

Supporting Information

Zn-Porphyrin propped with Hydantoin Anchor: Synthesis, Photophysics and Electron Injection/Recombination Dynamics

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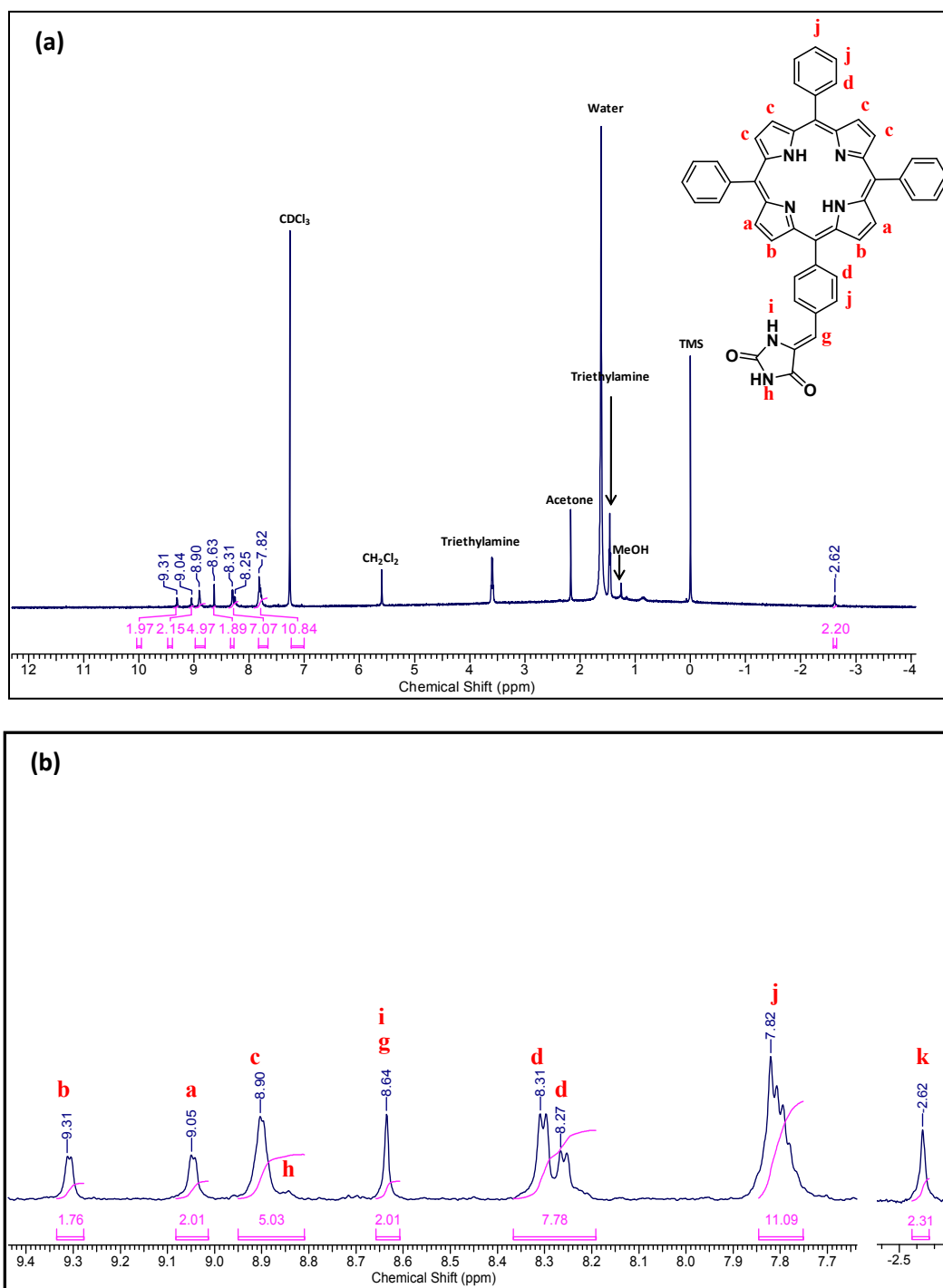


Figure S1: ^1H -NMR (CDCl_3 , 500 MHz) spectrum of free base porphyrin (PHy) (a) full spectrum (b) expanded view with proton assignments

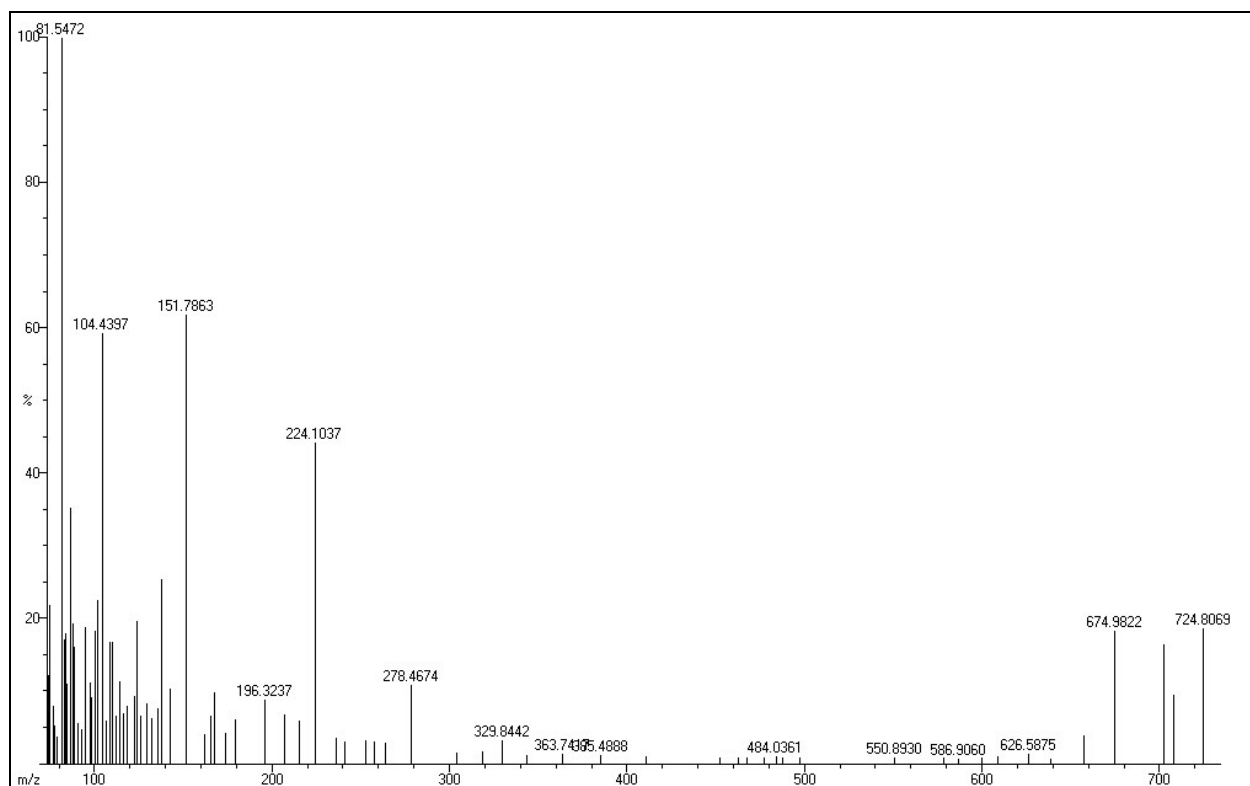


Figure S2: EI-MS spectrum of free base porphyrin (**PHy**)

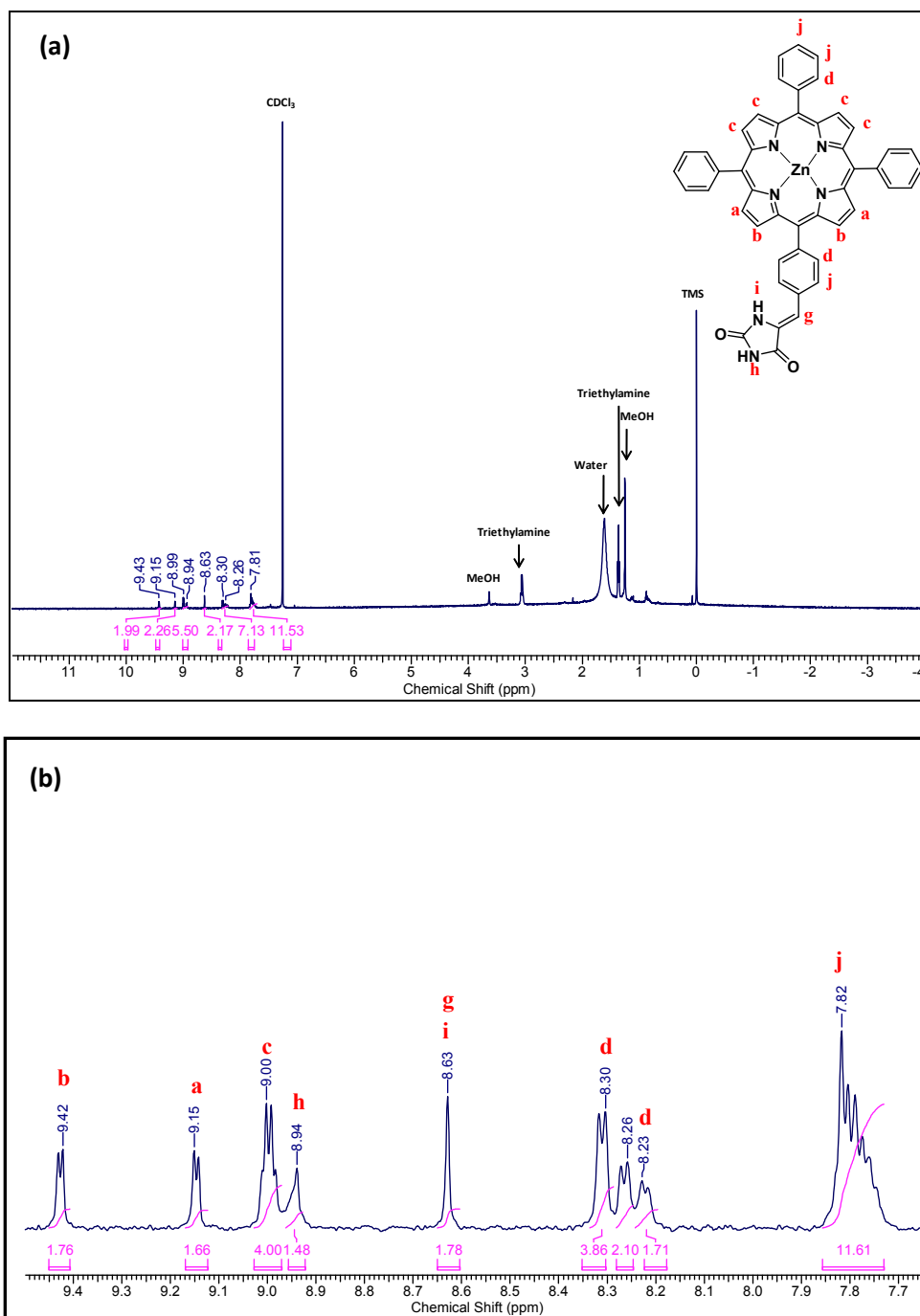


Figure S3: ^1H -NMR (CDCl_3 , 500 MHz) spectrum of zinc porphyrin (**ZnPhy**) (a) full spectrum (b) expanded view with proton assignments

Acquisition Time (sec)	0.5999	Comment	Zn-PHY	Date	19 Dec 2017 11:03:36	Date Stamp	19 Dec 2017 11:03:36
File Name	G:\18122017_Mothi\18122017_Mothi\26\data\1\1r	Frequency (MHz)	176.05	Nucleus	¹³ C	Number of Transients	24576
Origin	av700	Original Points Count	24997	Owner	sr	Points Count	131072
Receiver Gain	3.20	SW(cyclical) (Hz)	41666.67	Solvent	CHLOROFORM-d	Pulse Sequence	zgpg
Spectrum Type	STANDARD	Sweep Width (Hz)	41666.35	Temperature (degree C)	25.000	Spectrum Offset (Hz)	19609.0293

18122017_Mothi.026.001.1r.esp

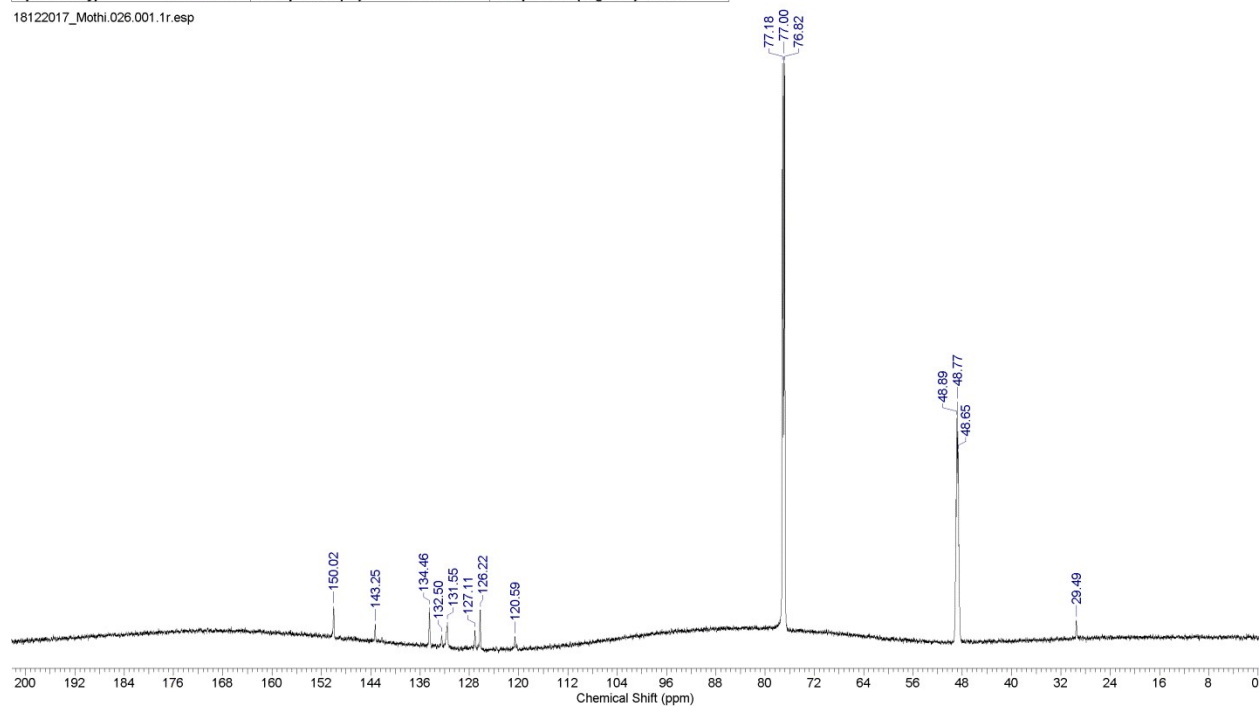


Figure S4: ¹³C-NMR (CDCl₃+ few drops CD₃OD, 176.05 MHz) spectrum of zinc porphyrin (ZnPHY)

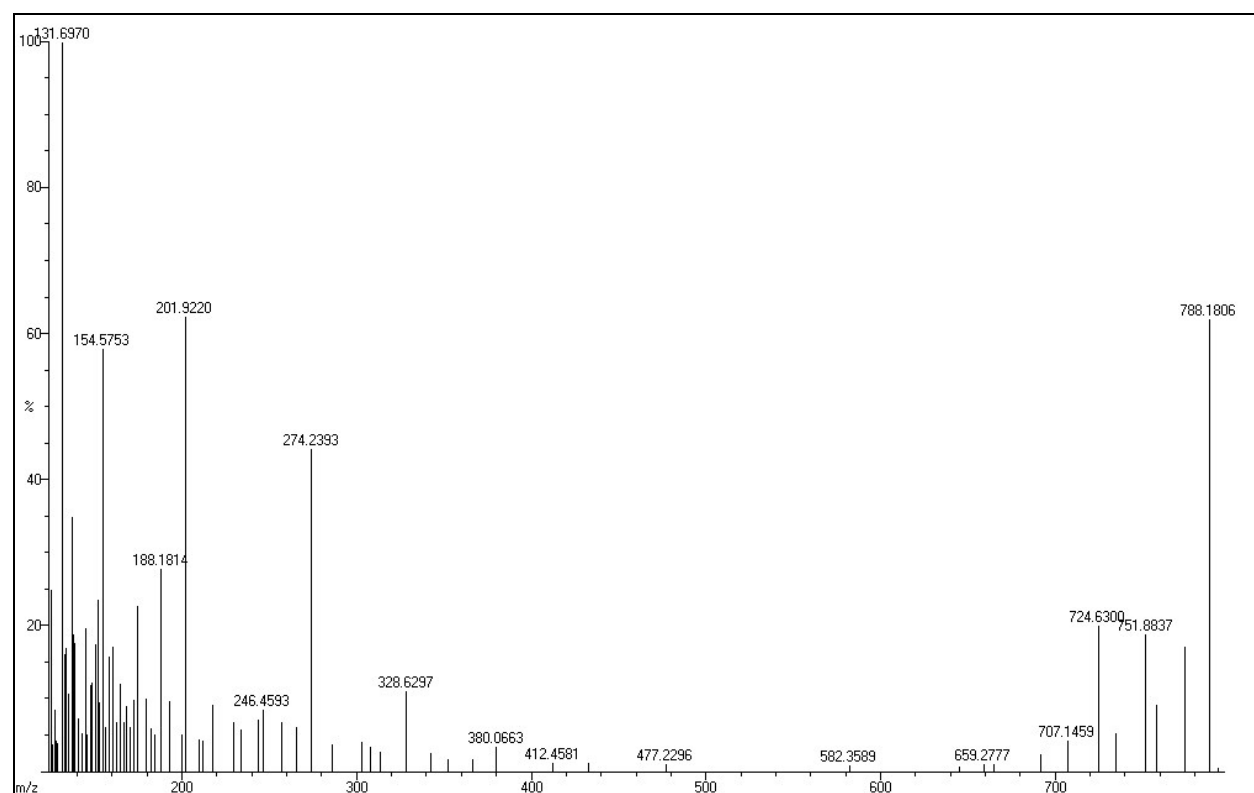


Figure S5: EI-MS spectrum of zinc porphyrin (**ZnPHy**)

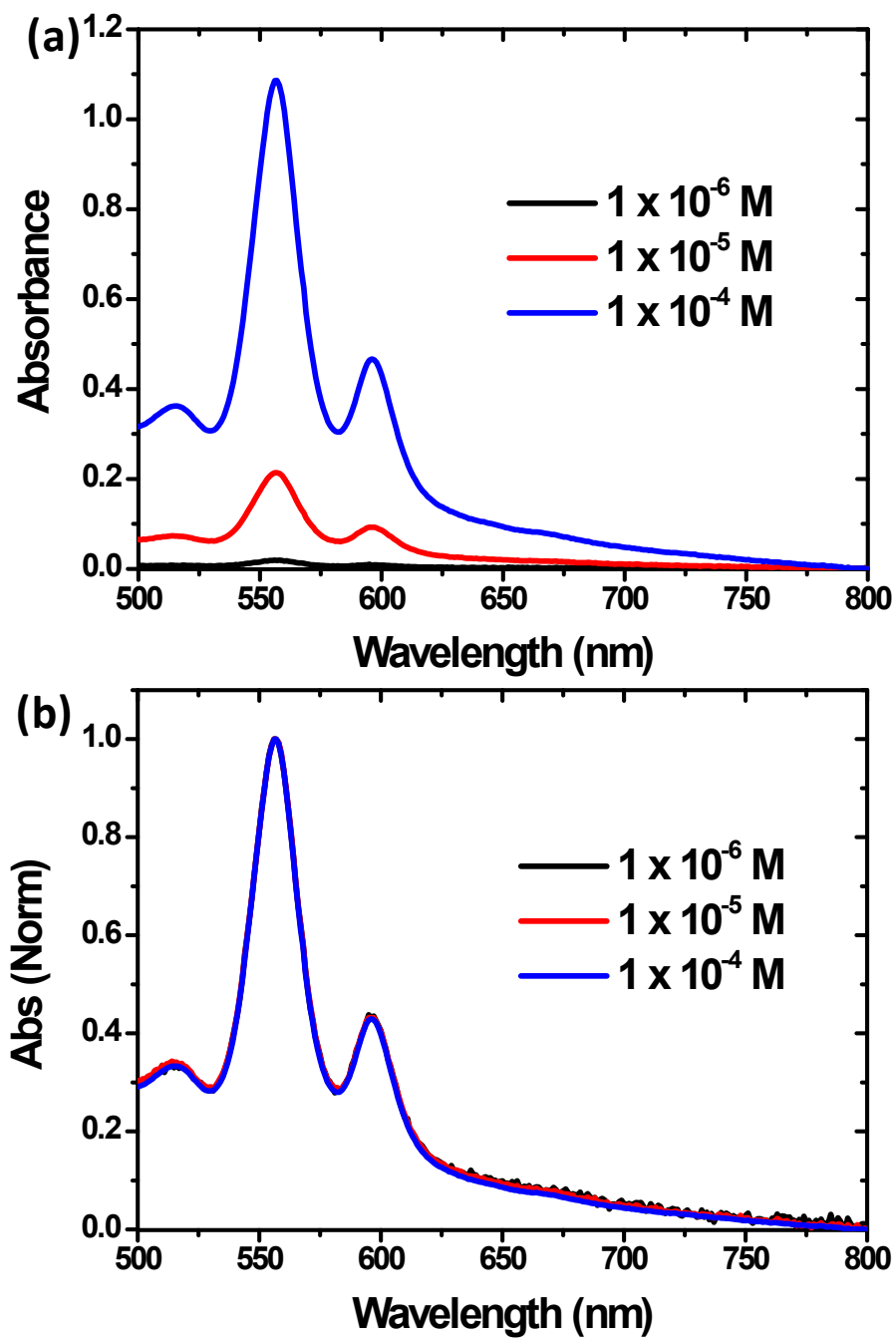


Figure S6: (a) Absorption spectra of ZnPhy in THF at various concentrations (b) Normalized absorption spectra of ZnPhy at various concentrations.

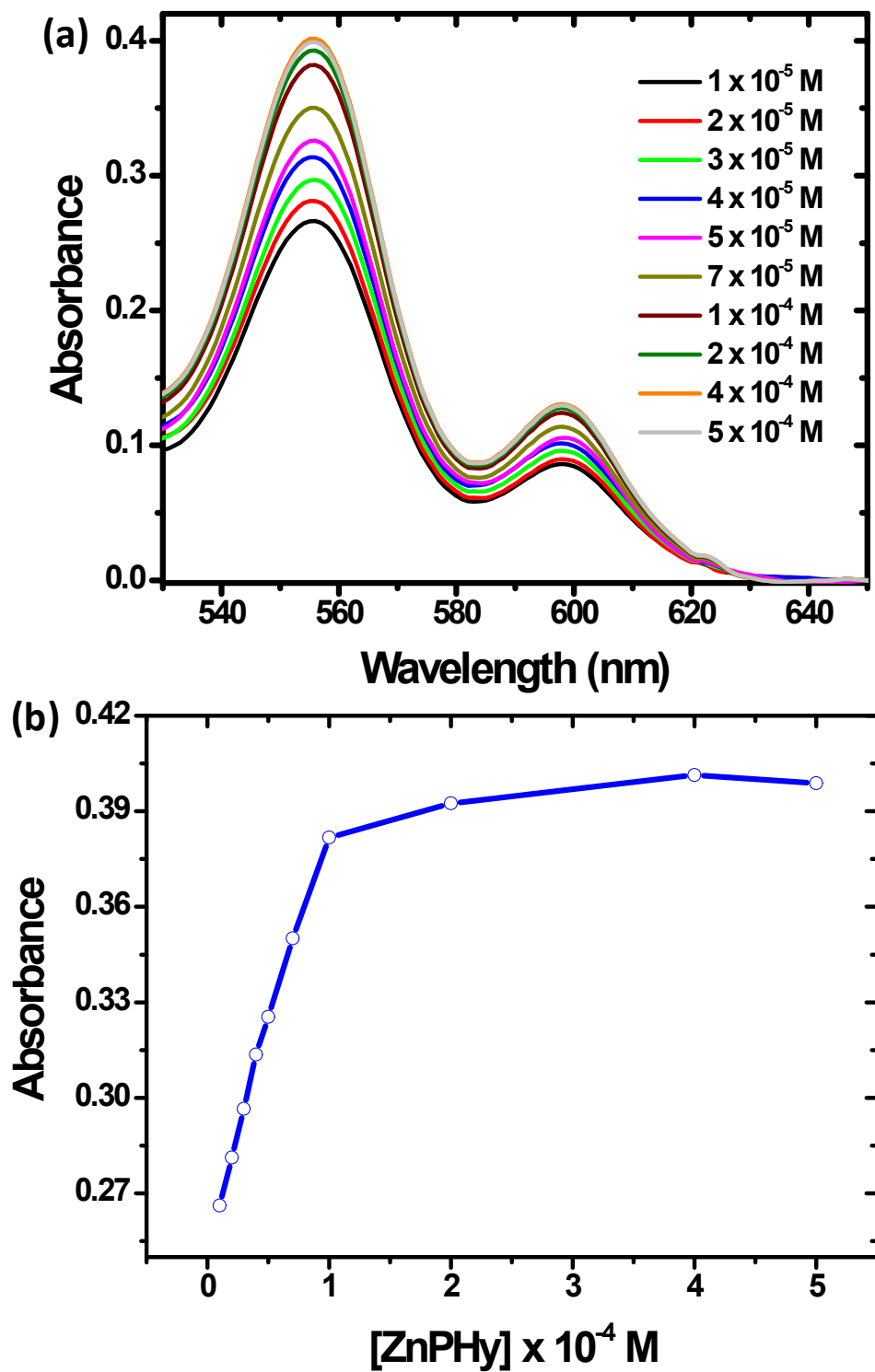


Figure S7: (a) Absorption spectra of ZnPHy/TiO₂ films, (b) A plot of absorbance vs. ZnPHy concentrations.

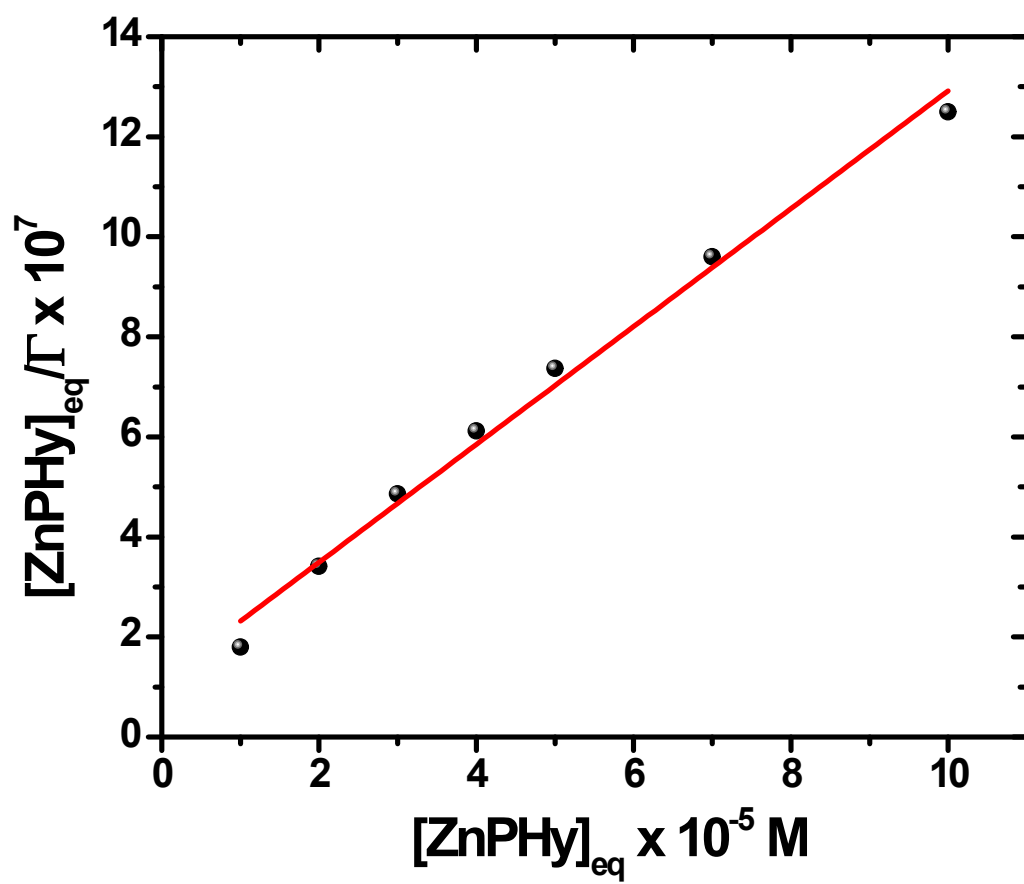


Figure S8: Plot of $[\text{ZnPHy}]_{\text{eq}} / \Gamma$ vs $[\text{ZnPHy}]_{\text{eq}}$.

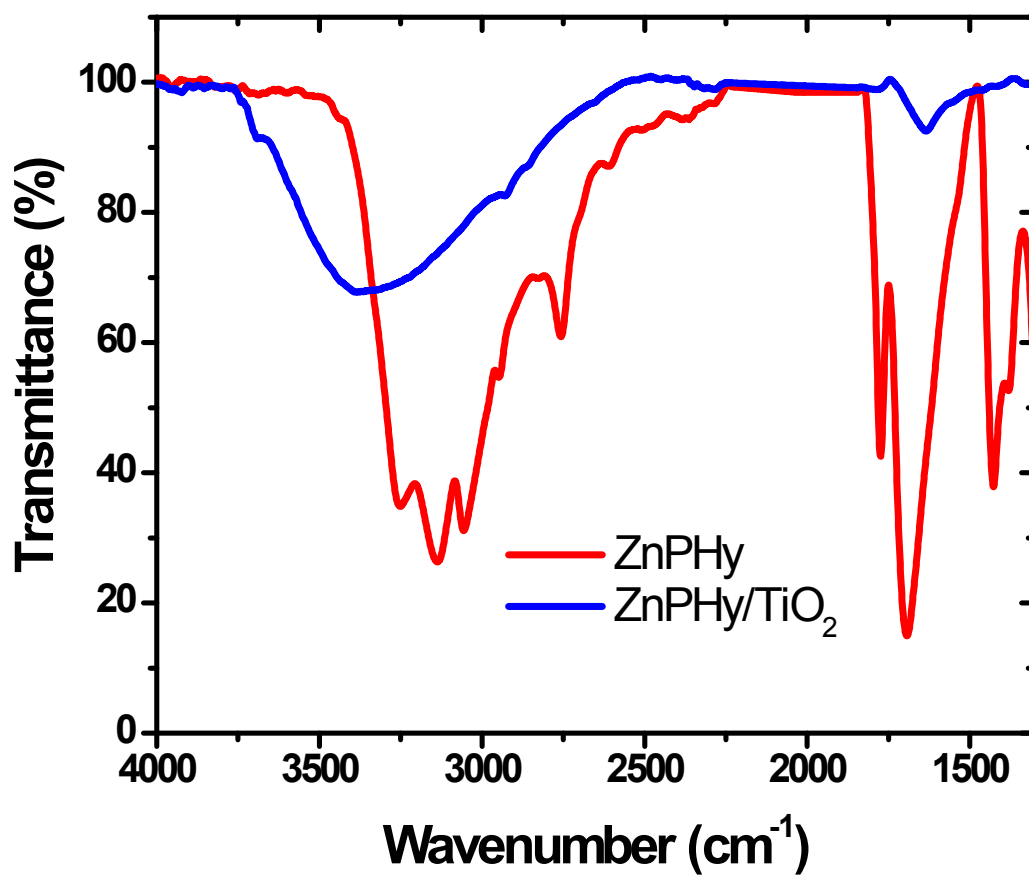


Figure S9: FT-IR spectra of ZnPHy and ZnPHy/TiO₂ film.

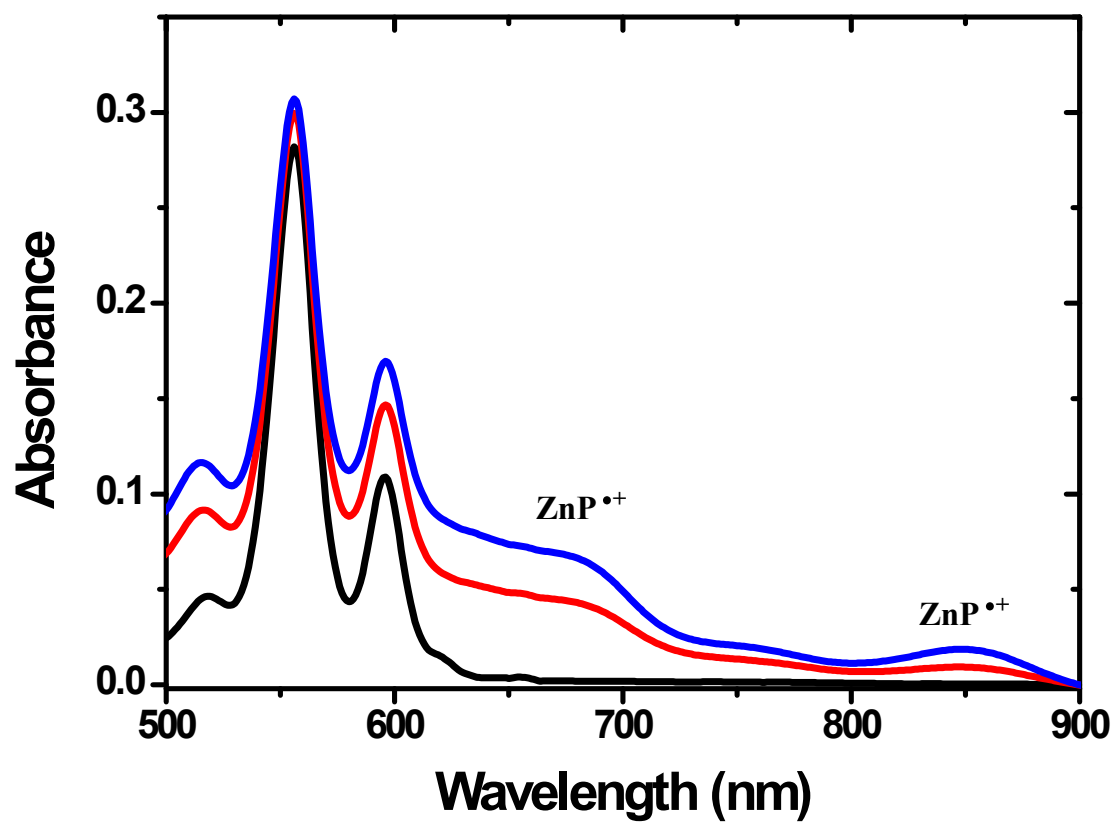


Figure S10: Chemical oxidation of ZnPHy using nitrosonium tetrafluoroborate.