1	Supporting Information for
2	Photoelectron Spectroscopy and Density Functional Theory Studies of $(FeS)_mH^-$ (m
3	= 2 - 4) Cluster Anions: Effects of the Single Hydrogen
4	Shi Yin, Elliot R. Bernstein*
5	Department of Chemistry, NSF ERC for Extreme Ultraviolet Science and Technology, Colorado State
6	University, Fort Collins, CO 80523, USA
7	AUTHOR INFORMATION
8	Corresponding Author
9	*Elliot R. Bernstein, E-mail. erb@colostate.edu
10 11 12	The following results are supplied as additional detailed information for these studies: 1.

13 The relative energy of  $(FeS)_{2-4}H^{-}$  cluster anions with different spin multiplicities at

14 BPW91/TZVP level in Figure S1 to S3, respectively; and 2. The first experimental and

15 calculated VDEs (in eV) for  $(FeS)_m$  (m = 2 - 8) cluster anions (taken from ref.[1]).



1 2 Figure S1. The relative energy of (FeS)<sub>2</sub>H<sup>-</sup> clusters with different spin multiplicities at BPW91/TZVP level

3 [Isomer I, SH-type] and [isomer II, FeH-type].



1 2 Figure S2. The relative energy of (FeS)<sub>3</sub>H<sup>-</sup> clusters with different spin multiplicities at BPW91/TZVP level

3 [isomer III, Fe<sub>3</sub>H-type].



Figure S3. The relative energy of (FeS)<sub>4</sub>H<sup>-</sup> clusters with different spin multiplicities at BPW91/TZVP level.

4	Table S1. The first calculated VDEs (in eV) for $(FeS)_m$ ( $m = 2 - 8$ ) at the BPW91/TZVP level, as well as the experimental
5	results for comparison. Different spin multiplicities of each cluster are displayed in order of their relative energy low to high
6	from top to bottom. All calculated and experimental data are taken from ref. [1].

Clusto	Spin	VDE (eV)	
Cluste	multiplicity	Calculate	Experimenta
r		d	Ι
(FeS)2 <sup>-</sup>	8	2.04	2.34
	2	1.88	-
(FeS) <sub>3</sub> -	10	3.24	3.57
	12	2.98	-
(FeS) <sub>4</sub> <sup>-</sup>	16	2.87	2.71
	8	2.20	_
	14	2.76	_
	10	1.83	-
	2	2.29	-
(FeS)5 <sup>-</sup>	12	3.08	3.54
	18	3.13	-
	10	3.06	
	20	3.26	

	16	3.02	
	4	3.04	
	8	2.56	
	6	2.86	
(FeS) <sub>6</sub> -	20	3.06	3.71
	22	3.08	
	16	4.33	
	2	4.47	
	8	6.03	
	6	3.45	
	24	3.27	
	18	2.85	
(FeS) <sub>7</sub> -	10	3.50	3.83
	8	3.59	
	6	3.63	
	14	3.84	
	2	3.73	
	16	3.47	
(FeS) <sub>8</sub> -	8	4.10	3.85
	4	3.68	
	10	4.64	

1 2

## 3 Reference

- 4 1. Yin, S.; Bernstein, E. R., Photoelectron Spectroscopy and Density Functional Theory Studies of Iron Sulfur (FeS)<sub>m</sub><sup>-</sup> (m = 2–8)
- 5 Cluster Anions: Coexisting Multiple Spin States. J. Phys. Chem. A **2017**, 121, 7362-7373.