(III)	-1	-87.6	-51.3	-120.4
EZ O2	Fe(II)	-2	-63.0	-11.9	-107.0
	Fe(III)	-1	-86.1	-49.7	-119.3
ZE O2	Fe(II)	-2	-90.2	-40.4	-133.6
	Fe(III)	-1	-88.0	-53.3	-113.0
ZZ O2	Fe(II)	-2	-66.4	-10.2	-127.5
	Fe(III)	-1	-122.3	-87.6	-141.9

Table S8. Doublet-Quartet Separations (def2-TZVPP, in kJ/mol), Corrected for Zero-Point Energies, for Compound I. (Positive number means that the quartet is favored).

Model	TPSSh-D3	B3LYP-D3	PBE-D3
EEcpdl	-5.4	-6.8	-1.0
EZcpdl	8.5	7.1	14.3
ZEcpdl	-4.2	-5.8	-0.4
ZZcpdl	-3.5	-3.1	4.0

Table S9. Electronic energies of fully geometry-relaxed BP86/def2_SVP potential energy surfaces for ZEO used for computing single-point TPSSh-D3/def2-TZVPP energies in Figure 2.

Torsion angle	2-vinyl		4-vinyl	
	BP86/def2-SVP	TPSSh-D3/def2-TZVPP	BP86/def2-SVP	TPSSh-D3/def2-TZVPP
-105	-3096.52825	-3098.58922	-3096.52952	-3098.59164
-90	-3096.52744	-3098.58592	-3096.52877	-3098.59083
-75	-3096.52819	-3098.58669	-3096.52947	-3098.58785
-60	-3096.52956	-3098.58824	-3096.53121	-3098.58974
-45	-3096.53099	-3098.58981	-3096.53272	-3098.59130
-30	-3096.53147	-3098.59031	-3096.53329	-3098.59174
-15	-3096.53105	-3098.59291	-3096.53284	-3098.59110
0	-3096.53047	-3098.59229	-3096.53207	-3098.59028
15	-3096.53141	-3098.59335	-3096.53282	-3098.59126
30	-3096.53178	-3098.59388	-3096.53328	-3098.59189
45	-3096.53121	-3098.59365	-3096.53261	-3098.59134
60	-3096.52957	-3098.58818	-3096.53117	-3098.58973
75	-3096.52744	-3098.58572	-3096.52945	-3098.58787
90	-3096.52715	-3098.58577	-3096.52867	-3098.58687
105	-3096.52764	-3098.58997	-3096.52951	-3098.58779
120	-3096.52983	-3098.59223	-3096.53113	-3098.58939
135	-3096.53163	-3098.59404	-3096.53289	-3098.59124
150	-3096.53285	-3098.59509	-3096.53390	-3098.59241
165	-3096.53324	-3098.59535	-3096.53453	-3098.59289
180	-3096.53320	-3098.59511	-3096.53424	-3098.59266
195	-3096.53313	-3098.59523	-3096.53442	-3098.59624
210	-3096.53272	-3098.59507	-3096.53393	-3098.59583
225	-3096.52958	-3098.59180	-3096.53308	-3098.59526
240	-3096.52828	-3098.59042	-3096.53083	-3098.59296
255	-3096.52709	-3098.58922	-3096.52930	-3098.59132

Table S10. Electronic energies of 5-coordinate systems geometry optimized at the BP86/def2-SVP level and thermodynamic terms and zero-point energies. The Fe(II) HS states were used for computing O₂-affinities as described in the main text.

Model	State	Q	M	E(BP86/def2-SVP) (a.u.)	ZPE (a.u.)	ΔE (kJ/mol)	ΔS (kJ/molK)	ΔG (kJ/mol)	ΔH (kJ/mol)
EE5	Fe(II) HS	-2	5	-3361.93481	0.6489	1814.53	1.147	1475.09	1817.03
	Fe(III) HS	-1	6	-3361.80462	0.6506	1817.93	1.133	1482.73	1820.43
EZ5	Fe(II) HS	-2	5	-3361.93556	0.6467	1811.11	1.195	1457.34	1813.61
	Fe(III) HS	-1	6	-3361.80520	0.6484	1812.4	1.147	1473.01	1814.9
ZE5	Fe(II) HS	-2	5	-3361.93603	0.6467	1811.33	1.199	1456.35	1813.83
	Fe(III) HS	-1	6	-3361.80566	0.6510	1818.53	1.127	1484.92	1821.03
ZZ5	Fe(II) HS	-2	5	-3361.93613	0.6462	1811.05	1.219	1449.99	1813.55
	Fe(III) HS	-1	6	-3361.80629	0.6520	1820.28	1.1150	1490.33	1822.78

Table S11. Electronic energies of 5-coordinate systems; The Fe(II) HS states were used for computing O₂-affinities as described in the main text.

Model	State	Q	M	TPSSh-D3	D3	B3LYP-D3	D3	PBE-D3	D3
EE5	Fe(II) HS	-2	5	-3364.29235	-0.09275	-3362.85481	-0.11255	-3361.35855	-0.06896
	Fe(III) HS	-1	6	-3364.15842	-0.09303	-3362.71857	-0.11285	-3361.22621	-0.06920
EZ5	Fe(II) HS	-2	5	-3364.29184	-0.09255	-3362.85377	-0.11233	-3361.35925	-0.06877
	Fe(III) HS	-1	6	-3364.15980	-0.09333	-3362.71993	-0.11326	-3361.22748	-0.06939
ZE5	Fe(II) HS	-2	5	-3364.29587	-0.09260	-3362.85673	-0.11239	-3361.35978	-0.06882
	Fe(III) HS	-1	6	-3364.15903	-0.09273	-3362.71900	-0.11250	-3361.22702	-0.06896
ZZ5	Fe(II) HS	-2	5	-3364.29377	-0.09133	-3362.85559	-0.11086	-3361.36112	-0.06786
	Fe(III) HS	-1	6	-3364.15951	-0.09124	-3362.71931	-0.11070	-3361.22786	-0.06782