Electronic Supplementary Information

Experimental and theoretical investigation of fluorescence

solvatochromism of dialkoxyphenyl-pyrene molecules

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Table of Contents

1. Methods	2
1.1 Materials	2
1.2 Synthesis and Characterization	2
Figures S1-S6 NMR spectra and MALDI-TOF MS	3
1.3 Techniques	7
2. Tables	8
Table S1	8
Table S2	9
Table S3	10
Table S4	11
Table S5	12
3. Figures	27
Figure S7	27
Figure S8	
Figure S9	29
4. References	

1. Methods

1.1 Materials

All starting materials and reagents, unless otherwise specified, were purchased from commercial suppliers (where noted) and used without further purification. 2,5-Dimethoxyphenylboronic acid,¹ and 2,6-dimethoxyphenylboronic acid² were synthesized and purified using reported methods. All reactions were performed under an argon atmosphere. Column chromatography was performed using Kanto Chemical silica gel 60 N (spherical, neutral). Spectroscopic grade solvents, dichloromethane (Wako Pure Chemical Industries, Ltd. (Wako)), toluene (Wako), chloroform (Wako), methanol (Wako), tetrahydrofuran (stabilizer free; Wako), *n*-hexane (Dojindo), acetonitrile (Dojindo) and dimethyl sulphoxide (Dojindo) were used for all spectroscopic studies without further purification.

1.2 Synthesis and characterization



General synthetic procedure for 1a and 3: A mixture of 1-bromopyrene (384 mg, 1.36 mmol, Wako), corresponding boronic acid (1.50 mmol), sodium carbonate (Na₂CO₃, 1.4 mL, 2 M a.q.), tetrakis(triphenylphosphine)palladium(0) (Pd(PPh₃)₄, 79 mg, 0.068 mmol, TCI) was refluxed in toluene-ethanol (54 mL-5.4 mL) under argon for 48–72 h until all starting material was consumed by TLC analysis. The reaction mixture was cooled, filtered and then evaporated to remove toluene. After evaporation, water (30 mL) was added and the aqueous layer was extracted with dichloromethane (3 × 30 mL). The combined organic layer was washed with brine (70 mL), dried over MgSO₄ and evaporated. The crude product was purified via column chromatography (SiO₂; dichloromethane/*n*-hexane, 1:5 v/v) and subsequent recycling HPLC with chloroform as the solvent to give the pure product.

1a: white powder (yield, 89%), $T_{\rm m}$: 134.7-137.5 °C; TLC (SiO₂; dichloromethane/*n*-hexane, 1:1 v/v): $R_{\rm f} = 0.39$; ¹H NMR (400 MHz, CDCl₃) δ (ppm): 8.23-8.14 (m, 3H), 8.09 (d, J = 2.0 Hz, 2H), 8.01-7.95 (m, 3H), 7.89 (d, J = 9.2 Hz, 1H), 7.04-7.00 (m, 3H), 3.83 (s, 3H), 3.64 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ (ppm): 153.55, 151.63, 134.10, 131.38, 131.00, 130.90, 130.71, 129.14, 127.89, 127.44, 127.31, 127.13, 125.85, 124.98, 124.71, 124.47, 118.09, 113.72, 112.56, 56.37, 55.82; MALDI-TOF MS (matrix: dithranol) calculated for C₂₄H₁₈O₂: 338.1, found: 338.7 [M]⁺.

3: white powder (yield, 87%), $T_{\rm m}$: 232.6-235.3 °C; TLC (SiO₂; dichloromethane/*n*-hexane, 4:1 v/v): $R_{\rm f}$ = 0.19; ¹H NMR (400 MHz, CDCl₃) δ (ppm): 8.23 (d, J = 8.0 Hz, 1H), 8.16-8.04 (m, 4H), 7.98-7.90 (m, 3H), 7.92 (d, J = 9.2 Hz, 1H), 7.44 (t, J = 8.0 Hz, 1H), 6.77 (d, J = 8.4 Hz, 2H), 3.64 (s, 6H); ¹³C NMR (100 MHz, CDCl₃) δ (ppm): 158.55, 131.33, 131.12, 130.58, 130.32, 129.76, 129.32, 128.79, 127.60, 127.01, 126.85, 125.84, 125.63, 124.99, 124.87, 124.75, 124.66, 124.52, 118.06, 104.28, 55.94; MALDI-TOF MS (matrix: dithranol) calculated for C₂₄H₁₈O₂: 338.1, found: 338.5 [M]⁺.



Figure S1. ¹H NMR (400 MHz, CDCl₃) spectrum of 1a.



Figure S2. ¹³C NMR (100 MHz, CDCI₃) spectrum of 1a.



Figure S3. MALDI-TOF MS of 1a. Matrix: dithranol.



Figure S4. ¹H NMR (400 MHz, CDCl₃) spectrum of 3.



Figure S5. ¹³C NMR (100 MHz, CDCl₃) spectrum of 3.



Figure S6. MALDI-TOF MS of 3. Matrix: dithranol.

1.3 Techniques

Recycling high performance liquid chromatography (HPLC) was performed at room temperature using a GPC column (YMC-GPC T30000 ϕ 20 × 600 mm) on a LC-9225NEXT system, equipped with RI and UV-Vis detectors. Melting points were determined on a Yanaco melting point apparatus MP-500P. ¹H NMR and ¹³C NMR spectra were recorded on a JEOL ECS-400 spectrometer at 400 MHz and 100 MHz, respectively, with tetramethylsilane used as the internal standard. Matrix-assisted laser desorption ionization time-of-flight mass spectra (MALDI-TOF MS) were obtained by a Shimadzu AXIMA-CFR Plus station.

UV-Vis absorption and fluorescence spectra were recorded on a JASCO V-670 spectrophotometer and a JASCO FP-8300 spectrophotometer, respectively. Absolute quantum yields were determined on a Hamamatsu Photonics absolute PL quantum yield spectrometer C11347.

Nanosecond time-resolved fluorescence lifetime measurements were carried out by using timecorrelated single photon counting (TCSPC) lifetime spectroscopy system HORIBA FluoroCube 3000U-UltraFast-SP spectrophotometer equipped with a nanosecond pulse LED (PB-280, 279 nm) and a nanosecond photon detection module (TBX). Decay analysis and the fitting routine to determine the lifetime(s) were performed using the DAS6 software provided by IBM. The quality of the fit has been judged by the fitting parameters such as chi-squared value χ^2 (< 1.20) as well as the visual inspection of the residuals.

A lab-built spectrometer equipped with a streak camera was used for recording picosecond timeresolved fluorescence spectra. Details of the spectrometer have been described elsewhere.^{3,4} In short, pump pulses were prepared by using output of a femtosecond Ti:sapphire laser system (Coherent Micra-5/Legend-Elite USP/OPerA Solo). The pump wavelength was set at 345 nm. Fluorescence from samples was dispersed with a spectrograph (Acton SP-2358) and detected with a streak camera (Hamamatsu Photonics C10627). The polarization of the collected fluorescence was selected with an α -BBO Glan-Taylor prism analyzer set at 54.7° with respect to the polarization of the pump pulse for avoiding effects of rotational relaxation on time-resolved spectra. The concentration of the sample solutions was 1 x 10⁻⁶ mol dm⁻³. The sample solutions were held in a 1-cm cuvette and rapidly stirred for avoiding accumulation of photodamage. The measurements were performed at 25 °C.

2. Tables

Solvent	Absorption Feature ^a	Fluorescence ^b	CIE	Fluoresc Deca	ence y ^b
Solvent	λ_{abs} , nm (ε , 10 ⁴ dm ³ mol ⁻¹ cm ⁻¹)	λ_{max} , nm $(oldsymbol{arPsi}_{FL})$	(x, y)	$ au$, ns c	χ ²
<i>n</i> -hexane	243 (4.38) 276 (3.08) 342 (2.50)	387 (0.09)	0.19, 0.11	7.7	1.05
Toluene	278 (0.31) 330 (1.85) 345 (2.53)	402 (0.43)	0.17, 0.06	8.5	1.13
CHCl₃	245 (4.01) 279 (3.02) 345 (2.55)	418 (0.50)	0.16, 0.07	5.0	1.19
THF	244 (4.81) 277 (3.31) 343 (2.59)	422 (0.48)	0.16, 0.08	5.3	1.07
CH_2CI_2	244 (4.24) 278 (2.99) 344 (2.37)	432 (0.42)	0.16, 0.10	3.9	1.17
CH₃OH	243 (4.19) 276 (2.89) 341 (2.35)	443 (0.18)	0.17, 0.15	3.5	1.06
CH₃CN	243 (4.28) 276 (2.89) 342 (2.33)	462 (0.14)	0.18, 0.21	2.7	1.07
DMSO	279 (3.86) 346 (2.49)	471 (0.20)	0.19, 0.24	3.0	1.10

Table S1. Photophysical parameters (λ : wavelength and Φ_{FL} : absolute fluorescence quantum yield) and fitting parameters (τ : decay time and χ^2 : chi-squared value) of the fluorescence decays of **1b** in different solvents

^a Concentration: 10 μM;

^{*b*} concentration: 1 μ M. λ_{ex} : 342 nm for *n*-hexane and CH₃CN; 345 nm for toluene and CHCl₃; 343 nm for THF; 344 nm for CH₂Cl₂; 341 nm for CH₃OH and 346 nm for DMSO;

^c monitored at their respective maximum emission wavelengths; λ_{ex} : 279 nm.

Solvent	Absorption Feature ^a	Fluorescence ^b	CIE	Fluoresc Deca	ence Y ^b
Joivent	λ_{abs} , nm (ε , 10 ⁴ dm ³ mol ⁻¹ cm ⁻¹)	λ _{max} , nm (Φ _{FL})	(x, y)	$ au$, ns c	χ^2
<i>n</i> -hexane	238 (4.18) 277 (3.58) 341 (3.00)	379 (0.03)	0.21, 0.16	8.3	1.11
Toluene	282 (2.50) 345 (2.80)	381 (0.15)	0.19, 0.09	15.3	1.16
CHCI ₃	245 (4.17) 280 (3.46) 345 (2.93)	381 (0.10)	0.19, 0.09	15.2	1.10
THF	243 (3.43) 279 (3.46) 343 (2.90)	380 (0.21)	0.18, 0.09	20.8	1.07
CH_2CI_2	243 (4.19) 280 (3.51) 344 (2.88)	381 (0.27)	0.18, 0.08	22.8	1.05
CH₃OH	241 (4.39) 277 (3.62) 341 (2.95)	379 (0.06)	0.21, 0.14	15.6	1.18
CH₃CN	240 (4.26) 278 (3.43) 342 (2.80)	380 (0.10)	0.20, 0.13	14.2	1.10
DMSO	281 (3.17) 346 (2.62)	382 (0.41)	0.17, 0.05	39.4	1.10

Table S2. Photophysical parameters (λ : wavelength and Φ_{FL} : absolute fluorescence quantum yield) and fitting parameters (τ : decay time and χ^2 : chi-squared value) of the fluorescence decays of **2a** in different solvents

^a Concentration: 10 µM;

^{*b*} concentration: 1 μ M. λ_{ex} : 341 nm for *n*-hexane and CH₃OH; 345 nm for toluene and CHCl₃; 343 nm for THF; 344 nm for CH₂Cl₂; 342 nm for CH₃CN and 346 nm for DMSO;

 $^{\circ}$ monitored at their respective maximum emission wavelengths; λ_{ex} = 279 nm.

Solvent	Absorption Feature ^a	Fluorescence ^b	CIE	Fluoresc Deca	ence Y ^b
Solvent	λ_{abs} , nm (ε , 10 ⁴ dm ³ mol ⁻¹ cm ⁻¹)	λ _{max} , nm (Φ _{FL})	(x, y)	$ au$, ns c	χ^2
n -hexane	240 (4.48) 279 (3.46) 341 (3.14)	379 (0.04)	0.21, 0.14	9.5	1.12
Toluene	282 (2.67) 345 (3.08)	381 (0.24)	0.18, 0.08	16.3	1.15
CHCl₃	244 (4.37) 281 (3.52) 345 (3.06)	381 (0.12)	0.19, 0.09	15.2	1.16
THF	241 (4.39) 279 (3.74) 343 (3.19)	380 (0.18)	0.18, 0.08	19.1	1.06
CH_2CI_2	242 (4.68) 280 (3.72) 344 (3.11)	381 (0.21)	0.18, 0.07	22.4	1.13
CH₃OH	241 (4.64) 277 (3.71) 341 (3.10)	380 (0.07)	0.19, 0.11	16.3	1.13
CH₃CN	241 (4.67) 278 (3.64) 342 (3.01)	380 (0.17)	0.19, 0.11	14.8	1.18
DMSO	281 (3.56) 346 (2.98)	382 (0.44)	0.17, 0.04	37.0	1.09

Table S3. Photophysical parameters (λ : wavelength and Φ_{FL} : absolute fluorescence quantum yield) and fitting parameters (τ : decay time and χ^2 : chi-squared value) of the fluorescence decays of **2b** in different solvents

^aConcentration: 10 µM;

^{*b*} concentration: 1 μ M. λ_{ex} : 341 nm for *n*-hexane and CH₃OH; 345 nm for toluene and CHCl₃; 343 nm for THF; 344 nm for CH₂Cl₂; 342 nm for CH₃CN and 346 nm for DMSO;

^cmonitored at their respective maximum emission wavelengths; λ_{ex} : 279 nm.

Solvent	Absorption Feature ^a	Fluorescence ^b	CIE	Fluoresc Deca	cence Y ^b
Solvent	λ_{abs} , nm (ε , 10 ⁴ dm ³ mol ⁻¹ cm ⁻¹)	λ _{max} , nm (Φ _{FL})	(x, y)	$ au$, ns c	χ ²
<i>n</i> -hexane	243 (5.15) 276 (3.72) 341 (3.49)	387 (0.02)	0.22, 0.17	7.6	1.07
Toluene	282 (1.71) 344 (3.24)	388 (0.07)	0.19, 0.13	14.3	1.03
CHCl₃	245 (4.57) 279 (3.36) 345 (3.13)	377 (0.07)	0.20, 0.13	14.2	1.10
THF	243 (4.98) 277 (3.69) 343 (3.37)	377 (0.11)	0.19, 0.11	35.6	1.18
CH_2CI_2	244 (5.15) 278 (3.71) 344 (3.40)	377 (0.13)	0.19, 0.11	20.8	1.07
CH₃OH	242 (5.41) 275 (3.76) 341 (3.48)	395 (0.05)	0.20, 0.15	14.3	1.10
CH₃CN	242 (5.24) 276 (3.60) 341 (3.33)	376 (0.05)	0.20, 0.14	13.6	1.15
DMSO	278 (3.51) 345 (3.16)	378 (0.23)	0.18, 0.07	43.5	1.08

Table S4. Photophysical parameters (λ : wavelength and Φ_{FL} : absolute fluorescence quantum yield) and fitting parameters (τ : decay time and χ^2 : chi-squared value) of the fluorescence decays of **3** in different solvents

^aConcentration: 10 µM.

^{*b*} concentration: 1 μ M. λ_{ex} : 341 nm for *n*-hexane, CH₃OH and CH₃CN; 344 nm for toluene and CH₂Cl₂; 345 nm for CHCl₃ and DMSO and 343 nm for THF;

^cmonitored at their respective maximum emission wavelengths; λ_{ex} = 279 nm.

 Table S5. Cartesian coordinates (in Å) of the optimized geometries.

1a :	min(GS) in C	CH ₂ Cl ₂ solution	on
С	-0.407341	2.077521	0.877392
С	-0.585473	0.794762	0.356149
С	0.542159	0.055020	-0.052586
С	1.838887	0.622441	0.089793
С	1.994817	1.925283	0.634422
С	0.855622	2.636685	1.017698
С	0.435130	-1.259280	-0.633727
С	2.995474	-0.113885	-0.318321
С	2.854273	-1.412474	-0.874800
С	1.530067	-1.954613	-1.022163
C	3.997056	-2.118772	-1.266727
C	5.261506	-1.560535	-1.115992
C	5.408883	-0.287818	-0.574361
C	4.291481	0.451258	-0.170985
C	4.412667	1.772215	0.389694
C	3.319357	2.4/31/3	0.771191
н	3.421690	3.470129	1.191300
н	5.406869	2.197079	0.498850
н	-0.551798	-1.690234	-0.760306
н	-1.281060	2.643083	1.18//83
п	1.424210	-2.944108	-1.408913
	0.139270	-2.120033	-1.424223
č	-1.973479	0.200047	1.006060
č	2 855757	0.871878	0.663138
č	-2.000707	-1 2520/8	0.000100
Ĉ	-4 175678	0 430950	-0 799561
н	-2 519919	1 704471	-1 273897
C	-4.617938	-0.637725	-0.032151
Ĥ	-5.632359	-1.009420	-0.110493
Н	3.886035	-3.112405	-1.692483
Н	6.399308	0.144561	-0.460364
Н	0.963948	3.635264	1.432309
0	-4.940658	1.108951	-1.706724
С	-6.289969	0.705987	-1.873002
Н	-6.711514	1.370165	-2.627201
Н	-6.353790	-0.330394	-2.223571
Н	-6.852395	0.811968	-0.938367
Н	-4.112973	-2.078528	1.465510
0	-1.533896	-1.347709	1.880987
C	-1.944666	-2.442078	2.682429
н	-2.237452	-3.298/12	2.004585
п	-1.079210	-2.7 10901	3.201131
П	-2.770119	-2.103003	3.339090
1a:	min(GS) in n	hexane soli	ution
С	-0.411652	2.078194	0.868158
С	-0.587211	0.795039	0.348360
С	0.541357	0.056186	-0.057749
С	1.837001	0.624436	0.086762
С	1.990535	1.927587	0.630358
С	0.850224	2.638197	1.009953
C	0.436392	-1.257614	-0.638947
С	2.994758	-0.111178	-0.317753

C	2.855684	-1.409750	-0.873926
C C	1.532408	-1.951864	-1.024537
Č	5.263187	-1.556689	-1.109114
C	5.408280	-0.284222	-0.567618
C	4.289924	0.454334	-0.167592
C	3.314043	2.475767	0.769782
Ĥ	3.414885	3.473304	1.189219
Н	5.402207	2.200934	0.503828
Н	-0.550365	-1.687711	-0.768964
п Н	1.428083	-2.941162	-1.462442
Н	6.142032	-2.116486	-1.415178
С	-1.974438	0.264285	0.215448
C	-2.427856	-0.816691	0.999561
C C	-2.002730	-1 260140	-0.860314
č	-4.182933	0.428480	-0.789879
Н	-2.533480	1.705149	-1.272236
С	-4.618814	-0.642727	-0.023569
п	-5.633438	-1.015204	-0.097049
н	6.398217	0.148969	-0.451230
Н	0.957232	3.637267	1.424108
0	-4.952948	1.108966	-1.690061
С	-6.298088	0.704890	-1.853356
H	-6.362529	-0.330768	-2.208449
Н	-6.859947	0.805429	-0.916840
Н	-4.101320	-2.088199	1.466748
C	-1.519785	-1.354559	2 668514
H	-2.218893	-3.304741	2.056109
Н	-1.046685	-2.713134	3.264260
Н	-2.744142	-2.167476	3.335906
1a :	min(ICT) in C	H ₂ Cl ₂ solutio	on
С	-0.484 <u>́</u> 566	2.175949	0.627241
C	-0.632525	0.830439	0.225309
C C	0.539446	0.090850	-0.140069
č	1.942315	2.030109	0.514756
С	0.755515	2.761296	0.773148
C	0.490273	-1.207073	-0.716867
C C	2 919909	-0.040708	-0.263283
č	1.628019	-1.915660	-1.028715
С	4.104532	-2.091532	-1.073324
C	5.350701	-1.511284	-0.854296
C	5.453825	-0.215586	-0.357052
č	4.372269	1.872883	0.437757
С	3.234112	2.589117	0.703472
Н	3.307380	3.610235	1.070247
п Н	-0.473646	∠.318869-1.651140	0.594037 -0.937621
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H H H C C C C C C H C H H H H C C C C C	-1.375297 1.546865 6.253258 -1.982790 -2.330253 -3.013566 -3.632410 -4.329431 -2.821847 -4.642331 -5.642344 4.029553 6.431079 0.834152 -5.214259 -6.574014 -7.077582 -6.636098 -7.028006 -3.869923 -1.353150 -1.561175 -1.842173 -0.605290 -2.329044	2.747693 -2.903106 -2.073038 0.267361 -0.985587 0.950887 -1.509327 0.436772 1.898706 -0.811177 -1.221197 -3.100225 0.232870 3.789359 1.213661 0.791230 1.585151 -0.145109 0.674748 -2.450122 -1.576144 -2.845200 -3.594714 -3.106699 -2.768947	0.873171 -1.475239 -1.078800 0.148276 0.761177 -0.483805 0.675370 -0.541634 -0.974500 0.027821 -0.012633 -1.470359 -0.198649 1.117658 -1.173742 -1.294971 -1.843023 -1.855557 -0.307388 1.155859 1.449738 2.071608 1.327480 2.520759 2.845892
1a :m	nin(ICT) in <i>n-</i>	hexane solu	tion
C C	-0.485617 -0.633285	2.183771 0.839183	0.606751 0.205307
C	0.537540	0.096578	-0.155831
C	1.939468	2.032262	0.504251
C	0.751562	2.767630	0.756397
C	0.489047	-1.200805	-0.264533
Č	2.917680	-1.377803	-0.787622
C	1.627712	-1.912996	-1.034210
C	4.102943	-2.094794	-0.839808
С	5.449756	-0.220784	-0.343959
C	4.295068	0.537247	-0.051015
c	3.229278	2.588512	0.699680
Н	3.301704	3.610037	1.065615
H H	5.346357	2.313837	0.603672
H	-1.376824	2.756285	0.851073
Н	1.547936	-2.898999	-1.484580
C	-1.982345	0.274417	0.138857
C	-2.323743	-0.975686	0.765033
C C	-3.019806	0.940701	-0.495891 0.695266
č	-4.336826	0.421333	-0.534947
Н	-2.836111	1.880688	-1.003943
С Н	-4.644064 -5.643240	-0.820402 -1.233537	0.049656

ннноснннносннн	4.028498	-3.103864	-1.460619
	6.426776	0.226609	-0.181548
	0.830830	3.796274	1.098492
	-5.227379	1.190966	-1.167522
	-6.584648	0.767946	-1.275316
	-7.092897	1.556100	-1.827807
	-6.651492	-0.174178	-1.826692
	-7.033656	0.660482	-0.283756
	-3.852474	-2.443543	1.189344
	-1.337270	-1.557109	1.444075
	-1.522234	-2.821631	2.073732
	-1.806910	-3.579303	1.338359
	-0.556230	-3.071258	2.507987
	-2.278686	-2.750555	2.860757
1 CCCCCCCCCCCCCCCCCCHHHHHCCCCCCHCHHHHOCHHHHOC	min(LE) in CF -0.520186 -0.654434 0.554826 1.829158 1.919938 0.701199 0.521136 3.022119 2.951968 1.685692 4.164369 5.394382 5.475965 4.297026 4.346658 3.178390 3.237679 5.313078 -0.435161 -1.419232 1.618152 6.306377 -1.988644 -2.317998 -3.037325 -3.614387 -4.335120 -2.827678 -4.626919 -5.625424 4.105549 6.444082 0.766281 -5.247996 -6.582290 -7.127239 -6.626335 -7.036334 -3.868393 -1.330008 -1.584452	12Cl2 solution 2.164102 0.802928 0.065627 0.679240 2.021891 2.757538 -1.199077 -0.032592 -1.347062 -1.894603 -2.048972 -1.460303 -0.172097 0.570107 1.883317 2.595013 3.617266 2.345966 -1.651971 2.731967 -2.08275 0.226683 -1.032988 0.956503 -1.523988 0.464903 1.914016 -0.790335 -1.207281 -3.051839 0.286273 3.787373 1.266109 0.798759 1.580887 -0.130293 0.640455 -2.478843 -1.652385 -2.925956	0.562053 0.200294 -0.149566 0.051693 0.526137 0.740442 -0.755008 -0.250013 -0.798744 -1.062822 -1.080512 -0.833313 -0.314066 -0.018129 0.487392 0.739215 1.102643 0.666684 -0.987940 0.780962 -1.513197 -1.052526 0.149754 0.7394471 -0.438700 0.650323 -0.508045 -0.903797 0.027020 -0.132030 -1.494472 -0.132030 -1.780123 -1.831294 -0.267488 1.093778 1.428302 1.998220

H H H	-1.870555 -0.648636 -2.368599	-3.652676 -3.235670 -2.867927	1.229806 2.462519 2.761042
	-2.308599 min(LE) <i>n</i> -he -0.519904 -0.654552 0.551356 1.826545 1.918056 0.702950 0.515263 3.017340 2.945620 1.676511 4.154416 5.385301 5.468257 4.293484 4.344273 3.179765 3.239854 5.311996 -0.442951 -1.419032 1.608669 6.296487 -1.987890 -2.309881 -3.041688 -3.604131 -4.337418 -2.839388 -4.622433 -5.619907 4.094261 6.437250 0.769823 -5.255096 -6.583561 -7.132515 -6.624820 -7.039031 -3.851130 -1.314186 -1.557222 -1.848188 -0.615352 -2.334510	-2.807927 xane solution 2.171570 0.809494 0.070951 0.680435 2.023664 2.761213 -1.194350 -0.034195 -1.348509 -1.892022 -2.052177 -1.467795 -0.179857 0.564195 1.879685 2.593727 3.616158 2.338582 -1.639924 2.739743 -2.877016 -2.018134 0.232023 -1.026176 0.953874 -1.523285 0.456226 1.908026 -0.796841 -1.217122 -3.054299 0.275545 3.791236 1.249582 0.776591 1.553726 -0.155745 0.620276 -2.476444 -1.635834 -2.898907 -3.639579 -3.198279 -2.833942	2.761042 0.547152 0.187753 -0.158608 0.045907 0.519825 0.728572 -0.767000 -0.251831 -0.800902 -1.070389 -1.070389 -1.070389 -1.070389 -0.015912 0.490183 0.736663 1.100009 0.672686 -1.006895 0.764402 -1.524336 -1.043568 0.144285 0.737556 -0.443005 0.663408 -0.015912 0.443005 0.764402 -1.524336 -1.043568 0.144285 0.737556 -0.443005 0.663408 -0.03429 -1.494135 -0.121979 1.067488 -1.128957 -1.246171 -1.778046 -1.821740 -0.261125 1.114693 1.427426 2.016325 1.261985 2.475672 2.786586
2a : C C C C C C C	min(GS) in C 0.016213 0.264202 -0.825826 -2.153423 -2.376536 -1.274442	H ₂ Cl ₂ solutio -2.204368 -0.903851 -0.064382 -0.550916 -1.870959 -2.682976	n 0.740957 0.293882 -0.019597 0.156032 0.630578 0.909595

000000	-0.658868	1.268307	-0.546188
	-3.274140	0.284733	-0.148481
	-3.070703	1.600794	-0.640928
	-1.721226	2.055658	-0.838218
	-4.178391	2.405614	-0.929541
	-5.470074	1.927842	-0.738959
C C C C C H H	-5.679500 -4.598480 -4.786088 -3.727742 -3.879440 -5.801341	0.638599 -0.197939 -1.538321 -2.334564 -3.346107 -1.899657	-0.261130 0.037954 0.528489 0.807695 1.174568 0.667916
H	0.345478	1.637709	-0.716609
H	0.859618	-2.845820	0.978475
H	-1.567077	3.054463	-1.237750
H	-6.320314	2.563800	-0.966523
C	1.681082	-0.458268	0.166629
C	2.152830	0.635993	0.892737
C C H C H C H C H C H C H C H C H C	2.556047 3.491184 1.498169 3.892597 2.211868 4.376598	-1.172568 1.011475 1.196481 -0.786691 -2.023542 0.308286	-0.651759 0.793568 1.551345 -0.746735 -1.229861
ОННННО	4.018345 -4.018345 -6.690919 -1.434931 4.668024	0.508250 0.603974 3.411932 0.268570 -3.695294 -1.538475	-0.102939 -1.307008 -0.116351 1.270307 -1.575237
0	3.860699	2.087066	1.541956
C	6.039885	-1.203388	-1.721557
H	6.450152	-1.928467	-2.423813
H	6.158810	-0.193731	-2.129333
H	6.569907	-1.281549	-0.766128
C	5.211487	2.521927	1.498319
H	5.496375	2.827356	0.485698
H	5.268965	3.381601	2.165335
H	5.890508	1.739343	1.853739
2a: C C C C C	min(GS) in 7 0.017767 0.264789 -0.825434 -2.152491 -2.374755	-nexane solu -2.201775 -0.902317 -0.064148 -0.550594 -1.869711	0.743741 0.294319 -0.020128 0.156468 0.632786
CCCCCC	-1.272418	-2.680269	0.912942
	-0.658552	1.267477	-0.548322
	-3.273226	0.284082	-0.148758
	-3.070150	1.599311	-0.642526
	-1.721104	2.053787	-0.840477
	-4.177945	2.403065	-0.931515
C C C C C C C H	-5.469168 -5.677942 -4.597123 -4.783786 -3.725441 -3.876795	1.925725 0.637432 -0.198383 -1.537833 -2.333076 -3.344356	-0.740419 -0.261252 0.038376 0.530362 0.810292 1.178409

T T T T T C C C C T C T C T T T T C C C C T C T C T T T T C C C T C T C T T T C C T T T C T T T C T T T C T T T	$\begin{array}{r} -5.798999\\ 0.346123\\ 0.862455\\ -1.567383\\ -6.319603\\ 1.681245\\ 2.153256\\ 2.555224\\ 3.490829\\ 1.500475\\ 3.891078\\ 2.211132\\ 4.375653\\ 5.411846\\ -4.017955\\ -6.689293\\ -1.432756\\ 4.665362\\ 3.860460\\ 6.032801\\ 6.441996\\ 6.153959\\ 6.567347\\ 5.208258\\ 5.499767\\ 5.266486\\ 5.886442\end{array}$	-1.899426 1.635403 -2.840817 3.052224 2.561387 -0.456558 0.637028 -1.170332 1.011917 1.196504 -0.785511 -2.018963 0.308773 0.604089 3.409117 0.267187 -3.692063 -1.537855 2.087395 -1.204668 -1.929803 -0.194655 -1.283012 2.517557 2.826626 3.375336 1.733034	0.670268 -0.719881 0.983268 -1.241390 -0.968519 0.166223 0.892877 -0.653258 0.793508 1.553891 -0.748306 -1.234432 -0.028321 -0.103781 -1.310038 -0.115687 1.275653 -1.578056 1.542829 -1.726779 -2.430140 -2.135391 -0.772877 1.505187 0.494565 2.175139 1.860638
2a : C	min(ICT) in C 0.061875	H ₂ Cl ₂ solutio -2.299574	on 0.333882
C C	0.302773 -0.842057	-0.925305 -0.069947	0.055174 -0.146644
C C	-2.155565 -2.342592	-0.598311 -1.965394	0.062878 0.412070
Č	-1.199401	-2.800775	0.514575
C	-0.747700 -3.306872	0.240459	-0.022844 -0.098491
C	-3.157592	1.597395	-0.513002
C	-4.300141	2.407363	-0.652459
C	-5.565190	1.892829	-0.396785
C	-4.616037	-0.278805	0.143702
C C	-4.753106	-1.650288	0.523861
H	-3.774386	-3.501322	0.914049
H H	-5.749569 0 223820	-2.042822	0.708404
Н	0.908051	-2.964794	0.478542
H H	-1.721745 -6.437670	3.084779 2.531437	-1.169151 -0.508555
С	1.667718	-0.457073	0.032855
C C	2.067720	0.820668	0.486005
-			
C	3.430382	1.243336	0.475520

нсннноосннсннн	$\begin{array}{c} 2.548909\\ 4.475517\\ 5.501120\\ -4.180168\\ -6.723083\\ -1.335217\\ 4.940188\\ 3.606708\\ 6.336873\\ 6.801120\\ 6.502502\\ 6.739446\\ 4.923985\\ 5.368564\\ 4.793893\\ 5.549425\end{array}$	-2.283354 0.423335 0.754335 3.440519 0.171621 -3.848791 -1.774173 2.472343 -1.476096 -2.378738 -0.641416 -1.245758 3.024731 3.104016 4.015093 2.410276	-0.770920 0.033665 0.014892 -0.967734 0.182065 0.770112 -0.829902 0.950007 -0.919988 -1.311152 -1.605461 0.069510 1.039542 0.044593 1.470223 1.691711
2°,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,	min(ICT) in n	-hexane solu	ution
	0.064470	-2.295697	0.335038
	0.301377	-0.923284	0.050327
	-0.842318	-0.067401	-0.150533
	-2.154321	-0.597194	0.062163
	-2.338468	-1.963806	0.414503
	-1.195028	-2.798083	0.519040
	-0.750742	1.269602	-0.628077
	-3.306357	0.239130	-0.098680
	-3.160285	1.595158	-0.514396
	-1.848226	2.069550	-0.800866
	-4.303941	2.403021	-0.653298
	-5.567108	1.887174	-0.395682
	-5.727774	0.563563	-0.004277
	-4.613598	-0.281551	0.145635
	-4.747710	-1.651532	0.527558
	-3.649659	-2.455128	0.643540
	-3.766439	-3.500582	0.920504
	-5.743517	-2.045061	0.713479
	0.218423	1.670344	-0.904243
	0.911733	-2.959024	0.484157
	-1.729739	3.082969	-1.175466
	-6.440903	2.524065	-0.507054
	1.666156	-0.454396	0.28779
	2.067819	0.822721	0.481258
	2.731115	-1.282183	-0.391798
	3.429454	1.243663	0.474832
	1.355040	1.514232	0.910804
	4.091387	-0.853104	-0.381022
	2.547411	-2.275689	-0.781863
	4.477416	0.423328	0.037395
	5.503409	0.753612	0.021231
	-4.185710	3.435978	-0.969944
	-6.720575	0.164747	0.186503
	-1.329966	-3.845030	0.778840
	4.941167	-1.775137	-0.831794
	3.605712	2.475016	0.950253
	6.334006	-1.479701	-0.918720
	6.799317	-2.381587	-1.311464
	6.504956	-0.643559	-1.602366

H C H H H	6.738592 4.918945 5.373359 4.788705 5.541211	-1.252321 3.024051 3.102145 4.016249 2.412064	0.071548 1.047409 0.055927 1.474745 1.706226
	min(LE) in CH 0.082861 0.320171 -0.838851 -2.150625 -2.333444 -1.173787 -0.735350 -3.291567 -3.140860 -1.852623 -4.300942 -5.559813 -5.721677 -4.598369 -4.735832 -3.622293 -3.749307 -5.726845 0.234203 0.934924 -1.724543 -6.431771 1.691584 2.041406 2.726901 3.368055 1.296684 4.045289 2.510810 4.391245 5.419258 -4.177603 -6.713781 -1.310939 4.959168 3.593796 6.325757 6.855247 6.467591 6.718641 4.927672 5.376489 4.862163 5.545610 min(LE) in <i>n</i> -1	I2CI2 solution -2.304161 -0.940857 -0.080791 -0.594666 -1.952863 -2.800987 1.215965 0.233925 1.573539 2.029036 2.395273 1.897758 0.586922 -0.272297 -1.608295 -2.431476 -3.467472 -1.998201 1.599943 -2.960410 3.034206 2.534059 -0.457444 0.813534 -1.288988 1.224162 1.466628 -0.855856 -2.259948 0.405416 0.736020 3.415530 0.200057 -3.840739 -1.727416 2.457458 -1.344952 -2.183814 -0.451155 -1.166436 2.936142 3.033339 3.919694 2.280130	0.345193 0.045746 -0.182381 0.068888 0.464435 0.560512 -0.708169 -0.106425 -0.573225 -0.887460 -0.725776 -0.433609 0.005280 0.172511 0.594732 0.722452 1.024267 0.808614 -0.999119 0.487610 -1.279155 -0.553879 0.024187 0.527125 -0.454503 0.549636 0.965140 -0.442054 -0.884890 0.059167 0.659167 0.067828 -1.077941 0.220994 0.843531 -0.953055 1.083599 -0.995159 -1.612264 0.011569 1.631133 1.789301
~		0.005050	0.050540

С	0.081977	-2.305376	0.350519
С	0.319004	-0.943453	0.045280
С	-0.836624	-0.085228	-0.184863
С	-2.149070	-0.595174	0.068672

СССССССССССТНЕНЕССССССССССССССССССССССС	$\begin{array}{c} -2.332447\\ -1.175891\\ -0.730327\\ -3.287487\\ -3.135224\\ -1.844412\\ -4.291742\\ -5.551106\\ -5.714343\\ -4.595499\\ -4.734448\\ -3.624248\\ -3.751840\\ -5.726514\\ 0.240670\\ 0.934868\\ -1.715465\\ -6.422125\\ 1.690632\\ 2.038036\\ 2.725078\\ 3.363432\\ 1.293664\\ 4.042144\\ 2.510163\\ 4.386832\\ 5.414613\\ -4.166624\\ -6.707256\\ -1.314062\\ 4.956632\\ 3.587195\\ 6.318779\\ 6.849094\\ 6.462045\\ 6.714981\\ 4.916536\\ 5.371100\\ 4.850109\\ 5.534498\end{array}$	-1.952302 -2.799205 1.211074 0.233923 1.572273 2.024502 2.394512 1.900547 0.590613 -0.268592 -1.605030 -2.428015 -3.462832 -1.991453 1.588988 -2.959779 3.028222 2.537930 -0.459129 0.809643 -1.286987 1.221933 1.458294 -0.852964 -2.255193 0.406563 0.738094 3.413722 0.206355 -3.837760 -1.721973 2.453740 -1.339913 -2.176685 -0.443307 -1.164737 2.930455 3.035834 3.911133 2.271020	0.470039 0.568913 -0.716538 -0.108820 -0.579371 -0.896083 -0.732832 -0.438636 0.003701 0.172608 0.600154 0.730276 1.036454 0.815907 -1.011201 0.495978 -1.291591 -0.559767 0.022930 0.531938 -0.461744 0.554744 0.977042 -0.447900 -0.898552 0.059411 0.069500 -1.087540 0.220615 0.856124 -0.963226 1.094092 -1.005216 -1.459875 -1.619425 0.001784 1.184979 0.192985 1.656185 1.805484
3: n CCCCCCCCCCCCCCCCCCC	hin(GS) in C⊢ -0.722875 -0.873848 0.271135 1.560752 1.689874 0.532259 0.189294 2.734308 2.617642 1.300738 3.777369 5.034364 5.157589 4.022720 4.117533	I ₂ CI ₂ solution 1.169482 0.414177 -0.028037 0.301834 1.071926 1.495482 -0.811324 -0.140390 -0.911226 -1.229864 -1.334930 -1.008427 -0.253964 0.189336 0.971884	1.804170 0.641395 -0.046281 0.455305 1.641249 2.299762 -1.253158 -0.231952 -1.419044 -1.903687 -2.077521 -1.580243 -0.418267 0.269933 1.475130

OTTTTTTCCCCCCTTTTTCCTTTTCCTTT	3.090867 5.105963 -0.793504 -1.611771 1.215144 5.925328 -2.246591 -2.892658 -2.925855 -4.186786 -4.220352 -4.831744 -5.837513 3.685973 6.142311 0.621126 -4.688320 -2.175406 -2.763296 -2.997188 -2.016853 -3.670928 -4.747475 -2.239272 -2.863901 -3.102294 -3.774600 -2.138757	1.031212 1.981202 1.219912 -1.063837 1.507227 -1.819796 -1.344497 0.091591 -1.084817 0.967673 -1.383221 0.683374 -0.491184 -0.717486 -1.924615 -0.001733 2.085242 -2.290409 -1.892900 -3.105406 -3.765873 -3.579517 -2.913871 1.357978 2.085746 3.020177 2.564676 3.429968 3.820516	2.120031 3.035486 1.852496 -1.637548 2.329358 -2.812377 -2.102252 0.153085 0.557452 -0.704578 0.122266 -1.148440 -0.726299 -1.067792 -2.985767 -0.034027 3.208268 0.433476 1.384165 1.831119 0.989310 2.467816 2.413460 -1.810596 -1.064328 -1.931587 -2.898671 -1.481841 -2.076487
3.000000000000000000000000000000000000	min(GS) in n- -0.722392 -0.874079 0.269121 1.559024 1.689489 0.532973 0.185222 2.731267 2.612819 1.295359 3.771444 5.028984 5.153716 4.020244 4.116365 3.007271 3.092288 5.105409 -0.798502 -1.611290 1.208305 5.919183 -2.246669 -2.905962 -2.911435 4.100203	hexane solut 0.299812 0.107008 -0.006418 0.077100 0.273445 0.382479 -0.205780 -0.205780 -0.232603 -0.312865 -0.341263 -0.258836 -0.066847 0.258836 -0.066847 0.246167 0.353874 0.504297 0.308753 -0.269301 0.386231 -0.462934 -0.345073 0.024482 -1.207187 1.178687 1.200640	ion 2.132016 0.759632 -0.051152 0.541371 1.941106 2.717285 -1.475310 -0.269106 -1.670398 -2.242478 -2.446679 -1.859743 -0.488367 0.323569 1.746025 2.514284 3.587349 2.192182 -1.928651 2.750250 -3.315464 -2.475865 0.180975 0.074406 -0.253574 0.449070

ССНННННОСНННОСННН	-4.204978	1.111936	-0.778861
	-4.830071	-0.125755	-0.868279
	-5.835174	-0.184002	-1.275665
	3.678549	-0.491439	-3.519096
	6.139021	-0.003251	-0.034108
	0.623341	0.533011	3.789935
	-4.710371	-2.241210	-0.531751
	-2.200629	-2.286217	0.506196
	-2.794581	-3.568251	0.418220
	-3.025415	-3.828948	-0.621039
	-2.054363	-4.265498	0.810374
	-3.706511	-3.626708	1.023495
	-4.720582	2.002535	-1.114429
	-2.211189	2.337334	-0.129641
	-2.811634	3.546035	-0.557104
	-3.043882	3.517751	-1.627854
	-3.723737	3.760021	0.011767
	-2.074901	4.326402	-0.367083
3 CCCCCCCCCCCCCCCCCHHHHHCCCCCCCHHHHHOCHH	nin(ICT) in CH -0.842286 -0.931843 0.282598 1.541831 1.593534 0.383257 0.293328 2.760346 2.732491 1.467141 3.947070 5.172722 5.215158 4.017336 4.030925 2.865298 2.893967 4.989978 -0.648683 -1.754954 1.429372 6.098651 -2.243344 -2.524971 -3.325191 -3.773103 -4.566453 -4.780652 -5.760644 3.919359 6.168295 0.413770 -3.960128 -1.507545 -1.631321 -1.849736 -0.665252	I2CI2 solution 1.896272 0.641797 -0.008246 0.592338 1.843090 2.480990 -1.212160 -0.046289 -1.282262 -1.828539 -1.900035 -1.313734 -0.116451 0.544122 1.784258 2.407984 3.357174 2.231788 -1.662384 2.400331 -2.743654 -1.803871 0.026949 -1.289875 0.697084 -1.905689 0.093245 -1.658152 -2.839413 0.340238 3.447377 -2.891425 -1.839186 -3.147283 -3.878187 -3.359705	0.915228 0.282107 -0.139594 0.173705 0.867454 1.205122 -0.883373 -0.212262 -0.932922 -1.270517 -1.295654 -0.960025 -0.269209 0.117579 0.821047 1.175329 1.705241 1.069192 -1.178144 1.214969 -1.855385 -1.248097 0.116982 0.642216 -0.523305 0.493844 -0.646421 -0.143096 -0.255582 -1.841206 -0.255582 -1.841206 -0.016199 1.702233 0.898504 1.298951 1.860524 1.077895 2.313272

HHOCHHH	-2.415217 -5.382626 -3.035272 -4.055742 -4.403701 -4.900416 -3.598269	-3.158649 0.600708 1.924616 2.647977 2.104017 2.850288 3.587868	2.622230 -1.143868 -1.014063 -1.687670 -2.572430 -1.020587 -1.993669
3 ΟΟΟΟΟΟΟΟΟΟΟΟΟΟΟΙΙΙΙΙΟΟΟΟΟΟΙΙΙΙΙΟΟΙΙΙΙΟΟΙΙΙΙ	$\begin{array}{r} \text{min(ICT) in n-h}\\ & -0.844915\\ & -0.935182\\ & 0.274715\\ & 1.531385\\ & 1.585996\\ & 0.368424\\ & 0.290315\\ & 2.754887\\ & 2.726326\\ & 1.463999\\ & 3.942900\\ & 5.157497\\ & 5.199963\\ & 4.008134\\ & 4.018714\\ & 2.852187\\ & 2.879372\\ & 4.977342\\ & -0.649274\\ & -1.759336\\ & 1.427059\\ & 6.086468\\ & -2.240314\\ & -2.464225\\ & -3.362545\\ & -3.692075\\ & -4.587412\\ & -4.749547\\ & -5.712626\\ & 3.915289\\ & 6.154525\\ & 0.401712\\ & -3.814488\\ & -1.416806\\ & -1.438702\\ & -1.642334\\ & -0.443527\\ & -2.189177\\ & -5.433464\\ & -3.162730\\ & -4.217460\\ & -4.538781\\ & -5.072599\\ & -3.808277\\ \end{array}$	exane soluti 1.961821 0.678278 0.024146 0.633898 1.902948 2.557159 -1.206349 -0.027644 -1.290446 -1.838407 -1.925963 -1.332081 -0.106647 0.568941 1.833210 2.471772 3.439403 2.285549 -1.652869 2.475766 -2.765903 -1.830789 0.038973 -1.292013 0.659374 -1.976463 -0.005847 -1.324608 -1.325623 -1.32756 -1.324608 -1.32562 -1.325765 -1.325765 -1.325765 -1.325765 -1.3257755 -1.3257755 -1.3257755 -1.3257755 -1.3257755 -1.3257755 -1.3257755 -1.3257755 -1.3257755 -1.3257755 -1.3257755 -1.3257755 -1.3257755 -1.357755 -1.357755 -1.357755	on 0.835480 0.255739 -0.153455 0.139143 0.784771 1.096959 -0.865786 -0.215697 -0.894305 -1.226249 -1.227312 -0.903935 -0.252407 0.102832 0.762573 1.081877 1.577595 1.003011 -1.73585 1.112916 -1.791682 -1.167648 0.130480 0.641413 -0.463846 0.511239 -0.565325 -0.086129 -0.183115 -1.742270 -0.011809 1.559944 0.915415 1.270550 1.789496 0.990819 2.196613 2.580350 -1.032417 -0.943781 -1.620115 -2.502794 -0.955782 -1.934364
3 :	min(LE) in CH_2	Cl ₂ solution	0 852300

С	-0.852237	1.922109	0.852309
С	-0.939450	0.632421	0.279611

ССССССССССССТТТТТССССССТТТТТОСТТТТОСТТТ	0.295902 1.549343 1.592713 0.348909 0.307348 2.766407 2.742029 1.496650 3.979786 5.187616 5.224653 4.019978 4.023047 2.830758 2.855808 4.972971 -0.633235 -1.771508 1.464105 6.118517 -2.244954 -2.513754 -3.326710 -3.763170 -4.579651 -4.783639 -5.759661 3.955939 6.176310 0.379454 -3.956640 -1.498484 -1.661919 -1.872857 -0.713208 -2.462528 -5.389382 -3.038697 -4.058831 -5.598016	-0.020220 0.589310 1.849444 2.509275 -1.197285 -0.042442 -1.270776 -1.813991 -1.893970 -1.311001 -0.105378 0.555362 1.782747 2.417954 3.376346 2.241357 -1.641440 2.430727 -2.735257 -1.797574 0.002939 -1.300776 0.689139 -1.300776 0.689139 -1.897450 0.097667 -1.192474 -1.654479 -2.831219 0.349146 3.475941 -2.887785 -1.882385 -3.204856 -3.904440 -3.465522 -3.253725 0.620179 1.922354 2.661926 2.137696 2.856721 3.605519	-0.135778 0.177391 0.846120 1.149176 -0.898439 -0.197366 -0.922660 -1.275527 -1.278179 -0.931623 -0.236706 0.140241 0.829439 1.163811 1.675258 1.089561 -1.203538 1.120117 -1.850793 -1.209034 0.125844 0.626180 -0.492849 0.476364 -0.633528 -0.153679 -0.266562 -1.827368 0.024865 1.644220 0.867626 1.307851 1.797310 0.981491 2.265403 2.542939 -1.125902 -0.978158 -1.633274 -2.527016 -0.960069 -1.923857
3 : mir	n(LE) in <i>n</i> -he	exane solutio	n
CCCCCCCCCCCCCCC	-0.855393 -0.941307 0.291844 1.545214 1.587303 0.346169 0.303070 2.761148 2.737497 1.490145 3.972660 5.180314 5.216416	1.937895 0.642906 -0.009746 0.597336 1.859978 2.523238 -1.187514 -0.037222 -1.266338 -1.806108 -1.891622 -1.312184 -0.105499	0.827649 0.264762 -0.148061 0.167237 0.832200 1.124682 -0.913068 -0.201059 -0.925093 -1.284323 -1.273446 -0.921750 -0.228028

С	4.015006	0.557265	0.141257
С	4.017026	1.788355	0.828954
С	2.827473	2.426617	1.154565
Н	2.851432	3.386554	1.663510
Н	4.967517	2.243974	1.092671
Н	-0.638595	-1.625168	-1.224719
Н	-1.774812	2.448774	1.089274
Н	1.458855	-2.726207	-1.861883
Н	6.111333	-1.801180	-1.194306
С	-2.243299	0.004035	0.128067
С	-2.493189	-1.302112	0.632575
С	-3.340326	0.676125	-0.478432
С	-3.735771	-1.915258	0.493215
С	-4.585849	0.068139	-0.609595
С	-4.770352	-1.224848	-0.129028
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Н	3.949181	-2.829041	-1.822498
Н	6.168270	0.346914	0.036585
Н	0.377057	3.493234	1.613219
Н	-3.912633	-2.907713	0.887425
0	-1.464081	-1.868446	1.304842
С	-1.601622	-3.187641	1.799968
Н	-1.811303	-3.895628	0.990016
Н	-0.643477	-3.432087	2.258055
Н	-2.393280	-3.246491	2.555447
Н	-5.406035	0.580930	-1.095020
0	-3.073522	1.915711	-0.960295
С	-4.096014	2.630066	-1.631303
Н	-4.435128	2.090682	-2.522894
Н	-4.948215	2.821015	-0.969415
Н	-3.647030	3.577442	-1.929138

3. Figures



Fig. S7 UV-vis absorption (left) and normalized fluorescence emission (right) spectra of (a) **1b** and (b) **2b** in various solvents (UV-vis: 10 μ M; FL: 1 μ M) at 298 K. The solvents used were: *n*-hexane, toluene, chloroform (CHCl₃), tetrahydrofuran (THF), dichloromethane (CH₂Cl₂), methanol (CH₃OH), acetonitrile (CH₃CN) and dimethyl sulphoxide (DMSO).



Fig. S8 Normalized excitation spectra of **1a** (1 μ M) in (a) *n*-hexane, (b) toluene, (c) CHCl₃, (d) THF, (e) CH₂Cl₂, (f) CH₃OH, (g) CH₃CN and (h) DMSO monitored at the wavelength of monomer emission (381 nm) and the respective maximum emission wavelengths in each solvent.



Fig. S9 Picosecond time-resolved fluorescence spectra of **2b** in (a) *n*-hexane, (b) CH_2CI_2 , (c) C_2H_5OH and (d) $1-C_4H_9OH$ with photoexcitation at 345 nm. * denotes Raman scattering signals from solvents, which are confirmed by the measurements of the blank solutions without the solute.

4. Supplementary References

- [1] S. Sagar and I. R. Green, Synthesis, 2009, 2009, 935–940.
- [2] T. Tu, Z. Sun, W. Fang, M. Xu and Y. Zhou, Org. Lett., 2012, 14, 4250–4253.
- [3] F. Lu, T. Takaya, K. Iwata, I. Kawamura, A. Saeki, M. Ishii, K. Nagura and T. Nakanishi, *Sci. Rep.*, 2017, 7, 3416.
- [4] Y. Nojima and K. Iwata, J. Phys. Chem. B, 2014, 118, 8631–8641.