## Supporting Information

# Phosphine-Pyridonate Ligands Containing Octahedral Ruthenium Complexes : Access to Esters and Formic Acid 

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## List of Contents

A. General Experimental Methods ..... S1
B. Synthesis of the Ligands ..... S1
C. Internal Standard Studies ..... S2
D. Analytical Data ..... S3
E. Internal Standard Studies for Formic Acid Production. ..... S7
G. NMR spectra ..... S8

## A. General Experimental Methods

All reactions were carried out under an inert argon atmosphere with standard schlenk techniques. Solvents were degassed and stored in argon atmosphere before use. Reagents were used as received without further purification, unless otherwise stated. ${ }^{1} \mathrm{H}$ NMR spectra were recorded using a Bruker Avance ${ }^{\text {III }} 300 \mathrm{MHz}$ and 400 MHz NMR spectrometers. All ${ }^{1} \mathrm{H}-\mathrm{NMR}$ data are reported in $\delta$ units, parts per million ( ppm ) and were calibrated relative to the reported residual solvent signals in the corresponding deuterated solvents. All ${ }^{13} \mathrm{C}-\mathrm{NMR}$ and ${ }^{31} \mathrm{P}-\mathrm{NMR}$ data are reported in ppm and were recorded with ${ }^{1} \mathrm{H}$ decoupling. The following abbreviations or combinations thereof were used to explain the multiplicities: $\mathrm{s}=$ singlet, $\mathrm{d}=$ doublet, $\mathrm{dd}=$ doublet of doublet, $\mathrm{dt}=$ doublet of triplet, $\mathrm{t}=$ triplet, $\mathrm{q}=$ quartet, quint $=$ quintet $, \mathrm{br}=\mathrm{broad}, \mathrm{m}$ $=$ multiplet. High resolution mass spectra (HRMS) were recorded on a Bruker microTOF mass spectrometer using ESI-TOF (electrospray ionization-time of flight).

## B. Synthesis of the Ligands

## 6-((Diphenylphosphaneyl)methyl)pyridin-2(1H)-one: (L1-H)

6-Methylpyridin-2-ol ( $1.0 \mathrm{~g}, 9.16 \mathrm{mmol}, 1.0$ equiv.) was dissolved in 10.0 mL THF, cooled at $0^{\circ} \mathrm{C}$ followed by the slow addition of $n-\mathrm{BuLi}(12 \mathrm{~mL}, 19.2 \mathrm{mmol}$, 2.1equiv.). This solution was stirred for an hour. Then it was cooled to $-78^{\circ} \mathrm{C}$ and was added to another solution containing Diphenylphosphinechloride $(1.6 \mathrm{~mL}, 9.16 \mathrm{mmol}, 1.0$ equiv.) in 2.5 mL of THF at $-78^{\circ} \mathrm{C}$. This solution was stirred at -78 ${ }^{\circ} \mathrm{C}$ for one hour. Then the solution was allowed to warm up naturally
 to room temperature and was stirred at room temperature for 16 hours. After the completion of reaction, solvent was evaporated followed by addition of 10 mL degassed water to dissolve the oily liquid. After 10 mL dicholoromethane was added and the solution was acidified slowly with $5 \% \mathrm{HCl}$ solution (degassed) till the pH was around 3. Addition of degassed acetone gradually precipitated the ligand as off-white solid. Further washing with degassed acetone afforded the expected ligand as white powder with $70 \%$ yield ( $1.9 \mathrm{~g}, 6.4 \mathrm{mmol}$ ).

Spectroscopic details are consistent with the reported literature. ${ }^{1}$

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## 6,6'-(phenylphosphinediyl)bis(methylene)dipyridin-2(1H)-one [P(NOH)2]: (L2-H)

Ligand L2-H is prepared by the reported procedure. ${ }^{2}$
6,6'-(tert-butylphosphinediyl)bis(methylene)dipyridin-2(1H)-one [ $\left.{ }^{[ } \mathbf{B u P}(\mathbf{N O H})_{2}\right]:$ (L3-H)
Ligand L3-H is prepared by the reported procedure. ${ }^{2}$

## C. Internal Standard Studies



Internal standard graph for calculation of conversion of benzyl alcohol


Internal standard graph for calculation of yield of benzyl benzoate

[^1]

Internal standard graph for calculation of yield of benzaldehyde

## D. Analytical Data

Benzyl benzoate (3a) ${ }^{4}$ : Obtained by the reaction of benzyl alcohol ( 0.5 mmol ) in presence of complex Ru-1 and NaOH in toluene at $150^{\circ} \mathrm{C}$ for 16 h under argon atmosphere. Purification by silica gel column chromatography (pet. ether : $\mathrm{EtOAc}=19: 1$ ) afforded the product as a colourless liquid
 with $94 \%$ yield. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $88.11-8.09$ (m, $\mathrm{ArCH}, 2 \mathrm{H}$ ), 7.59-7.55 (m, ArCH , $1 \mathrm{H}), 7.48-7.34(\mathrm{~m}, \mathrm{ArCH}, 7 \mathrm{H}), 5.39\left(\mathrm{~s}, \mathrm{OCH}_{2}, 2 \mathrm{H}\right) ;{ }^{13} \mathrm{C}$ NMR ( $101 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 166.6$ (quat-C), 136.2 (quat-C), 133.1 ( ArCH ), 130.3 (quat-C), 129.8 ( ArCH ), 128.7 ( ArCH ), 128.5 $(\mathrm{ArCH}), 128.4(\mathrm{ArCH}), 128.3(\mathrm{ArCH}), 66.8\left(\mathrm{OCH}_{2}\right)$.

2-Methylbenzyl 2-methylbenzoate (3b) ${ }^{3}$ : Obtained by the reaction of 2-methylbenzyl alcohol ( 0.5 mmol ) in presence of complex $\mathbf{R u}-1$ and NaOH in toluene at $150^{\circ} \mathrm{C}$ for 20 h under argon atmosphere. Purification by silica gel column chromatography (pet. ether : $\mathrm{EtOAc}=25: 1$ ) afforded the product as a colourless oil with $72 \%$ yield. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 7.97(\mathrm{~d}, J=7.9 \mathrm{~Hz}$,
 $\mathrm{ArCH}, 1 \mathrm{H}), 7.46-7.40(\mathrm{~m}, \mathrm{ArCH}, 2 \mathrm{H}), 7.32-7.24$ (m, ArCH, 5H), 5.39 (s, $\mathrm{OCH}_{2}, 2 \mathrm{H}$ ), 2.64 (s, $\mathrm{CH}_{3}, 3 \mathrm{H}$ ), 2.45 ( $\mathrm{s}, \mathrm{CH}_{3}, 3 \mathrm{H}$ ); ${ }^{13} \mathrm{C}$ NMR ( $101 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 167.4$ (quat-C), 140.5 (quat-C), 137.1 (quat-C), 134.2 (quat-C), 132.2 ( ArCH ), 131.9 ( ArCH ), $130.8(\mathrm{ArCH}), 130.5(\mathrm{ArCH})$, 129.6 (quat-C), $129.4(\mathrm{ArCH}), 128.6(\mathrm{ArCH}), 126.2(\mathrm{ArCH}), 125.8(\mathrm{ArCH}), 65.0(\mathrm{OCH}), 21.9$

[^2]$\left(\mathrm{CH}_{3}\right), 19.1\left(\mathrm{CH}_{3}\right)$.
3-Methylbenzyl 3-methylbenzoate (3c) ${ }^{4}$ : Obtained by the reaction of 3-methylbenzyl alcohol ( 0.5 mmol ) in presence of complex $\mathbf{R u - 1}$ and NaOH in toluene at $150^{\circ} \mathrm{C}$ for 20 h under argon atmosphere. Purification by silica gel column chromatography (pet. ether : $\mathrm{EtOAc}=25: 1$ ) afforded the
 product as a colourless oil with $63 \%$ yield. ${ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): ~ \delta 7.90-7.88(\mathrm{~m}$, $\mathrm{ArCH}, 2 \mathrm{H}$ ), 7.38-7.27 (m, $\mathrm{ArCH}, 4 \mathrm{H}$ ), 7.25-7.06 (m, ArCH, 2H), 5.33 ( $\mathrm{s}, \mathrm{OCH}_{2}, 2 \mathrm{H}$ ), 2.40 ( s , $\mathrm{CH}_{3}, 3 \mathrm{H}$ ), $2.38\left(\mathrm{~s}, \mathrm{CH}_{3}, 3 \mathrm{H}\right) ;{ }^{13} \mathrm{C}$ NMR ( $101 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 166.8$ (quat-C), 138.4 (quat-C), 138.3 (quat-C), 136.2 (quat-C), $133.9(\mathrm{ArCH}), 130.4(\mathrm{ArCH}), 130.2$ (quat-C), $129.1(\mathrm{ArCH})$, $129.1(\mathrm{ArCH}), 128.6(\mathrm{ArCH}), 128.4(\mathrm{ArCH}), 127.0(\mathrm{ArCH}), 125.4(\mathrm{ArCH}), 66.8\left(\mathrm{OCH}_{2}\right), 21.5$ $\left(\mathrm{CH}_{3}\right), 21.4\left(\mathrm{CH}_{3}\right)$.

4-Methylbenzyl 4-methylbenzoate (3d) ${ }^{4}$ : Obtained by the reaction of 4-methylbenzyl alcohol ( 0.5 mmol ) in presence of complex $\mathbf{R u}-1$ and NaOH in toluene at $150{ }^{\circ} \mathrm{C}$ for 20 h under argon atmosphere. Purification by silica gel column chromatography (pet. ether : $\mathrm{EtOAc}=25: 1$ )
 afforded the product as a colourless oil with $89 \%$ yield. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 7.99$ (d, $J=8.2 \mathrm{~Hz}, \mathrm{ArCH}, 2 \mathrm{H}$ ), 7.36 (d, $J=7.9 \mathrm{~Hz}, \mathrm{ArCH}, 2 \mathrm{H}$ ), 7.23 (dd, $J_{l}=10.8 \mathrm{~Hz}, J_{2}=8.2 \mathrm{~Hz}$, $\mathrm{ArCH}, 4 \mathrm{H}$ ), 5.33 ( $\mathrm{s}, \mathrm{OCH}_{2}, 2 \mathrm{H}$ ), $2.42\left(\mathrm{~s}, \mathrm{CH}_{3}, 3 \mathrm{H}\right), 2.38\left(\mathrm{~s}, \mathrm{CH}_{3}, 3 \mathrm{H}\right) ;{ }^{13} \mathrm{C}$ NMR ( 101 MHz , $\left.\mathrm{CDCl}_{3}\right): \delta 166.6$ (quat-C), 143.7 (quat-C), 138.1 (quat-C), 133.3 (quat-C), 129.8 ( ArCH ), $129.4(\mathrm{ArCH}), 129.2(\mathrm{ArCH}), 128.4(\mathrm{ArCH}), 127.6$ (quat-C), $66.6\left(\mathrm{OCH}_{2}\right), 21.8\left(\mathrm{CH}_{3}\right), 21.3$ $\left(\mathrm{CH}_{3}\right)$.

4-Fluorobenzyl 4-fluorobenzoate (3e) ${ }^{4}$ : Obtained by the reaction of 4-fluorobenzyl alcohol ( 0.5 mmol ) in presence of complex $\mathbf{R u}-\mathbf{1}$ and NaOH in toluene at $150{ }^{\circ} \mathrm{C}$ for 24 h under argon atmosphere. Purification by silica gel column chromatography (pet. ether : $\operatorname{EtOAc}=9: 1$ ) afforded the product as a colourless oil with $78 \%$ yield. ${ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ : $\delta 8.09-8.06(\mathrm{~m}, \mathrm{ArCH}, 2 \mathrm{H}), 7.44-7.41(\mathrm{~m}, \mathrm{ArCH}, 2 \mathrm{H}), 7.12-7.05(\mathrm{~m}, \mathrm{ArCH}, 4 \mathrm{H}), 5.32(\mathrm{~s}$, $\left.\mathrm{OCH}_{2}, 2 \mathrm{H}\right) ;{ }^{13} \mathrm{C}$ NMR ( $101 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 166.0\left(\mathrm{~d},{ }^{1} J_{\mathrm{C}-\mathrm{F}}=255 \mathrm{~Hz}\right.$, quat-C), 165.5 (quat-C), $162.8\left(\mathrm{~d},{ }^{1} J_{\mathrm{C}-\mathrm{F}}=247 \mathrm{~Hz}\right.$, quat-C), $132.3\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{F}}=9.3 \mathrm{~Hz}, \mathrm{ArCH}\right), 131.9\left(\mathrm{~d},{ }^{4} J_{\mathrm{C}-\mathrm{F}}=3.2 \mathrm{~Hz}\right.$, quat-C), $130.4\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{F}}=8.3 \mathrm{~Hz}, \mathrm{ArCH}\right), 126.4\left(\mathrm{~d},{ }^{4} J_{\mathrm{C}-\mathrm{F}}=3.0 \mathrm{~Hz}\right.$, quat-C), $115.7\left(\mathrm{~d},{ }^{2} J_{\mathrm{C}-\mathrm{F}}=\right.$ $21.8 \mathrm{~Hz}, \mathrm{ArCH}), 66.2\left(\mathrm{OCH}_{2}\right)$.

[^3]4-Bromobenzyl 4-bromobenzoate (3f) ${ }^{4}$ : Obtained by the reaction of 4-bromobenzyl alcohol ( 0.5 mmol ) in presence of complex $\mathbf{R u - 1}$ and NaOH in toluene at $150^{\circ} \mathrm{C}$ for 24 h under argon atmosphere. Purification by silica gel column chromatography (pet. ether : $\mathrm{EtOAc}=9: 1$ ) afforded the product as a white solid with $83 \%$ yield. ${ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta$
 7.94-7.89 (m, ArCH, 2H), 7.60-7.50 (m, ArCH, 4H), 7.33-7.2 (m, $\mathrm{ArCH}, 2 \mathrm{H}), 5.30\left(\mathrm{~s}, \mathrm{OCH}_{2}\right.$, 2 H ); ${ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 165.7$ (quat-C), 134.9 (quat-C), 131.9 ( ArCH ), 131.9 $(\mathrm{ArCH}), 130.1(\mathrm{ArCH}), 128.9$ (quat-C), 128.5 (quat-C), 122.6 (quat-C), $66.3\left(\mathrm{OCH}_{2}\right)$.

Octyl octanoate ( $\mathbf{3 g})^{3}$ : Obtained by the reaction of 1-octanol ( 0.5 mmol ) in presence of complex Ru-1 and NaOH in toluene at $150{ }^{\circ} \mathrm{C}$ for 24 h under argon atmosphere. Purification by silica gel column chromatography (pet. ether : $\mathrm{EtOAc}=19: 1$ ) afforded the product
 as a white solid with $85 \%$ yield. ${ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ : $\delta$ $4.05\left(\mathrm{t}, J=6.7 \mathrm{~Hz}, \mathrm{OCH}_{2}, 2 \mathrm{H}\right), 2.28\left(\mathrm{t}, J=7.5 \mathrm{~Hz}, \mathrm{CH}_{2} \mathrm{CO}, 2 \mathrm{H}\right), 1.64-1.57(\mathrm{~m}, 4 \mathrm{H}), 1.29-1.27$ $(\mathrm{m}, 18 \mathrm{H}), 0.89-0.85(\mathrm{~m}, 6 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $101 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 174.1$ (quat-C), $64.5\left(\mathrm{OCH}_{2}\right)$, $34.6\left(\mathrm{CH}_{2}\right), 31.9\left(\mathrm{CH}_{2}\right)$, $31.8\left(\mathrm{CH}_{2}\right)$, $29.4\left(\mathrm{CH}_{2}\right)$, $29.3\left(\mathrm{CH}_{2}\right)$, $29.3\left(\mathrm{CH}_{2}\right)$, $29.1\left(\mathrm{CH}_{2}\right), 28.8$ $\left(C H_{2}\right)$, $26.1\left(\mathrm{CH}_{2}\right), 25.2\left(\mathrm{CH}_{2}\right), 22.8\left(\mathrm{CH}_{2}\right), 22.7\left(\mathrm{CH}_{2}\right), 14.2\left(\mathrm{CH}_{3}\right)$, $14.2\left(\mathrm{CH}_{3}\right)$.

Hexyl hexanoate ( $\mathbf{3 h})^{5}$ : Obtained by the reaction of 1-hexanol ( 0.5 mmol ) in presence of complex Ru-1 and NaOH in toluene at $150{ }^{\circ} \mathrm{C}$ for 24 h under argon atmosphere. Purification by silica gel column chromatography (pet. ether : $\mathrm{EtOAc}=19: 1$ ) afforded the product as a white solid with $77 \%$ yield. ${ }^{1} \mathrm{H}$ NMR (400 $\mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 4.05\left(\mathrm{t}, J=6.7 \mathrm{~Hz}, \mathrm{OCH}_{2}, 2 \mathrm{H}\right), 2.28(\mathrm{t}, J=7.5 \mathrm{~Hz}$,
 $\left.\mathrm{CH}_{2} \mathrm{CO}, 2 \mathrm{H}\right), 1.63-1.59(\mathrm{~m}, 4 \mathrm{H}), 1.30(\mathrm{br}, 10 \mathrm{H}), 0.90-0.87(\mathrm{~m}, 6 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( 101 MHz , $\left.\mathrm{CDCl}_{3}\right): \delta 174.1\left(\right.$ quat-C), $64.5\left(\mathrm{OCH}_{2}\right), 34.5\left(\mathrm{CH}_{2}\right), 31.6\left(\mathrm{CH}_{2}\right), 31.5\left(\mathrm{CH}_{2}\right), 28.8\left(\mathrm{CH}_{2}\right), 25.7$ $\left(\mathrm{CH}_{2}\right)$, $24.9\left(\mathrm{CH}_{2}\right)$, $22.7\left(\mathrm{CH}_{2}\right)$, $22.5\left(\mathrm{CH}_{2}\right)$, $14.1\left(\mathrm{CH}_{3}\right)$, $14.0\left(\mathrm{CH}_{3}\right)$.

3-Phenylpropyl 3-phenylpropanoate (3i) ${ }^{4}$ : Obtained by the reaction of 3-phenyl-propan-1-ol ( 0.5 mmol ) in presence of complex $\mathbf{R u}-1$ and NaOH in toluene at $150{ }^{\circ} \mathrm{C}$ for 24 h under argon atmosphere. Purification by silica gel column chromatography (pet. ether : $\operatorname{EtOAc}=9: 1$ ) afforded the product as a white solid with $78 \%$ yield. ${ }^{1} \mathrm{H}$ NMR ( 300 MHz ,

[^4]$\left.\mathrm{CDCl}_{3}\right): ~ \delta 7.35-7.17(\mathrm{~m}, \mathrm{ArCH}, 10 \mathrm{H}), 4.12\left(\mathrm{t}, J=6.5 \mathrm{~Hz}, \mathrm{OCH}_{2}, 2 \mathrm{H}\right), 2.99(\mathrm{t}, J=7.8 \mathrm{~Hz}$, $\left.\mathrm{CH}_{2} \mathrm{CO}, 2 \mathrm{H}\right), 2.69-2.64(\mathrm{~m}, 4 \mathrm{H}), 1.96(\mathrm{q}, \mathrm{J}=6.3 \mathrm{~Hz}, 2 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 173.0$ (quat-C), 141.3 (quat-C), 140.6 (quat-C), $128.6(\mathrm{ArCH}), 128.5(\mathrm{ArCH}), 128.5(\mathrm{ArCH}), 128.4$ $(\mathrm{ArCH}), 126.4(\mathrm{ArCH}), 126.1(\mathrm{ArCH}), 63.9\left(\mathrm{OCH}_{2}\right), 36.0\left(\mathrm{CH}_{2}\right), 32.3\left(\mathrm{CH}_{2}\right), 31.1\left(\mathrm{CH}_{2}\right), 30.3$ $\left(\mathrm{CH}_{2}\right)$.

3,7-Dimethyloct-6-en-1-yl 3,7-dimethyloct-6-enoate (3j): Obtained by the reaction of Citronellol ( 0.5 mmol ) in presence of complex Ru-1 and NaOH in toluene at $150{ }^{\circ} \mathrm{C}$ for 24 h under argon atmosphere. Purification by silica gel
 column chromatography (pet. ether : $\mathrm{EtOAc}=9: 1$ ) afforded the product as a colourless oil with $82 \%$ yield. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 5.08(\mathrm{t}, J=6.6 \mathrm{~Hz}, 2 \mathrm{H}), 4.14-4.04(\mathrm{~m}, 2 \mathrm{H})$, $2.29\left(\mathrm{dd}, J_{1}=14.6 \mathrm{~Hz}, J_{2}=5.9 \mathrm{~Hz}, 1 \mathrm{H}\right), 2.09\left(\mathrm{dd}, J_{1}=14.5 \mathrm{~Hz}, J_{2}=8.2 \mathrm{~Hz}, 1 \mathrm{H}\right), 2.01-1.91$ $(\mathrm{m}, 5 \mathrm{H}), 1.67(\mathrm{~s}, 3 \mathrm{H}), 1.59(\mathrm{~s}, 3 \mathrm{H}), 1.57-1.13(\mathrm{~m}, 7 \mathrm{H}), 0.93(\mathrm{~d}, J=6.6 \mathrm{~Hz}, 3 \mathrm{H}), 0.90(\mathrm{~d}, J=$ $6.6 \mathrm{~Hz}, 3 \mathrm{H}$ ); ${ }^{13} \mathrm{C}$ NMR ( $101 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 173.5$ (d, quat-C), 131.5 (d, quat-C), 124.6 (d, $C H), 62.8\left(\mathrm{~d}, \mathrm{CH}_{2}\right), 42.0\left(\mathrm{~d}, \mathrm{CH}_{2}\right), 37.1\left(\mathrm{CH}_{2}\right), 36.9\left(\mathrm{CH}_{2}\right), 35.7\left(\mathrm{CH}_{2}\right), 30.2(\mathrm{CH}), 29.6(\mathrm{~d}$, $C H), 25.8\left(\mathrm{~d}, \mathrm{CH}_{3}\right), 25.5\left(\mathrm{~d}, \mathrm{CH}_{2}\right), 19.8\left(\mathrm{CH}_{3}\right), 19.5\left(\mathrm{CH}_{3}\right), 17.8\left(\mathrm{CH}_{3}\right)$. T wo adjacent peaks are due to the two diastereoisomers. HRMS(ESI-TOF): calc'd for $\mathrm{C}_{20} \mathrm{H}_{36} \mathrm{O}_{2} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+}$ 331.26075; found 331.2608.

4-Methyltetrahydro-2H-pyran-2-one (3k) ${ }^{\mathbf{6}}$ : Obtained by the reaction of 3-methyl-pentane-1,5-diol ( 0.5 mmol ) in presence of complex $\mathbf{R u} \mathbf{- 1}$ and NaOH in toluene ( 2 mL ) at 150 ${ }^{\circ} \mathrm{C}$ for 24 h under argon atmosphere. Purification by silica gel column chromatography (pet. ether : $\mathrm{EtOAc}=2: 1$ ) afforded the product as a colourless oil with $81 \%$ yield. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 4.40$ (ddd, $J_{1}=11.4 \mathrm{~Hz}, J_{2}$ $\left.=4.9 \mathrm{~Hz}, J_{3}=4.0 \mathrm{~Hz}, 1 \mathrm{H}\right), 4.25\left(\mathrm{td}, J_{1}=11.0 \mathrm{~Hz}, J_{2}=3.8 \mathrm{~Hz}, 1 \mathrm{H}\right), 2.70-2.62$
 $(\mathrm{m}, 1 \mathrm{H}), 2.13-2.03(\mathrm{~m}, 2 \mathrm{H}), 1.94-1.87(\mathrm{~m}, 1 \mathrm{H}), 1.55-1.46(\mathrm{~m}, 1 \mathrm{H}), 1.05(\mathrm{~d}, J=6.3 \mathrm{~Hz}, 3 \mathrm{H})$; ${ }^{13} \mathrm{C}$ NMR ( $101 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 171.3$ (quat-C), $68.7\left(\mathrm{CH}_{2}\right), 38.4\left(\mathrm{CH}_{2}\right), 30.8\left(\mathrm{CH}_{2}\right), 26.7(\mathrm{CH})$, $21.6\left(\mathrm{CH}_{3}\right)$.

[^5]
## E. Internal Standard Studies for Formic Acid Production

$$
\begin{gathered}
{\left[\mathrm{HCO}_{2} \mathrm{H}\right]=0.3268 *\left(\text { int. } \mathrm{HC}\left(\mathrm{HCO}_{2} \mathrm{H}\right) /\right.} \\
\text { int. } \mathrm{HCO}(\mathrm{DMF}))
\end{gathered}
$$



Internal standard curve for the calculation of Formic acid concentration









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| 130 | 110 | 90 | 70 | 50 | 30 | 10 | －10 | －30 | －50 | －70 | －90 | －110 | －130 | －1！ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |



| . 90 | 180 | 170 | 160 | 150 | 140 | 130 | 120 | 110 <br> f1 (ppm) | 100 | 90 | 80 | 70 | 60 | 50 | 40 | 31 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |




| J0 | 190 | 180 | 1/0 | 160 | 150 | 140 | 130 | 120 | 110 | 100 | 90 | 80 | 70 | 60 | 50 | 40 | 30 | 20 | 10 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  |  |  |  |  |  |  |  | (ppm) |  |  |  |  |  |  |  |  |  |











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\(-5.328\)
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$-174.147$






4
8-\infty
8-\infty
$\xrightarrow{2}$












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