## 7-Cyanoindole Fluorescence as a Local Hydration Reporter: Application to Probe the Microheterogeneity of Nine Water-Organic Binary Mixtures

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## **Supporting Information**

**Table S1**: Organic mole fraction ( $\chi_{OM}$ ) at which the fluorescence intensity ( $S_{Fmax}$ ) of 7-CNI is the largest, and  $\lambda_{max}$  is the peak wavelength of the corresponding fluorescence spectrum of 7-CNI obtained at  $\chi_O$ . Also shown for comparison are the peak emission wavelength ( $\lambda_O$ ) of 7-CNI, and the respective peak wavenumber value ( $\omega_O$ ), obtained in the corresponding pure organic solvent.

Organic Solvent	χом	S <sub>Fmax</sub>	$\lambda_{\max}$ (nm)	$\lambda_{\rm O}$ (nm)	$\omega_0 (cm^{-1})$
ACN	0.5	8.4	391	378	26455
DMSO	0.3	5.9	393	-	-
THF	0.2	10.8	391	370	27027
Dioxane	0.3	10.7	392	402	27248
МеОН	0.7	6.8	391	387	25840
EtOH	0.6	8.6	391	381	26247
IPA	0.5	9.6	392	379	26385
TBA	0.4	10.1	391	379	26385
EtG	1.0	9.7	393	-	-

**Table S2**: Dependence of the fluorescence lifetime ( $\tau_F$ ) of 7-CNI on the organic component and its mole fraction ( $\chi_O$ ) in the water-organic solvent binary system. Also listed are the  $\chi^2$  values resultant from fitting the corresponding fluorescence decay kinetics to a single-exponential function.

Organic Solvents	χο	$ au_{\mathrm{F}}(\mathrm{ns})$	$\chi^2$
DMSO	0.10	7.7	1.17
DMSO	0.15	11.7	1.11
DMSO	0.30	16.2	1.33
DMSO	0.80	14.6	1.19
THF	0.04	3.8	0.99
THF	0.10	10.4	0.99
THF	0.20	13.3	1.04
THF	0.40	13.2	1.23
THF	0.60	9.9	1.23
THF	0.80	8.4	1.06
THF	1.00	6.8	1.05
ACN	0.05	3.2	1.00
ACN	0.15	7.9	1.00
ACN	0.40	12.3	1.11
ACN	0.50	12.1	1.06
ACN	1.00	7.9	1.06
DIO	0.05	4.8	0.99
DIO	0.10	8.8	1.08
DIO	0.25	14.3	1.07
DIO	0.50	12.4	1.26
DIO	0.80	10.4	1.06
EtOH	0.00	1.9	0.95
EtOH	0.10	3.7	0.94
EtOH	0.20	7.4	1.06
EtOH	0.70	10.9	1.16
EtOH	1.00	8.8	0.96
МеОН	0.10	2.3	1.06
МеОН	0.40	6.7	1.04
МеОН	0.80	9.2	0.99
MeOH	1.00	8.2	1.01
IPA	0.05	3.1	1.08
IPA	0.15	9.3	1.17
IPA	0.50	11.7	1.01
IPA	0.90	9.7	1.14
EtG	0.10	2.9	0.92
EtG	0.50	8.4	1.16
EtG	1.00	12.3	0.96
TBA	0.10	9.0	1.22
TBA	0.40	12.1	1.30
TBA	1.00	9.4	1.15



Figure S1. Absorption spectra of L-Trp, L-Tyr, L-Phe, and 7-CNI in water. For each solution, the concentration of the solute was  $\sim 122 \mu$ M.



**Figure S2.** Top panel: normalized absorption spectra of 7-CNI (~10  $\mu$ M) in representative waterorganic binary mixtures and at different organic mole fractions, as indicated. Bottom panel: corresponding fluorescence excitation spectra of 7-CNI for each case measured at  $\lambda_{max}$ .



**Figure S3.** Normalized fluorescence spectra of 7-CNI in different water-organic binary systems, as indicated. In each panel, the numbers indicate the mole fractions of the organic solvent.



**Figure S4.** Normalized fluorescence intensity ( $S_F$ ) of 5-CNI as a function of  $\chi_O$  in different waterorganic binary mixtures, as indicated. In each case, the smooth line serves to guide the eyes.



**Figure S5.** Representative fluorescence decay kinetics (red) of 7-CNI obtained in different waterorganic solvent binary mixtures with the indicated mole fraction of the corresponding organic component. In each case, the black line represents the best fit of the kinetics to a single-exponential function with the resultant residues shown in the top panel.



**Figure S6.** Representative fluorescence decay kinetics (red) of 7-CNI obtained in different wateralcohol binary mixtures with the indicated alcohol mole fraction. In each case, the black line represents the best fit of the kinetics to a single-exponential function with the resultant residues shown in the top panel.



**Figure S7.** Dependence of  $S_{\text{Fmax}}$  on  $\omega_{\text{O}}$ .