

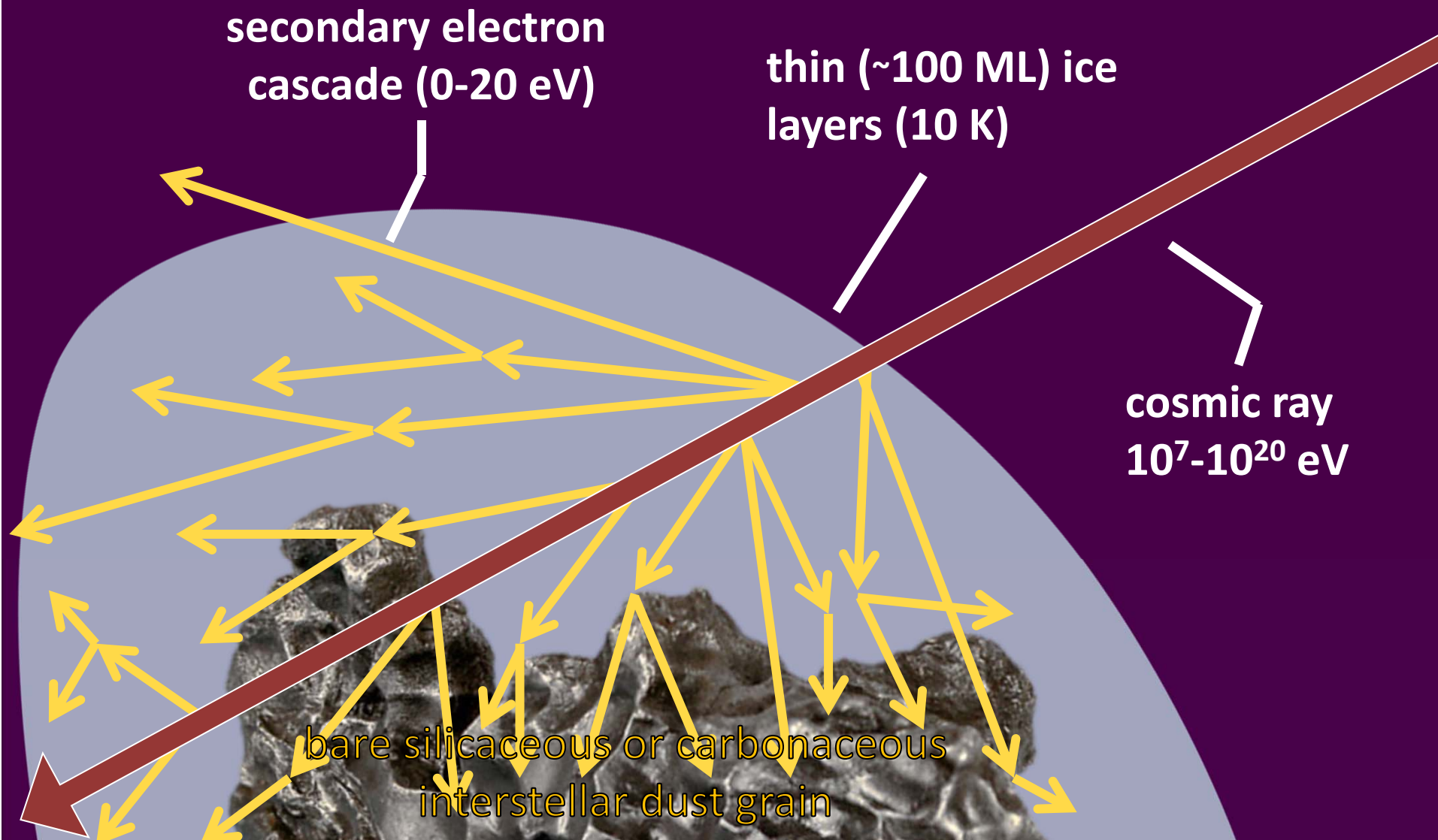
# Annie Jump Cannon (1863-1941)

- Graduated from Wellesley College 1884

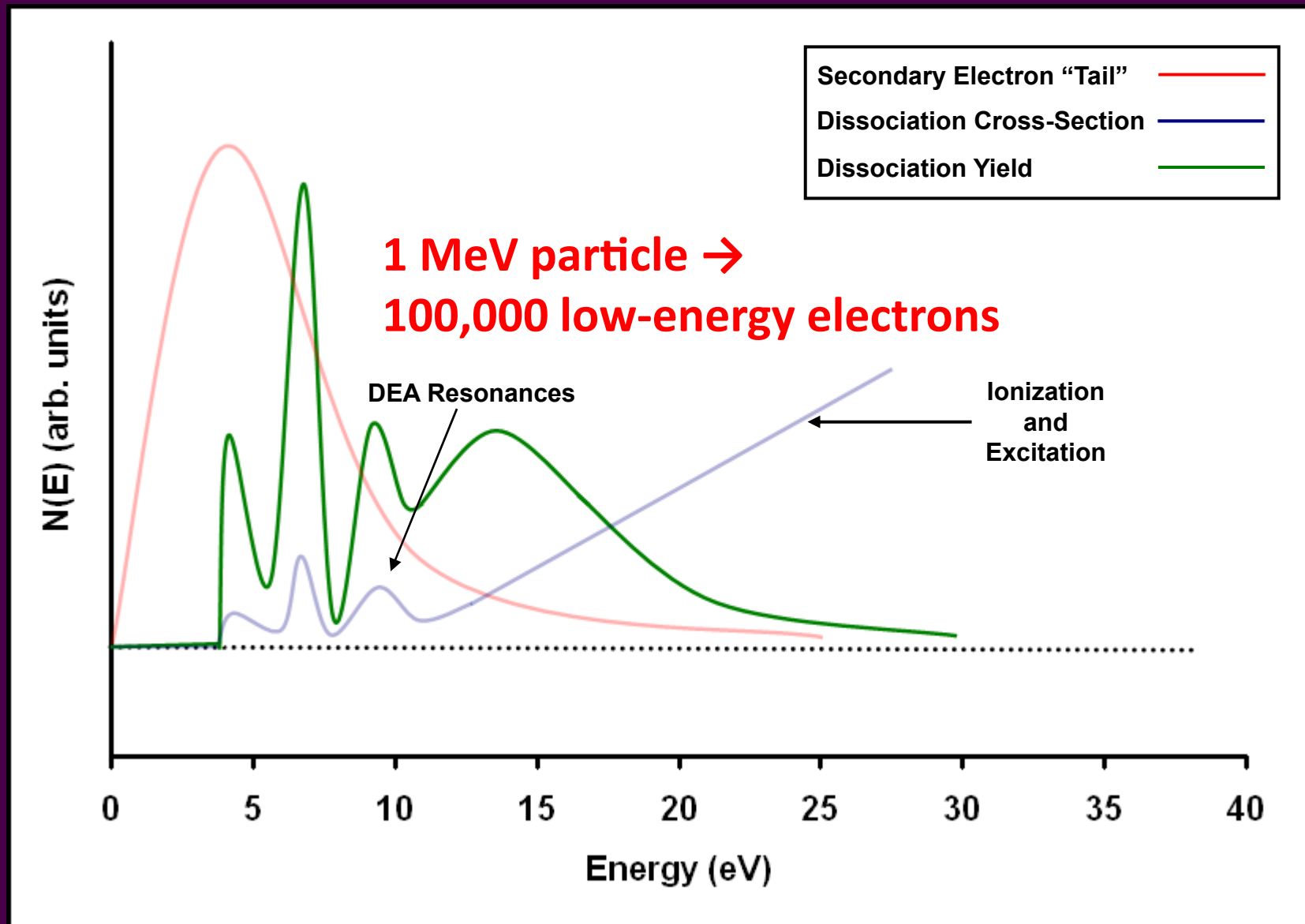


- Likely discoverer of the still enigmatic spectral diffuse interstellar bands (DIB) (~ 1918)
- Stellar classification: "OBAFGKM"
- Classified 350,000 stars
- Became a "special student" of astronomy at Radcliffe College (1894)
- First woman to receive an honorary degree from Oxford (1925)

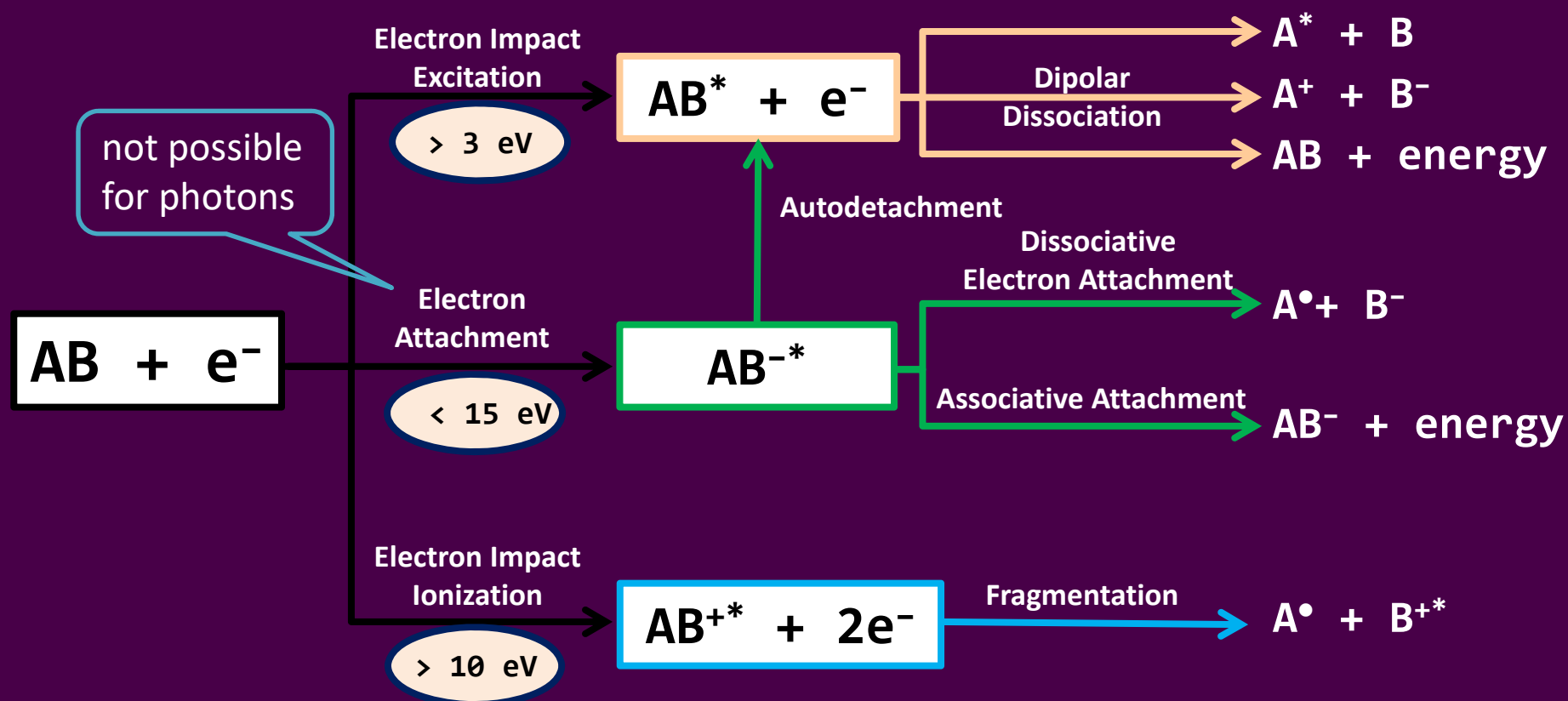
# Formation of Secondary Electrons in Cosmic Ices and Dust Grains



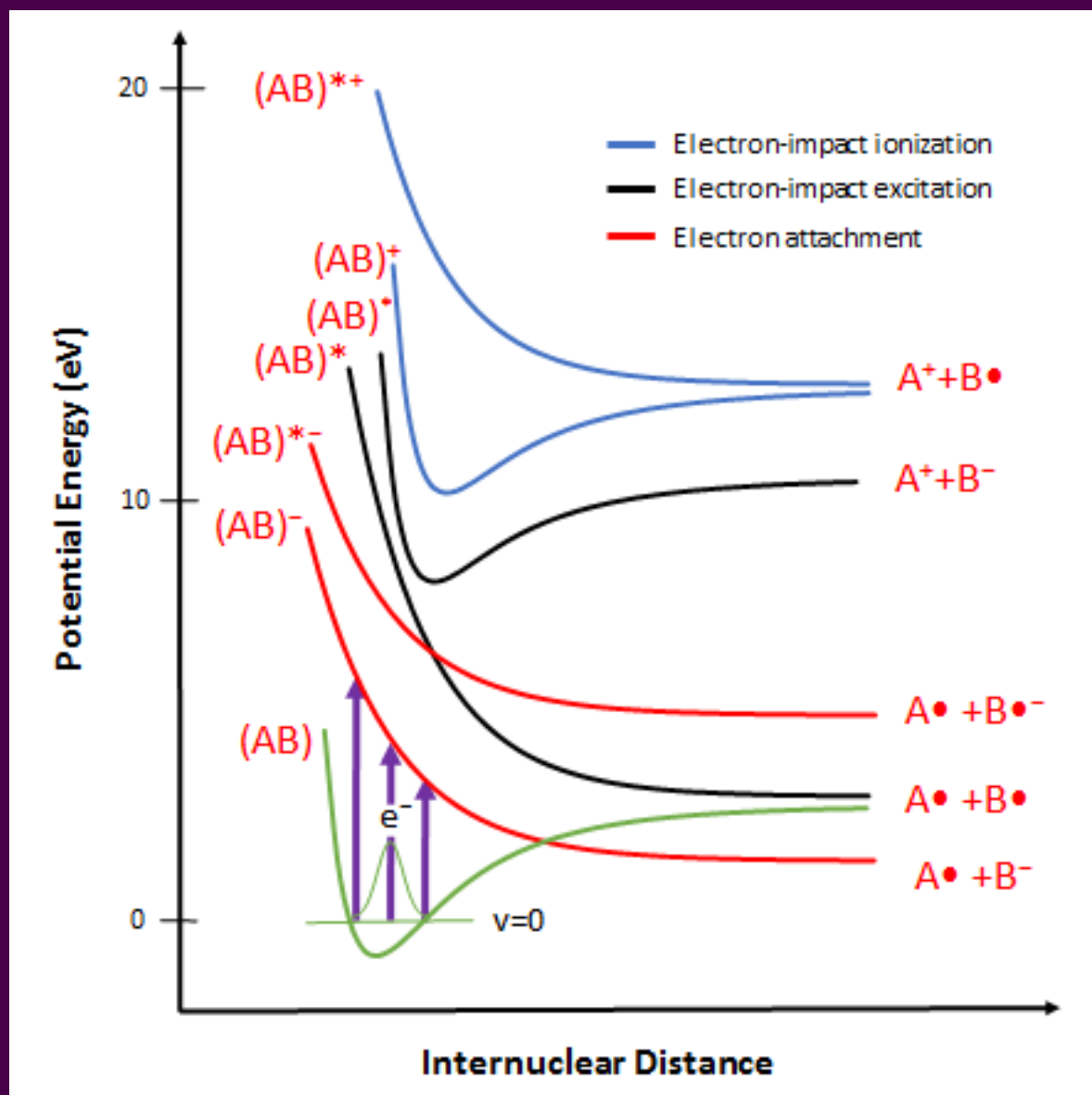
# Importance of Low-Energy Electrons



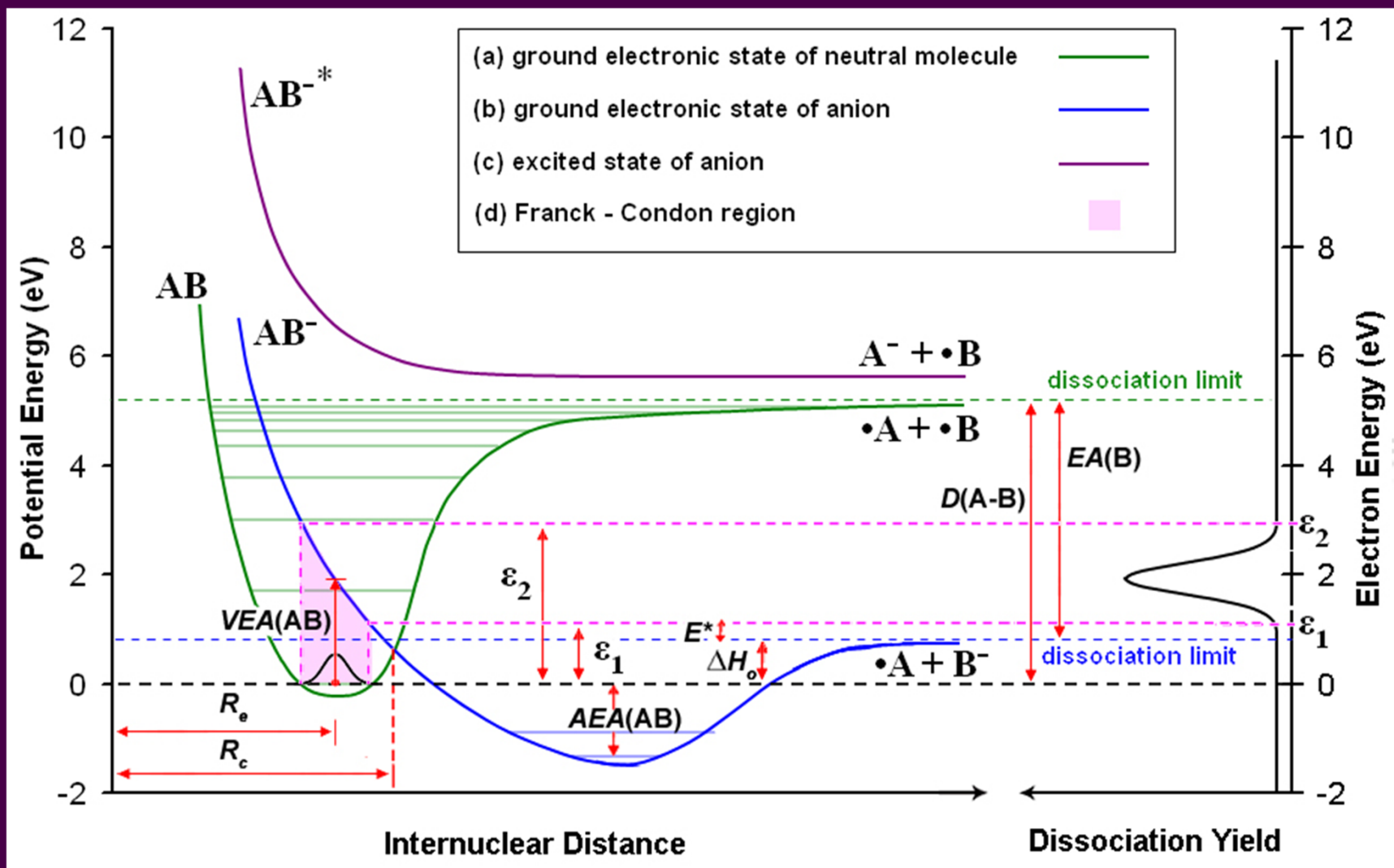
# Electron-induced Dissociation Mechanisms



# Electron-induced Dissociation Mechanisms

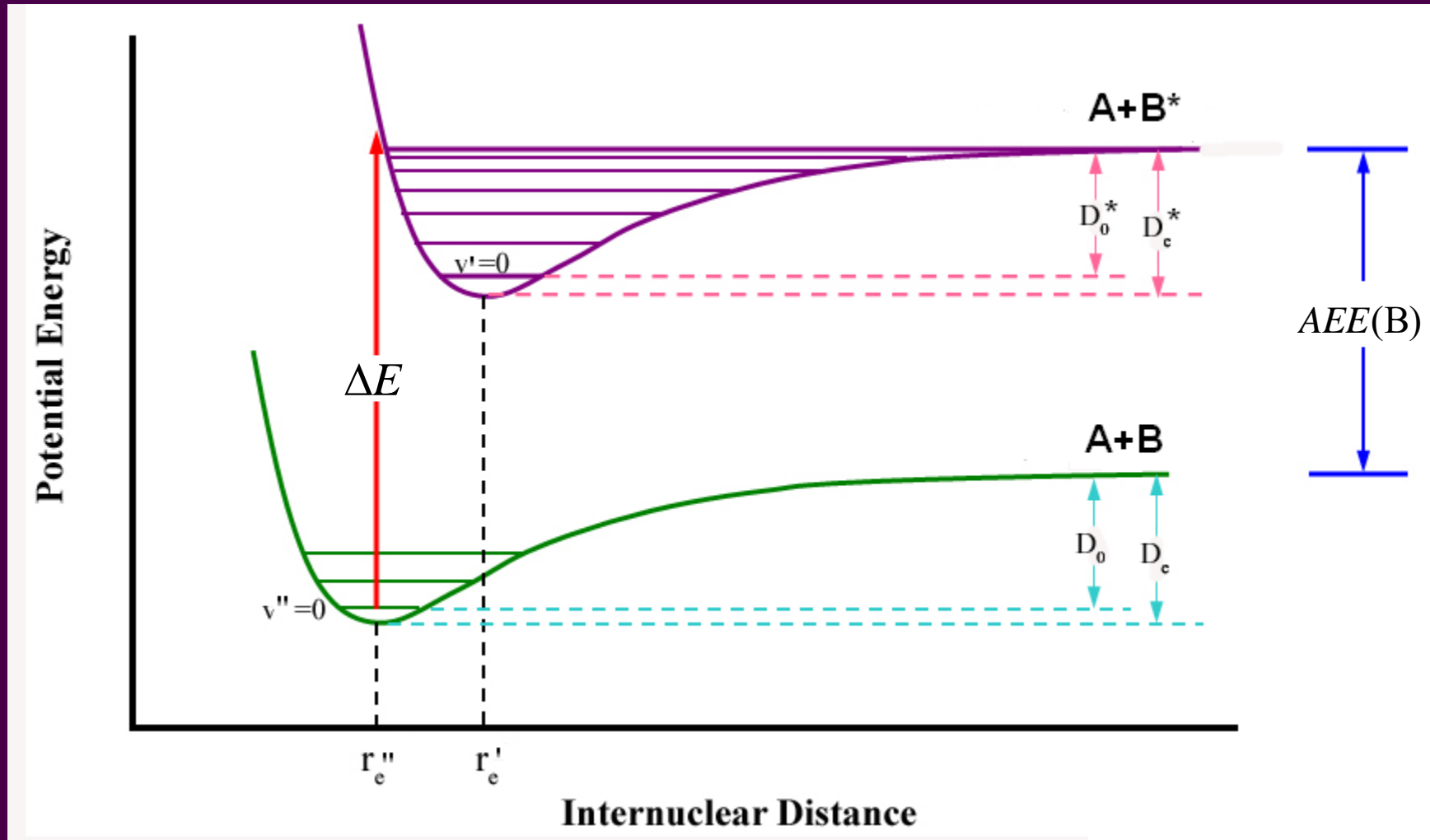


# Breaking a 3.5 eV Bond with a 0 eV Electron



Threshold Energy:  $\Delta H_0(B^{-}) = D(A - B) - EA(B)$

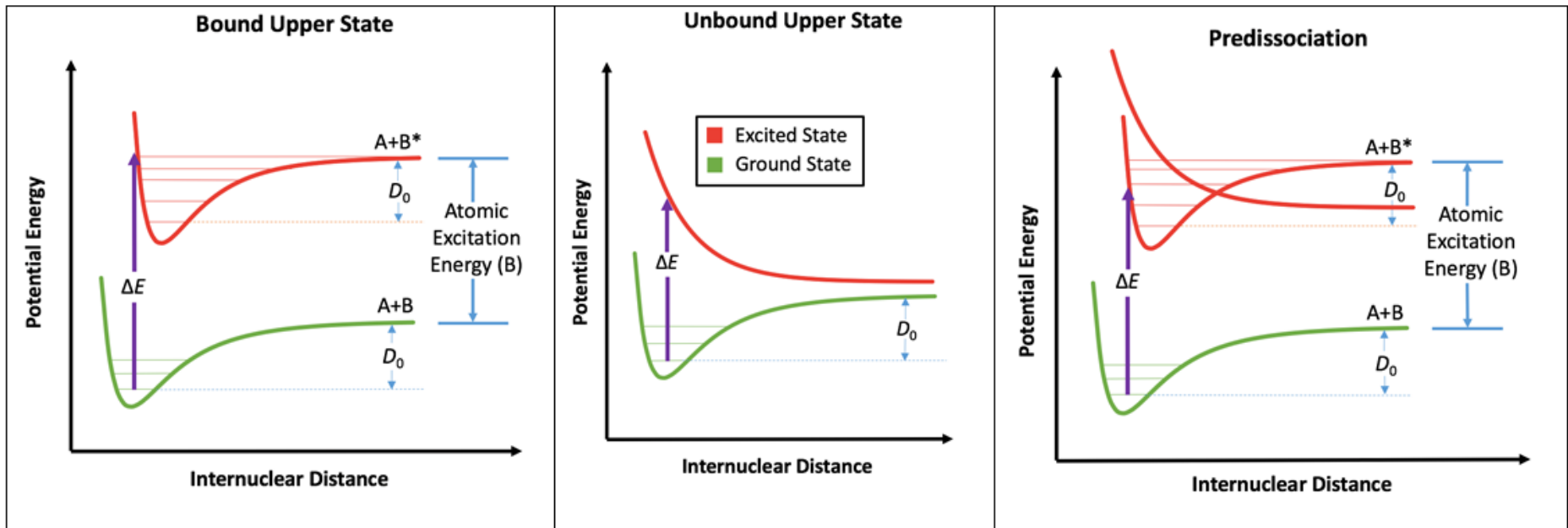
# Photon-Induced Dissociation



Threshold Energy:  $\Delta E = D_0(A - B) + AEE(B)$

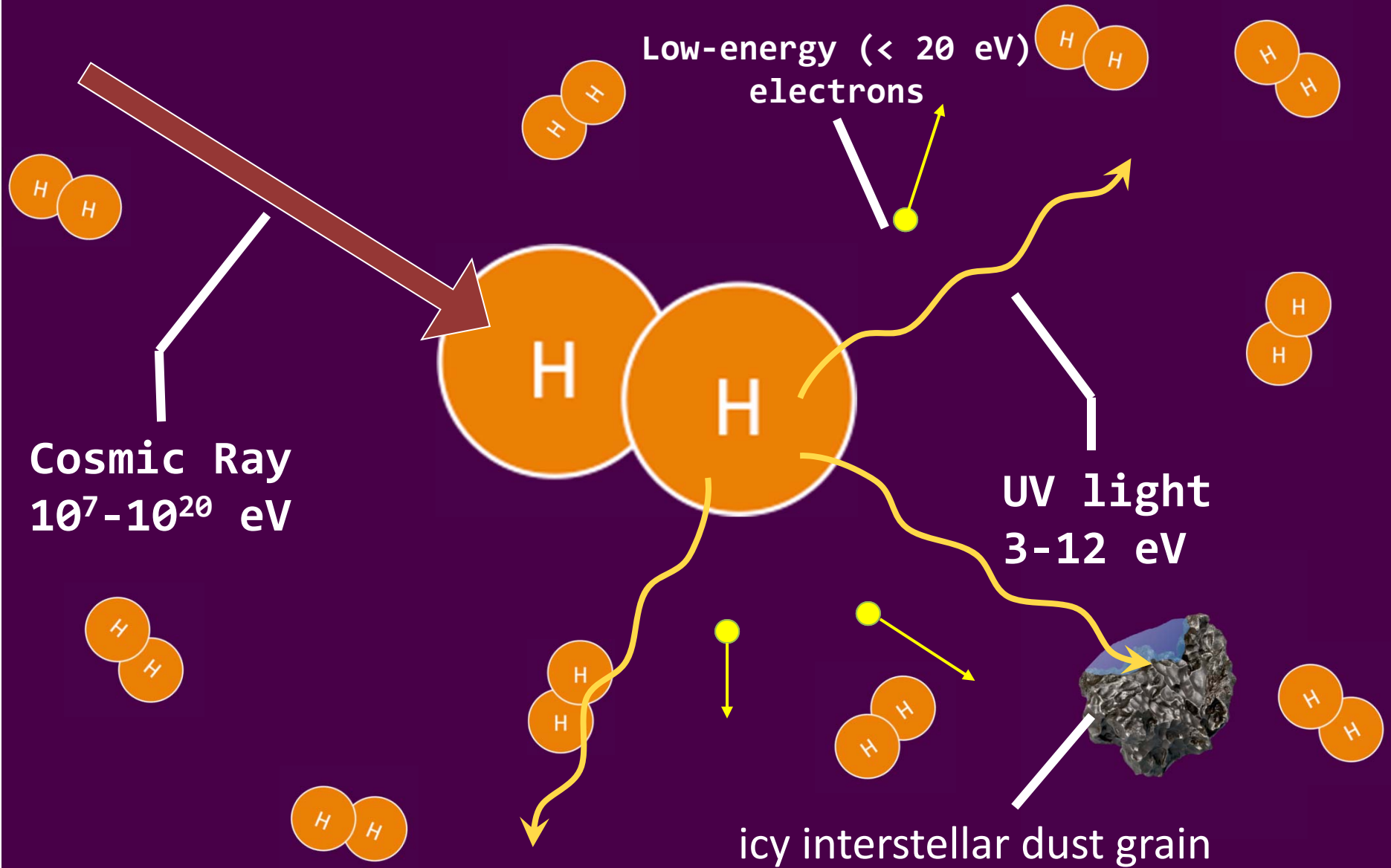
# Photon-Induced Dissociation

## Three Scenarios for Photon-Induced Dissociation

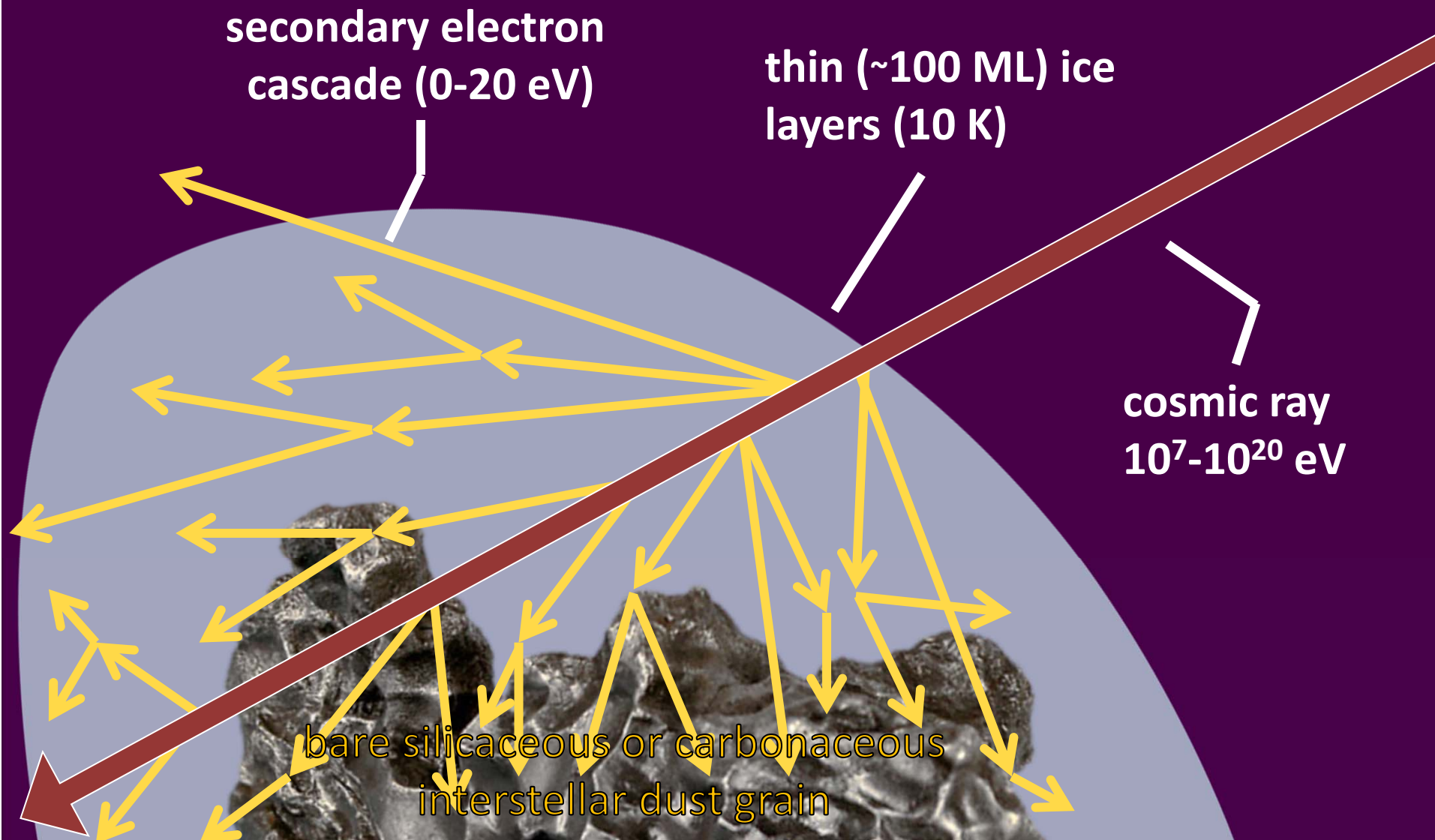




# UV light formation within dark, dense molecular clouds



# Formation of Secondary Electrons in Cosmic Ices and Dust Grains

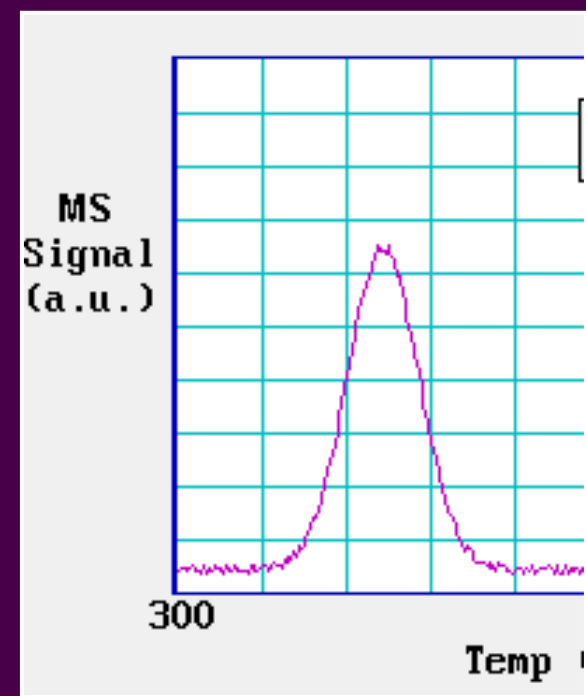
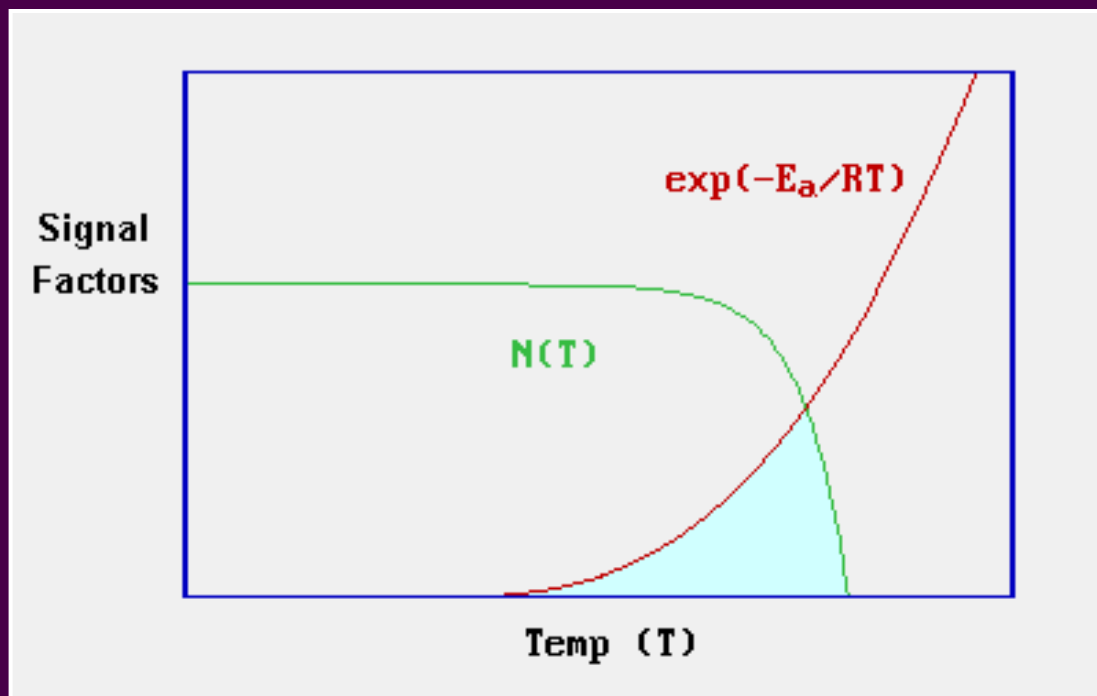


# UHV Chamber at Wellesley College



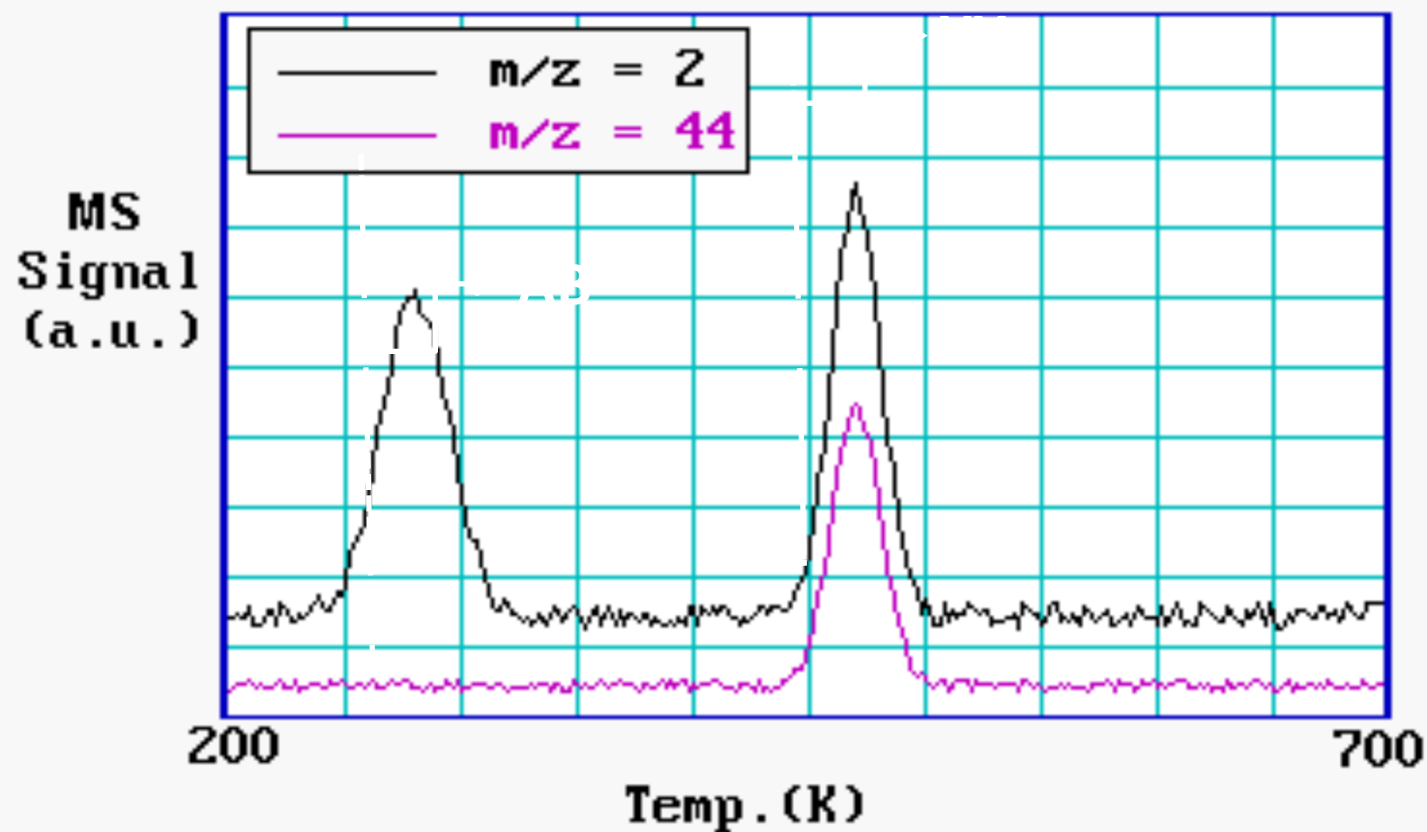
# Temperature-Programmed Desorption (TPD)

$$\text{MS signal} = R = kN = A e^{\frac{-E_a}{RT}} N$$



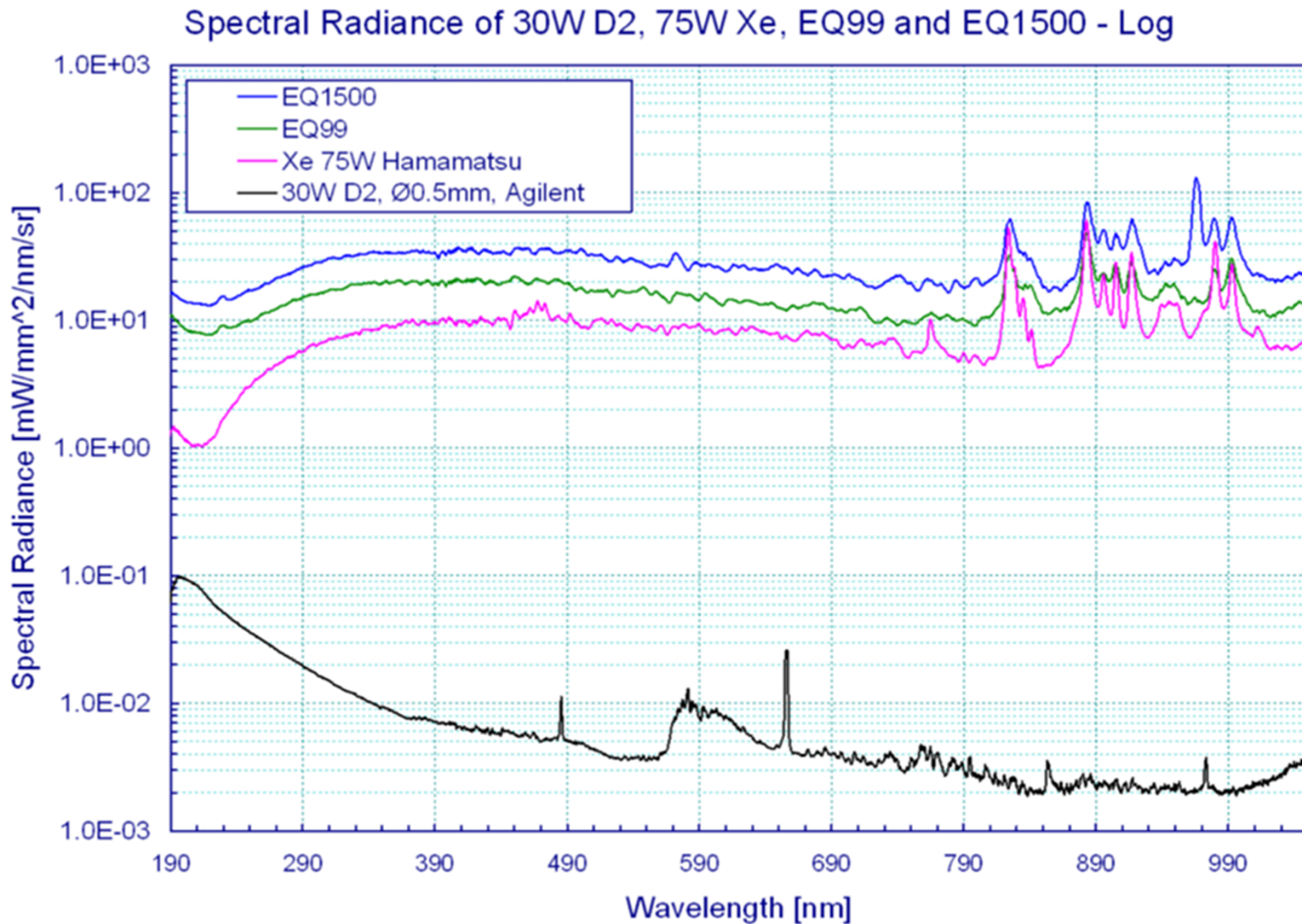
Figures: [http://www.chem.qmul.ac.uk/surfaces/scc/scat5\\_6.htm](http://www.chem.qmul.ac.uk/surfaces/scc/scat5_6.htm)

# Temperature Programmed Desorption How to Identify Desorption Peaks

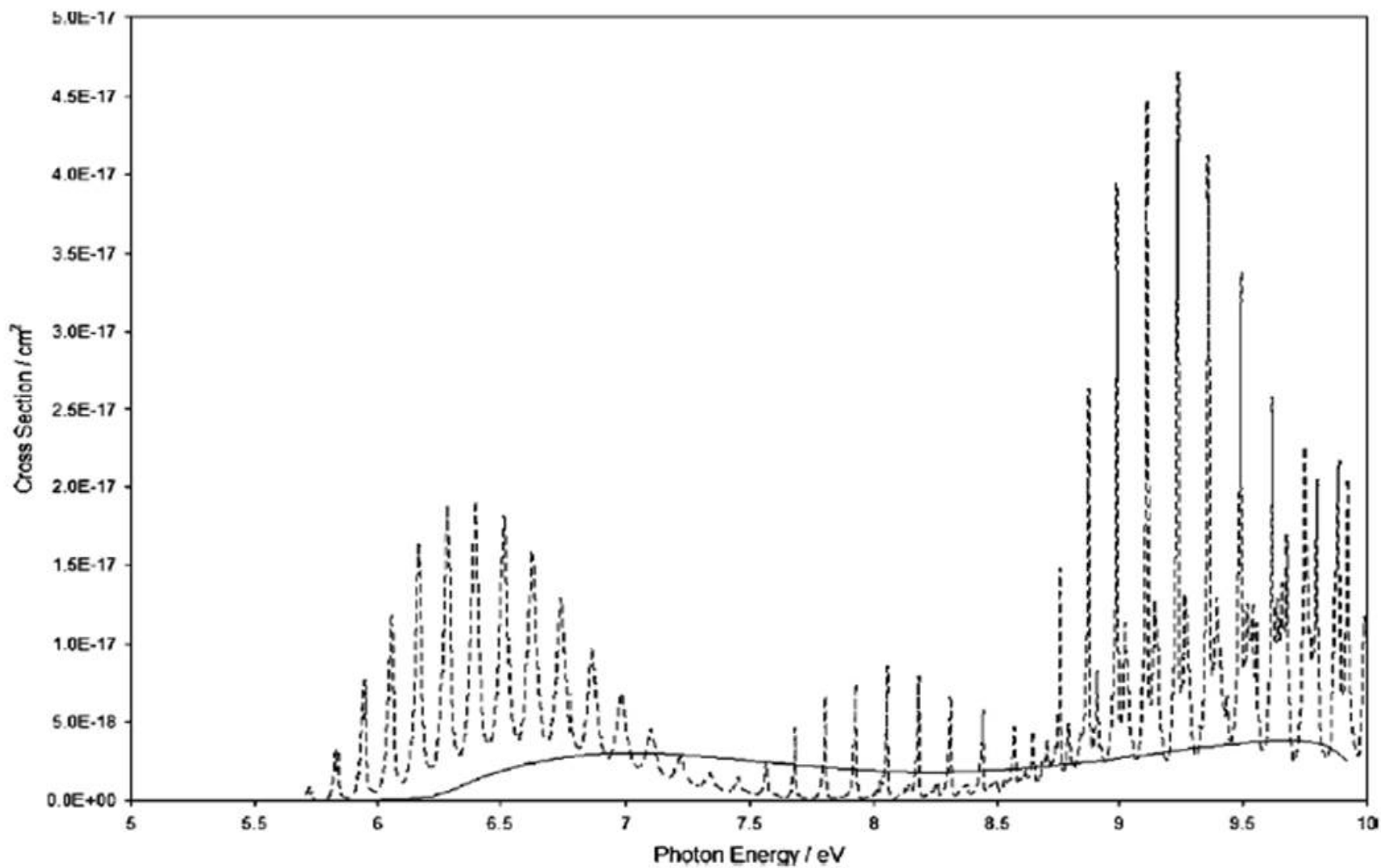


Figures: [http://www.chem.qmul.ac.uk/surfaces/scc/scat5\\_6.htm](http://www.chem.qmul.ac.uk/surfaces/scc/scat5_6.htm)

# Extremely Bright < 8 eV Photon Source: Laser-Driven Plasma: EQ 1500



# UV Absorption Spectrum of Condensed Ammonia



# Mean Free Path Calculation for a 7 eV Photon

$$\begin{aligned}
 \lambda &= \frac{1}{n\sigma} \\
 &= \frac{1}{(\text{number of molecules per cm}^3)(\text{photon absorption cross section in cm}^2)} \\
 &= \frac{1}{\left(\frac{\text{density in g/cm}^3}{\text{molar mass in g/mole}}\right) \left(\frac{\text{number of molecules}}{\text{per mole}}\right) (\text{photon absorption cross section in cm}^2)} \\
 &= \frac{1}{\left(\frac{0.68 \text{ g/cm}^3}{17 \text{ g/mole}}\right) (6.02 \times 10^{23} / \text{mole}) (2 \times 10^{-18} \text{ cm}^2)} \\
 &= 0.2 \text{ microns}
 \end{aligned}$$

Most, if not all, of the ice mantle surrounding dust grains will be susceptible to photochemistry



Low-energy electron-  
induced radiolysis in  
cosmic ices

radical-radical reactions

radical formation

$H\cdot, \cdot NH_2$

$N_2H_4$

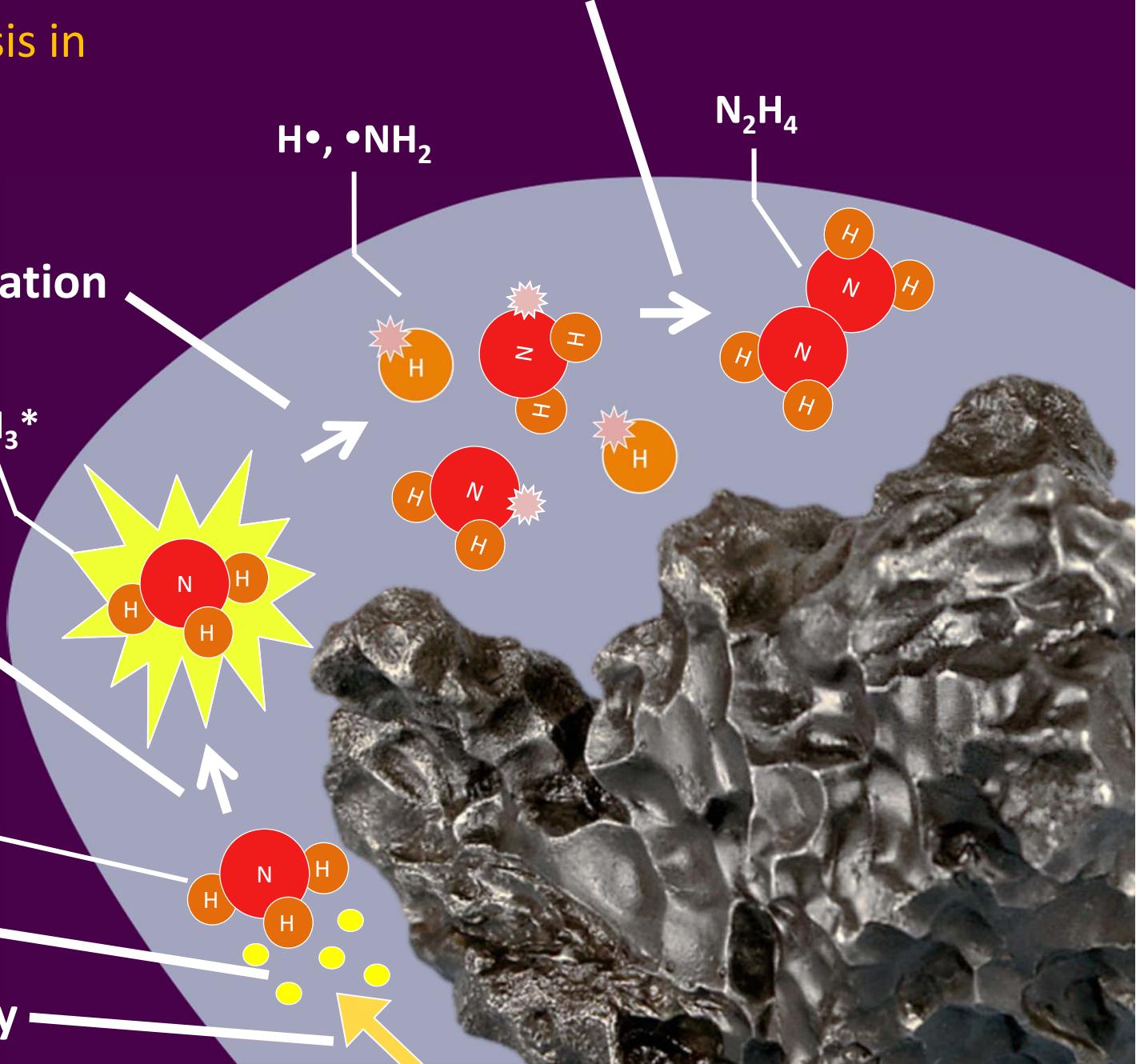
excitation

$NH_3^*$

low-energy  
electrons

$NH_3$

cosmic ray



# Role of low-energy electrons in the synthesis of prebiotic molecules

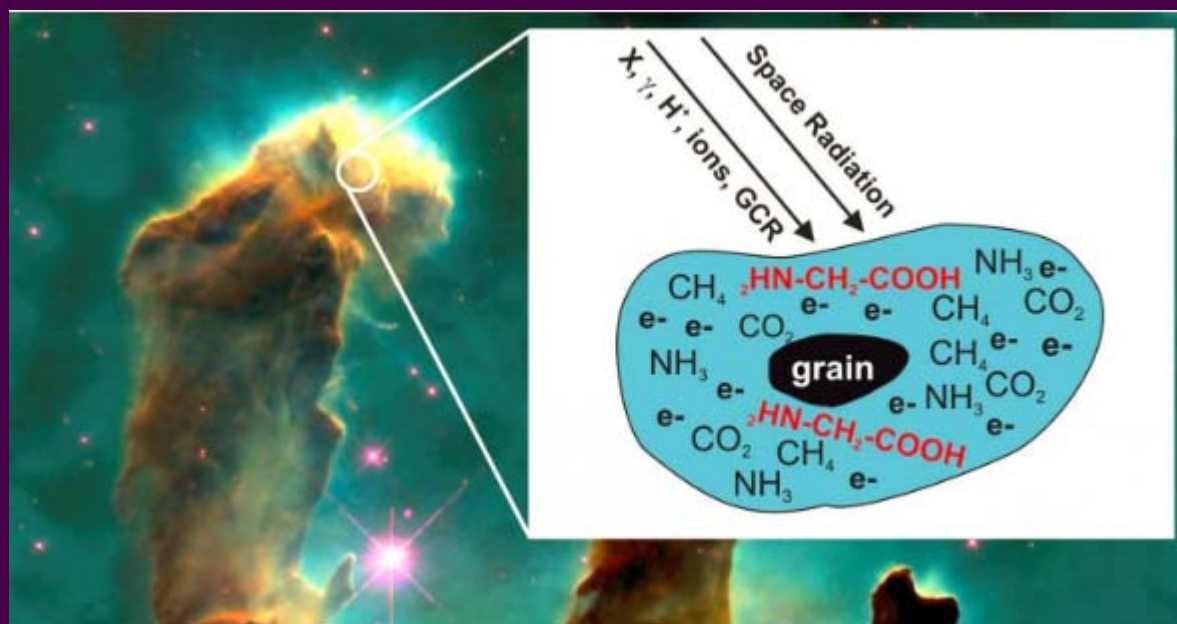
THE JOURNAL OF CHEMICAL PHYSICS **148**, 164702 (2018)



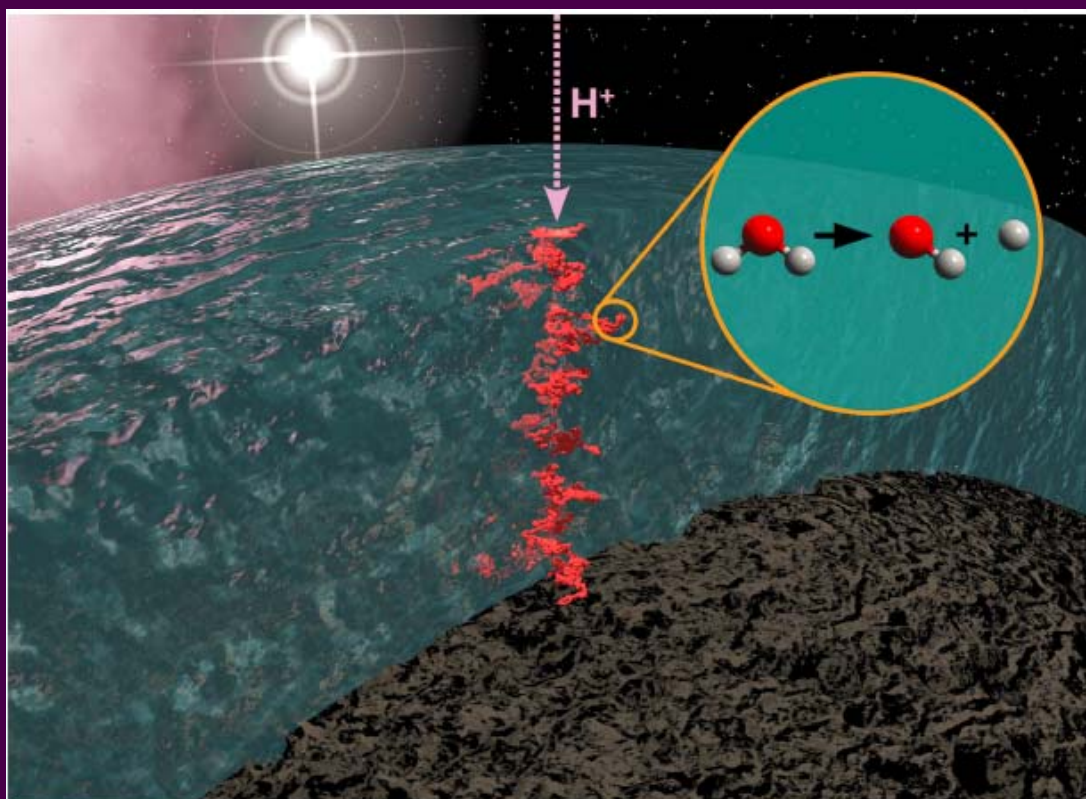
## Glycine formation in $\text{CO}_2:\text{CH}_4:\text{NH}_3$ ices induced by 0-70 eV electrons

Sasan Esmaili, Andrew D. Bass, Pierre Cloutier, Léon Sanche, and Michael A. Huels<sup>a)</sup>  
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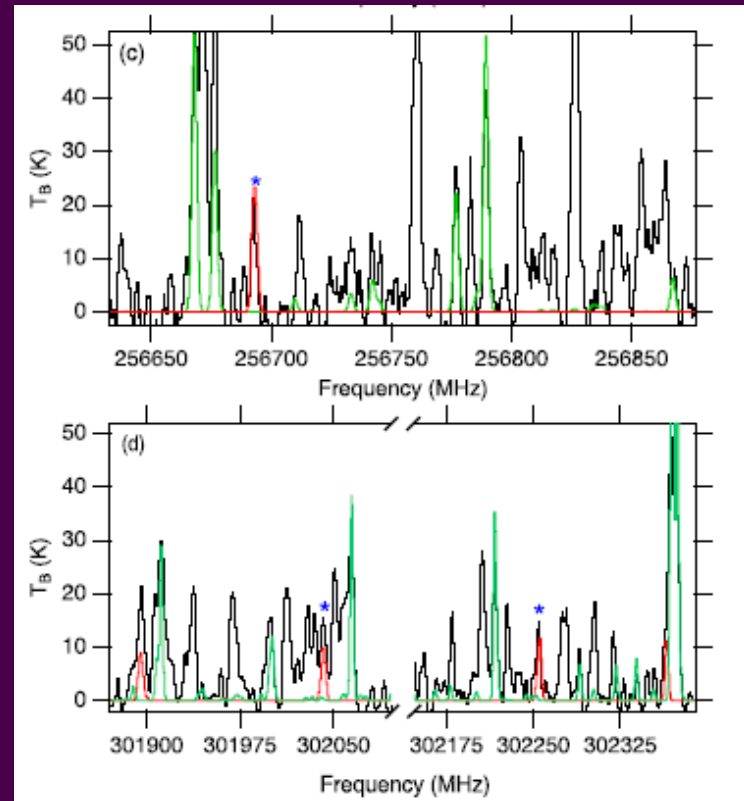
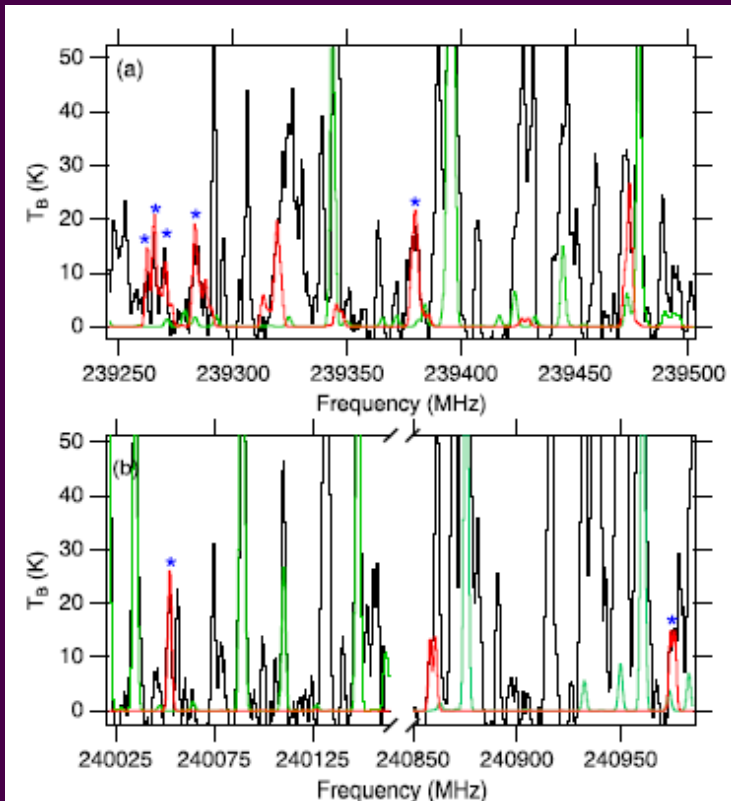


## A general method for the inclusion of radiation chemistry in astrochemical models

Cite this: *Phys. Chem. Chem. Phys.*,  
2018, 20, 5359

Christopher N. Shingledecker \*<sup>ab</sup> and Eric Herbst<sup>ab</sup>

# December 2017: Identification of Methoxy methanol in the ISM [ $\sim 10$ quadrillion miles away]



Black: Overall microwave spectrum of NGC 6334I

Red: Simulated rotational spectrum of methoxymethanol

Green: Simulations of species that are major contributors to the overall spectrum