

Supplementary Information

Enhanced visible light harnessing and oxygen vacancy promoted N, S co-doped CeO₂ nanoparticle:

A challenging photocatalyst for Cr(VI) reduction

S. Mansingh^a, D. K. Padhi^{bc} and K. M. Parida^{a*}

^a Centre for Nano Science and Nano Technology SOA University, Bhubaneswar—751 030, Odisha, India.

^b Environment & Sustainability Department, CSIR-Institute of Minerals and Materials Technology, Bhubaneswar – 751 013, Odisha, India.

^c Academy of Scientific and Innovative Research (AcSIR), Council of Scientific and Industrial Research, AnusandhanBhawan, 2 Rafi Marg, New Delhi-110 001, India.

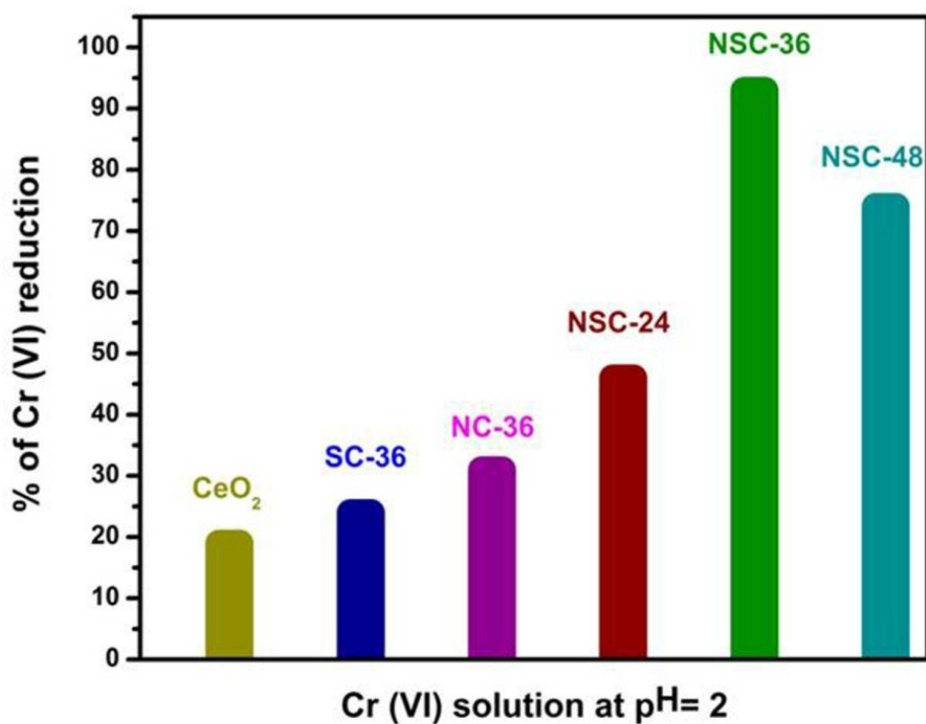


Figure S1. Photo-reduction of Cr(VI) by neat CeO₂, SC-36, NC-36, NCS-24, NCS-36 and NCS-48 under Visible-light irradiation [catalyst dose = 1g/L; [Cr(VI)] = 50 mg/L; time = 120 min]

Mott-Schottky

Mott-Schottky measurements was carried out to know the flat band edge potential and type of semiconducting material (n or p), which are important parameter of photochemical reaction because of their impact on charge transfer mechanism and recombination probability. The electrochemical flat band potential of the as prepared photocatalysts were calculated from by the following equation:

$$\frac{1}{C^2} = \frac{2}{\epsilon \epsilon_0 N_D} \left(E - E_{fb} - \frac{k_b T}{q} \right) \dots \dots \dots (2)$$

Where the symbols have specific notations such as C is the space charge capacitance, ϵ and ϵ_0 are the dielectric constants of free space and material electrodes respectively, N_D is the donor density, k_b is the Boltz's man constant, q is the electronic charge and T is the absolute temp.

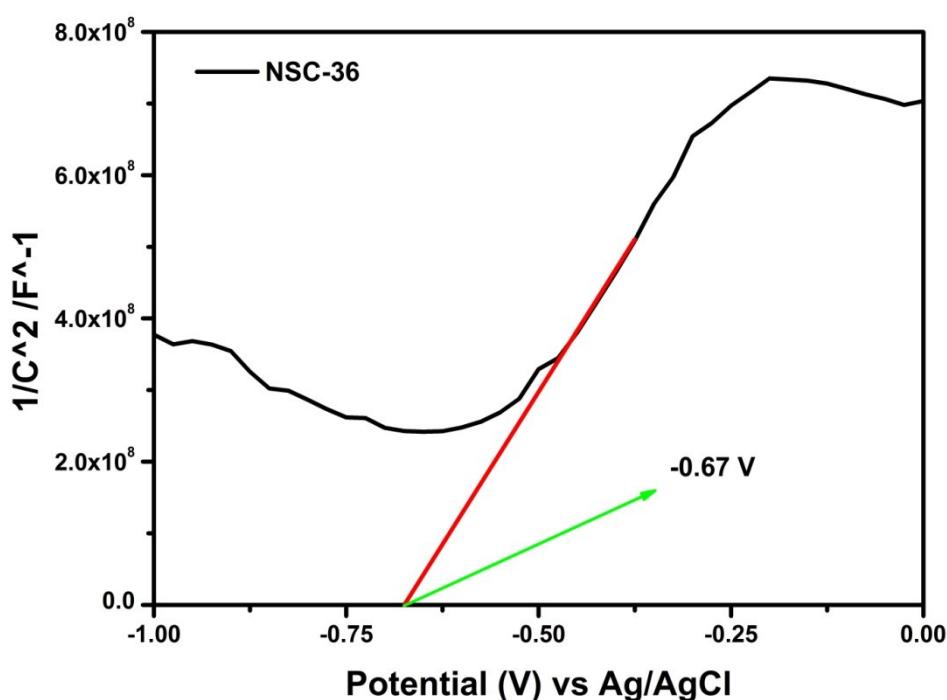


Figure S2. Mott-Schotky plot of NSC-36 photocatalyst measured 1k Hz.

The flat band (E_{fb}) potential of the semiconductor was determined from the intercept of the straight line on potential axis as shown in fig(1) and the nature of tangent indicates the type of extrinsic material i.e. positive slope for n-type and negative for p-type materials. In this case the E_{fb} value for the n-type doped

CeO₂ is computed to be -0.67 V vs Ag/AgCl electrode. Hence, the conduction bandage potential of NSC-36 was determined to -0.41 eVvs NHE. Accordingly, its corresponding valence band potential (VB) was calculated to be 1.78 eVvs NHE.

Reference

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