Supporting Information for

Participation of Interfacial Hydroxyl Groups in the Water-Gas Shift

Reaction over Au/MgO Catalysts

Yanran Cui^a, Zhenglong Li^{a,b}, Zhijian Zhao^a, Viktor J. Cybulskis^a, Kaiwalya D. Sabnis^a, Chang Wan Han^c, Volkan Ortalan^c, William F. Schneider^d, Jeffrey Greeley^a, W. Nicholas Delgass^a, and Fabio H. Ribeiro^a

^a Charles D. Davidson School of Chemical Engineering, Purdue University, West Lafayette, IN 47907(USA)

^b Energy and Transportation Sciences Division, Oak Ridge National Laboratory, Oak Ridge, TN 37831 (USA)

^c School of Materials Engineering, Purdue University, West Lafayette, IN 47907(USA)

^d Department of Chemical and Biomolecular Engineering, University of Notre Dame, Notre Dame, IN 46556 (USA)

[§] Co-first authors, ^{§§} Co-corresponding authors

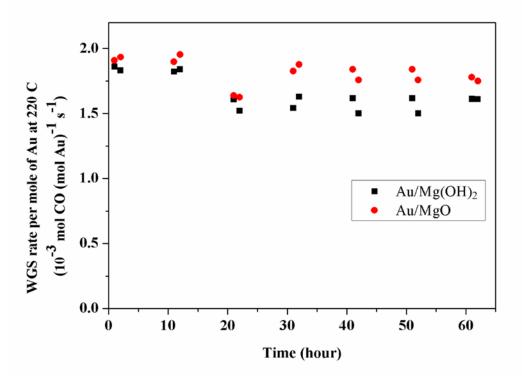


Figure S1. Deactivation plot during the measurement of WGS kinetics (standard conditions, 6.8% CO, 21.9% H₂O, 8.5% CO₂, 37.4% H₂, and balance Ar).

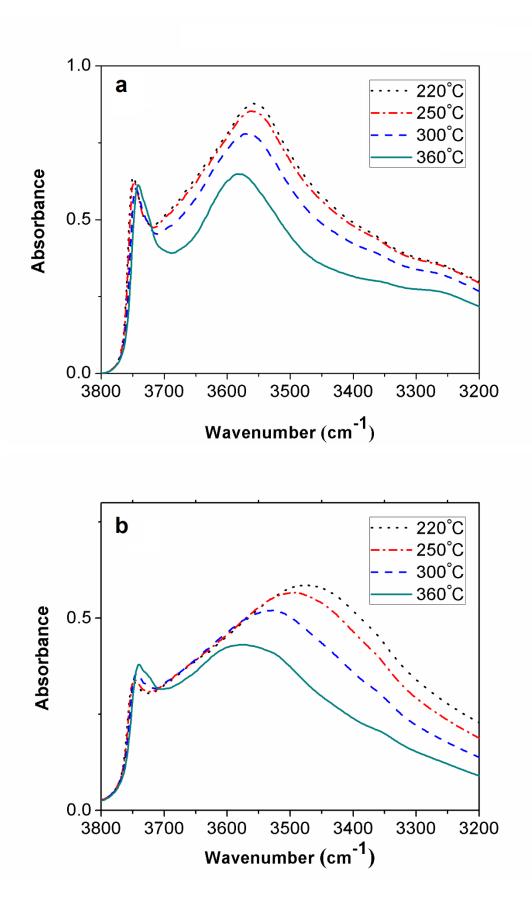


Figure S2. N₂ TPD (ramp rate: 5 °C min⁻¹) after exposure to 11% H₂O/N₂: a) Au/MgO b) MgO Support. Both samples were pretreated by reduction in 25% H₂/N₂ at 150 °C for 2 hours (flow rate: 50 ml min⁻¹, ramp rate: 2 °C min⁻¹) followed by annealing in N₂ at 400 °C for 2 hours (flow rate: 50 ml min⁻¹, ramp rate: 5 °C min⁻¹).

Table S1. Simulated relative WGS rates by manually tuned (D-OD)*# binding energy

$\Delta BE^{a} / eV$	0	+0.02	+0.03	+0.04
$r_{ m H}/r_{ m D}$	1.02	1.42	1.77	2.22

^a: Manually tuned (D-OD)*# binding energy (eV). Positive number indicates weakening (D-OD)*# binding strength.