

Supporting Information for

**Nitrile hydroboration reactions catalysed by simple nickel salts,
bis(acetylacetonato)nickel(II) and its derivatives**

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Shimada).

Contents

Experimental details and compound characterization data.....	S2
Figure 1. Molecular structure of <i>t</i> BuCOCHC(<i>t</i> Bu)OBcat with 50% probability ellipsoids.	S19
Table S1. Crystal data and details of the crystal structure determination for <i>t</i> BuCOCHC(<i>t</i> Bu)OBcat.....	S20
References.....	S20

Experimental Details and Compound Characterization Data

General considerations

Unless otherwise noted, all manipulations were performed under a nitrogen atmosphere using Schlenk techniques or a glove box. Benzene, toluene, hexane, and THF were purified by a solvent purification system (MBraun SPS-800 or Glass Contour Ultimate Solvent System). Other solvents (1,2-dichloroethane, benzene-*d*₆) were dried over CaH₂ or sodium benzophenone ketyl and distilled. All reagents were purchased from commercial suppliers and used without further purification unless otherwise noted. Catecholborane was purchased from Sigma-Aldrich Ltd. and purified by distillation. ¹H, ¹¹B, and ¹³C{¹H} NMR spectra (¹H, 400 MHz; ¹¹B, 128 MHz; ¹³C, 101 MHz) were recorded using a Bruker AVANCE 600 spectrometer. Chemical shifts are reported in δ (ppm) and are referenced to the residual solvent signals for ¹H and ¹³C, and to boron trifluoride diethyl ether complex (BF₃·OEt₂, 0.0 ppm) as an external reference for ¹¹B.

Catalytic Hydroborations

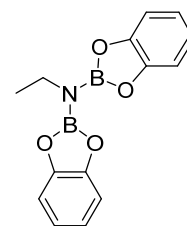
A typical procedure (Table 2, entry 1) is as follows. All reactions were carried out under nitrogen atmosphere. To a stirred solution of bis(2,2,6,6-tetramethyl-3,5-heptanedionato)nickel(II) (**3**) (0.001 mmol) in benzene (0.5 mL), was added benzonitrile (20.6 mg, 0.20 mmol) at 25 °C. After the mixture was stirred for 1 min, catecholborane (52.8 mg, 0.44 mmol) was added, and then the solution was stirred at room temperature for 18 hours. PhSiMe₃ (13.1 mg, 0.087 mmol) as an internal standard was added to the reaction mixture, and ¹H NMR was measured to determine the NMR yield of PhCH₂N(Bcat)₂ (>99%). The resulting solution was then diluted by benzene (30 mL) and filtered to remove precipitates. The filtrate was concentrated to dryness to give analytically pure PhCH₂N(Bcat)₂ (65.9 mg, 0.19 mmol, 93%).

Compound Characterization Data

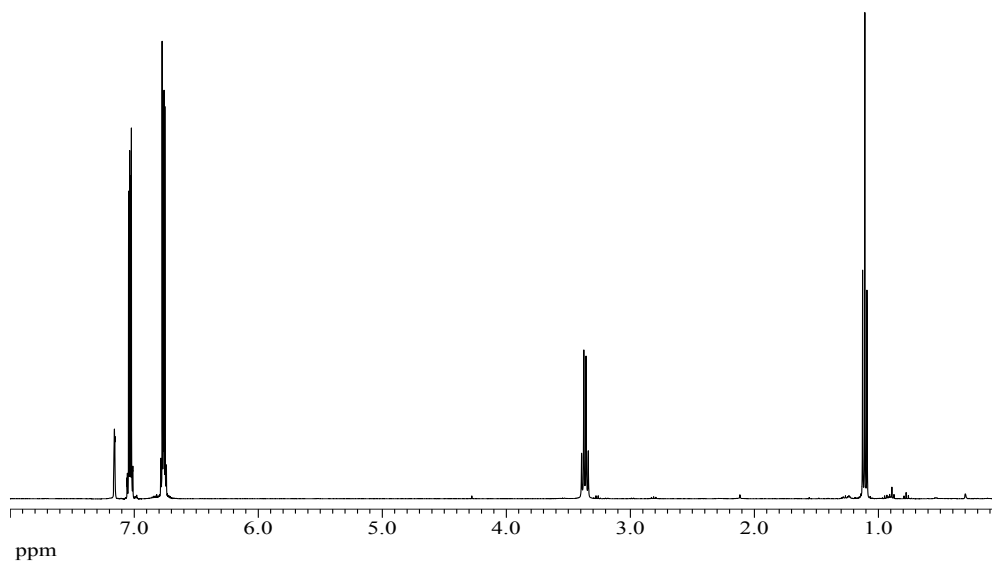
The final product was characterized by ¹H, ¹³C{¹H} and ¹¹B{¹H} NMR due to the instability of the hydroborated products under air. CH₃CH₂N(Bcat)₂ and PhCH₂N(Bcat)₂ were identified by comparing their ¹H, ¹¹B, and ¹³C{¹H} NMR data with those previously reported.¹

$CH_3CH_2(Bcat)_2$

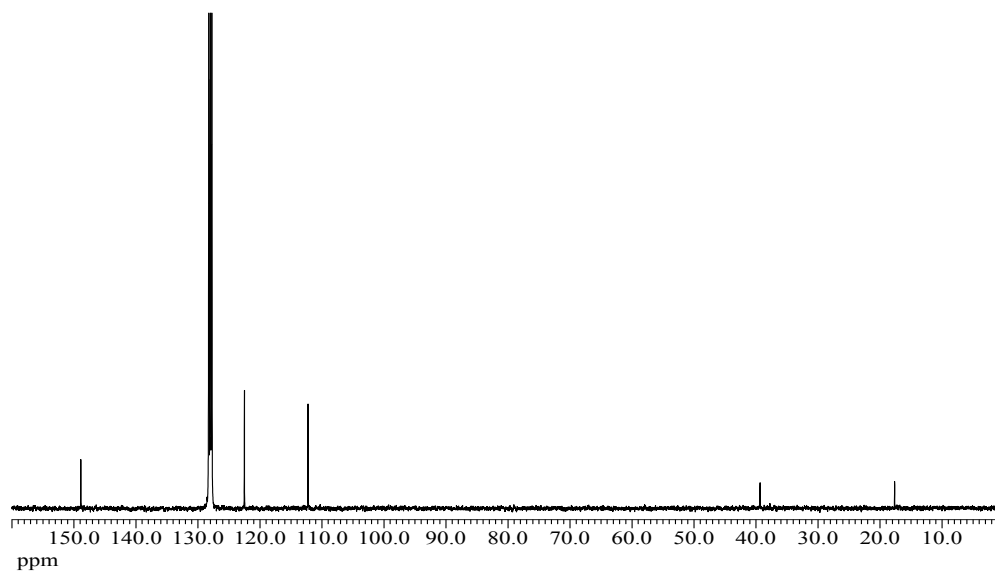
1H NMR (C_6D_6 , 25 °C): 7.03 (m, 4H, *Bcat*), 6.76 (m, 4H, *Bcat*), 3.34 (q, 2H, CH_2N , $J = 7.2$ Hz), 1.11 (t, 3H, CH_3 , $J = 7.2$ Hz). ^{11}B NMR (C_6D_6 , 25 °C): 26.8 (bs, *Bcat*). $^{13}C\{^1H\}$ NMR (C_6D_6 , 25 °C): 148.9, 122.5, 112.3, 39.4, 17.7.



1H NMR:

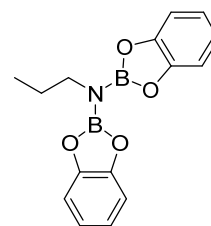


$^{13}C\{^1H\}$ NMR:

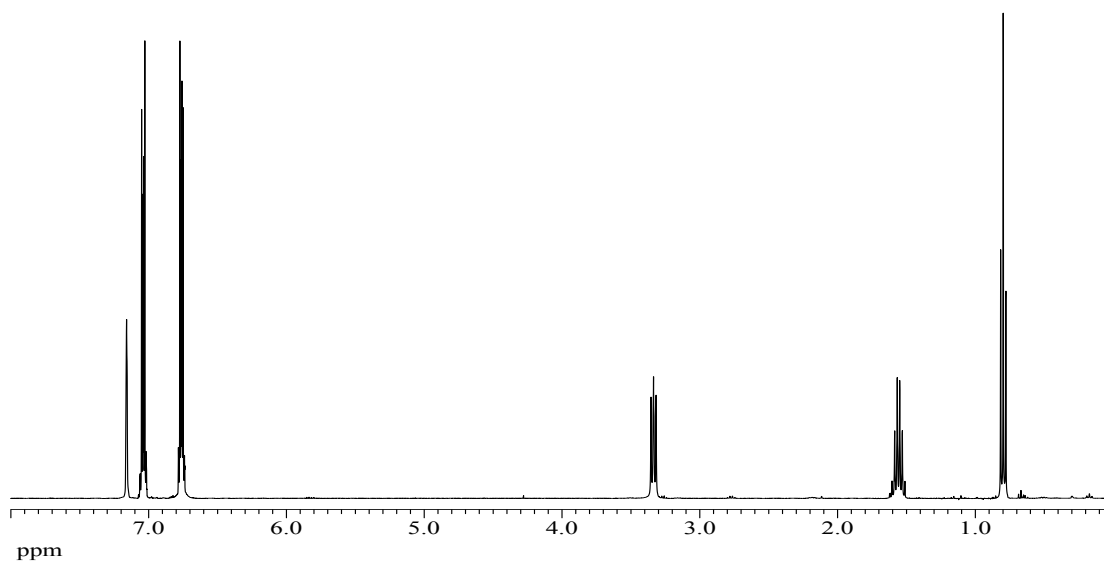


$CH_3CH_2CH_2N(Bcat)_2$

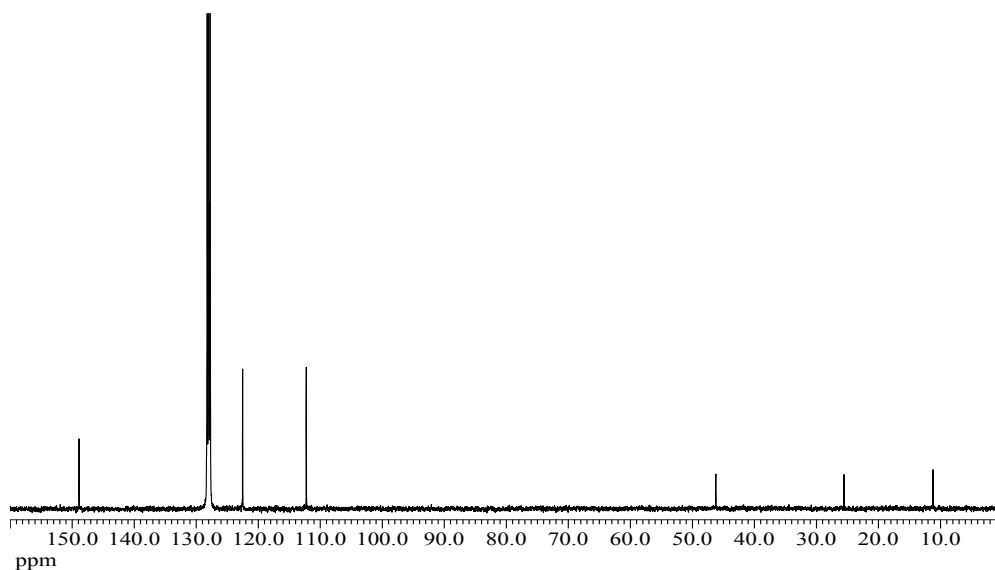
1H NMR (C_6D_6 , 25 °C): 7.04 (m, 4H, *Bcat*), 6.76 (m, 4H, *Bcat*), 3.34 (t, 2H, CH_2N , $J = 7.6$ Hz), 1.56 (m, 2H, $NCH_2-CH_2-CH_3$), 0.80 (t, 3H, CH_3 , $J = 7.2$ Hz). ^{11}B NMR (C_6D_6 , 25 °C): 27.0 (bs, *Bcat*). $^{13}C\{^1H\}$ NMR (C_6D_6 , 25 °C): 148.9, 122.5, 112.3, 46.2, 25.6, 11.2.



1H NMR:



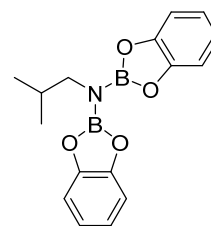
$^{13}C\{^1H\}$ NMR:



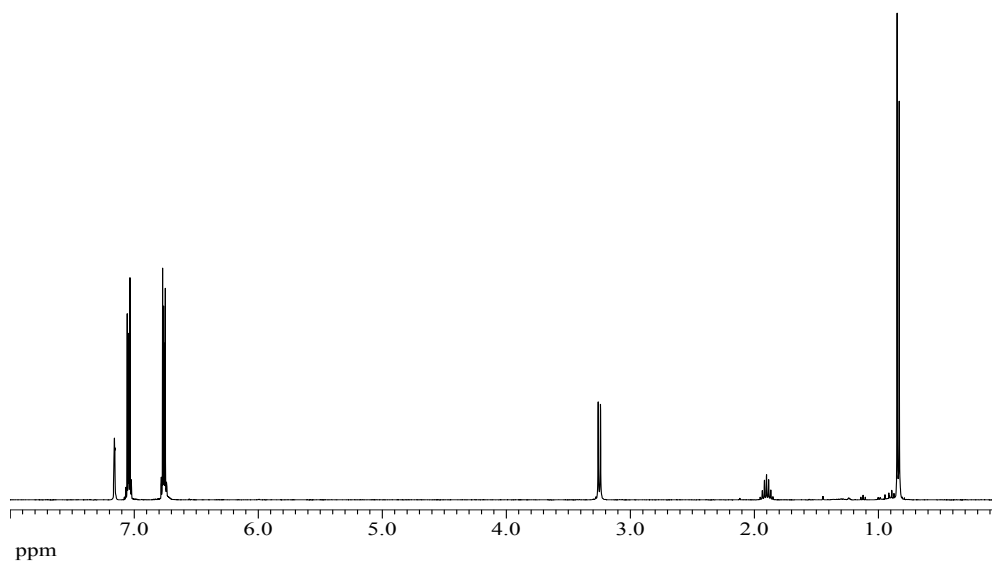
$i\text{PrCH}_2\text{N}(\text{Bcat})_2$

^1H NMR (C_6D_6 , 25 °C): 7.05 (m, 4H, *Bcat*), 6.76 (m, 4H, *Bcat*), 3.25 (d, 2H, CH_2N , $J = 7.6$ Hz), 1.90 (m, 1H, $(\text{CH}_3)_2\text{CH}$), 0.84 (d, 6H, CH_3 , $J = 6.4$ Hz).

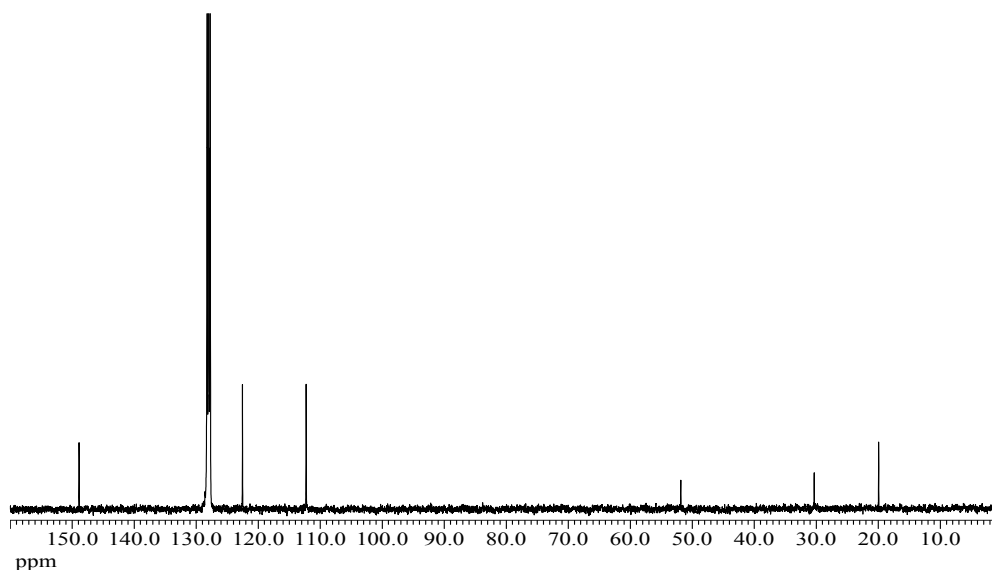
^{11}B NMR (C_6D_6 , 25 °C): 26.9 (bs, *Bcat*). $^{13}\text{C}\{^1\text{H}\}$ NMR (C_6D_6 , 25 °C): 148.9, 122.6, 112.3, 51.9, 30.4, 20.0.



^1H NMR:

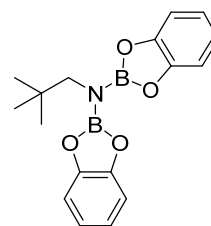


$^{13}\text{C}\{^1\text{H}\}$ NMR:

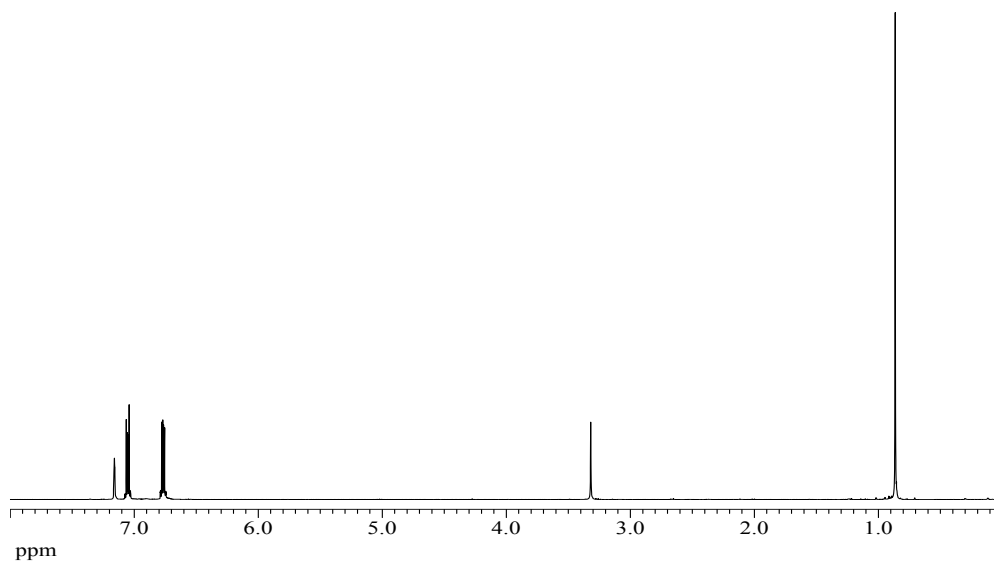


$t\text{BuCH}_2\text{N}(\text{Bcat})_2$

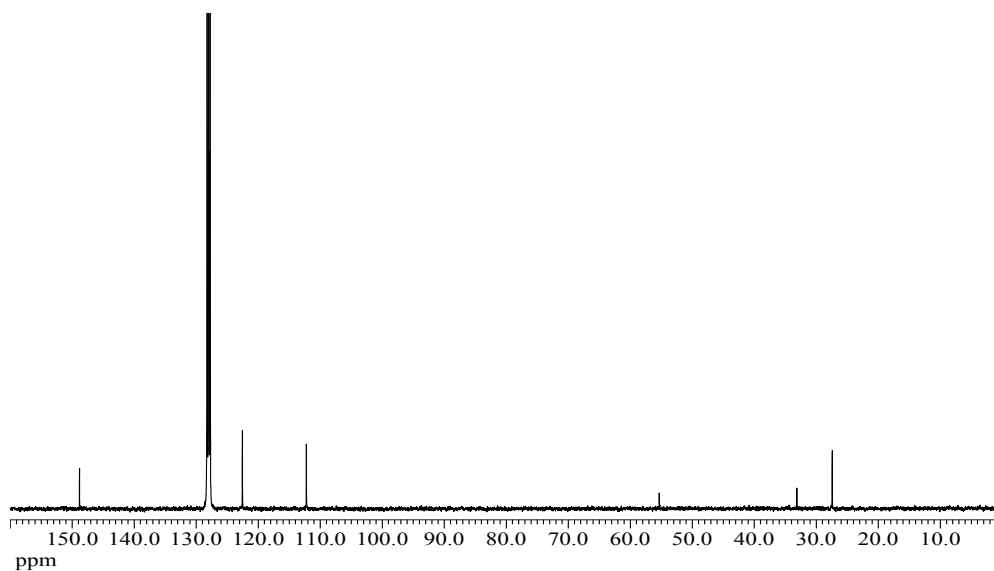
^1H NMR (C_6D_6 , 25 °C): 7.05 (m, 4H, *Bcat*), 6.77 (m, 4H, *Bcat*), 3.32 (s, 2H, CH_2N), 0.86 (s, 9H, CH_3). ^{11}B NMR (C_6D_6 , 25 °C): 27.0 (bs, *Bcat*). $^{13}\text{C}\{^1\text{H}\}$ NMR (C_6D_6 , 25 °C): 148.8, 122.6, 112.2, 55.3, 33.2, 27.4.



^1H NMR:

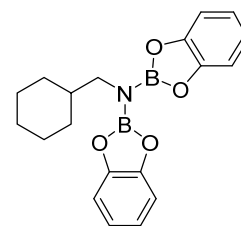


$^{13}\text{C}\{^1\text{H}\}$ NMR:

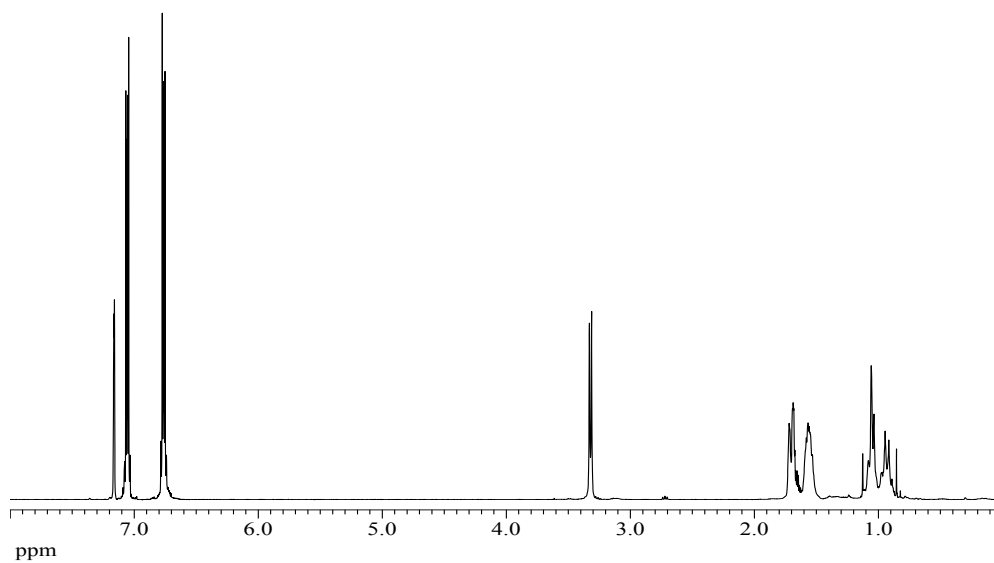


$CyCH_2N(Bcat)_2$

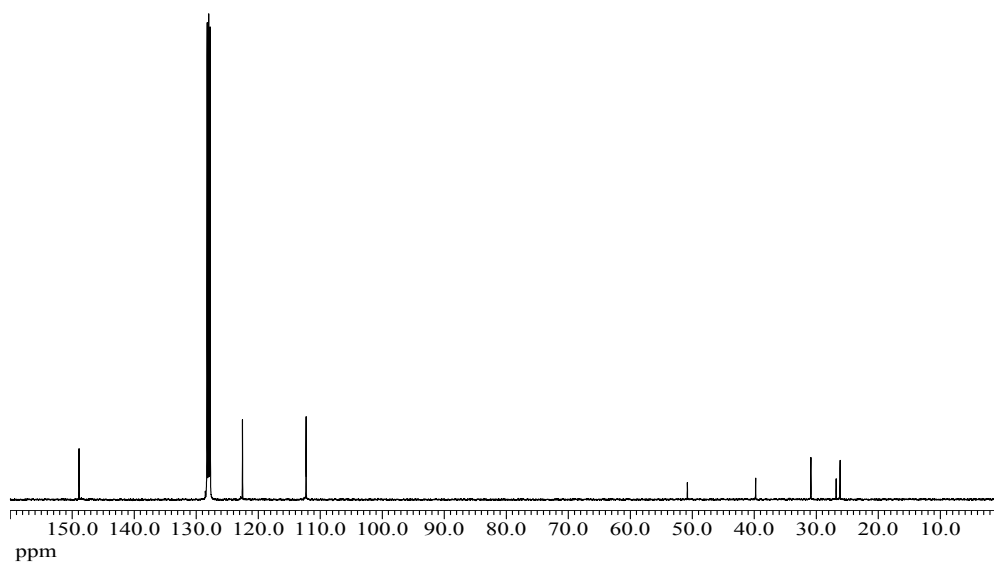
1H NMR (C_6D_6 , 25 °C): 7.06 (m, 4H, *Bcat*), 6.76 (m, 4H, *Bcat*), 3.32 (d, 2H, CH_2N , $J = 7.2$ Hz), 1.63 (m, 6H, *Cy-H*), 0.99 (m, 5H, *Cy-H*). ^{11}B NMR (C_6D_6 , 25 °C): 26.9 (bs, *Bcat*). $^{13}C\{^1H\}$ NMR (C_6D_6 , 25 °C): 148.9, 122.55, 112.3, 50.8, 39.8, 30.9, 26.8, 26.2.



1H NMR:

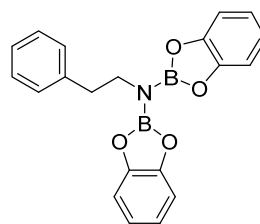


$^{13}C\{^1H\}$ NMR:

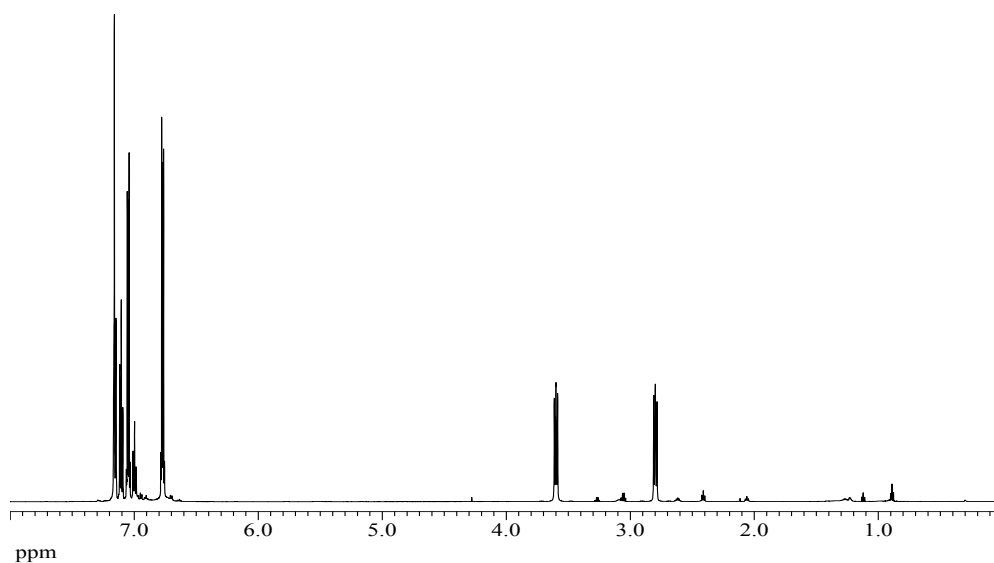


PhCH₂CH₂N(Bcat)₂

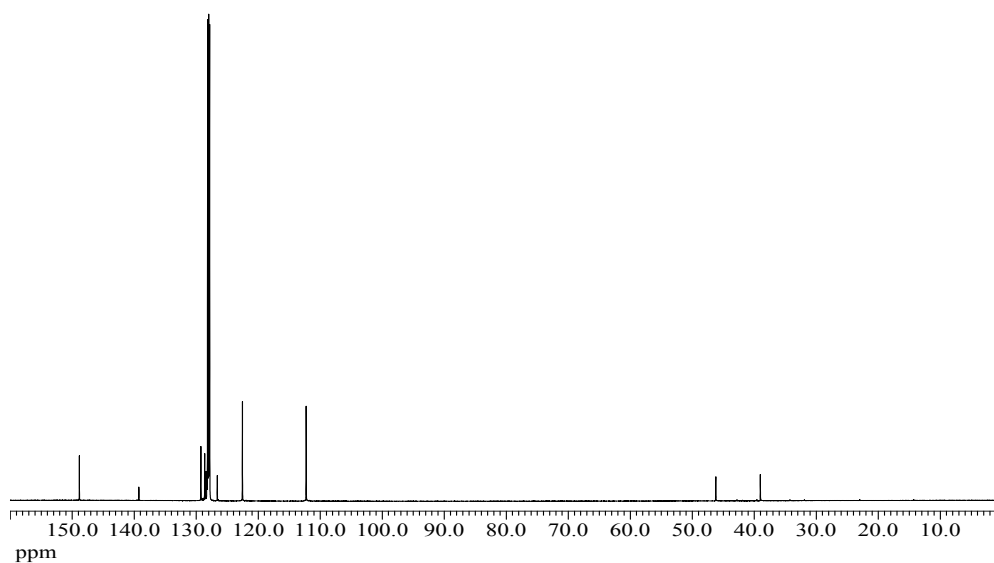
¹H NMR (C₆D₆, 25 °C): 7.15 (d, 2H, *J* = 7.2 Hz, Ar-*H*), 7.10 (t, 2H, *J* = 7.2 Hz, Ar-*H*), 7.05 (m, 4H, *Bcat*), 7.00 (t, 1H, *J* = 7.2 Hz, Ar-*H*), 6.77 (m, 4H, *Bcat*), 3.60 (t, 2H, *J* = 7.6 Hz, CH₂N), 2.80 (t, 2H, *J* = 7.6 Hz, PhCH₂). ¹¹B NMR (C₆D₆, 25 °C): 27.0 (bs, *Bcat*). ¹³C{¹H} NMR (C₆D₆, 25 °C): 148.8, 139.3, 129.3, 128.7, 128.5, 128.3, 122.6, 112.3, 46.2, 39.0.



¹H NMR:

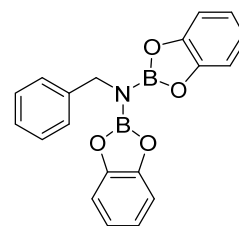


¹³C{¹H} NMR:

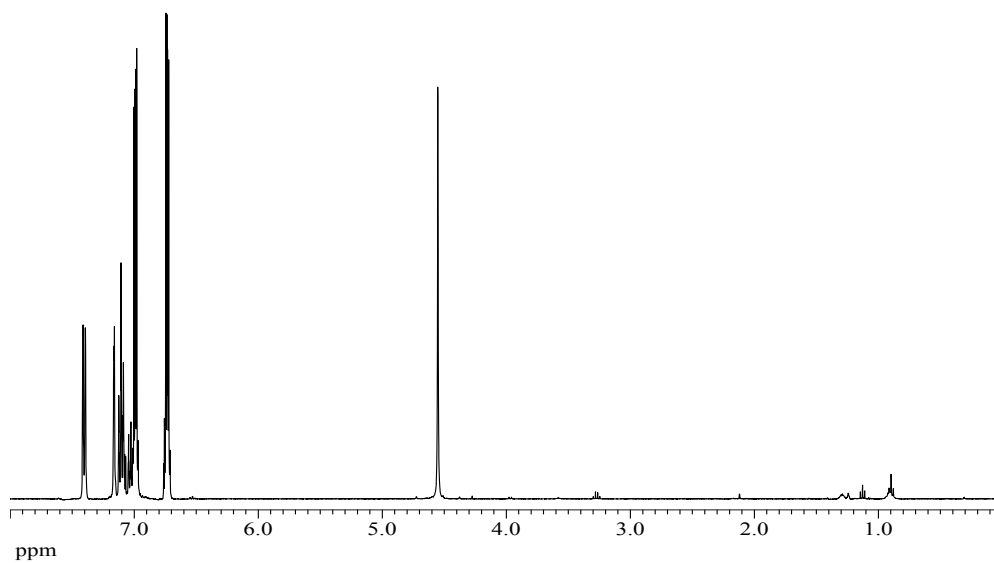


PhCH₂N(Bcat)₂

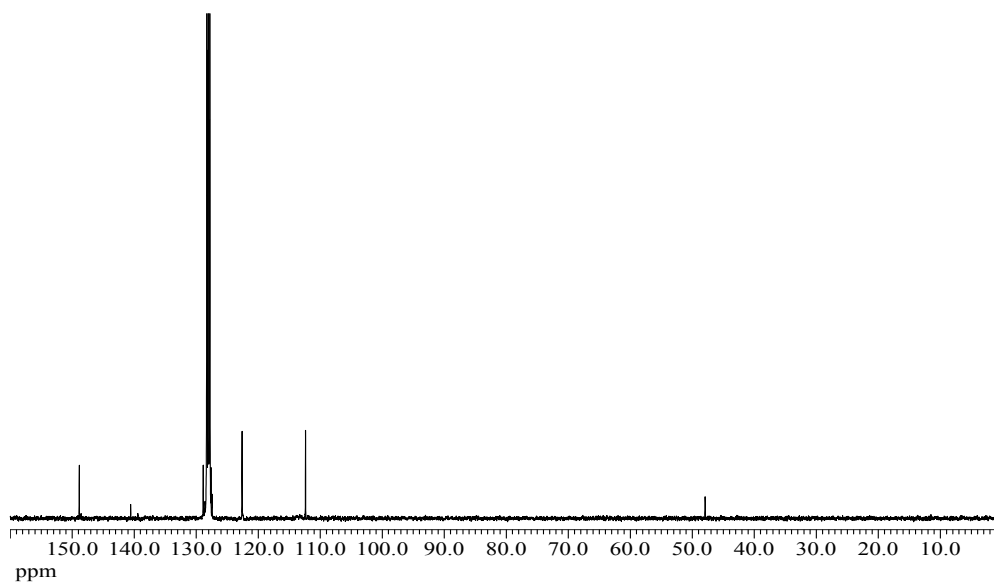
¹H NMR (C₆D₆, 25 °C): 7.40 (d, 2H, *J* = 7.2 Hz, Ar-*H*), 7.11 (t, 2H, *J* = 7.6 Hz, Ar-*H*), 7.05 (m, 1H, Ar-*H*), 6.99 (m, 4H, *Bcat*), 6.73 (m, 4H, *Bcat*), 4.55 (s, 2H, CH₂N). ¹¹B NMR (C₆D₆, 25 °C): 27.2 (bs, *Bcat*). ¹³C{¹H} NMR (C₆D₆, 25 °C): 148.8, 140.6, 139.4, 128.8, 127.4, 122.6, 112.3, 47.9.



¹H NMR:

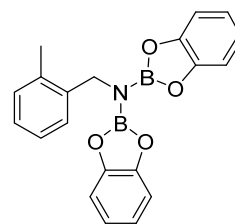


¹³C{¹H} NMR:

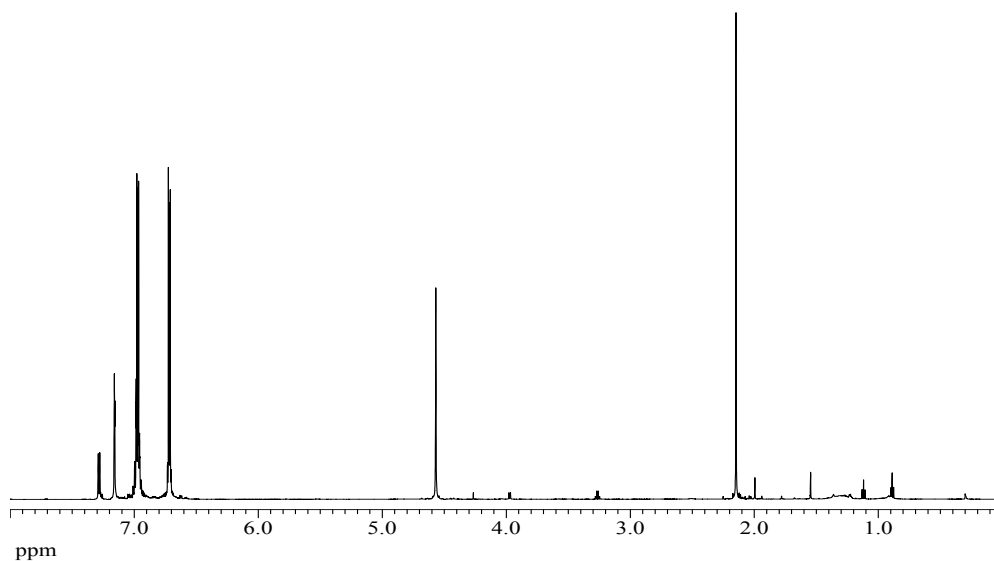


(*o*-tolyl)CH₂N(Bcat)₂

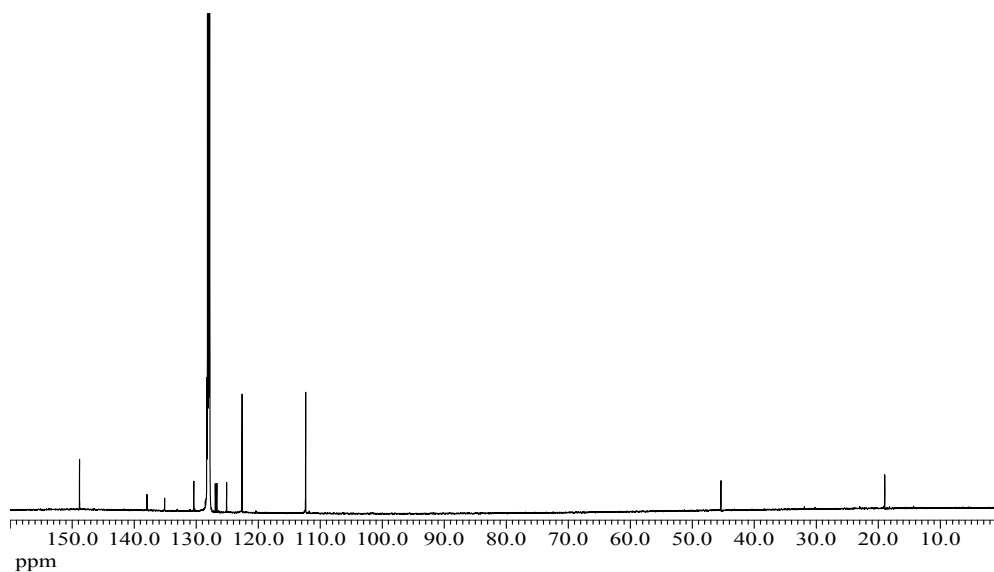
¹H NMR (C₆D₆, 25 °C): δ 7.28 (d, 2H, *J* = 8.0 Hz, Ar-*H*), 6.98 (m, 6H, Ar-*H* and Bcat), 6.88 (d, 1H, *J* = 7.6 Hz, Ar-*H*), 6.72 (m, 4H, Bcat), 4.57 (s, 2H, CH₂N), 2.15 (s, 3H, CH₃). ¹¹B NMR (C₆D₆, 25 °C): δ 27.1 (bs, Bcat). ¹³C{¹H} NMR (C₆D₆, 25 °C): δ 148.8, 138.0, 135.1, 130.4, 128.3, 126.9, 126.7, 125.1, 122.6, 112.4, 45.38, 19.0.



¹H NMR:

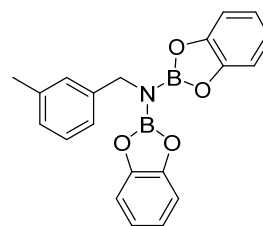


¹³C{¹H} NMR:

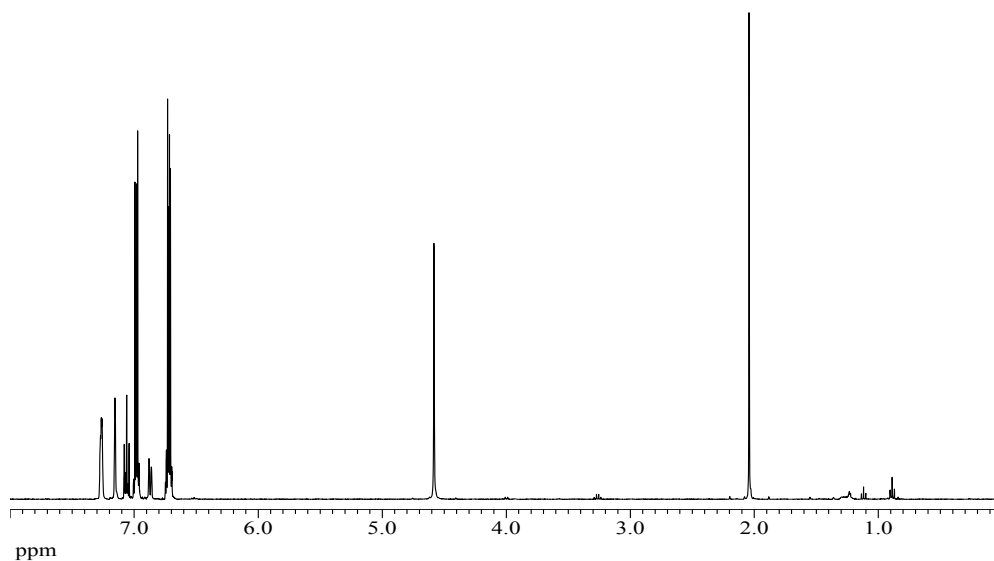


(m-tolyl)CH₂N(Bcat)₂

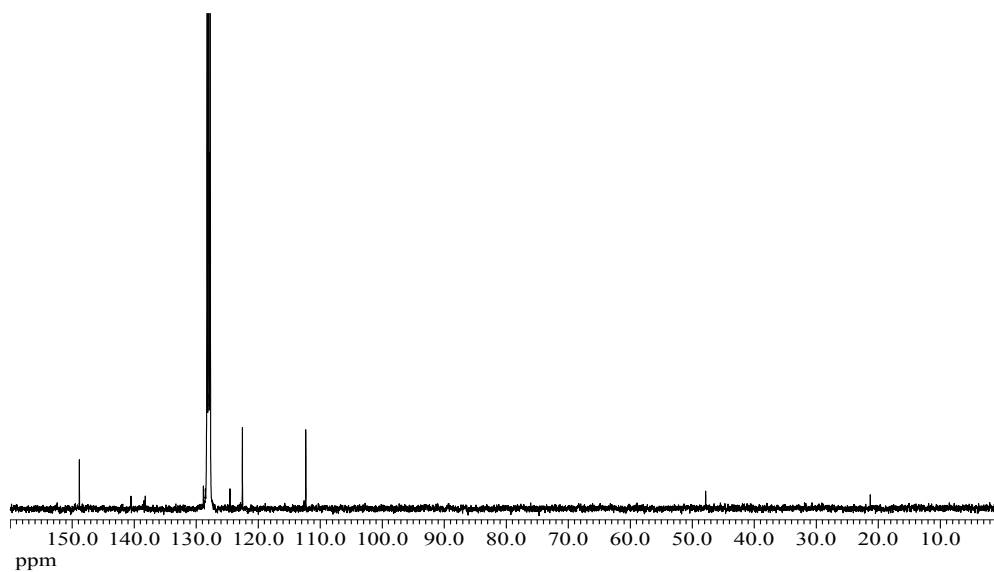
¹H NMR (C₆D₆, 25 °C): δ 7.27 (m, 2H, Ar-*H*), 7.11 (m, 1H, Ar-*H*), 6.99 (m, 4H, Bcat), 6.88 (d, 1H, *J* = 7.6 Hz, Ar-*H*), 6.72 (m, 4H, Bcat), 4.59 (s, 2H, CH₂N), 2.05 (s, 3H, CH₃). ¹¹B NMR (C₆D₆, 25 °C): δ 27.0 (bs, Bcat). ¹³C{¹H} NMR (C₆D₆, 25 °C): δ 148.8, 140.5, 138.2, 128.9, 127.9, 124.6, 122.6, 112.3, 47.9, 21.3.



¹H NMR:

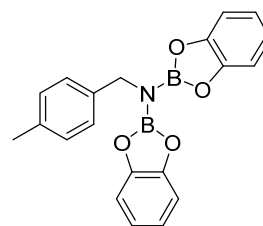


¹³C{¹H} NMR:

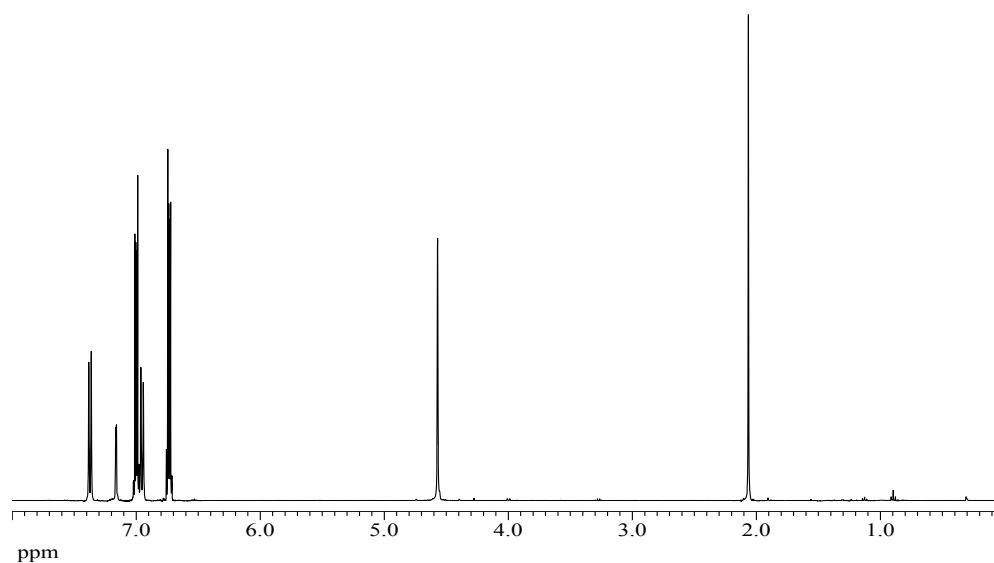


(p-tolyl)CH₂N(Bcat)₂

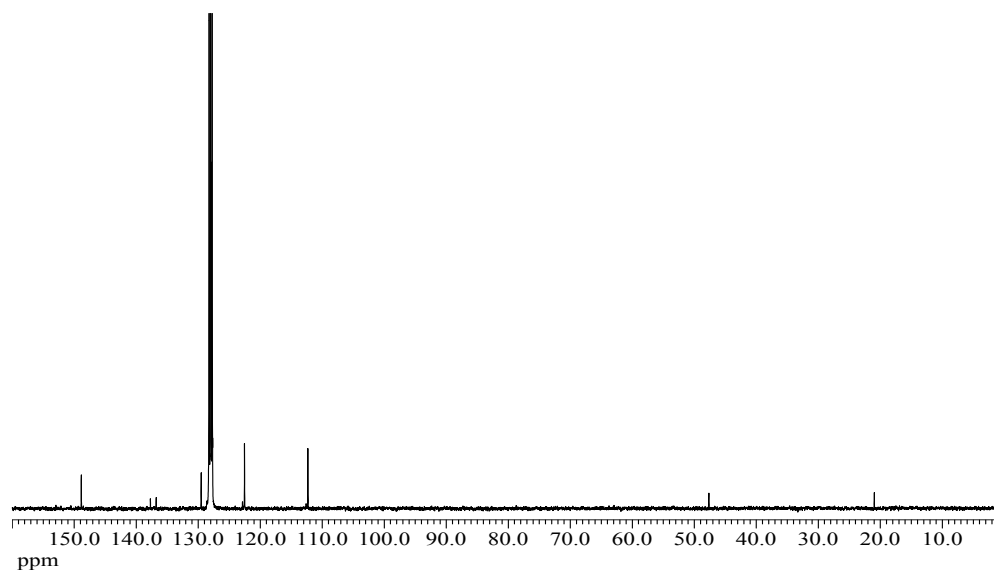
¹H NMR (C₆D₆, 25 °C): 7.37 (d, 2H, *J* = 8.0 Hz, Ar-*H*), 7.00 (m, 4H, Bcat), 6.95 (d, 2H, *J* = 8.0 Hz, Ar-*H*), 6.72 (m, 4H, Bcat), 4.57 (s, 2H, CH₂N), 2.06 (s, 3H, CH₃). ¹¹B NMR (C₆D₆, 25 °C): 27.1 (bs, Bcat). ¹³C{¹H} NMR (C₆D₆, 25 °C): 148.9, 137.7, 136.8, 129.5, 122.5, 112.3, 47.7, 21.0.



¹H NMR:

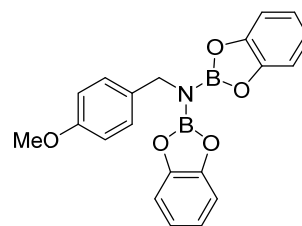


¹³C{¹H} NMR:

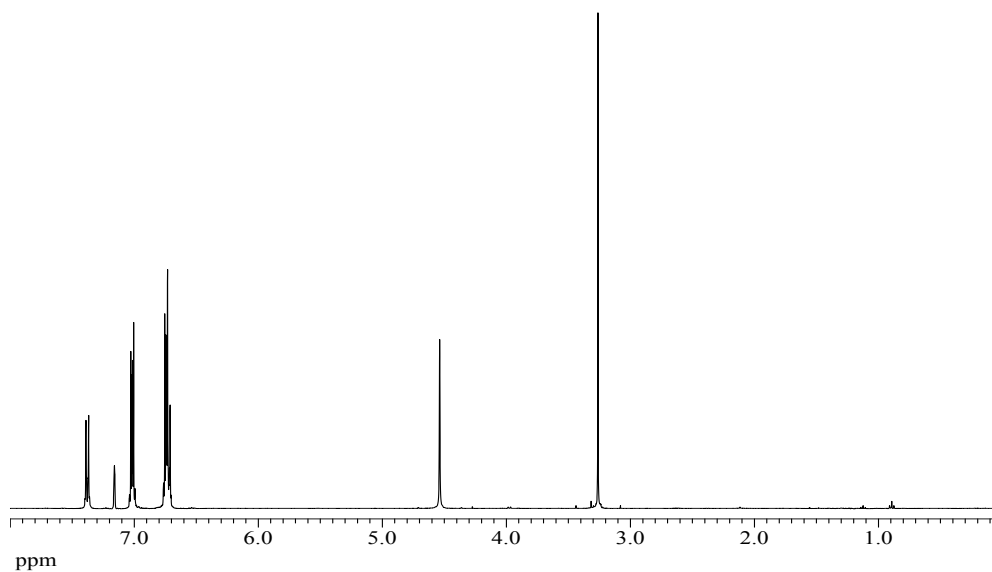


(p-MeOC₆H₄)CH₂N(Bcat)₂

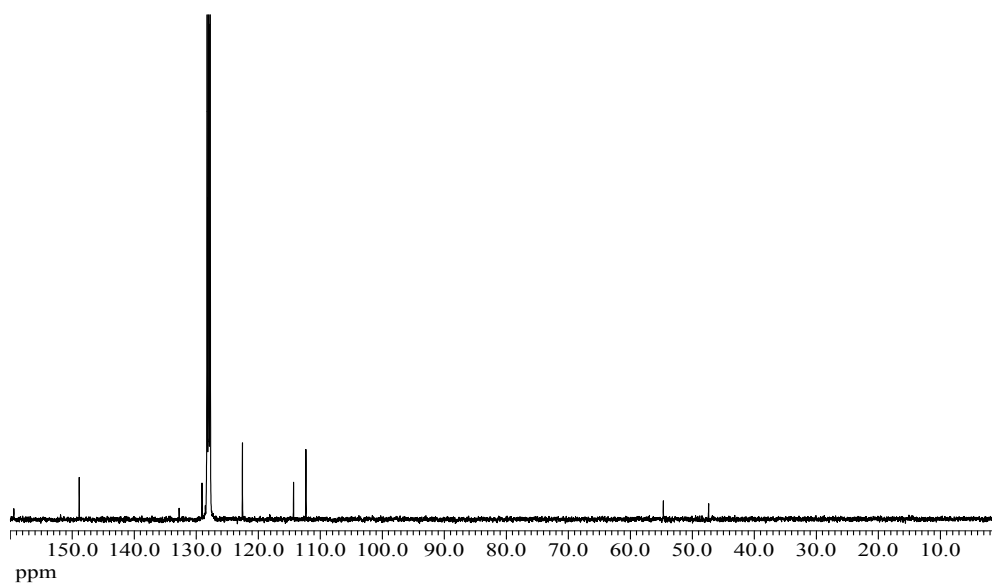
¹H NMR (C₆D₆, 25 °C): 7.38 (d, 2H, *J* = 8.8 Hz, Ar-*H*), 7.02 (m, 4H, Bcat), 6.73 (m, 6H, Ar-*H* and Bcat), 4.54 (s, 2H, CH₂N), 3.26 (s, 3H, OCH₃). ¹¹B NMR (C₆D₆, 25 °C): 27.2 (bs, Bcat). ¹³C{¹H} NMR (C₆D₆, 25 °C): 159.4, 148.9, 132.8, 136.8, 129.1, 122.6, 114.2, 112.3, 54.8, 47.4.



¹H NMR:

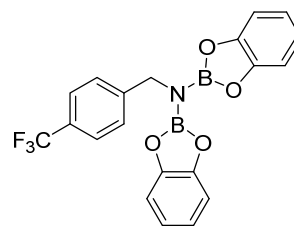


¹³C{¹H} NMR:

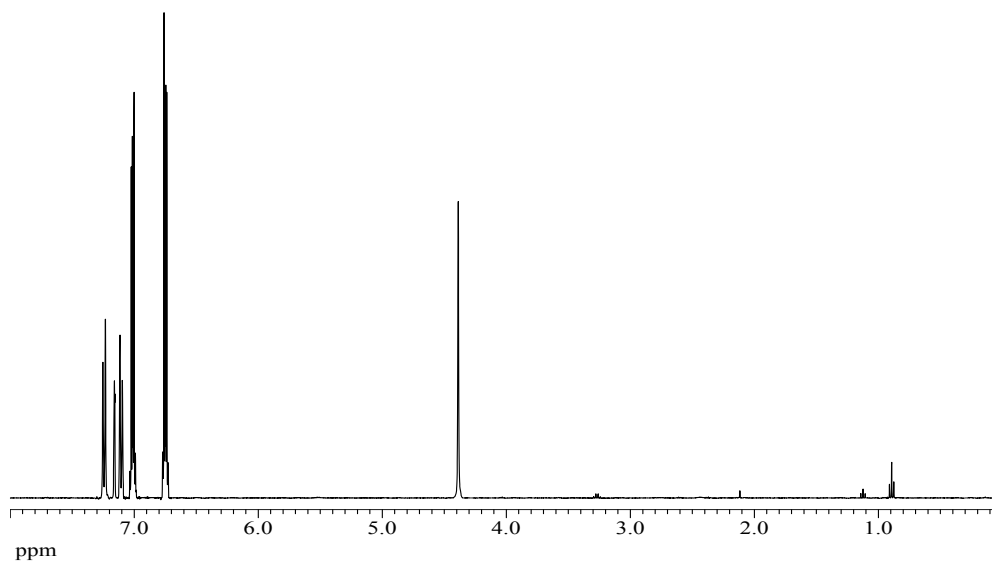


$(p\text{-F}_3\text{CC}_6\text{H}_4)\text{CH}_2\text{N}(\text{Bcat})_2$

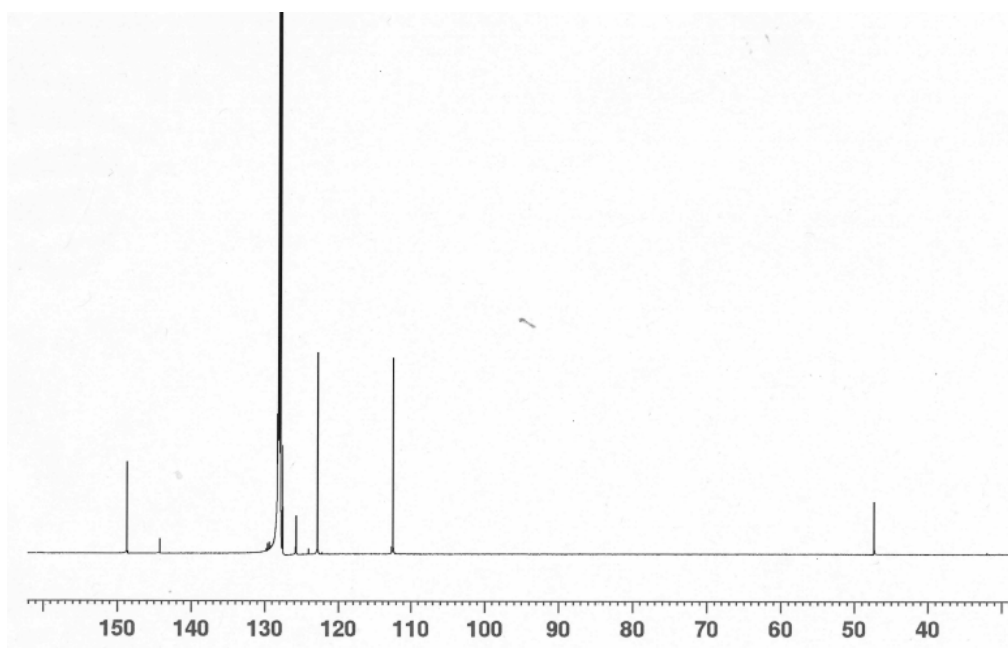
^1H NMR (C_6D_6 , 25 °C): 7.24 (d, 2H, $J = 8.0$ Hz, Ar- H), 7.11 (d, 2H, $J = 8.0$ Hz, Ar- H), 7.01 (m, 4H, Bcat), 6.75 (m, 4H, Bcat), 4.39 (s, 2H, CH_2N). ^{11}B NMR (C_6D_6 , 25 °C): 26.8 (bs, Bcat). $^{13}\text{C}\{^1\text{H}\}$ NMR (C_6D_6 , 25 °C): 148.7, 144.2, 136.8, 129.5, 125.8, 127.5, 122.8, 112.4, 47.3.



^1H NMR:

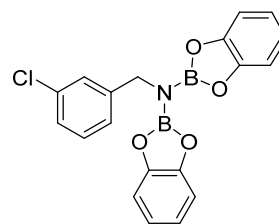


$^{13}\text{C}\{^1\text{H}\}$ NMR:

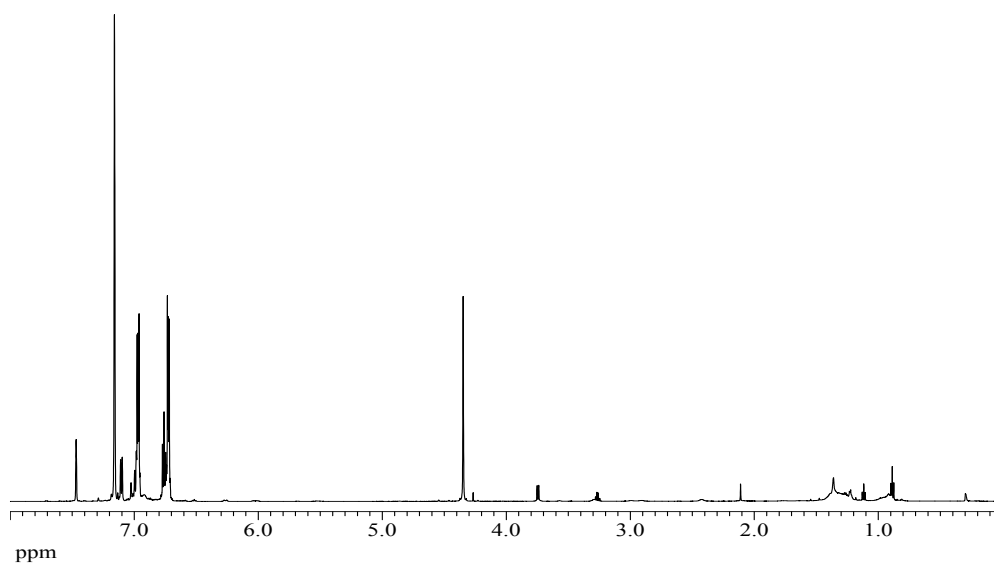


(m-ClC₆H₄)CH₂N(Bcat)₂

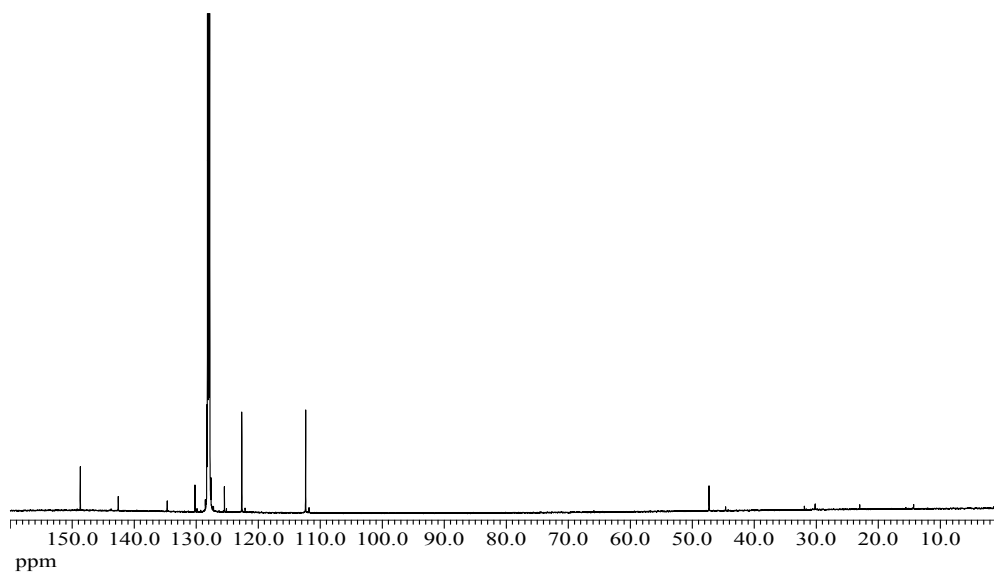
¹H NMR (C₆D₆, 25 °C): 7.47 (s, 1H, Ar-*H*), 7.10 (d, 1H, *J* = 8.0 Hz, Ar-*H*), 6.98 (m, 5H, Ar-*H* and Bcat), 6.74 (m, 5H, Ar-*H* and Bcat), 4.35 (s, 2H, CH₂N). ¹¹B NMR (C₆D₆, 25 °C): 26.9 (bs, Bcat). ¹³C {¹H} NMR (C₆D₆, 25 °C): 148.7, 142.6, 134.7, 130.2, 128.3, 125.5, 122.7, 112.4, 47.3.



¹H NMR:



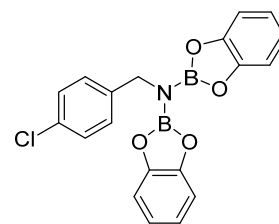
¹³C {¹H} NMR:



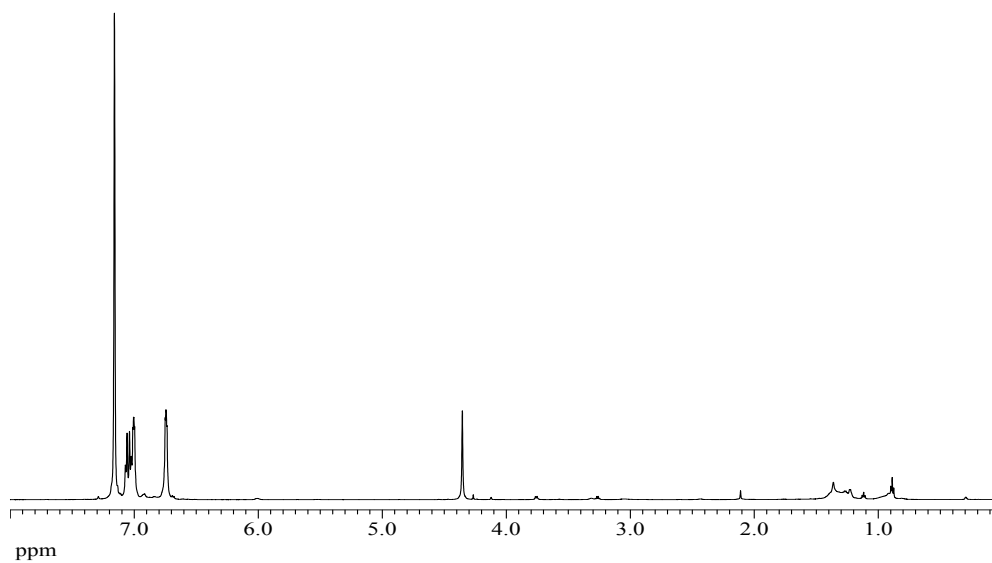
(p-ClC₆H₄)CH₂N(Bcat)₂

¹H NMR (C₆D₆, 25 °C): 7.04 (m, 8H, Ar-*H* and *Bcat*), 6.74 (m, 4H, *Bcat*), 4.35 (s, 2H, CH₂N). ¹¹B NMR (C₆D₆, 25 °C): 26.9 (bs, *Bcat*).

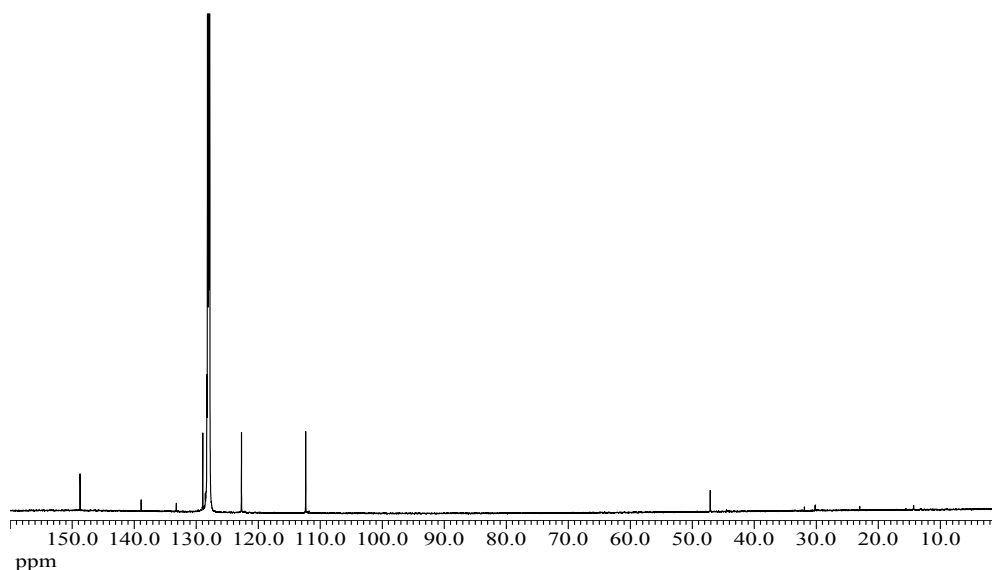
¹³C{¹H} NMR (C₆D₆, 25 °C): 148.7, 138.9, 133.2, 129.0, 128.3, 122.7, 112.4, 47.1.



¹H NMR:



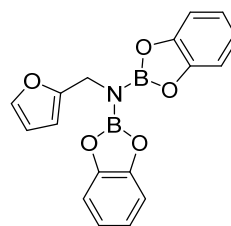
¹³C{¹H} NMR:



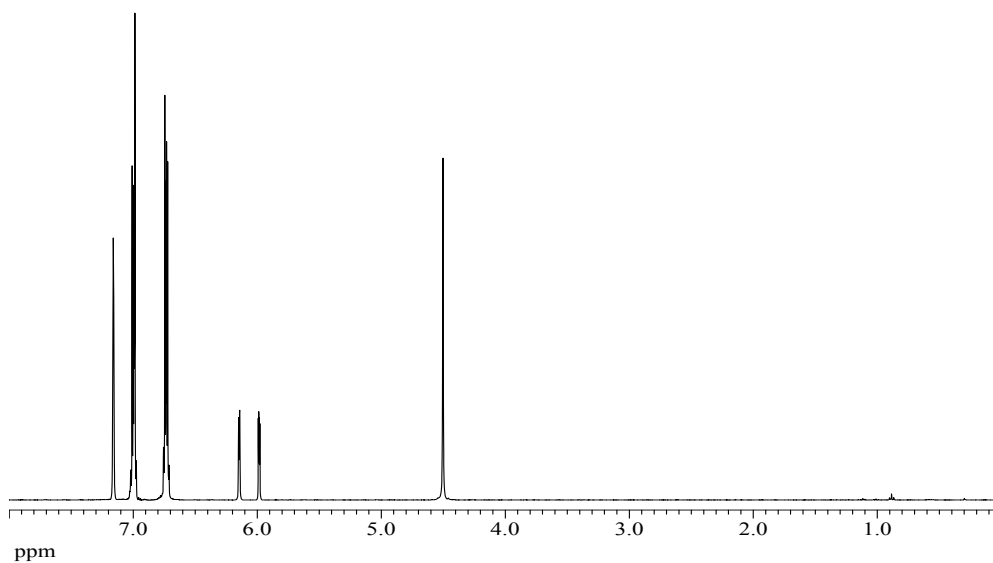
(2-furyl)CH₂N(Bcat)₂

¹H NMR (C₆D₆, 25 °C): 7.00 (m, 5H, furyl-*H* and Bcat), 6.73 (m, 4H, Bcat), 6.15 (d, 1H, *J* = 3.2 Hz, furyl-*H*), 5.98 (m, 1H, furyl-*H*), 4.50 (s, 2H, CH₂N).

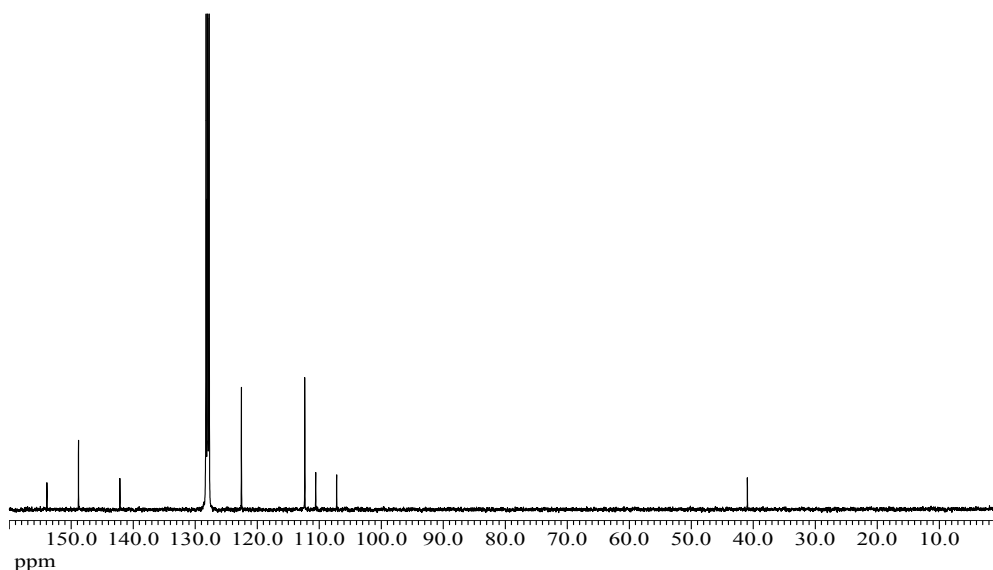
¹¹B NMR (C₆D₆, 25 °C): 27.0 (bs, Bcat). ¹³C{¹H} NMR (C₆D₆, 25 °C): 159.9, 148.8, 142.2, 122.6, 112.3, 110.6, 107.2, 41.0.



¹H NMR:

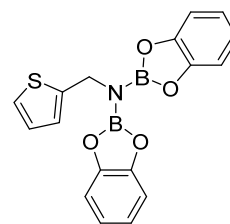


¹³C{¹H} NMR:

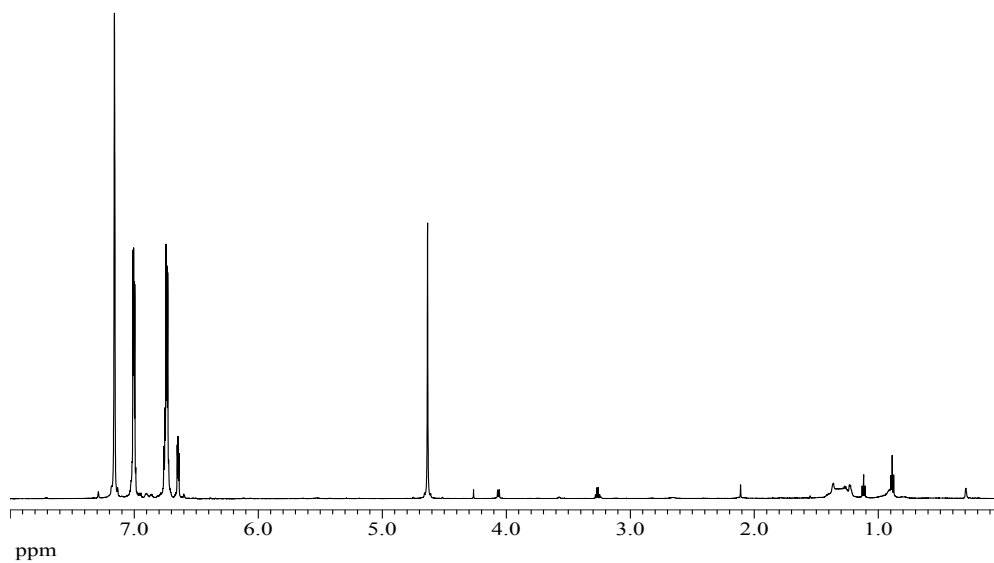


(2-thienyl)CH₂N(Bcat)₂

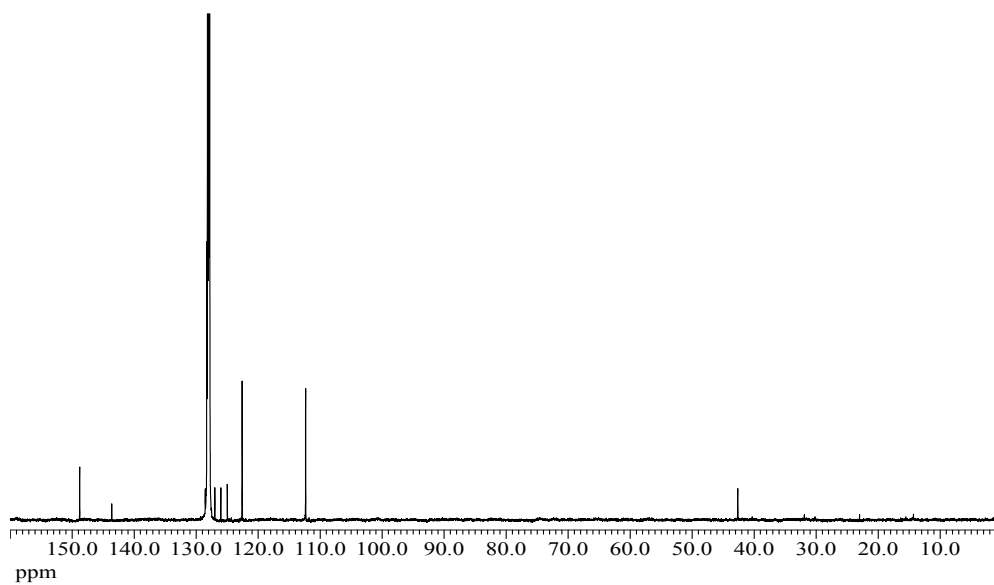
¹H NMR (C₆D₆, 25 °C): 7.00 (m, 4H, Bcat), 6.74 (m, 6H, thienyl-*H* and Bcat), 6.65 (m, 1H, thienyl-*H*), 4.63 (s, 2H, CH₂N). ¹¹B NMR (C₆D₆, 25 °C): 26.8 (bs, Bcat). ¹³C{¹H} NMR (C₆D₆, 25 °C): 148.8, 143.6, 127.0, 126.0, 125.0, 122.6, 112.4, 42.7.



¹H NMR:



¹³C{¹H} NMR:



Reaction of 3 with HBcat. A J-young NMR tube was charged with a C₆D₆ solution (0.4 mL) of **3** (5.0 mg, 0.012 mmol) and HBcat (5.5 mg, 0.046 mmol). After 15 min at room temperature, formation of black precipitate was observed. Formation of [tBuCOCHC(tBu)O-κO,κO']Bcat was supported by both ¹H and ¹¹B{¹H} NMR although all the signals appeared as significantly broad signals due to the existence of paramagnetic nickel species. In addition, formation of several unidentified products, which exhibit broad signals as δ 28.9, 23.3, 18.7 ppm in the ¹¹B NMR spectrum, was also confirmed. After filtration, slow evaporation of the filtrate afforded single crystals of [tBuCOCHC(tBu)O-κO,κO']Bcat.

[tBuCOCHC(tBu)O-κO,κO']Bcat: ¹H NMR (C₆D₆, 25 °C): 7.11 (dd, 2H, *J* = 5.4, 3.6 Hz, *Bcat*), 6.84 (dd, 2H, *J* = 5.4, 3.6 Hz, *Bcat*), 5.82 (s, 1H, CH), 0.84 (s, 18H, *t*Bu). ¹¹B NMR (C₆D₆, 25 °C): 9.7 (bs, *Bcat*). ¹³C{¹H} NMR (C₆D₆, 25 °C): 201.3, 151.9, 120.1, 110.2, 92.9, 39.5, 26.9.

Single-crystal X-ray diffraction studies. The single crystal X-ray diffraction measurements of [tBuCOCHC(tBu)O-κO,κO']Bcat was performed under a cold nitrogen stream on a Rigaku XtaLAB P200 diffractometer with a Pilatus 200K detector using multi-layer mirror monochromated Mo Kα radiation. The determination of crystal systems and unit cell parameters and data processing were performed with the *CrystalClear* program package. The data sets were corrected for Lorentz and polarization effects and absorption. The structure was solved by direct methods using SIR97 program,² and refined by full-matrix least squares calculations on *F*² for all reflections (SHELXL-97)³. The structure was not fully refined due to the bad quality of the crystal and the final R values remain 0.1208 (*R*₁) and 0.2771 (*wR*₂).

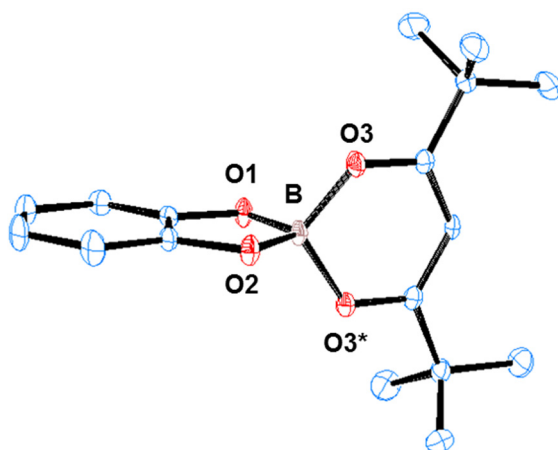


Figure 1. Molecular structure of [tBuCOCHC(tBu)O-κO,κO']Bcat with 50% probability ellipsoids. Hydrogen atoms are omitted for clarity. Selected bond distances (Å) and angles (deg): B–O1 1.463(3), B–O2 1.452(3), B–O3 1.490(2), O1–B–O2 107.1(2), O3–B–O3* 108.9(2).

Table S1. Crystal data and details of the crystal structure determination for *t*BuCOCHC(*t*Bu)OBcat.

formula	C ₁₇ H ₂₃ O ₄ B	Z	4
fw	302.17	<i>D</i> _{calcd} (g/cm ³)	1.200
T (K)	93(2)	<i>R</i> 1, <i>wR</i> 2 [<i>I</i> > 2σ(<i>I</i>)]	0.1208, 0.2771
cryst system	Orthorhombic	<i>R</i> 1, <i>wR</i> 2 (all data)	0.1386, 0.2945
space group	<i>P</i> bnm (#62)	GOF	1.371
<i>a</i> (Å)	9.246(3)		
<i>b</i> (Å)	12.853(4)		
<i>c</i> (Å)	14.074(4)		
α (deg)	90		
β (deg)	90		
γ (deg)	90		
<i>V</i> (Å ³)	1672.5(9)		

References

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- 3) Sheldrick, G. M. *Acta Cryst.* **2008**, A64, 112.