Bonding and Optical Properties of Spirocyclicphosphazene Derivatives. A DFT Approach

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Supplementary Information



Figure S1. In the figure are shown the NBO Natural charges of the corresponding systems.



Figure S2. In the figure are shown the NBO Natural charges of the corresponding systems.



Figure S3. The TD-DFT UV-vis spectra for molecules in study.





Figure S4. Plotted molecular orbital functions involved at the electronic transitions of spiro-cyclic phosphazene derivatives.





Figure S4. Plotted molecular orbital functions involved at the electronic transitions of spiro-cyclic phosphazene derivatives.

	1a	2a	2b	2c	6(3b)	4a	^a Exp.
P ₁ -N ₁	1.581	1.581	1.583	1.580	1.586	1.580	
P ₁ -N ₂	1.582	1.585	1.583	1.580	1.587	1.581	1.585,1.568
P ₂ -N ₂	1.582	1.578	1.578	1.580	1.572	1.584	1.585,1.582
P ₂ -N ₃	1.582	1.579	1.581	1.581	1.581	1.581	1.585,1.586
P ₃ -N ₁	1.581	1.581	1.578	1.580	1.573	1.584	1.585
P ₃ -N ₃	1.580	1.584	1.581	1.580	1.582	1.589	
P ₁ -O ₁	1.605	1.606	1.610	1.608	1.611	1.608	1.586
P ₁ -O ₂	1.604	1.604	1.600	1.604	1.592	1.608	1.581
P ₂ -O ₃	1.605	1.609	1.603	1.608	1.610	1.601	1.584
P ₂ -O ₄	1.604	1.609	1.601	1.602	1.611	1.605	
C-C _{ar}	1.391	1.392	1.391	1.388	1.391	1.392	1.393
C-C	1.480	1.480	1.480	1.479	1.480	1.480	
C-C _{ar2}	1.392	1.388	1.388	1.388	1.388	1.401	
C-C ₂	1.480	1.480	1.478	1.478	1.479	1478	
Br-C	-	1.913	1.913	1.912	-	-	
O ₂ N-C	-	-	-	-	1.479	-	
H ₂ N-C	-	-	-	-	-	1.398	
O-C _{ar}	1.393	1.387	1.390	1.390	1.384	1.391	1.395
P ₁ -N ₂ -P ₂	122.2	121.9	122.3	122.6	122.7	121.9	122.9,120.8
P ₁ -N ₁ -P ₃	122.3	122.5	122.5	122.3	122.6	122.1	122.9,121.2
N ₁ -P ₁ -N ₂	117.6	117.4	117.2	117.6	116.4	118.2	118.5,118.6
N ₁ -P ₃ -N ₃	117.5	117.9	117.7	117.5	118.3	117.3	118.5,118.3
O ₁ -P ₁ -O ₂	102.0	102.3	102.4	102.0	103.2	101.7	102.4
O ₃ -P ₂ -O ₄	102.1	101.6	101.8	102.0	101.0	102.5	103.0
P ₁ -O ₁ -C ₉	123.7	122.0	122.7	122.6	121.1	122.9	121.8
P ₂ -O ₃ -C ₅	123.7	123.9	122.2	121.99	120.3	124.8	

 Table S1. Selected structural parameters of phosphazene derivatives

aRef. 24

CSF	Donor	Acceptor	ΔΕ2	εj-εi
Electron W-D			(Kcal/mol)	a.u.
1				
	BD ⁽²⁾ C1-C2	BD* ⁽²⁾ C3-C6	19.74	0.28
	BD ⁽²⁾ C1-C2	BD* ⁽²⁾ C5-C7	20.92	0.29
Overall	BD ⁽²⁾ C-C	BD* ⁽²⁾ C-C	729.63	
2c	BD ⁽²⁾ C1-C2	BD* ⁽²⁾ C3-C6	19.43	0.28
	BD ⁽²⁾ C1-C2	BD* ⁽²⁾ C5-C7	21.41	0.27
Overall	BD ⁽²⁾ C-C	BD*(2)C-C	725.36	
	BD ⁽¹⁾ C1-C2	BD*(1)C5-Br71	5.03	0.79
	BD ⁽¹⁾ C6-C7	BD* ⁽¹⁾ C5-Br71	4.68	0.81
	LP ⁽³⁾ Br71	BD* ⁽²⁾ C5-C 7	9.78	0.31
	LP ⁽²⁾ Br71	BD* ⁽¹⁾ C2-C 5	2.88	0.86
	LP ⁽²⁾ Br71	BD* ⁽¹⁾ C5-C7	2.98	0.87
3c				
	BD ⁽²⁾ C1-C2	BD* ⁽²⁾ C3-C6	21.64	0.27
	BD ⁽²⁾ C1-C2	BD* ⁽²⁾ C5-C7	20.98	0.28
Overall	BD ⁽²⁾ C-C	BD* ⁽²⁾ C-C	735.78	
	BD ⁽²⁾ C18-C19	BD* ⁽²⁾ N76-O77	26.93	0.15
	BD (2) C-C7	BD*(2) N73-O74	26.27	0.15
	BD* ⁽²⁾ N73-O74	BD* ⁽²⁾ C5-C7	15.03	0.14
	BD* ⁽²⁾ N76-O77	BD* ⁽²⁾ C18-C19	14.5	0.14
4c	BD ⁽²⁾ C1-C2	BD* ⁽²⁾ C3-C6	15.73	0.29
	BD ⁽²⁾ C1-C2	BD* ⁽²⁾ C5-C7	22.1	0.26
Overall	BD ⁽²⁾ C-C	BD* ⁽²⁾ C-C	717.03	
	LP ⁽¹⁾ N67	BD* ⁽²⁾ C5-C7	29.32	0.32
	LP ⁽¹⁾ N68	BD* ⁽²⁾ C18-C19	27.19	0.32
	BD (1)C1-C2	BD* ⁽¹⁾ C5-N67	3.83	1.15
	BD ⁽¹⁾ C6 - C 7	BD* ⁽¹⁾ C5-N67	3.64	1.16
	BD ⁽¹⁾ C19-C23	BD* (1)C18-N68	3.51	1.13
	BD ⁽¹⁾ C20-C24	BD* (1)C18-N68	3.82	1.14

Table S2. Second order stabilization energies, ΔE^2 , ϵj - ϵi from NBO analysis for the selected systems.