

Electronic Supplementary Information

High CO₂ Absorption Capacity by Chemisorption at Cations and Anions in Choline-Based Ionic Liquids

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Contents

1. NMR spectroscopic characterization data
2. ¹H and ¹³C NMR of Ionic liquids
3. ¹H-¹³C HMBC spectra of [N_{1,1,6,2OH}][4-Triz]⁺¹³CO₂ in D₂O
4. CO₂ absorption capacities of Ionic Liquids
5. Viscosity of the synthesized choline based N-heterocyclic ionic liquids as a function of temperature
6. Thermogravimetric analysis (TGA) of [N_{1,1,6,2OH}][4-Triz], [N_{1,1,6,2O4}][4-Triz] and [N_{1,1,6,2O4}][Pyrz]
7. Comparative ¹³C NMR data of [N_{1,1,6,2OH}][4-Triz] and [N_{1,1,6,2OH}][4-Triz]⁺CO₂
8. FT-IR spectra of [N_{1,1,6,2OH}][4-Triz] and [N_{1,1,6,2OH}][4-Triz]⁺CO₂
9. Computational data

1. Spectroscopic Characterization data:

[N_{1,1,6,2O4}][4-Triz] : light brown colour semi-solid; yield 83%; ¹H NMR (400 MHz, D₂O): δ 0.86-0.92 (m, 6H), 1.31-1.38 (m, 8H), 1.54-1.61 (m, 2H), 1.72-1.78 (m, 2H), 3.10 (s, 6H) 3.32-3.36 (m, 2H), 3.52-3.58 (m, 4H), 3.90 (bs, 2H), 8.09 (s, 2H); ¹³C NMR (100.6 MHz, D₂O) δ 15.90, 16.03, 21.60, 24.55, 24.73, 27.98, 33.27, 33.57, 54.19, 65.63, 66.66, 68.28, 73.91, 153.45.

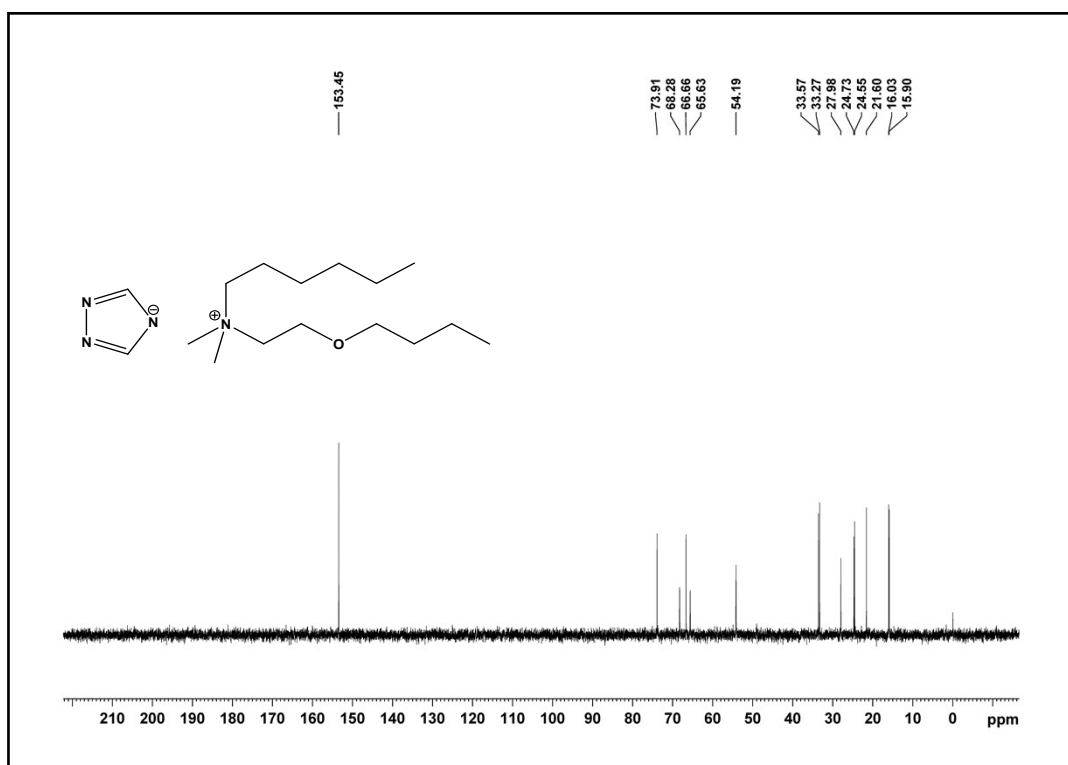
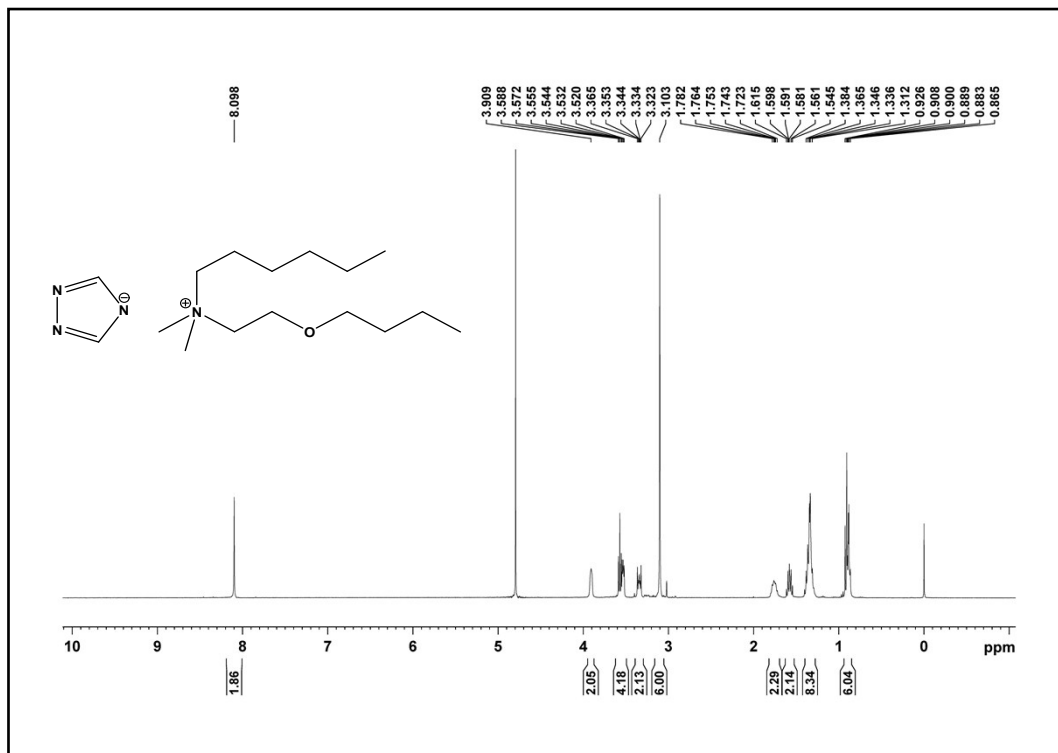
[N_{1,1,6,2O4}][Pyrz] : light brown colour solid; yield 81%; ¹H NMR (400 MHz, D₂O): δ 0.87-0.93 (m, 6H), 1.33-1.40 (m, 8H), 1.54-1.61 (m, 2H), 1.76 (bs, 2H), 3.10 (s, 6H) 3.32-3.36 (m, 2H), 3.52-3.59 (m, 4H), 3.91 (bs, 2H), 6.43 (s, 1H), 7.71 (s, 2H); ¹³C NMR (100.6 MHz, D₂O) δ 15.78, 15.90, 21.47, 24.42, 24.60, 27.85, 33.14, 33.44, 54.07, 65.51, 66.53, 68.14, 73.78, 107.65, 137.33.

[N_{1,1,6,2OH}][4-Triz]: brown colour oil; yield 91%; ¹H NMR (400 MHz, D₂O): δ 0.88 (t, 3H, *J*= 6.8 Hz), 1.33 (s, 6H), 1.76-1.78 (m, 2H), 3.10 (s, 6H), 3.34 (t, 2H, *J*=8.4 Hz), 3.44-3.47 (m, 2H), 4.02 (s, 2H), 8.09 (s, 2H); ¹³C NMR (100.6 MHz, D₂O) δ 153.42, 68.21, 67.52, 58.03, 53.99, 53.95, 33.09, 27.81, 24.52, 24.38, 15.87.

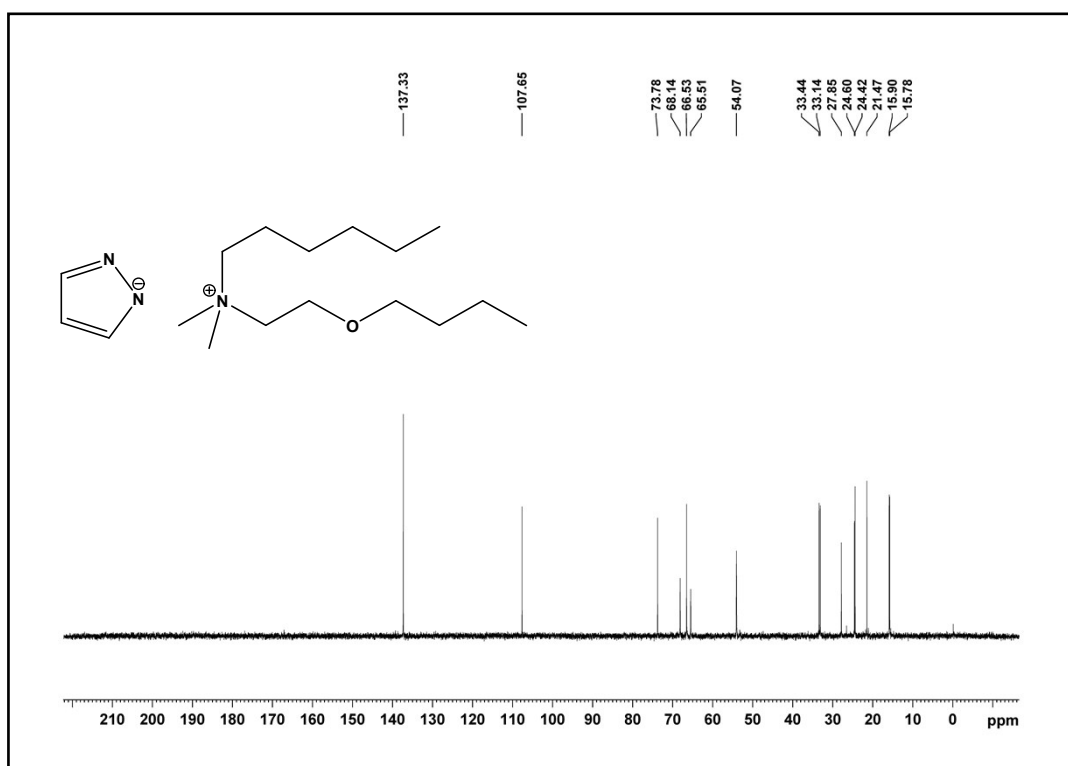
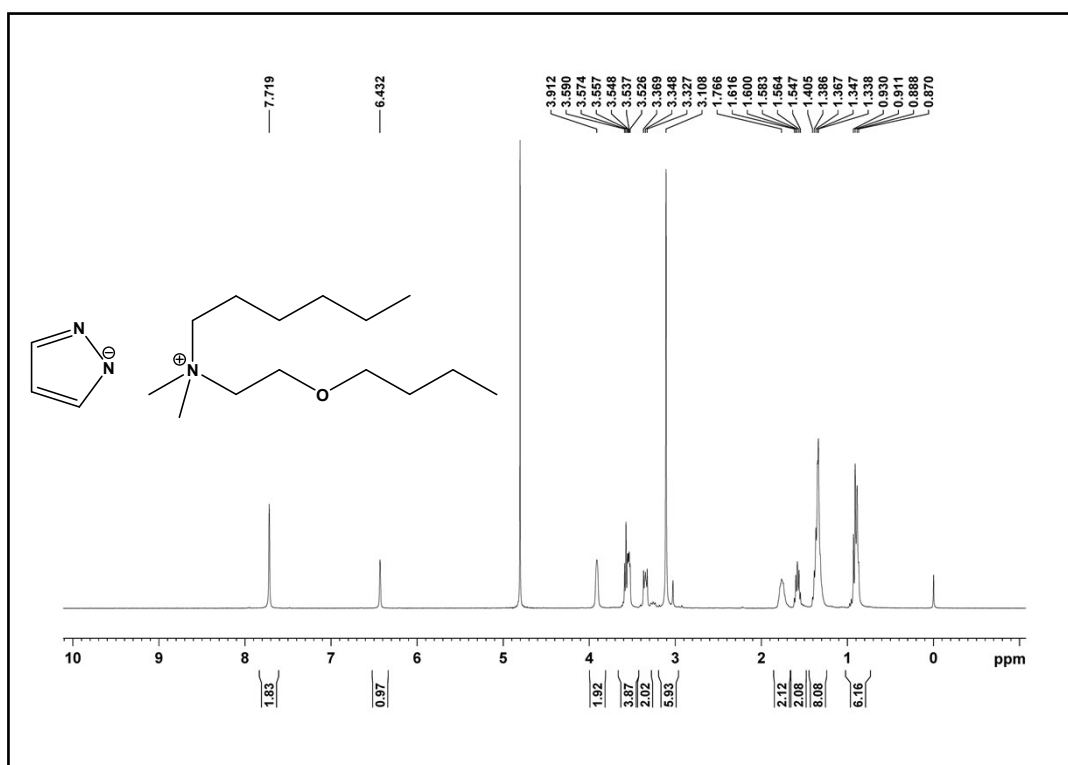
[N_{1,1,6,2OH}][Imi]: brown colour oil; yield 92%; ¹H NMR (400 MHz, D₂O): δ 0.87 (t, 3H, *J*= 6.8 Hz), 1.28-1.37 (m, 6H), 1.74-1.78 (m, 2H), 3.10 (s, 6H), 3.31-3.35 (m, 2H), 3.42 (t, 2H, *J*=5.2 Hz), 4.00 (s, 2H), 7.12 (s, 2H), 7.75 (s, 1H); ¹³C NMR (100.6 MHz, D₂O) δ 15.87, 24.38, 24.53, 27.82, 33.09, 53.91, 53.94, 53.98, 58.08, 67.94, 68.15, 124.67, 139.13

[N_{1,1,6,2OH}][Succ]: light brown colour viscous oil; yield 94%; ¹H NMR (400 MHz, D₂O): δ 0.89 (t, 3H, *J*= 6.8Hz), 1.30-1.39 (m, 6H), 1.77-1.82 (m, 2H), 2.44-2.53 (m, 4H) 3.13 (s, 6H), 3.35-3.39 (m, 2H), 3.48-3.50 (m, 2H), 4.04 (m, 2H); ¹³C NMR (100.6 MHz, D₂O) δ 15.87, 24.38, 24.54, 27.82, 33.09, 34.41, 35.63, 54.01, 58.05, 67.52, 68.23, 181.69, 183.77.

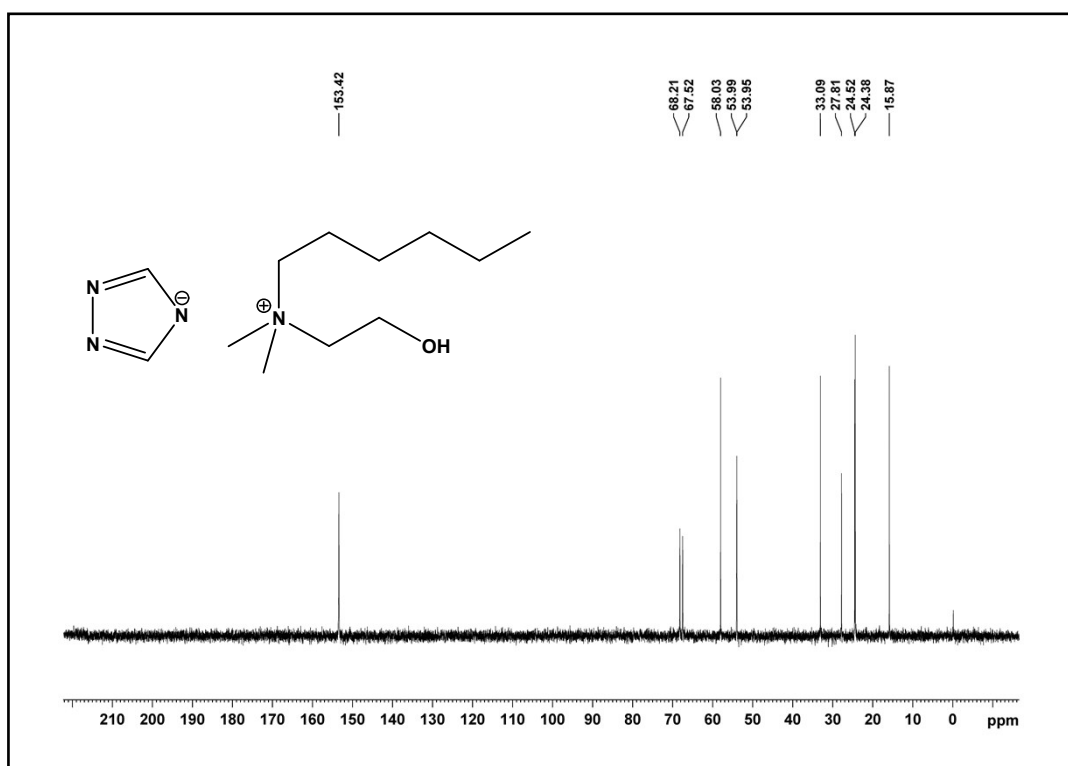
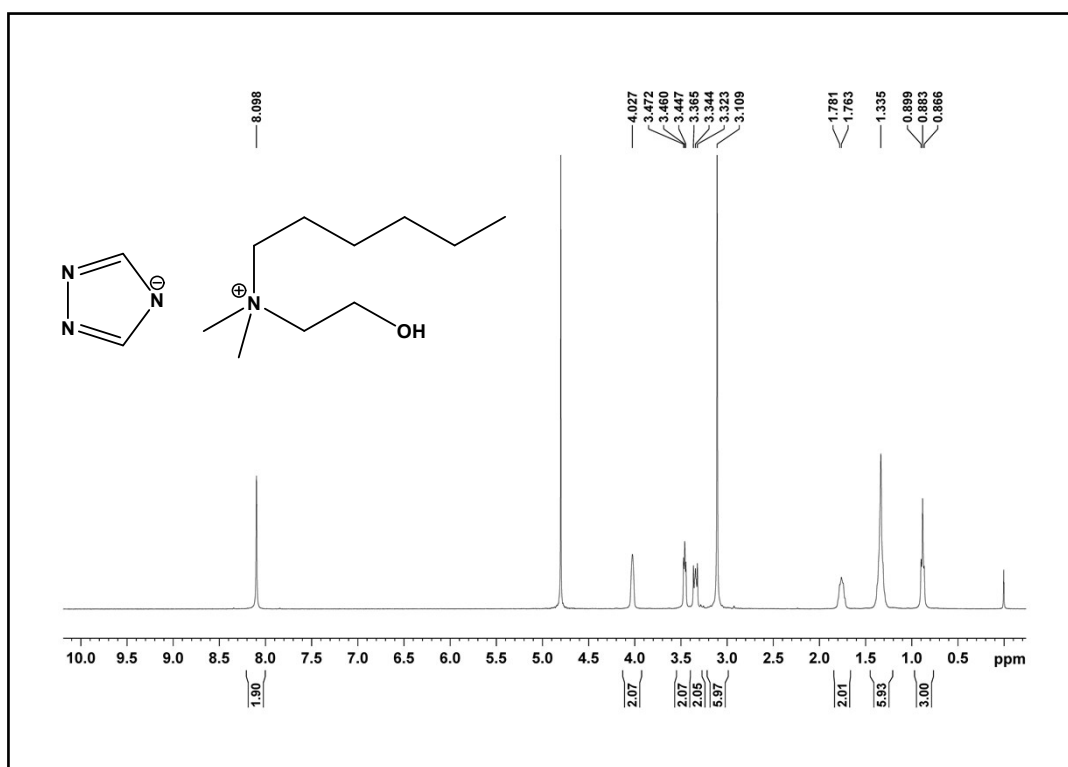
2. ^1H and ^{13}C NMR of all ILs



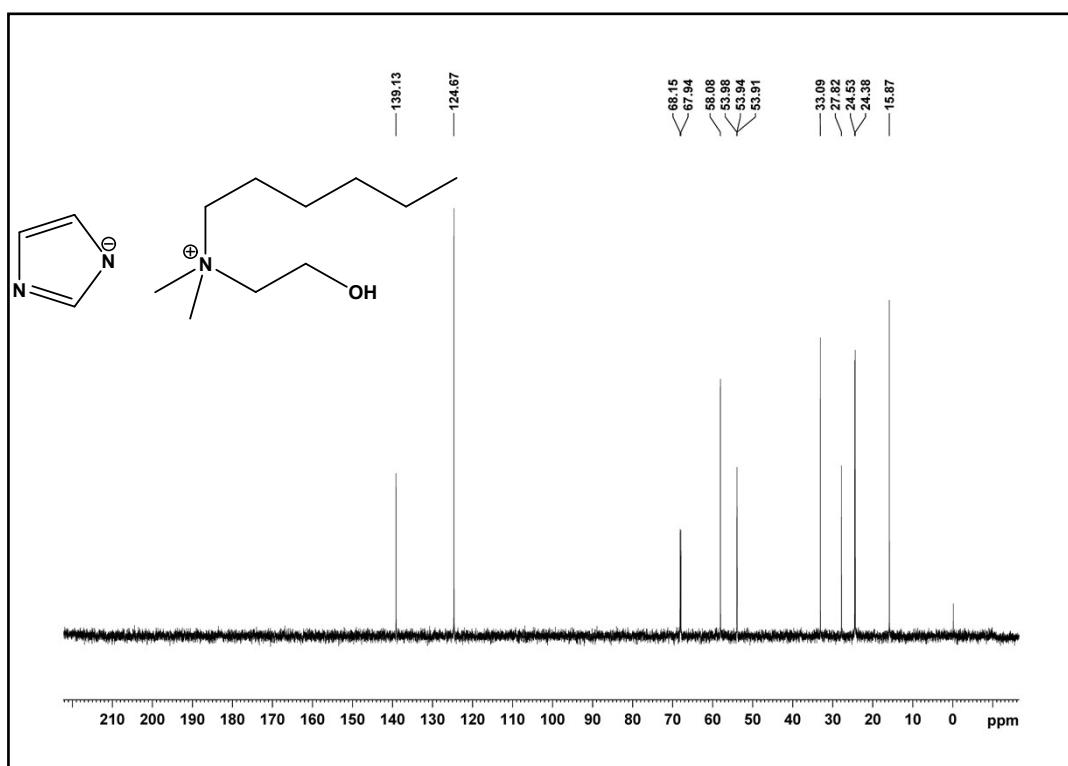
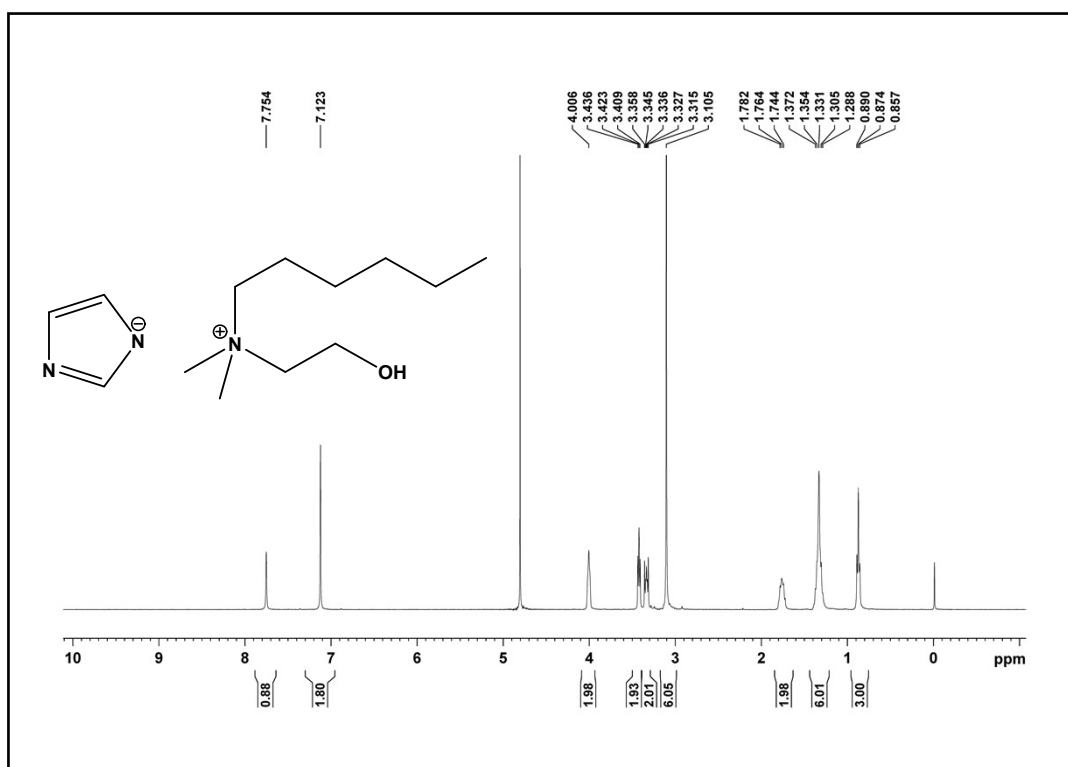
^1H and ^{13}C NMR of $[\text{N}_{1,1,6,204}][4\text{-Triz}]$



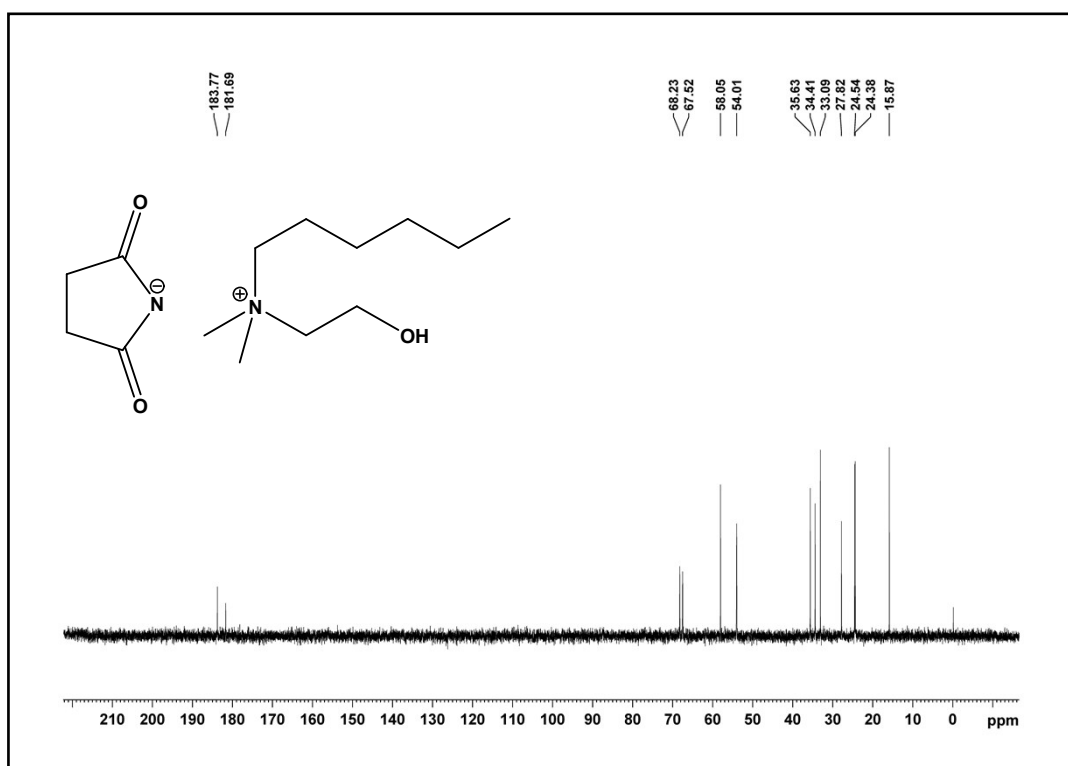
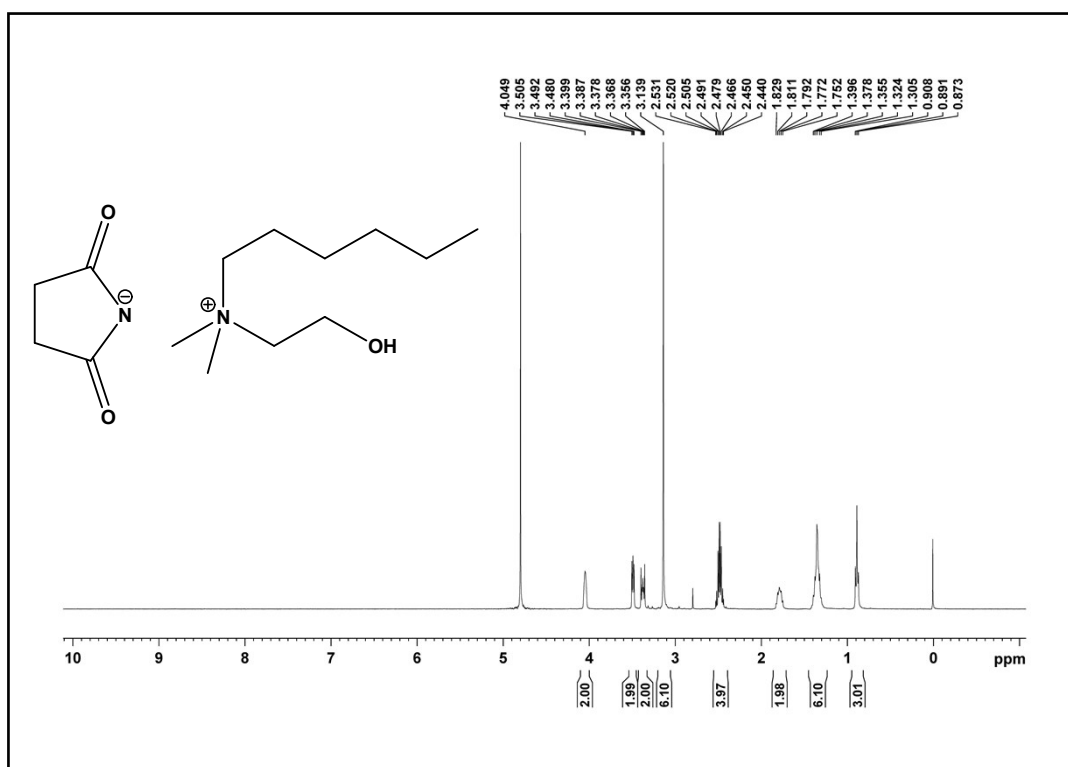
¹H and ¹³C NMR of [N_{1,1,6,2}O₄][Pyrz]



¹H and ¹³C NMR of [N_{1,1,6,2OH}][4-Triz]

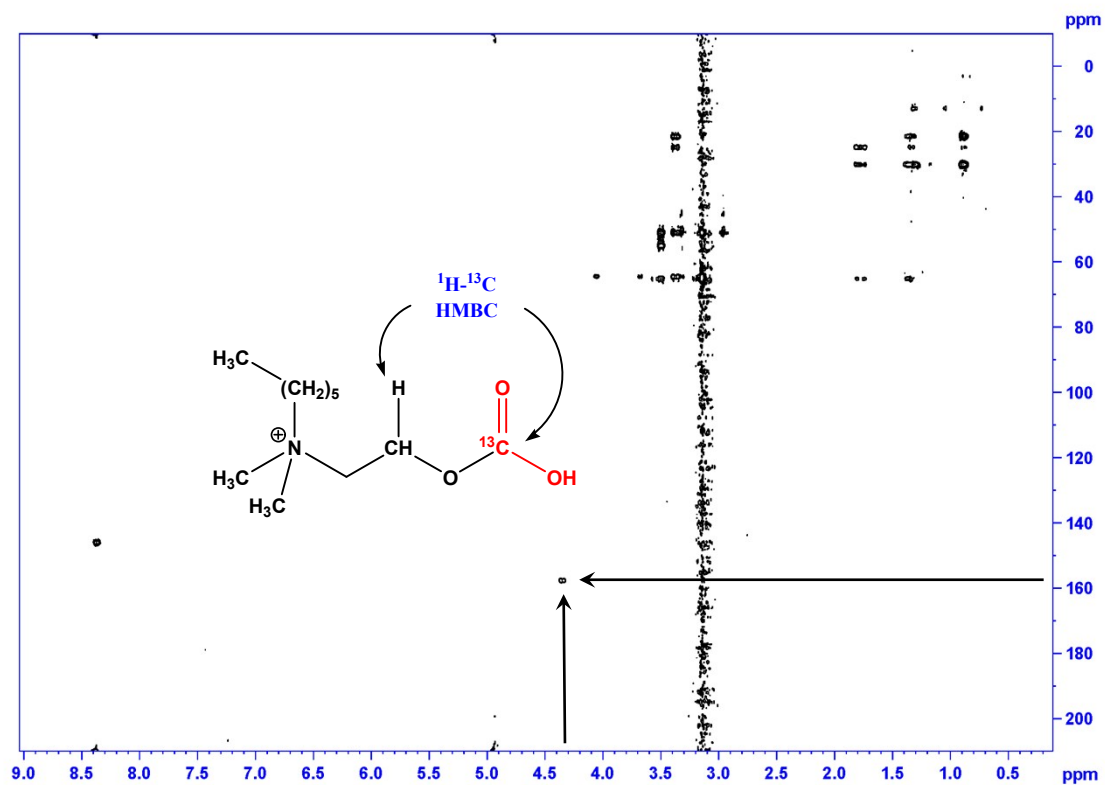


¹H and ¹³C NMR of [N_{1,1,6,2OH}][Imi]

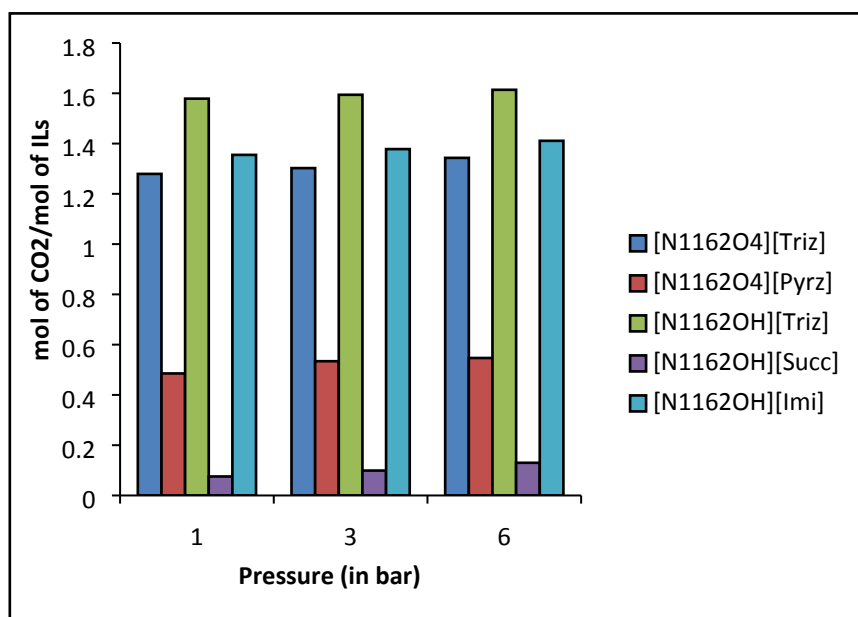


¹H and ¹³C NMR of [N_{1,1,6,2}OH][Succ]

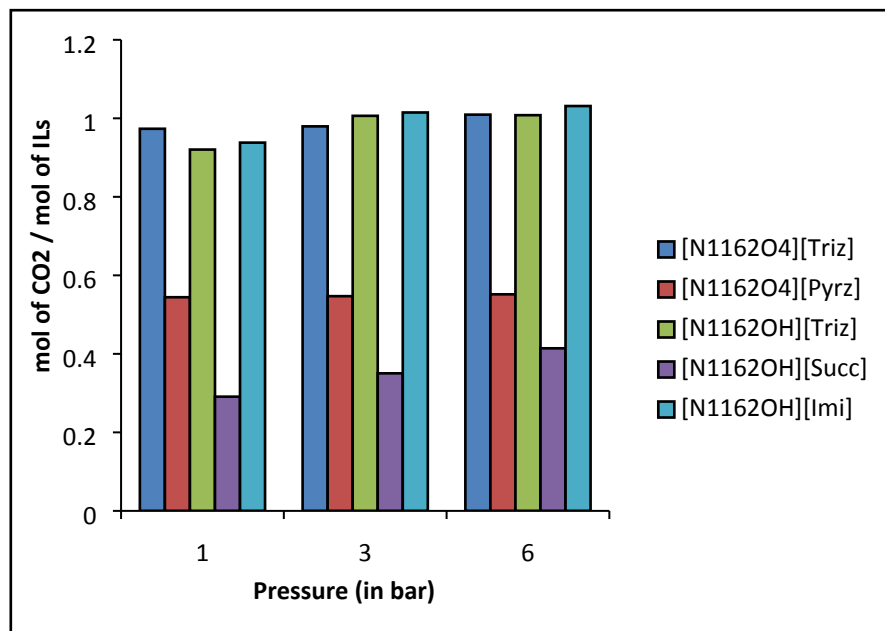
3. ^1H - ^{13}C HMBC spectra of $[\text{N}_{1,1,6,20\text{H}}][4\text{-Triz}]^{+13}\text{CO}_2$ in D_2O



4. CO₂ absorption capacities of Ionic Liquids

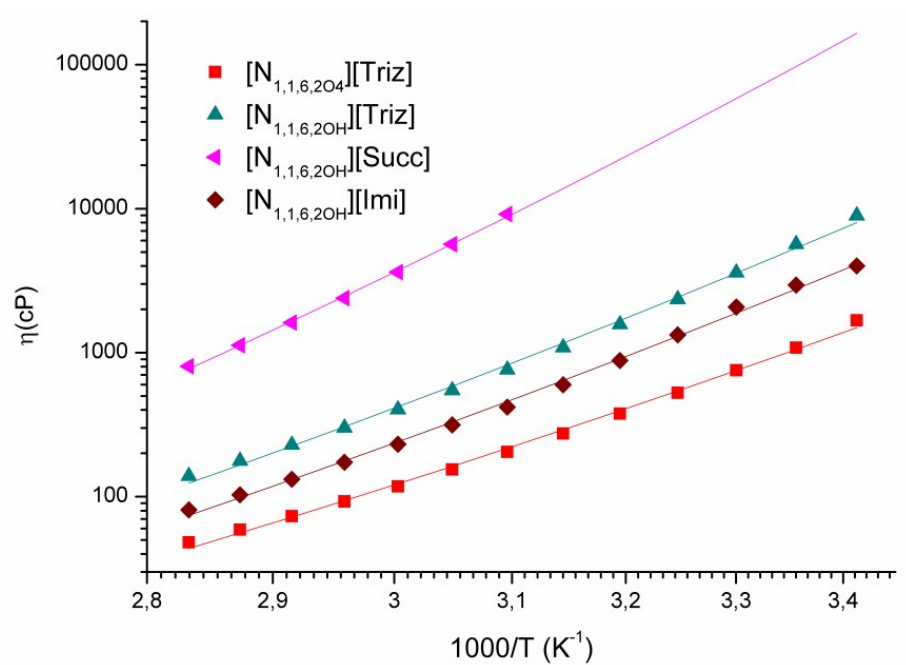


CO₂ capture capacity of ILs at 20°C

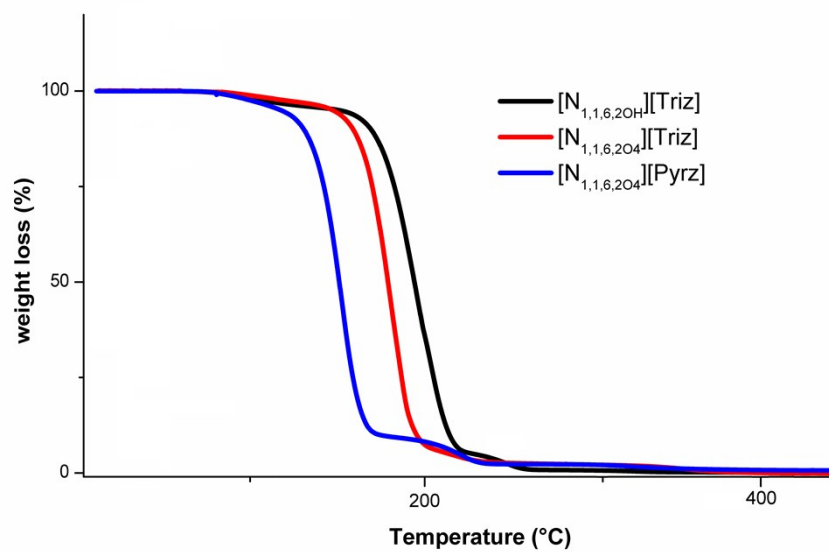


CO₂ capture capacity of ILs at 40°C

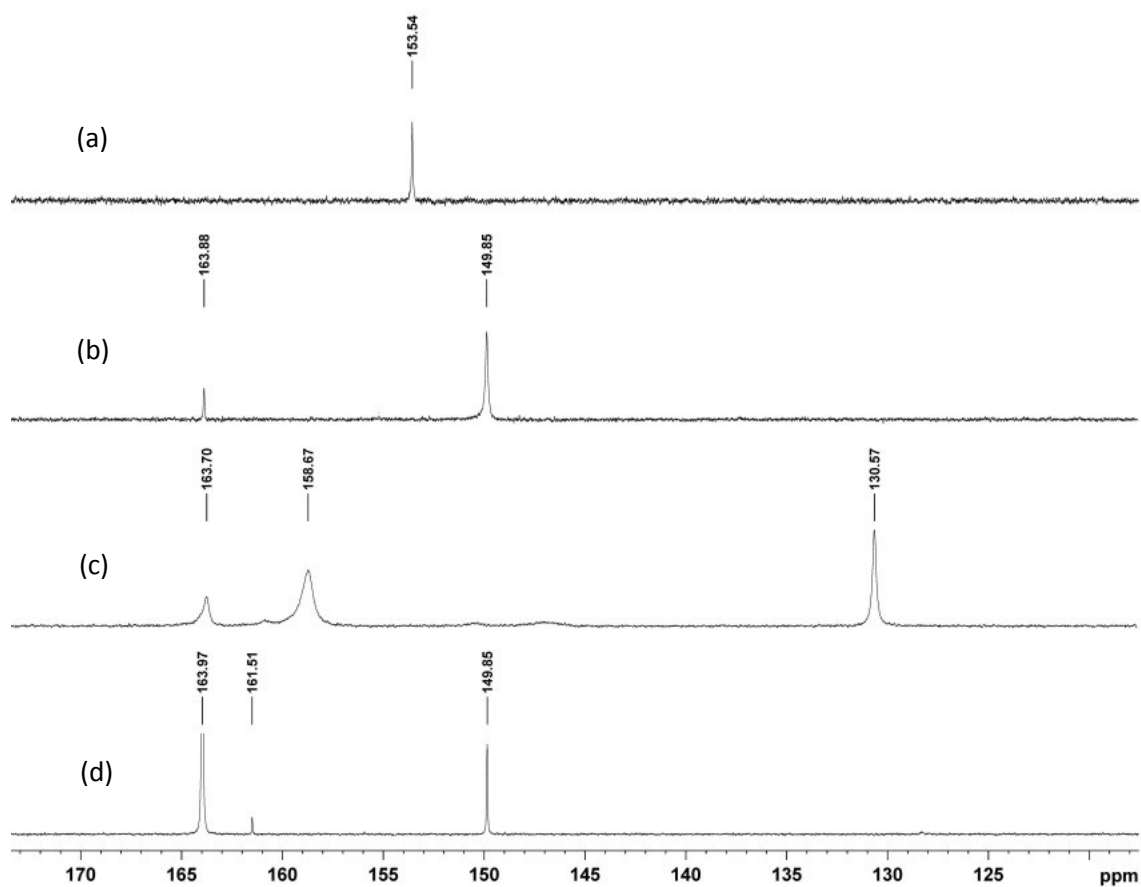
5. Viscosity of the synthesized choline based *N*-heterocyclic ionic liquids as a function of temperature



6. Thermogravimetric analysis (TGA) of $[N_{1,1,6,2OH}][4\text{-Triz}]$, $[N_{1,1,6,2O4}][4\text{-Triz}]$ and $[N_{1,1,6,2O4}][\text{Pyrz}]$

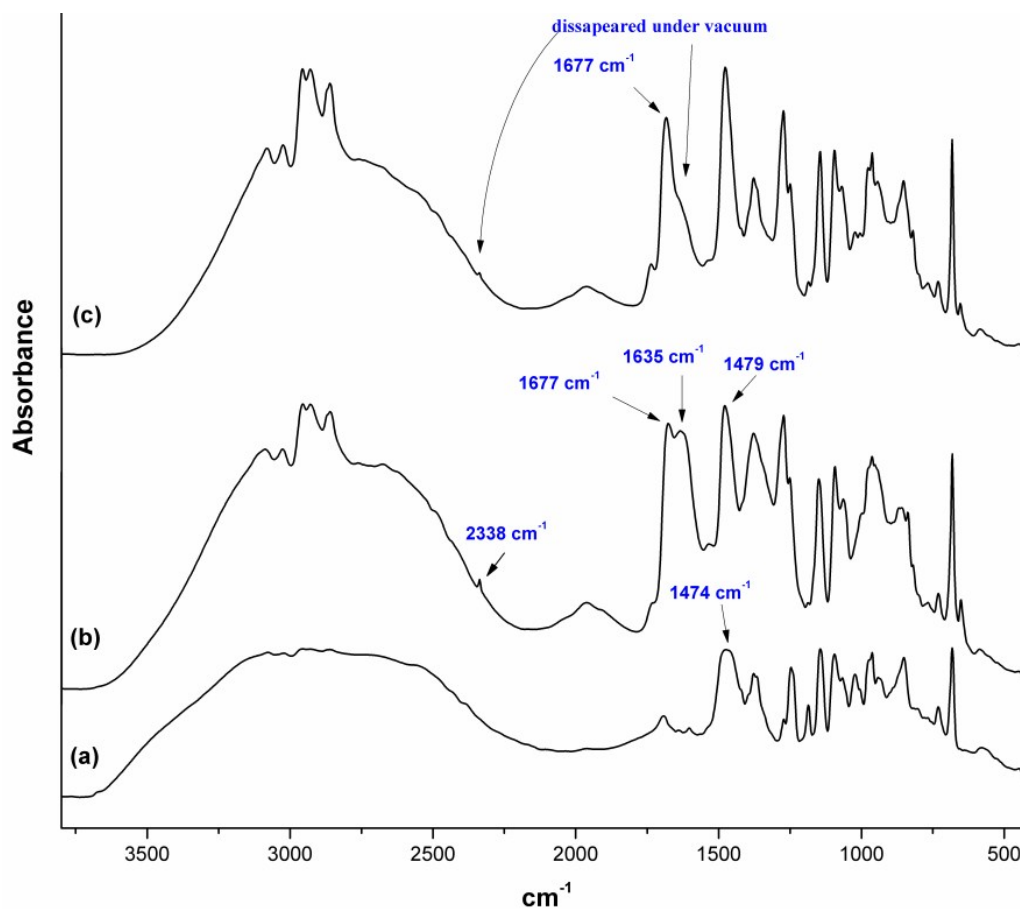


7. Comparative ^{13}C NMR data of $[\text{N}_{1,1,6,2\text{OH}}][4\text{-Triz}]$ and $[\text{N}_{1,1,6,2\text{OH}}][4\text{-Triz}]+\text{CO}_2$ reaction



(a) IL before CO_2 absorption dissolved in D_2O , (b) IL after CO_2 absorption dissolved in D_2O , (c) neat IL after $^{13}\text{CO}_2$ absorption with D_2O as an external lock, and (d) IL after $^{13}\text{CO}_2$ absorption dissolved in D_2O .

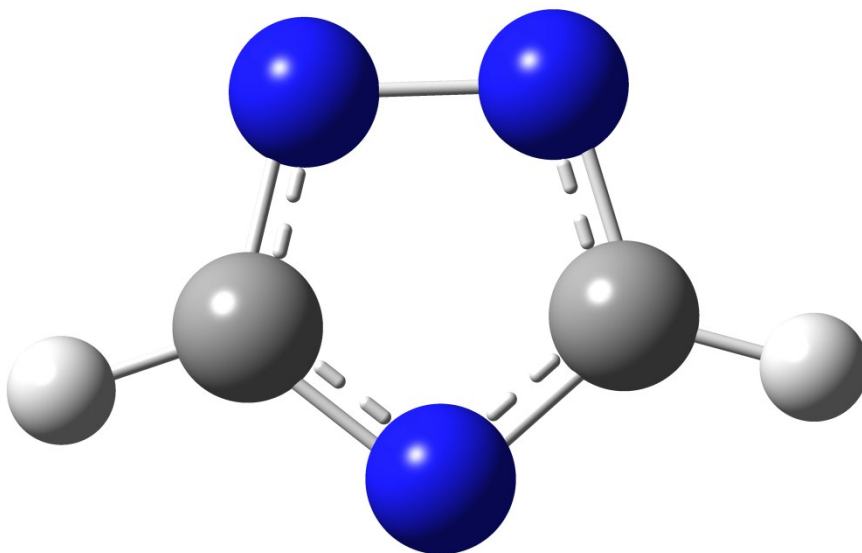
8. FT-IR spectra of $[N_{1,1,6,2OH}][4\text{-Triz}]$ and after CO_2 reaction



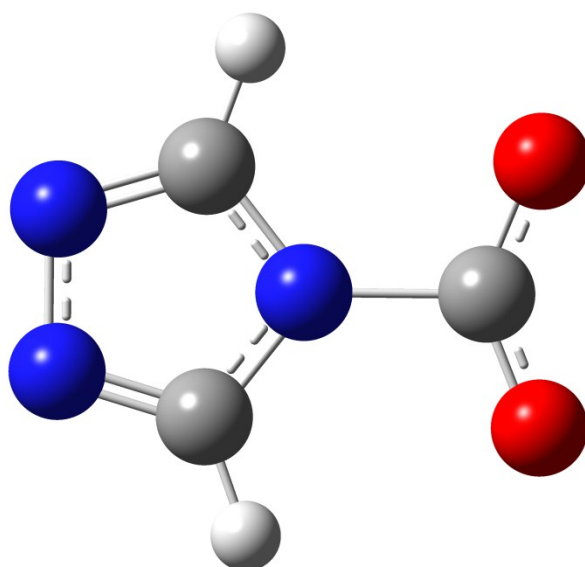
FT-IR spectra (under vacuum) of (a) IL $[N_{1,1,6,2OH}][4\text{-Triz}]$, (b) $[N_{1,1,6,2OH}][4\text{-Triz}]\text{-CO}_2$, (c) $[N_{1,1,6,2OH}][4\text{-Triz}]\text{-CO}_2$ after 5 minutes.

9. Computational data

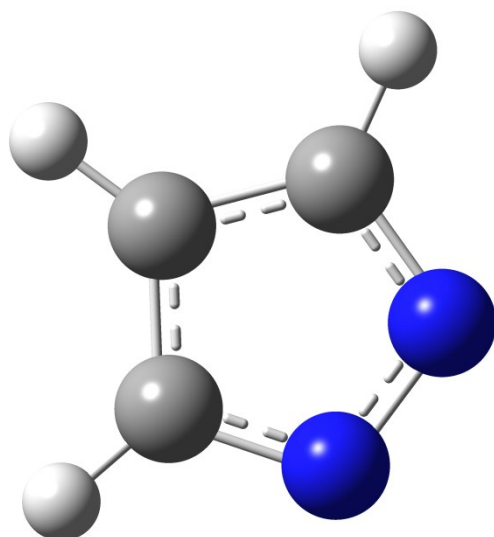
B3LYP/6-311+G(d,p) optimized geometry of the 4-triazole, pyrazole and Imidazole anion as well as their carbamate anion with Cartesian coordinate.



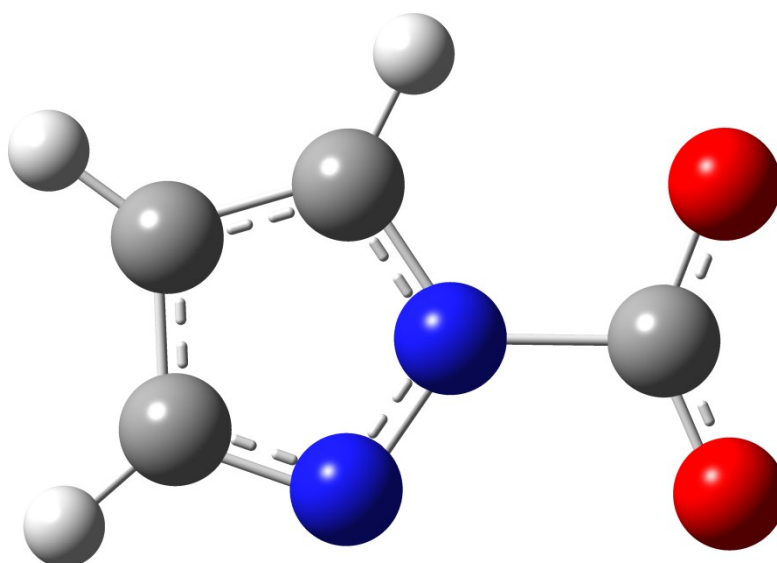
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C	0.37276400	1.02757000	0.00000000
N	1.19945300	-0.02127900	0.00000000
H	-1.73526800	-1.28348900	0.00000000
H	0.73724700	2.04909900	0.00000000
N	0.38026200	-1.12554200	0.00000000
N	-0.94369400	0.71262800	0.00000000



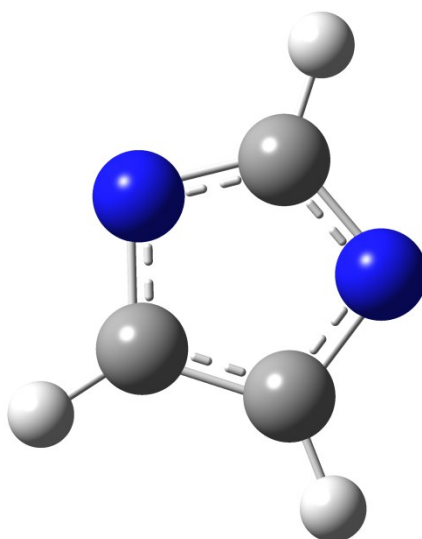
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H	-0.45573900	2.06847900	0.00009200
N	0.00197000	-0.02037400	0.00003000
C	1.56270700	-0.03487800	0.00034000
O	2.03369200	1.10036800	0.00051900
O	2.01255600	-1.17866300	0.00034400
N	-2.09175900	0.69337500	-0.00033300
C	-0.84813300	-1.08234300	-0.00021600
H	-0.49475200	-2.10032500	-0.00022100
N	-2.10474200	-0.69485600	-0.00043800



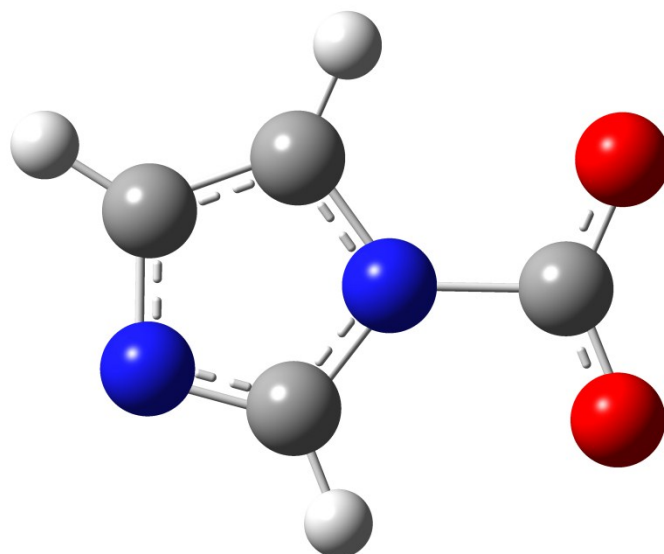
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N	0.36496800	-1.10722300	0.00000000



C	-2.06778600	-0.67528500	-0.00042800
C	-2.12568800	0.73681700	-0.00033500
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H	-2.88363500	-1.38682300	-0.00064000
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H	-0.31333300	2.08498000	0.00012000
N	-0.04430700	-0.00614600	0.00001900
C	1.52546900	-0.02438700	0.00033400
O	1.97141300	1.12791500	0.00051000
O	2.00787300	-1.14747400	0.00034800
N	-0.81000500	-1.11551500	-0.00021300



C	0.42658200	-1.09936200	0.00000000
C	-0.89800500	-0.67227700	0.00000000
C	0.38029500	1.02413600	0.00000000
N	1.25233700	-0.00516600	0.00000000
H	0.82071100	-2.10995100	0.00000000
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C	-0.81746100	-1.05133000	-0.00020900
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N	0.01869600	0.02208200	0.00003400
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O	2.05513100	1.12415300	0.00052800
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N	-2.09462200	-0.70494500	-0.00043500