

Electronic Supplementary Information for

Efficient chromium-based catalysts for ethylene tri-/tetramerization switched by silicon-bridged/N, P-based ancillary ligands: A structural, catalytic and DFT study

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Table S1. The data of molecular orbital energies for analysis.

Entry	Cyclic intermediates	Energies of LUMO+1 (kcal/mol)	Energy gaps (kcal/mol)
1	C1a	-99.05	68.78
2	C2b	-94.31	73.51

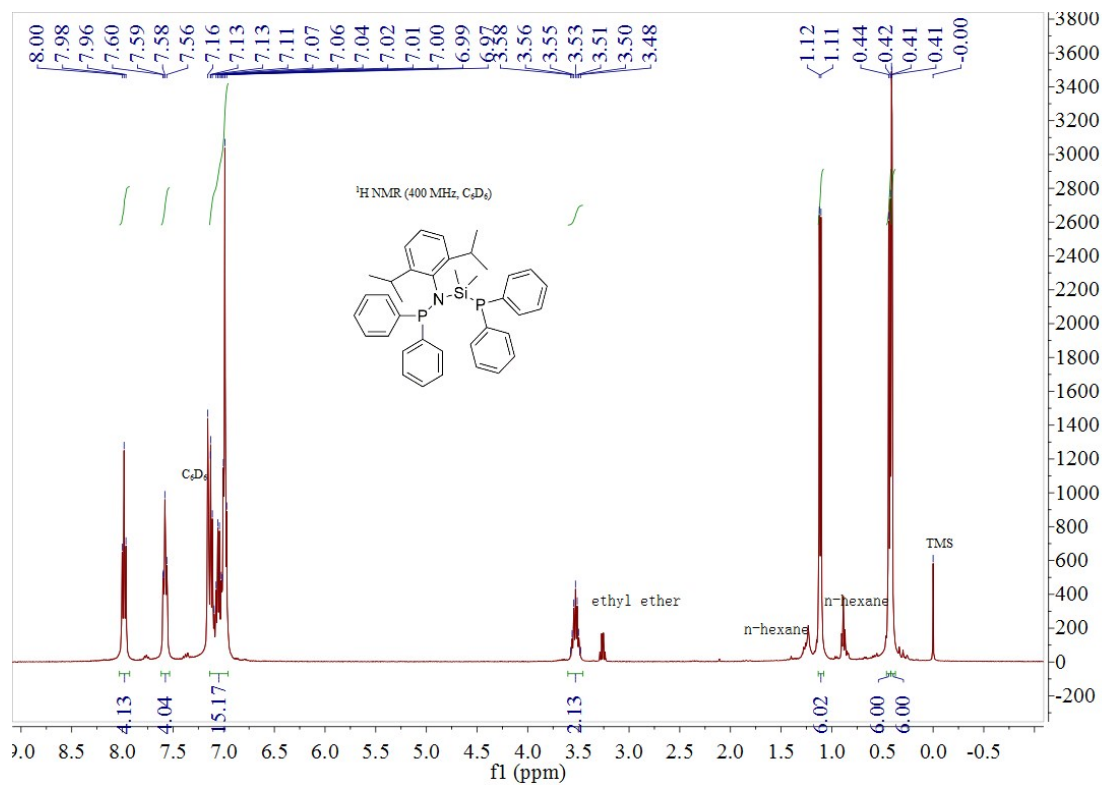


Figure S1. ¹H NMR spectrum of L1 (400MHz, C₆D₆)

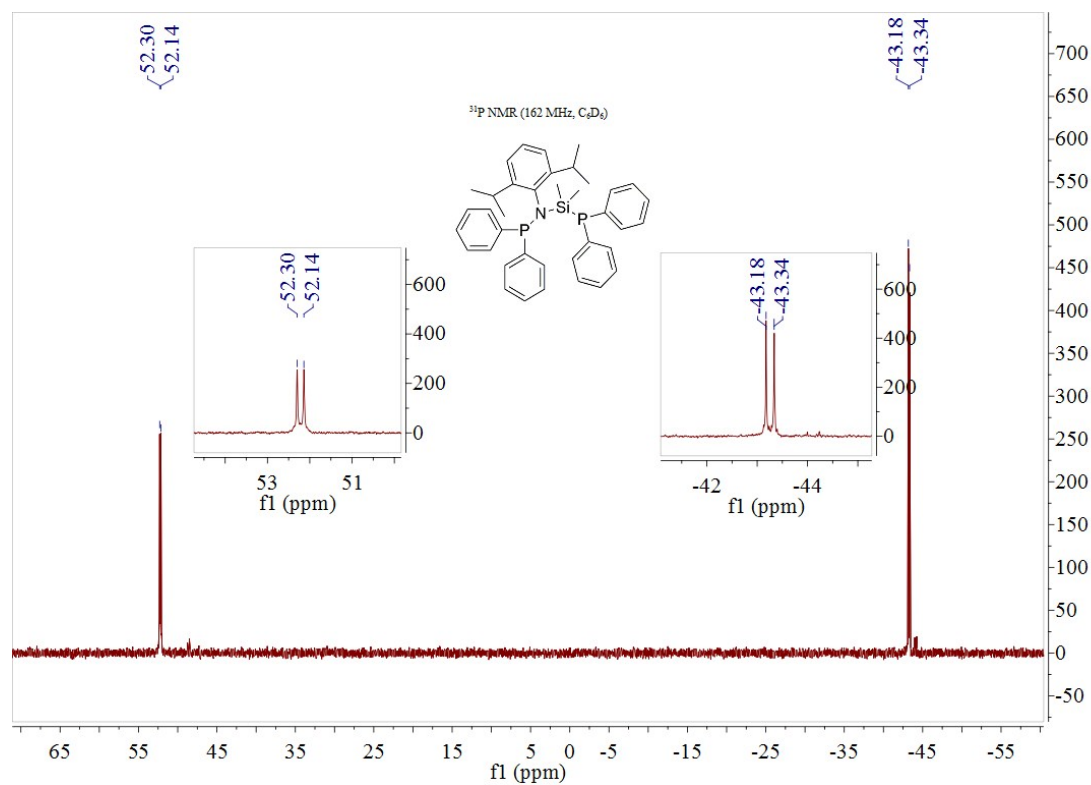


Figure S2. ³¹P NMR spectrum of L1 (162MHz, C₆D₆)

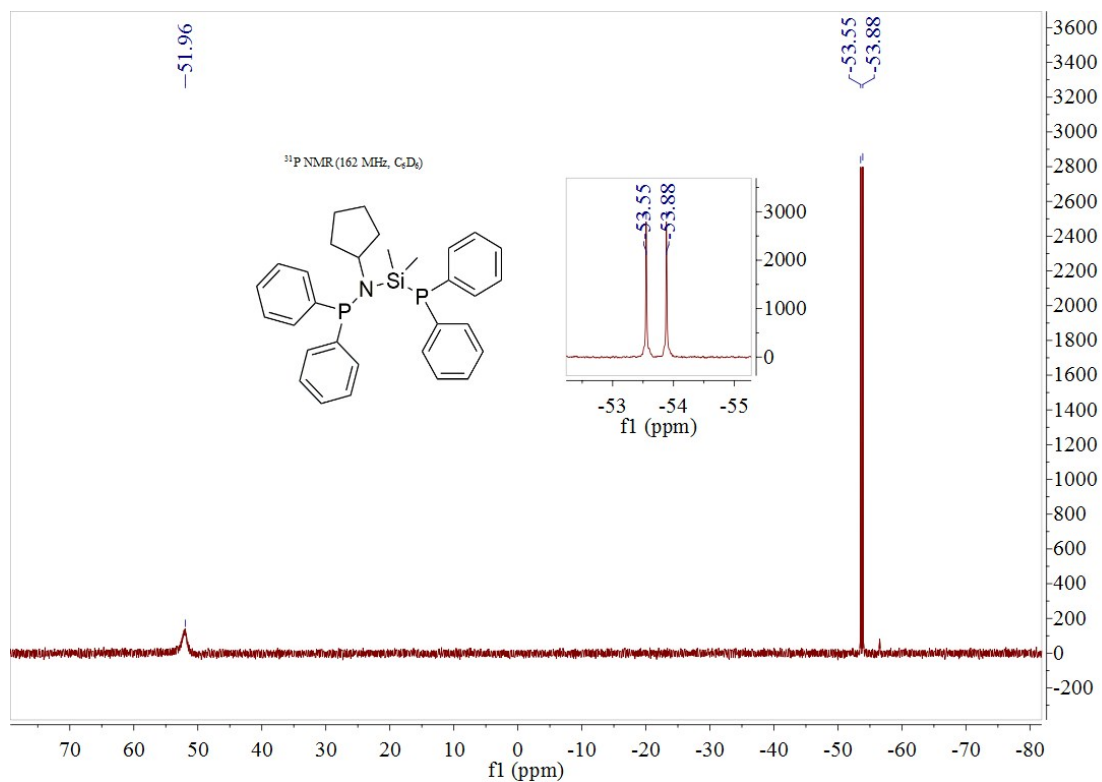


Figure S5. ³¹P NMR spectrum of L2 (162MHz, C₆D₆)

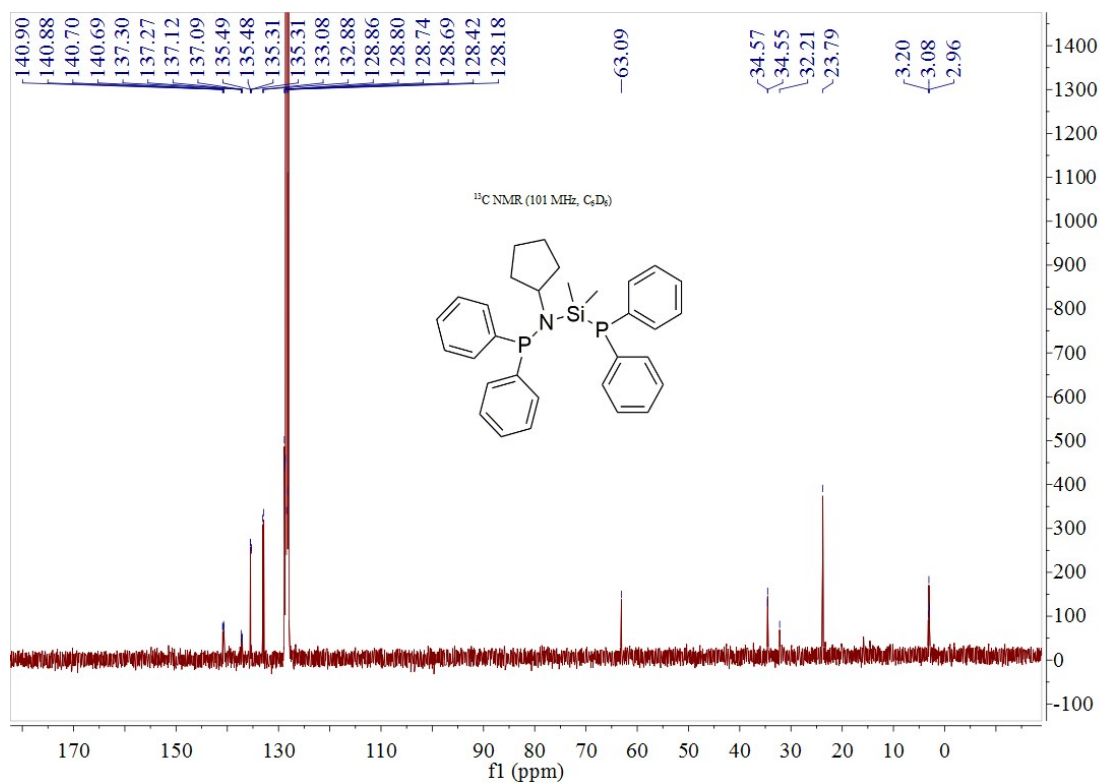


Figure S6. ¹³C NMR spectrum of L2 (101MHz, C₆D₆)

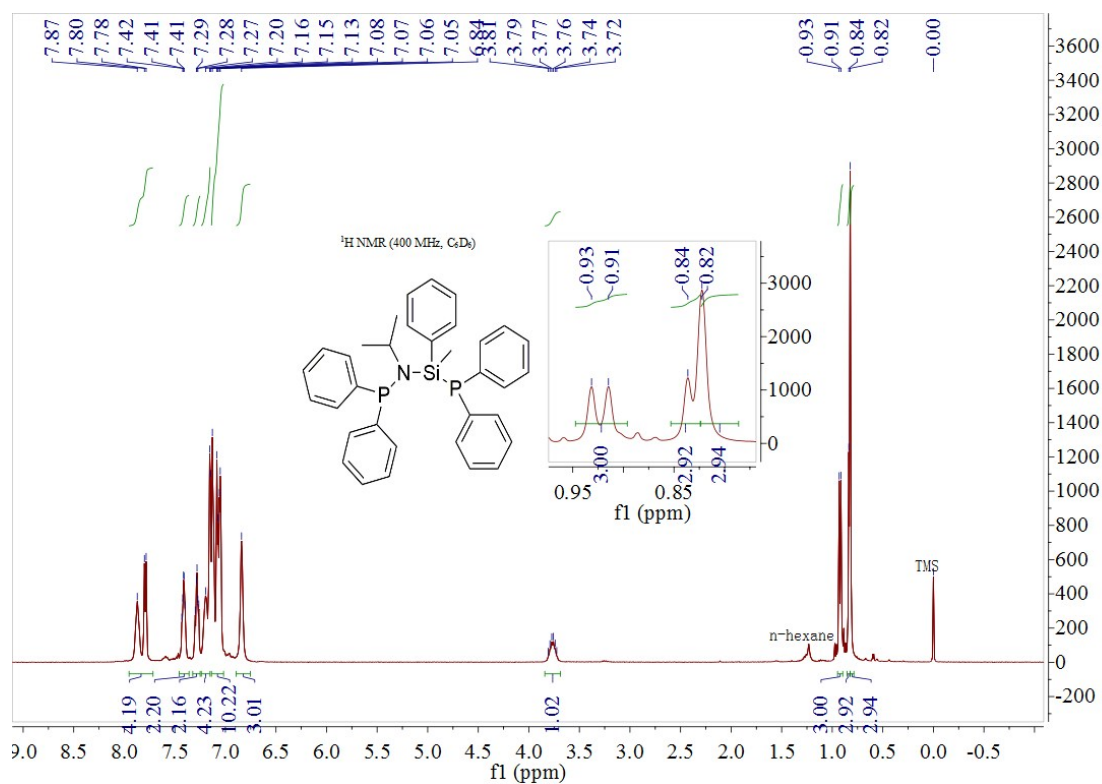


Figure S7. ¹H NMR spectrum of L3 (400MHz, C₆D₆)

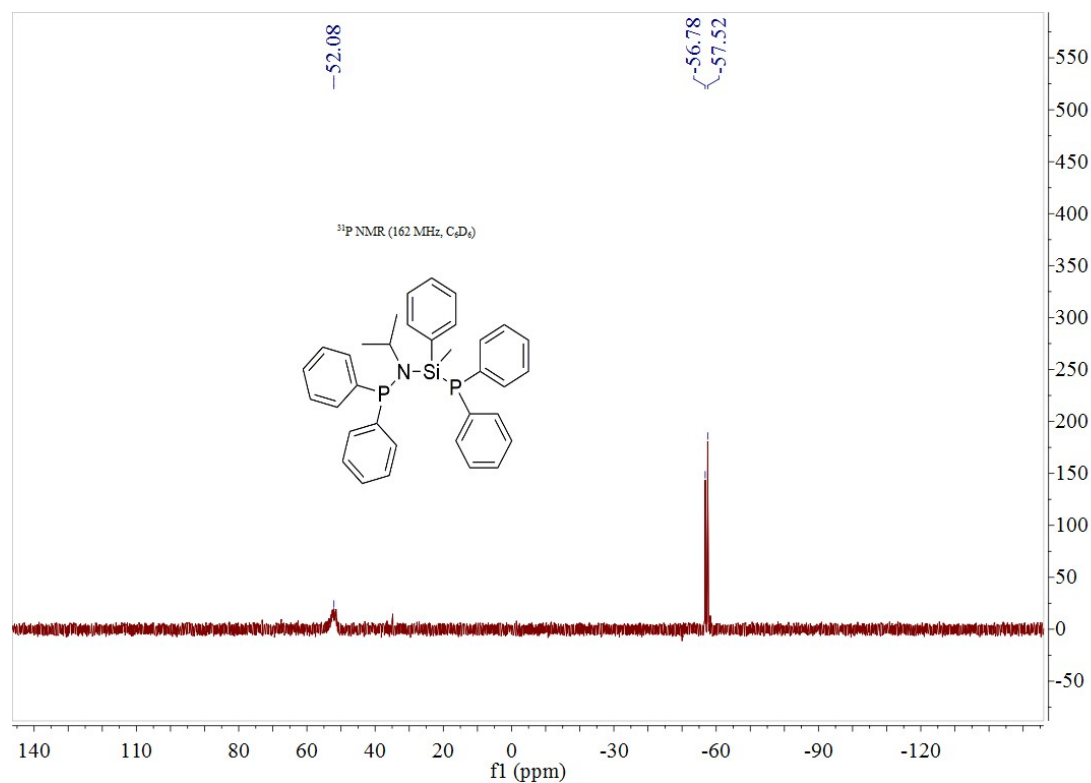


Figure S8. ³¹P NMR spectrum of L3 (162MHz, C₆D₆)

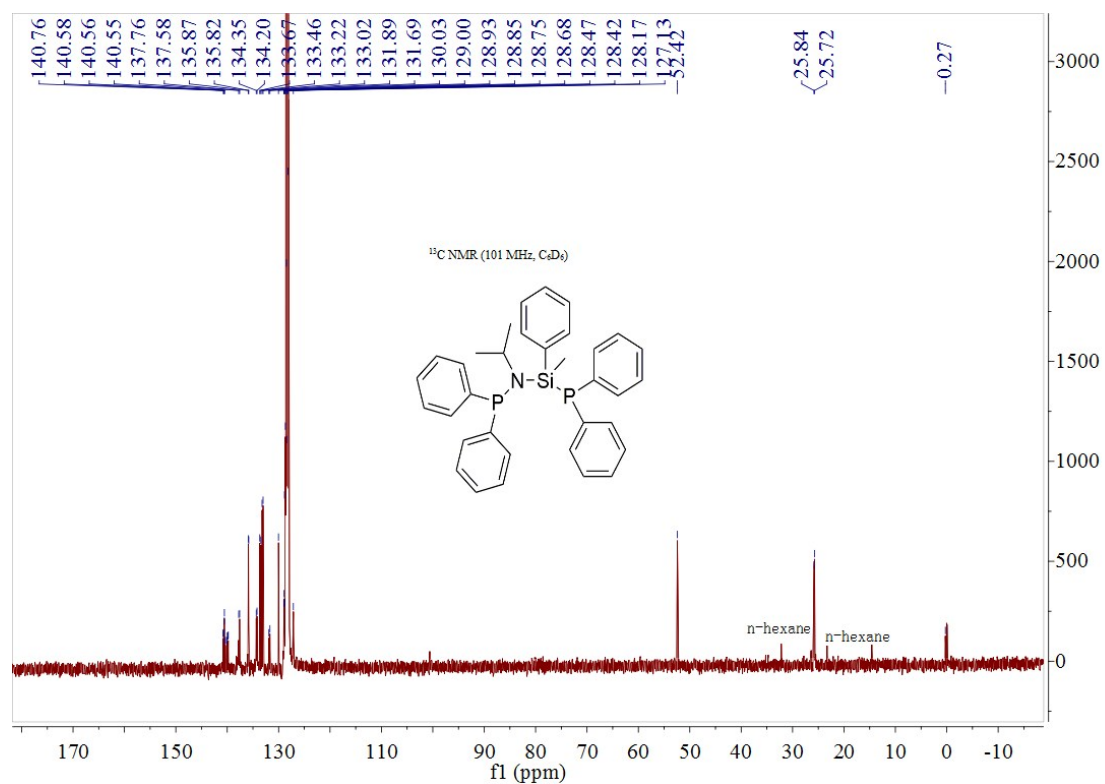


Figure S9. ¹³C NMR spectrum of L3 (101MHz, C₆D₆)

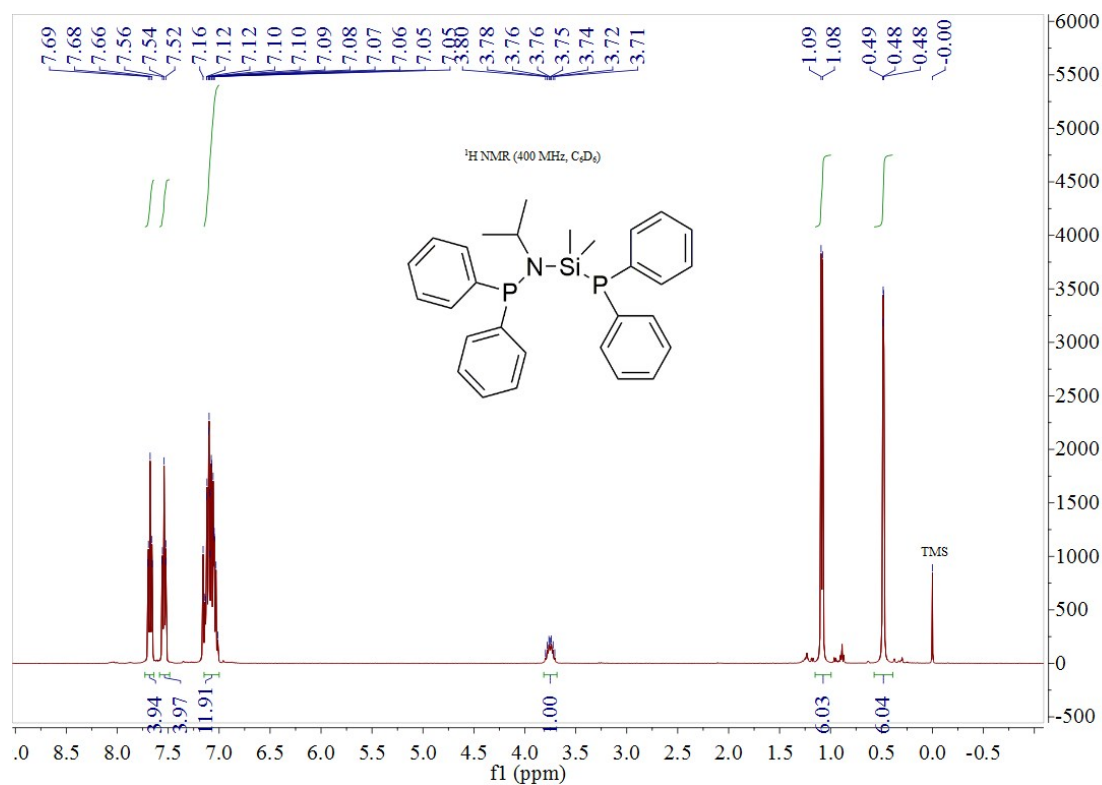


Figure S10. ¹H NMR spectrum of L4 (400MHz, C₆D₆)

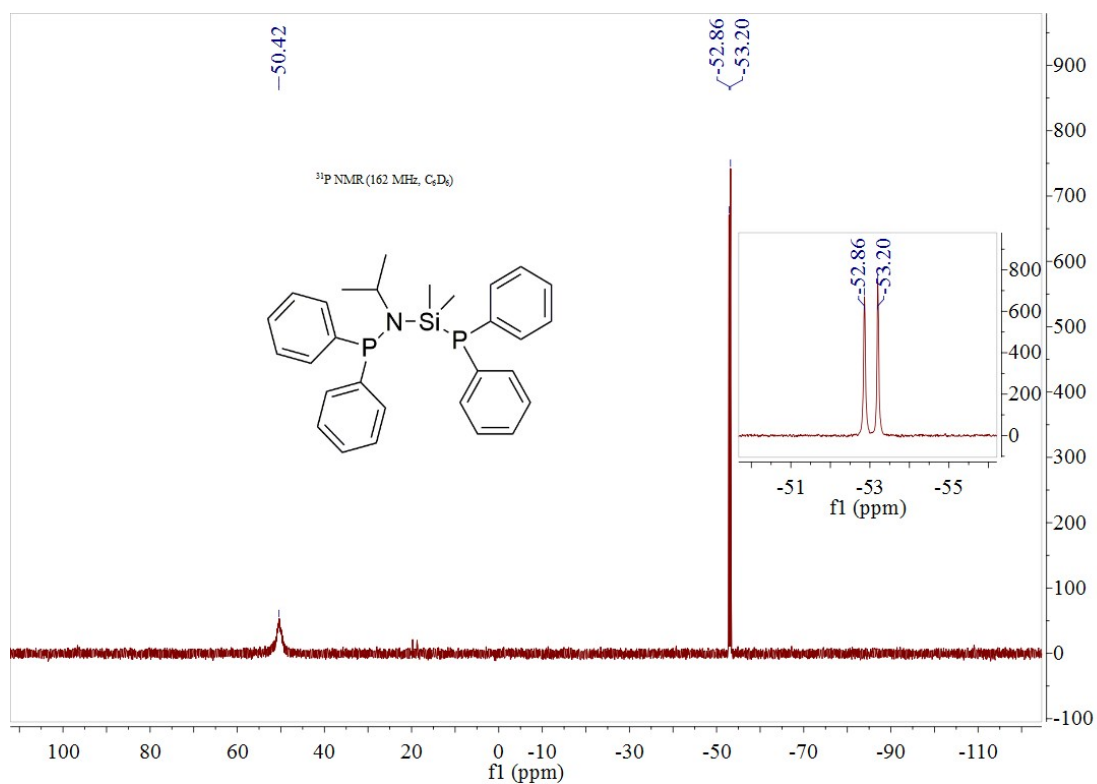


Figure S11. ³¹P NMR spectrum of L4 (162MHz, C₆D₆)

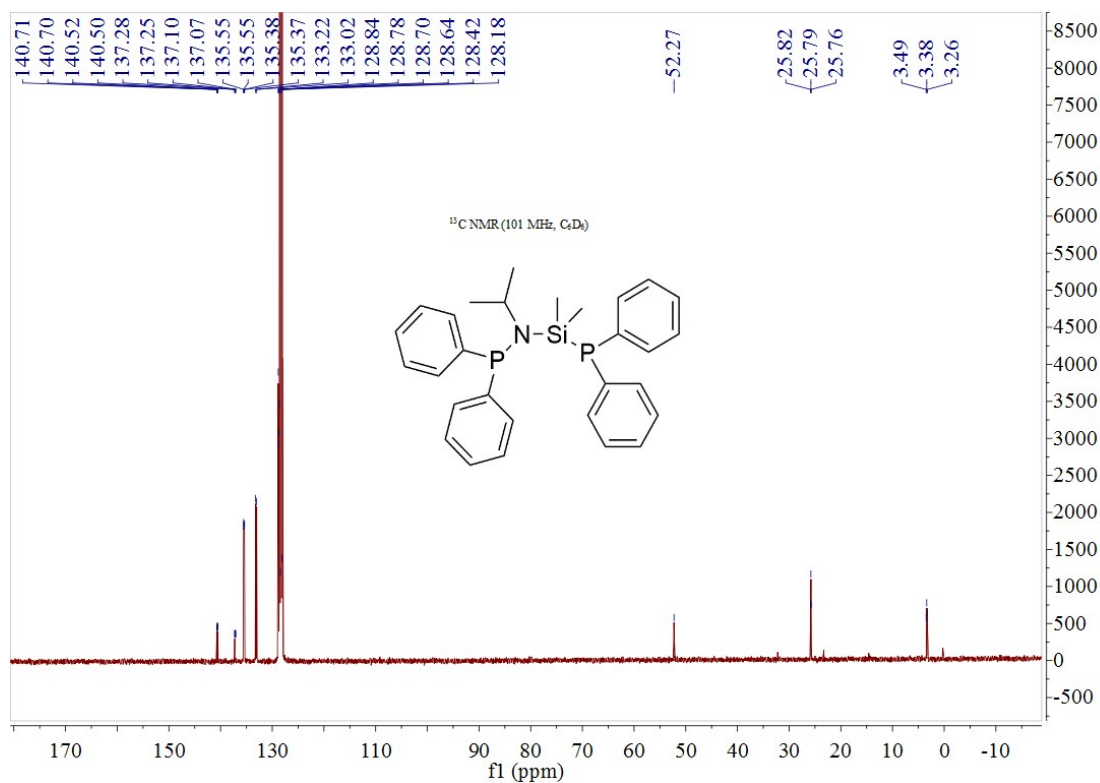


Figure S12. ¹³C NMR spectrum of L4 (101MHz, C₆D₆)

Table S1. Absolute energies in hartrees of the intermediates

	Intermediates					
	C1a	C1b	C1c	C2a	C2b	C2c
Absolute energies (hartrees)	-2744.33438	-2822.96657	-2901.58220	-2472.75898	-2551.42484	-2630.01498