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## Supporting Information

## Enhanced and Durable Electrocatalytic Performance of Thin Layer PtRu Bimetallic

## Alloys on Pd-Nanocubes for Methanol Oxidation Reactions

Ammar Bin Yousaf <sup>a,b\*</sup>, M. Imran <sup>b</sup>, Peter Kasak <sup>a</sup>, Fathima Sifani Zavahir <sup>a</sup>, Syed Javaid

Zaidi <sup>a</sup> and Carlos Fernandez <sup>c</sup>

<sup>a</sup> Center for Advanced Materials, Qatar University, Doha 2713, Qatar

<sup>b</sup> Hefei National Laboratory for Physical Science at Microscale, School of Chemistry and

Materials Sciences, University of Science and Technology of China, Hefei 230026, P. R.

China.

<sup>c</sup> School of Pharmacy and Life Sciences, Sir Ian Wood Building, Robert Gordon University,

AB107GJ, Aberdeen, UK

Corresponding Author's emails: <u>ammar@mail.ustc.edu.cn</u>, <u>muhammad.ammar@qu.edu.qa</u>

ammar.chemist18@gmail.com (A. B. Yousaf)



Fig. S1: Base CV for commercial PtRu/C in 0.1 M HClO<sub>4</sub> solution at 50 mV/s and CV for commercial PtRu/C in 0.1 M HClO<sub>4</sub> + 1M CH<sub>3</sub>OH solution at 50 mV/s



Fig. S2: Nyquist plots for electrochemical impedance spectroscopic curves at three different potentials from 0.10 V to 0.70 V vs. RHE for all the catalysts.



Fig. S3: Electrochemical CO oxidation of (a) Pd-NCs and (b) PtRu@Pd-NCs catalysts, in 0.1 M HClO<sub>4</sub>, the scan rate is 50 mV s<sup>-1</sup>.

Sr. No.	Catalyst	j / mA cm <sup>-2</sup> (w.r.t ECSA)	j / mA cm <sup>-2</sup> (w.r.t geometric surface area)
01	PtRu@Pd-NCs	11.44	44.29
02	PtRu/C	0.65	10.38

Table S1: Measurements of current densities of PtRu@Pd-NCs and PtRu/C catalysts by normalizing current with ECSA and geometric surface area