## **Supporting information**

## Seed-induced and additive-free synthesis of oriented nanorodassembled meso/macroporous zeolites: toward efficient and costeffective catalysts for the MTA reaction

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**Figure S1.** SEM images of (a, c) C-ZSM-5 and (b, d) C-ZSM-11 under different magnification. For comparisons, conventional ZSM-5 and ZSM-11 (denoted as C-ZSM-5 and C-ZSM-11, respectively) were also synthesized by using TPABr or TBABr as the templates under seed-free crystallization conditions. The molar composition of the synthesis mixture for C-ZSM-5 and C-ZSM-11 were 3.5 NaOH :  $60 \text{ SiO}_2 : 4 \text{ NaAlO}_2 : 2500 \text{ H}_2\text{O} : 8 \text{ TPABr}$  and 3.5 NaOH :  $60 \text{ SiO}_2 : 4 \text{ NaAlO}_2 : 2500 \text{ H}_2\text{O} : 8 \text{ TBABr}$ , respectively.



**Figure S2.** SEM images of silicalite-1 (a) and silicalite-2 seeds (b). It was obvious that the as-prepared silicalite-1 and silicalite-2 seeds had similar particle size of about 150 nm but with slightly different crystal morphologies of near-sphere shape and olive shape, respectively.



**Figure S3.** Low-resolution TEM images of (a) N-ZSM-5 and (b) N-ZSM-11, which showed the existence of zeolite nanorods that assembled into hedgehog-shaped submicron particles, in which substantial mesopores were formed due to the stacking of these nanorods, resulting in quite different structure properties from C-ZSM-5 and C-ZSM-11.



Figure S4. Nitrogen adsorption/desorption isotherms of Zn/N-ZSM-5 and Zn/N-ZSM-11.



**Figure S5.** TG curves of Zn/N-ZSM-5, Zn/N-ZSM-11, Zn/C-ZSM-5, and Zn/C-ZSM-11 after MTA reaction tested at 748 K ( $W_{cat} = 0.7$  g; GHSV = 0.75 h<sup>-1</sup>). Clearly, both of Zn/N-ZSM-5 and Zn/N-ZSM-11 have much slower coke formation rate (about 4.41 mg  $g_{cat}^{-1}$  h<sup>-1</sup>) than Zn/C-ZSM-5 and Zn/C-ZSM-11 (26.33 mg  $g_{cat}^{-1}$  h<sup>-1</sup>).

Sample	$\mathbf{S}_{\text{BET}}$	$S_{Micro}^{a}$	S <sub>Meso</sub>	V <sub>Micro</sub> <sup>a</sup>	$V_{Meso}{}^{b}$	
name	$(m^2g^{-1})$	$(m^2 g^{-1})$	$(m^2 g^{-1})$	$(cm^3 g^{-1})$	$(cm^3 g^{-1})$	
Zn/N-ZSM-5	415	371	44	0.15	0.23	
Zn/N-ZSM-11	417	369	48	0.15	0.18	

Table S1. Textural properties and compositions of various samples.

<sup>*a*</sup> *t*-plot method.

 ${}^{b}V_{meso} = V_{tot} - V_{micro.}$