2 0 / degree

Supporting Information

The evolution of Fe phases for fused iron catalysts during the



reduction and Fischer-Tropsch synthesis



2 0 / degree



Figure. S2. Fe⁰ crystalline size in the catalysts after reduction (Reduction condition: P=0.2MPa, 10K for GHSV=10000h⁻¹, 6K for GHSV=6000h⁻¹, 2K for GHSV=2000h⁻¹,)



Figure. S3. X-ray diffraction (XRD) patterns of the catalysts after 48h FTS (Reduction condition: P=0.2MPa, (a) GHSV=6000h⁻¹, (b) GHSV=2000h⁻¹)



Figure. S4. X-ray diffraction (XRD) patterns of the catalysts after 500h FTS (Reduction condition: P=0.2MPa, (a) GHSV=6000h⁻¹, (b) GHSV=2000h⁻¹)



Fig. S5. CO+H₂ conversion versus time on stream for fused iron catalysts reduced under different temperature





Fig. S6. CH₄ selectivity versus time on stream for fused iron catalysts reduced under different temperature



(Reduction condition: P=0.2MPa, (a) GHSV=6000h⁻¹, (b) GHSV=2000h⁻¹)

Fig. S7. C₅+ selectivity versus time on stream for fused iron catalysts reduced under different temperature



(Reduction condition: P=0.2MPa, (a) GHSV=6000h-1, (b) GHSV=2000h-1)

Fig. S8. C₂₋₄ olefin to paraffin ratio versus time on stream for fused iron catalysts reduced under different temperature

(Reduction condition: P=0.2MPa, (a) GHSV=6000h⁻¹, (b) GHSV=2000h⁻¹)

H₂ adsorption



Fig.S9. Evolution of different iron phases over fused iron catalyst reduced at 6000 GHSV during reduction and reaction (The content of individual phase is calculated by XRD)



Fig.S10. Evolution of different iron phases over fused iron catalyst reduced at 2000 GHSV during reduction and reaction (The content of individual phase is calculated by XRD)