

Supporting Information

Investigation of the highly tunable selectivity to linear α -olefins in Fischer-Tropsch synthesis over silica supported Co and CoMn catalysts by carburization-reduction pretreatment

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Supplementary Figures and Tables

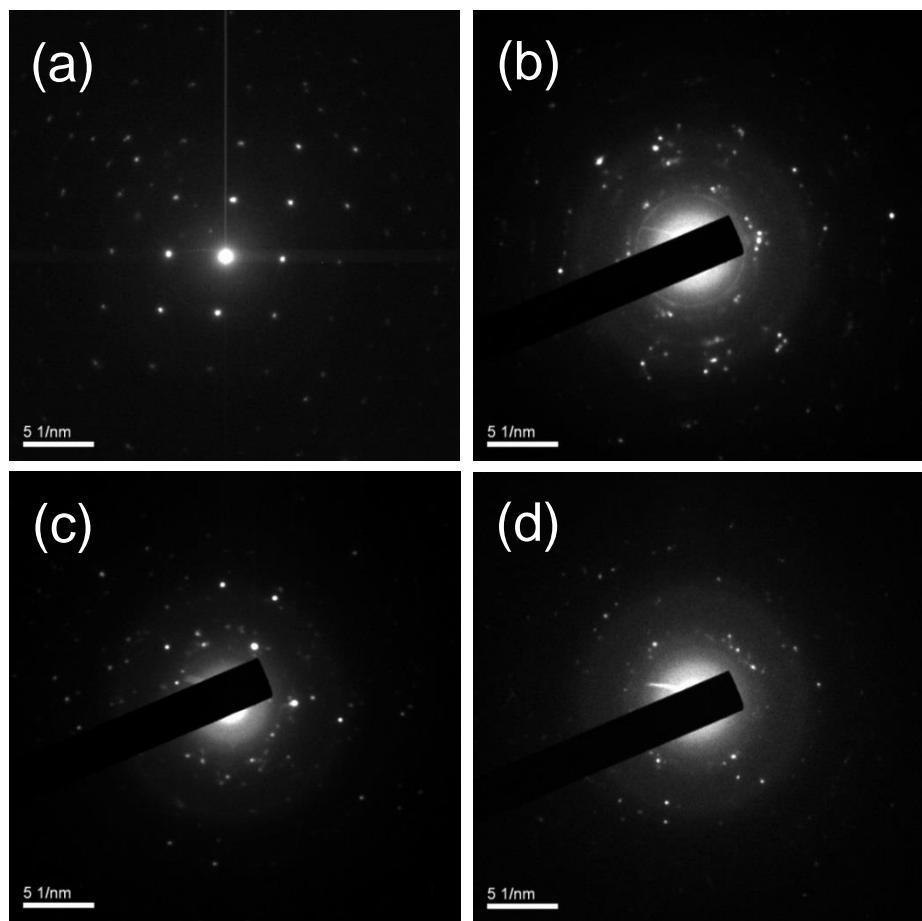


Fig. S1 Rotation electron diffraction (ED) patterns of the 15Co/SiO₂ catalysts: (a) calcined 15Co/SiO₂, (b) 15Co/SiO₂-R, (c) 15Co/SiO₂-RC, (d) 15Co/SiO₂-RCR.

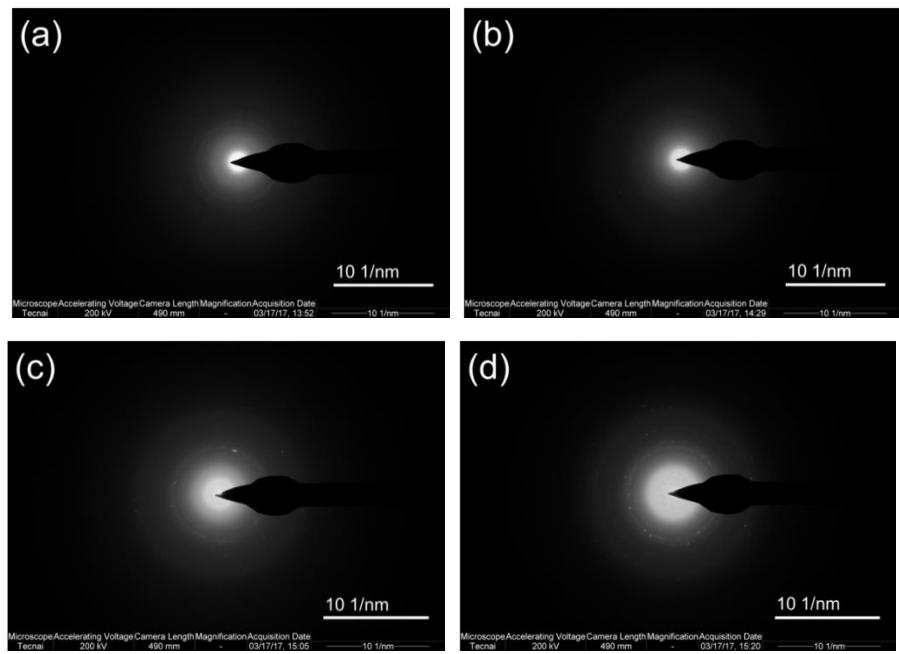


Fig. S2 Rotation electron diffraction (ED) patterns of (a) calcined 15Co3.7Mn/SiO₂, (b) 15Co3.7Mn/SiO₂-R, (c) 15Co3.7Mn/SiO₂-RC, and (d) 15Co3.7Mn/SiO₂-RCR catalysts.

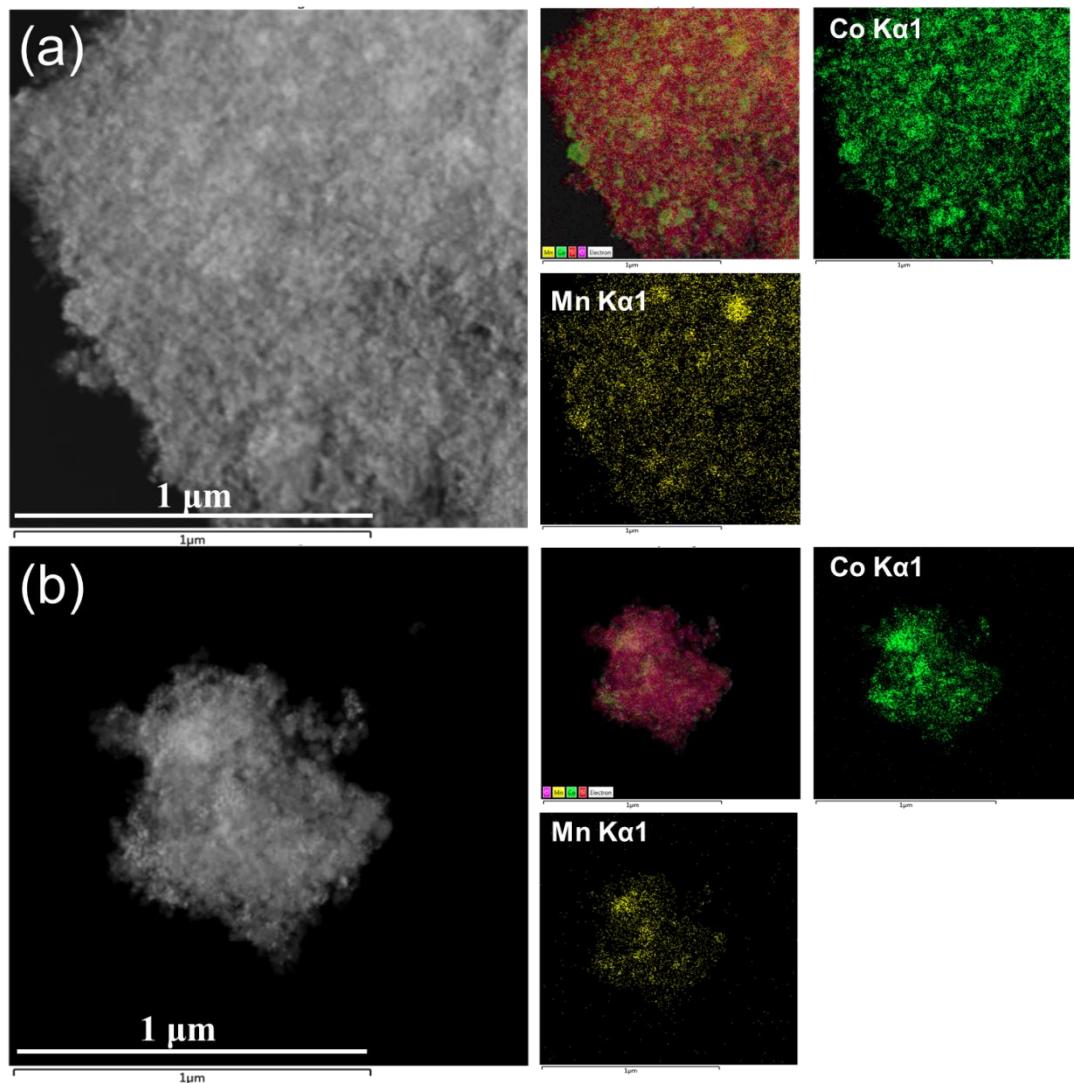


Fig. S3 EDX mapping of the calcined 15Co3.7Mn/SiO₂ (a), and 15Co3.7Mn/SiO₂-R (b).

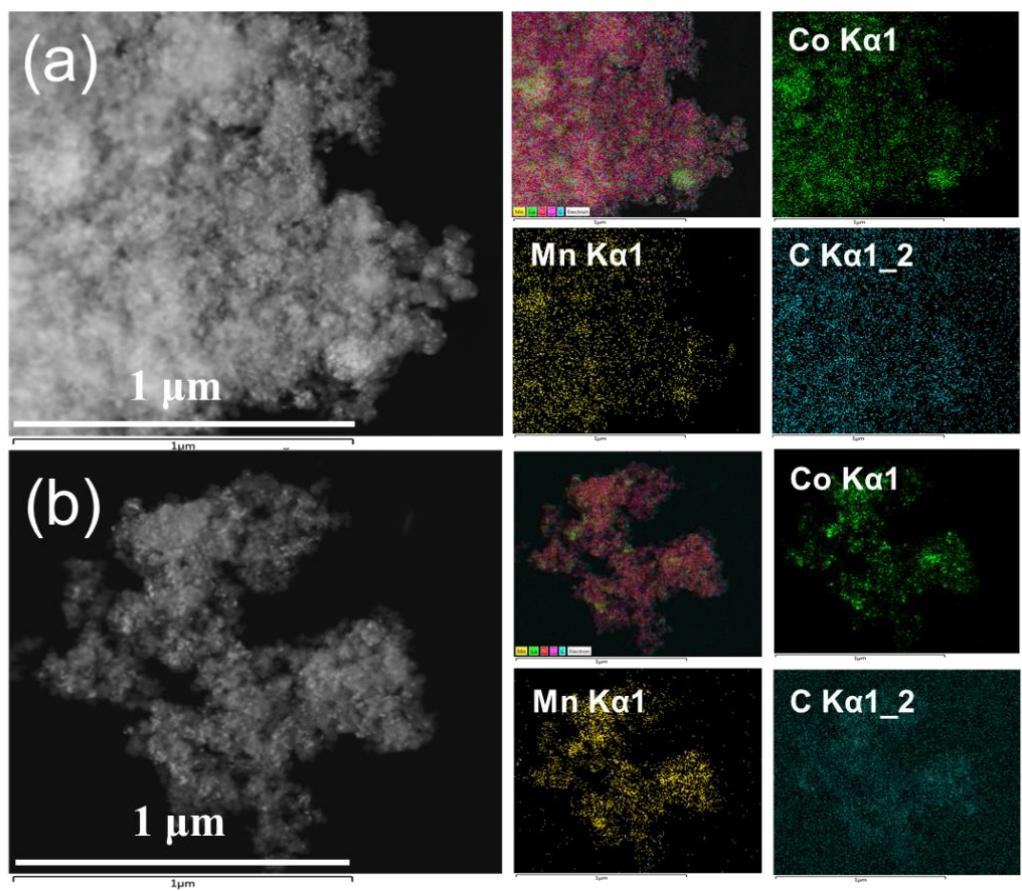


Fig. S4 EDX mapping of (a) 15Co3.7Mn/SiO₂-RC, and (b) 15Co3.7Mn/SiO₂-RCR catalysts.

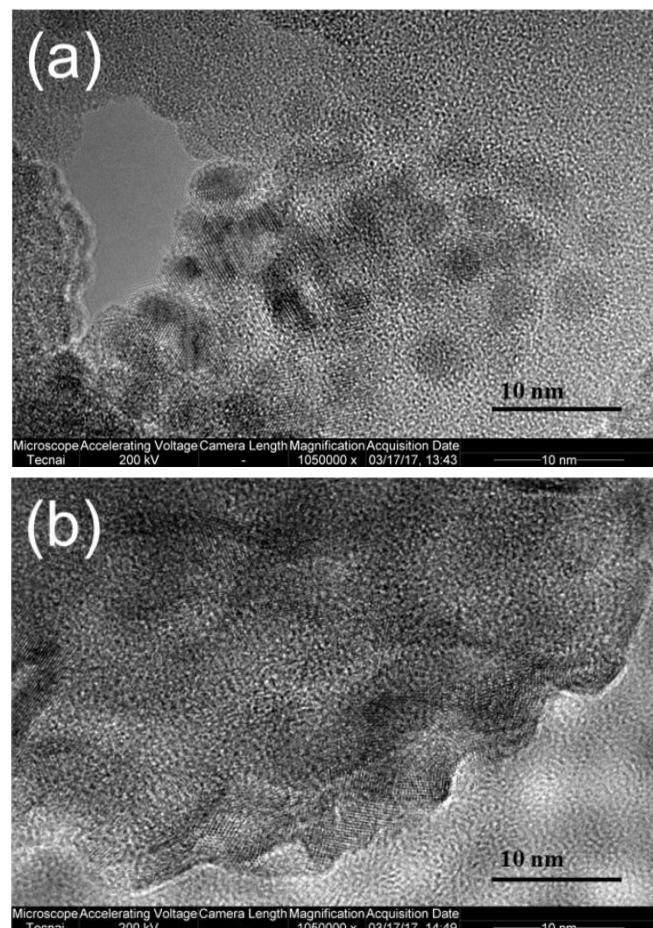


Fig. S5 High resolution TEM images of (a) calcined 15Co3.7Mn/SiO₂ and (b) 15Co3.7Mn/SiO₂-RC catalysts.

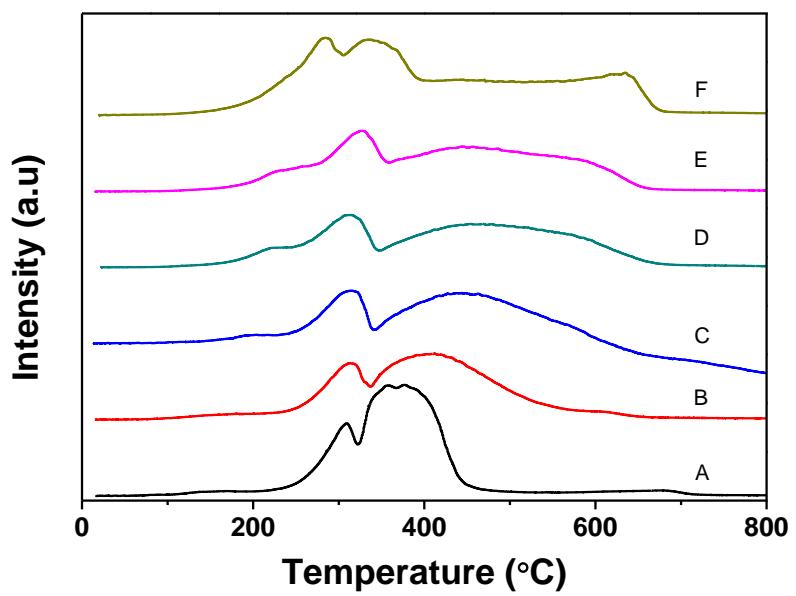


Fig. S6 H₂-TPR profiles for the Mn-promoted 15Co/SiO₂ catalysts: (A) 15Co/SiO₂, (B) 15Co0.8Mn/SiO₂, (C) 15Co1.7Mn/SiO₂, (D) 15Co3.7Mn/SiO₂, (E) 15Co5Mn/SiO₂, (F) 15Co15Mn/SiO₂.

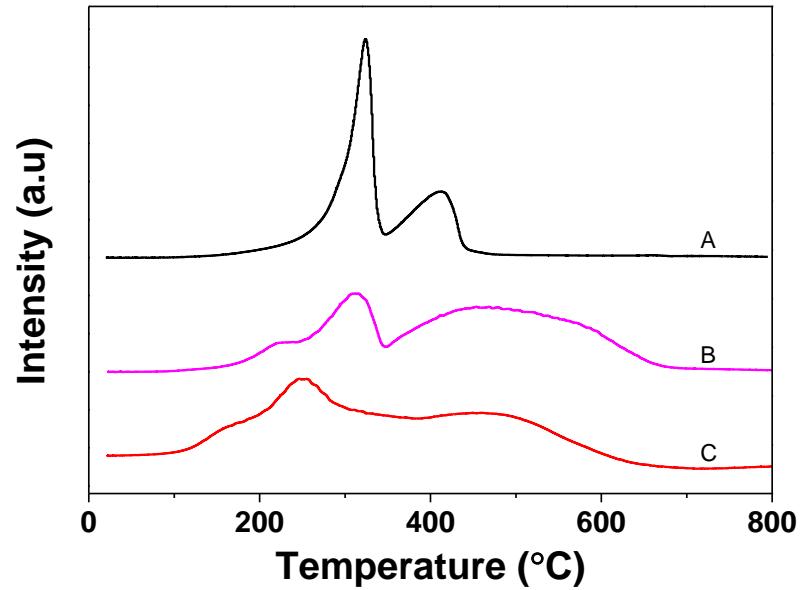


Fig. S7 H₂-TPR profiles for (A) calcined 15Mn/SiO₂, (B) calcined 15Co3.7Mn/SiO₂, and (C) 15Co3.7Mn/SiO₂-RC catalysts.

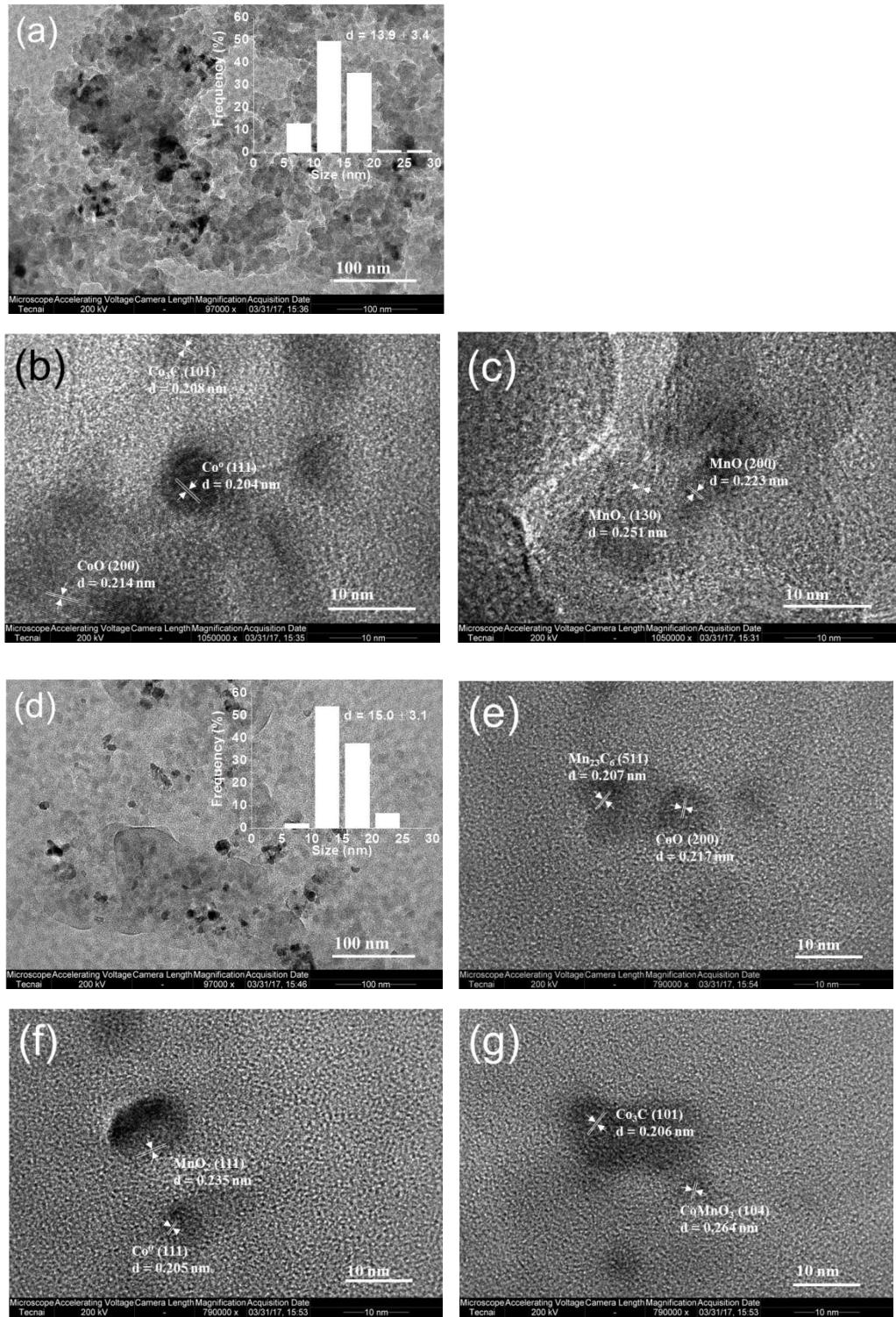


Fig. S8 Low and high resolution TEM images for the spent 15Co3.7Mn/SiO₂ catalyst with TOS under the reaction conditions of T = 230 °C, P = 10 bar, H₂/CO = 2, and GHSV = 4.5 L g_{cat}⁻¹ h⁻¹: (a-c) t = 2 h; (d-g) t = 20 h.

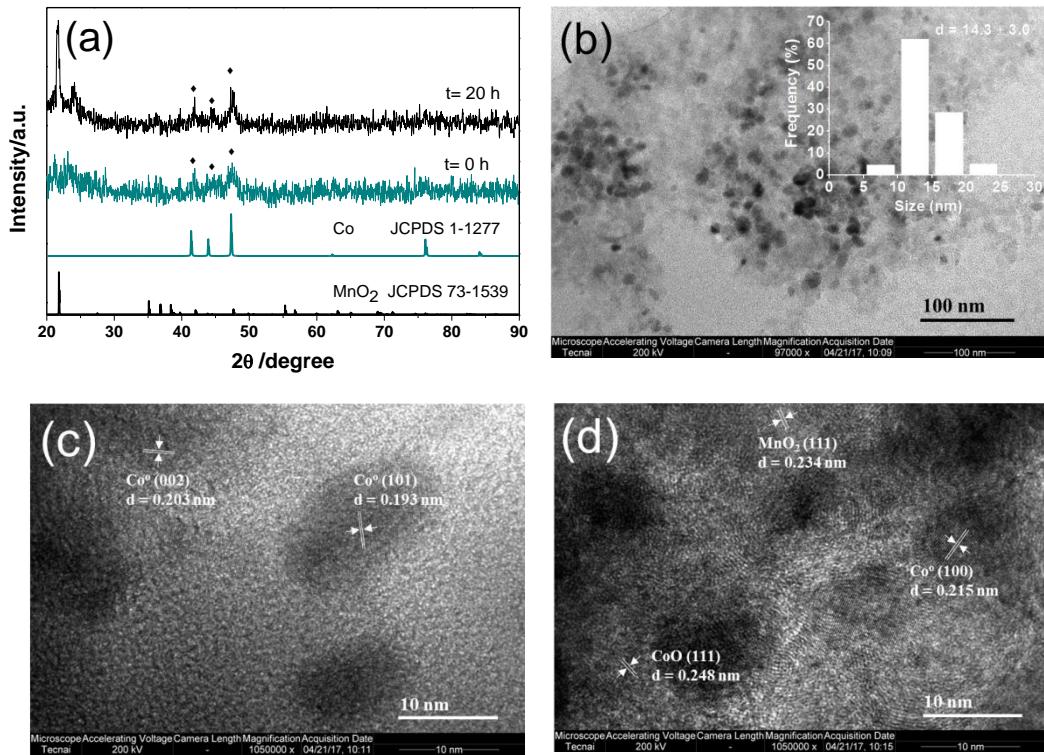


Fig. S9 XRD diffraction patterns (a, $t = 0, 20$ h) and low and high resolution TEM images (b-d, $t = 20$ h) of the 15Co3.7Mn/SiO₂-RCR catalyst with TOS under the reaction conditions of $T = 230$ °C, $P = 10$ bar, GHSV = 4.5 L g_{cat}⁻¹ h⁻¹, and H₂/CO = 2.

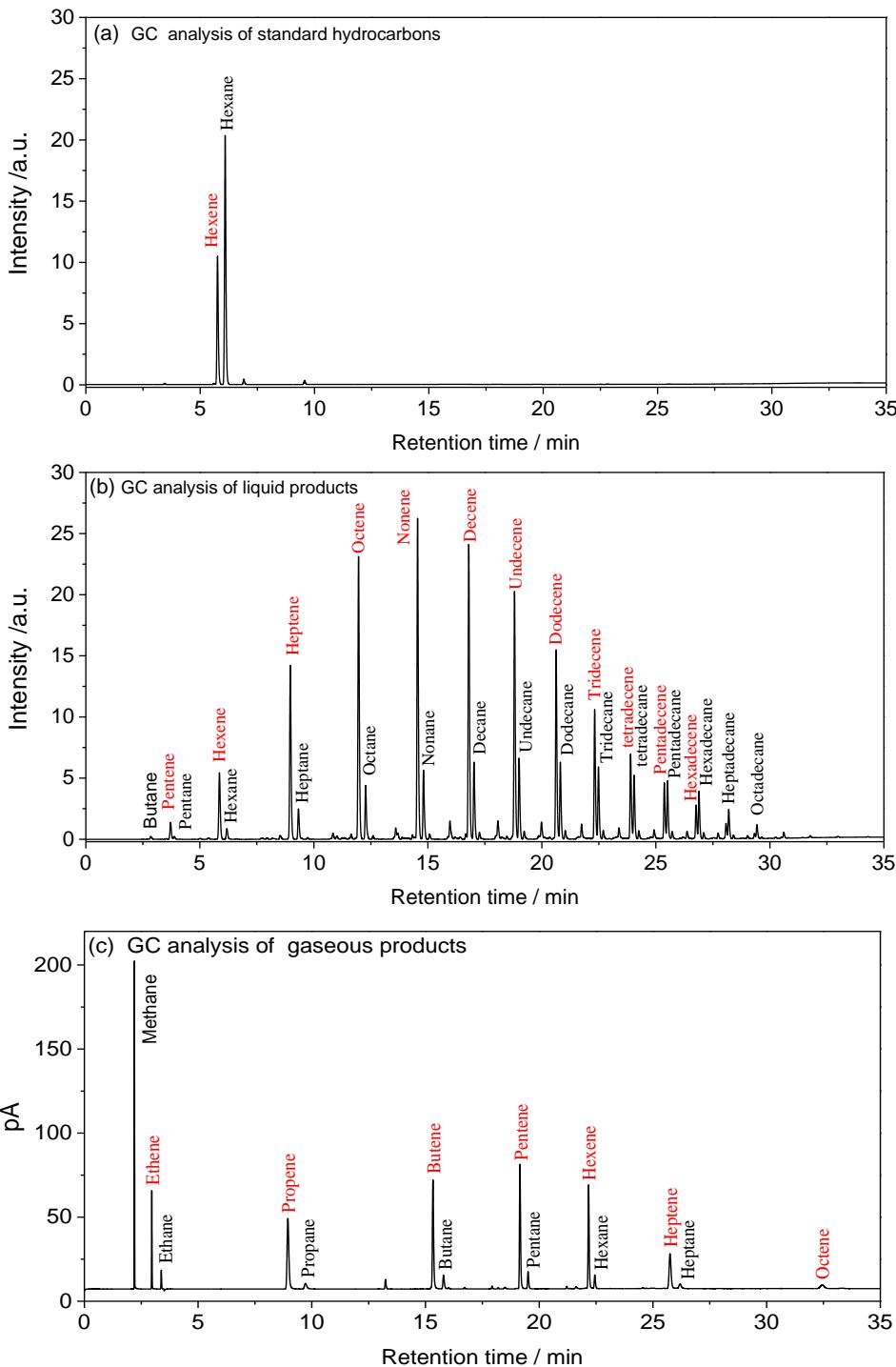


Fig. S10 GC chart of (a) the standard hydrocarbons, (b) the FT liquid products, and (c) the FT gaseous products obtained from the FTS reaction over the 15Co3.7Mn/SiO₂ catalyst under the conditions of 240 °C, 10 bar, and H₂/CO = 0.5.

It is clearly shown in the GC chart that, in the case of the tested Co-based catalysts, the FT products mostly consist of the n-paraffins and α -olefins, including very small amount of the iso-hydrocarbons and almost no internal olefins.

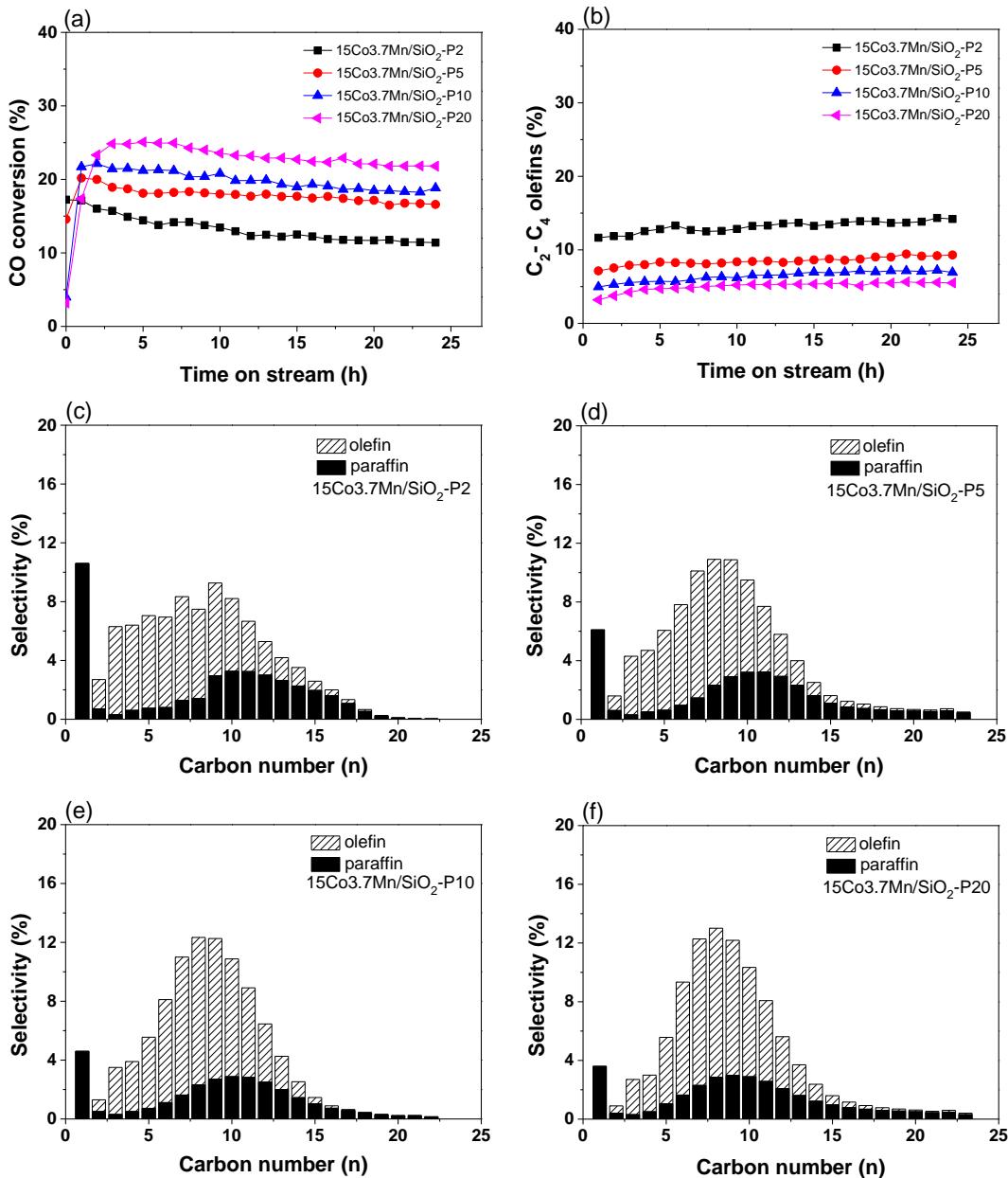


Fig. S11 Influence of the reaction pressure on the CO conversion (a) and C₂–C₄ selectivity (b)

with time on stream, and products distribution (c-f) for the 15Co3.7Mn/SiO₂ catalyst.

Table S1 Textural properties of the support and prepared cobalt catalysts.

Sample	S_{BET} (m^2/g)	D_{pore} (nm)	V_{pore} (cm^3/g)
Q-15	186.8	25.1	1.17
15Co/SiO ₂	153.5	25.2	0.90
15Co/SiO ₂ -R	153.8	22.0	0.83
15Co3.7Mn/SiO ₂	158.3	21.8	0.83
15Co3.7Mn/SiO ₂ -R	154.7	21.5	0.81

Table S2 The influence of pretreatment on the crystal facet and phase state of cobalt over the 15Co/SiO₂ catalyst.

Sample	Nanoparticle	Crystal facet	Lattice distance (nm)		
			From XRD ^a	From HRTEM ^b	From ED ^c
15Co/SiO ₂	Co ₃ O ₄	(311)	0.243	0.252	0.244
15Co/SiO ₂ -R	Co	(111)	0.205	0.207	0.212
15Co/SiO ₂ -RC	Co ₂ C	(111)	0.212	0.205	0.220
15Co/SiO ₂ -RCR	Co	(101)	0.192	0.189	0.185
	Co	(002)	0.203	0.204	0.203

^a corresponding to Figure 1; ^b corresponding to Figure 3; ^c corresponding to Figure S1.

Table S3 Cobalt particle size calculated from XRD patterns and TEM images of the pretreated 15Co/SiO₂ catalysts.

Sample	Nanoparticle	Particle size (nm)	
		By XRD	By TEM
15Co/SiO ₂	Co ₃ O ₄	17.5 (13.1) ^a	16.1±5.0 (12.1±5.0) ^a
15Co/SiO ₂ -R	Co	15.0	15.5±3.4
15Co/SiO ₂ -RC	Co ₂ C	17.1	17.8±4.6
15Co/SiO ₂ -RCR	Co	19.7	20.3±6.4

^a Calculated from Co₃O₄ particle, $d(\text{Co}) = 0.75 \times d(\text{Co}_3\text{O}_4)$

Table S4 Cobalt particle size calculated from XRD patterns and TEM images of the pretreated 15Co3.7Mn/SiO₂ catalysts.

Sample	Nanoparticle	Particle size (nm)	
		By XRD	By TEM
15Co3.7Mn/SiO ₂	Co ₃ O ₄	13.0 (9.8) ^a	10.0±2.3 (7.5±2.3) ^a
15Co3.7Mn/SiO ₂ -R	Co	14.6	13.2±2.5
15Co3.7Mn/SiO ₂ -RC	Co ₂ C	13.9	12.7±2.9
15Co3.7Mn/SiO ₂ -RCR	Co	11.8	12.4±2.6

^a Calculated from Co₃O₄ particle, $d(\text{Co}) = 0.75 \times d(\text{Co}_3\text{O}_4)$

Table S5 The influence of pretreatment on the crystal facet and phase state of cobalt over the 15Co3.7Mn/SiO₂ catalyst.

Sample	Nanoparticle	Crystal facet	Lattice distance (nm)		
			From XRD ^a	From HRTEM ^b	From ED ^c
15Co3.7Mn/SiO ₂	Co ₃ O ₄	(311)	0.243	0.246	0.248
15Co3.7Mn/SiO ₂ -R	Co	(111)	0.205	0.209	0.197
	CoO	(200)	0.213	0.214	---
15Co3.7Mn/SiO ₂ -RC	Co ₂ C	(111)	0.212	0.212	0.227
15Co3.7Mn/SiO ₂ -RCR	Co	(101)	0.192	0.190	0.172
	Co	(002)	0.206	0.204	---
	Co	(100)	0.218	0.216	---

^acorresponding to Figure 2; ^bcorresponding to Figure 4; ^ccorresponding to Figure S2.

Table S6 H₂ uptake and calculated cobalt reducibility over the Mn-promoted 15Co/SiO₂ catalysts.

Sample	Co (wt%)	Mn (wt%)	H ₂ uptake (mmol) ^a	Reducibility (%) ^b
15Mn/SiO ₂	0	15	0.064	46.7 ^c
15Co/SiO ₂	15	0	0.099	58.5
15Co0.8Mn/SiO ₂	15	0.8	0.074	43.7 ^d
15Co1.7Mn/SiO ₂	15	1.7	0.058	34.3 ^e
15Co3.7Mn/SiO ₂	15	3.7	0.057	---
15Co5Mn/SiO ₂	15	5	0.063	---
15Co15Mn/SiO ₂	15	15	0.093	---

^a50 mg sample; ^bcobalt reducibility; ^ccalculated from the MnO₂ to MnO; ^{d,e}regardless of the reduction of MnO₂.

Table S7 The evolution of crystal facet and phase state of Co (Mn) over the 15Co3.7Mn/SiO₂ catalyst with TOS.

Reaction time (h)	Nanoparticle	Crystal facet	Lattice distance (nm)	
			From XRD data ^a	From HRTEM ^b
Reduced sample (t = 0)	Co	(111) major	0.205	0.209
	CoO	(200)	0.213	0.214
Spent sample (t = 2)	Co	(111) major	0.205	0.204
	Co ₃ C	(101) minor	0.205	0.208
	CoO	(200)	0.213	0.214
	MnO	(200)	0.222	0.223
Spent sample (t = 20)	MnO ₂	(130)	0.255	0.251
	Co	(111) minor	0.205	0.205
	Co ₃ C	(101) major	0.205	0.206
	CoO	(200)	0.213	0.217
	MnO	(200)	0.222	0.223
	MnO ₂	(111)	0.234	0.235
	Mn ₂₃ C ₆	(511)	0.204	0.207
	CoMnO ₃	(104)	0.267	0.264

Reaction conditions: T = 230 °C, P = 10 bar, H₂/CO = 2, GHSV = 4.5 L g_{cat}⁻¹ h⁻¹. ^acorresponding to Figure 7 (a); ^bcorresponding to Figure 4 (c, d) and Figure S8 (a-g).