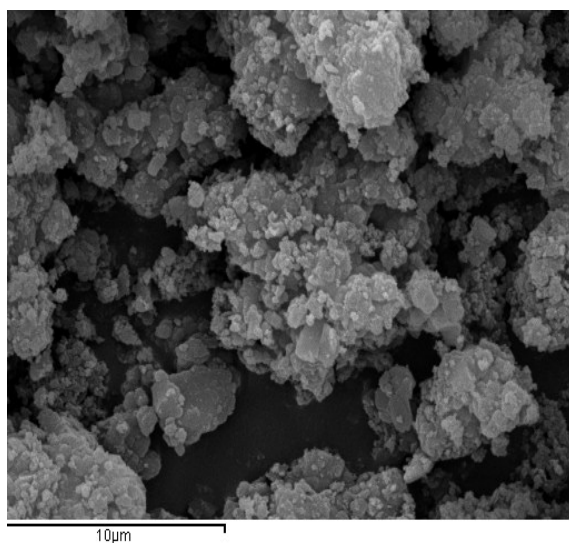


Supplementary Information

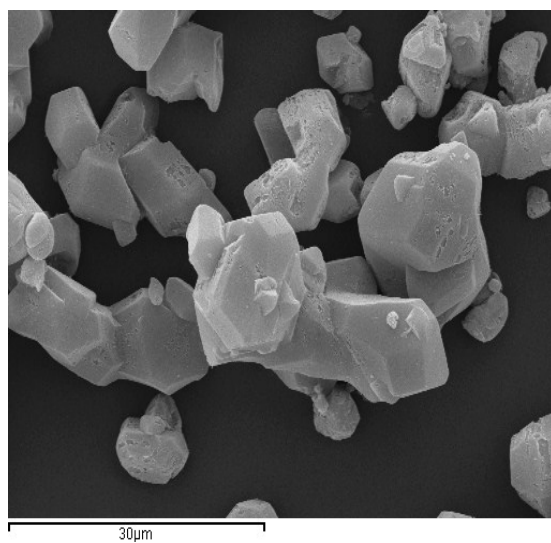
Polymers from biomass: One pot two steps synthesis of furilydenepropanenitrile derivatives with MIL-100 (Fe) catalyst

Anastasia Rapeyko, Karen S. Arias, Maria J. Climent, Avelino Corma*, Sara Iborra*

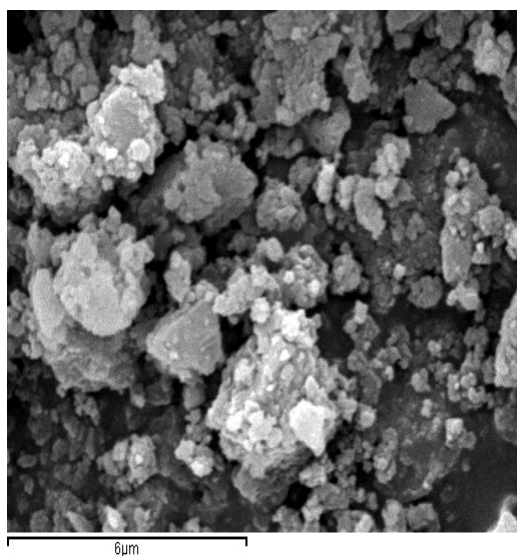
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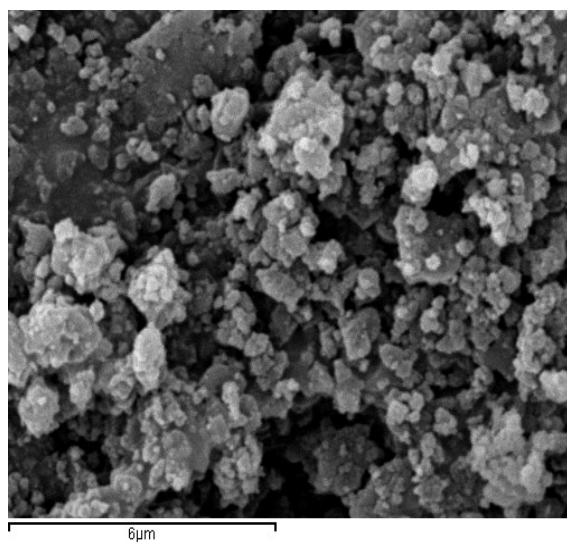
Fe(BTC)



Cu₃(BTC)₂



MIL100 (Fe)



MIL100 (Fe)-NH₄F

Figure S1. SEM image analysis of different MOFs

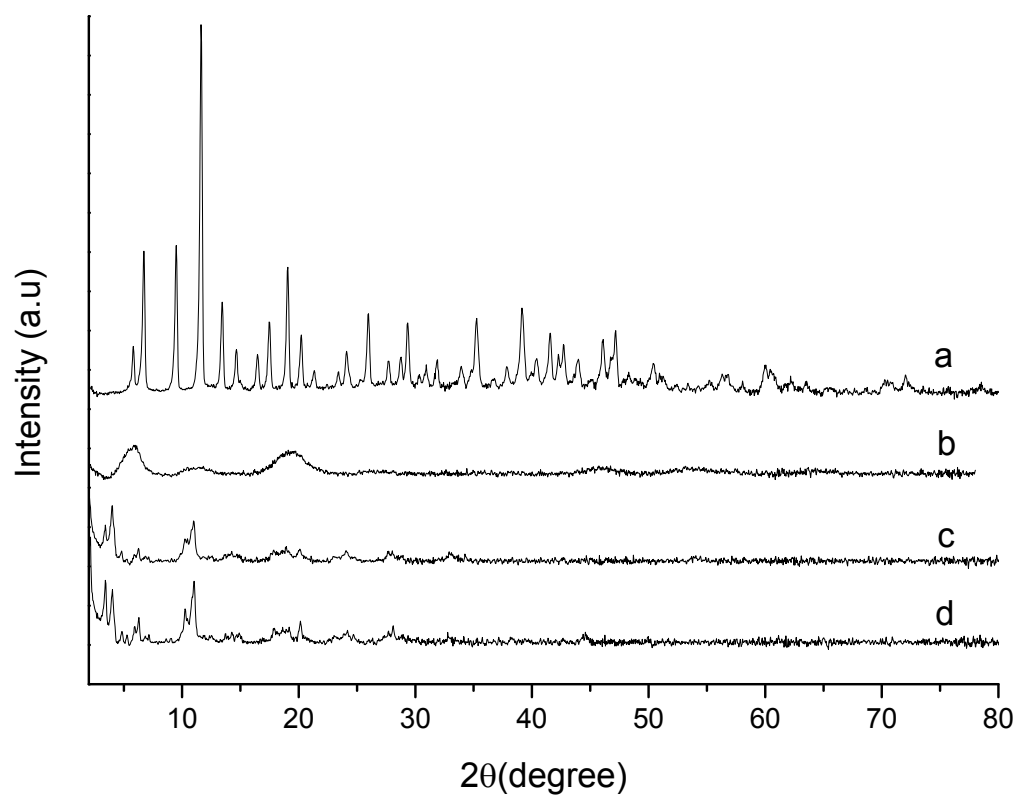


Figure S2. XRD patterns of (a) $\text{Cu}_3(\text{BTC})_2$, (b) $\text{Fe}(\text{BTC})$, (c) $\text{MIL-100}(\text{Fe})$ and (d) $\text{MIL-100}(\text{Fe})\text{-NH}_4\text{F}$.

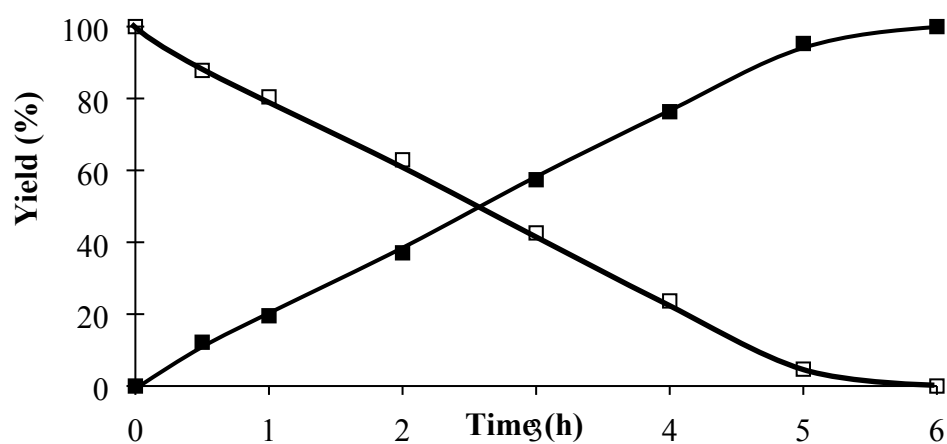


Figure S3. DFF yield versus time plot for the aerobic oxidation of HMF to DFF catalyzed by MIL-100 (Fe)-NH₄F. Reaction Conditions: HMF (1 mmol, 126 mg), catalyst (45 mg, 0.17 mmol of Fe); TEMPO (0.076 mmol, 12 mg); NaNO₂ (0.14 mmol; 10 mg); CH₃CN (5 mL) at 75 °C under atmospheric pressure of oxygen. (□) HMF; (■) DFF.

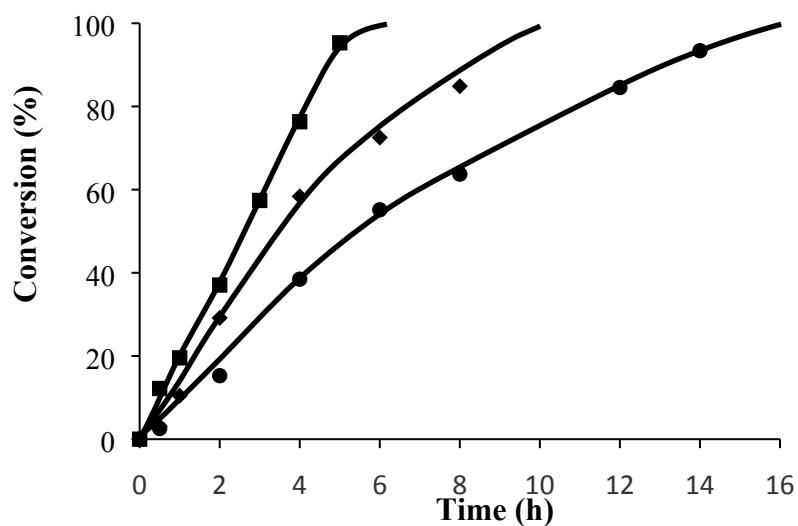


Figure S4. Conversion of 5-HMF versus time plot for oxidation of 5-HMF during reuses of MIL-100 (Fe)-NH₄F. Reaction conditions: 5-HMF (1 mmol, 126 mg); MIL-100 (Fe)-NH₄F (45 mg); TEMPO (0.076 mmol; 12 mg); NaNO₂ (0.14 mmol, 10 mg); at 75 °C, in CH₃CN -5ml under atmospheric pressure of oxygen. 1st cycle (■), 2nd cycle (◆), 3th cycle(●).

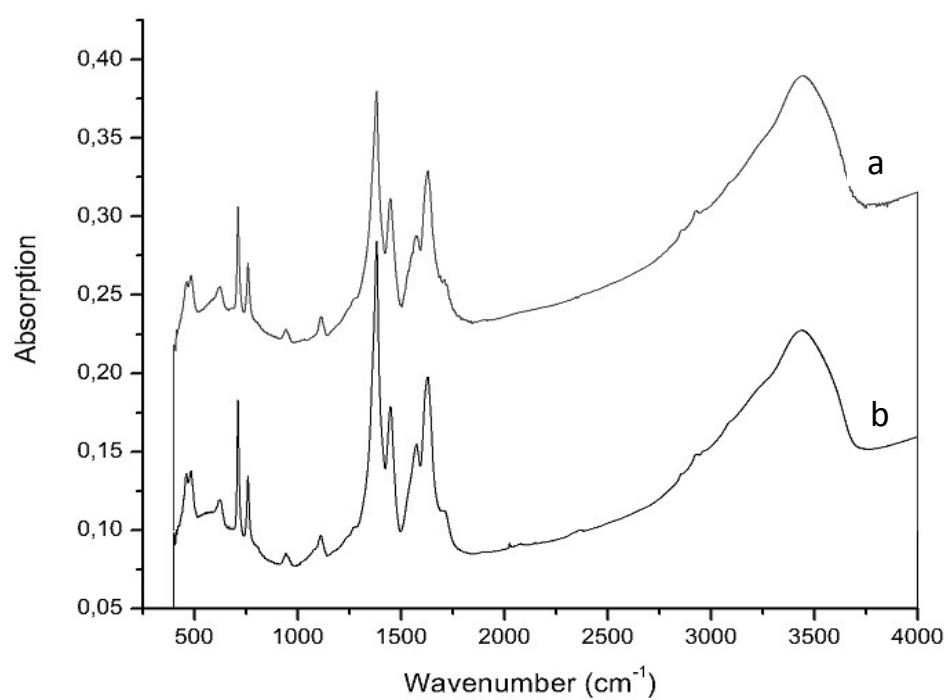


Figure S5. IR spectra of MIL-100(Fe)-NH₄F a) fresh and b) reused catalyst

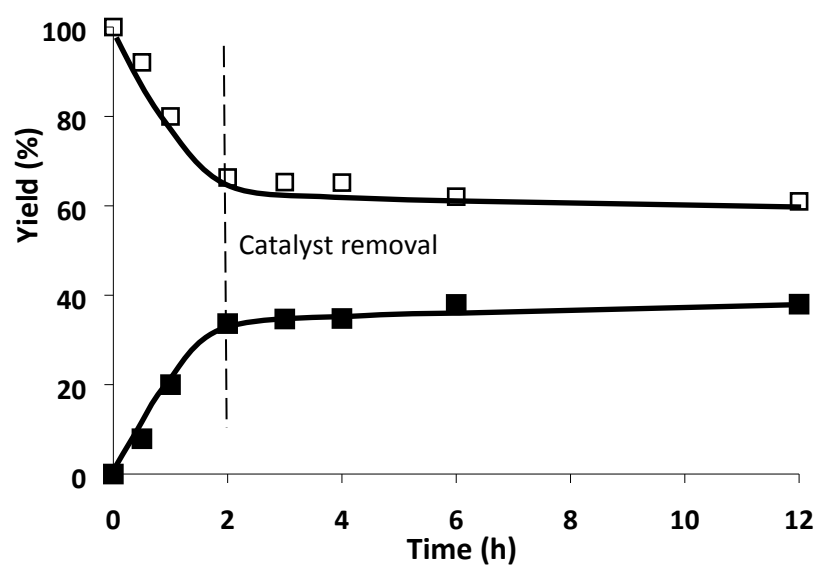


Figure S6. Leaching test of MIL-100 (Fe)-NH₄F catalysts.

Table S1. Physical and chemical properties of different studied catalysts

Catalyst	Metal content (wt %)	S_{BET} (m^2/g)	Total pore volume V_{total} (cm^3/g)	Crystal size (μm)
Fe(BTC)	21	613	0.36	0.48
MIL-100 (Fe)- NH_4F	21	1370	0.84	0.57
MIL-100 (Fe)	20	993	0.69	0.47
$\text{Cu}_3(\text{BTC})_2$	25	1341	0.75	8.70
HY-Fe	5.3	544	0.46	-

