

Electronic Supplementary Information (ESI)

Highly active Pd-Ni nanocatalysts supported on multicharged polymer matrix

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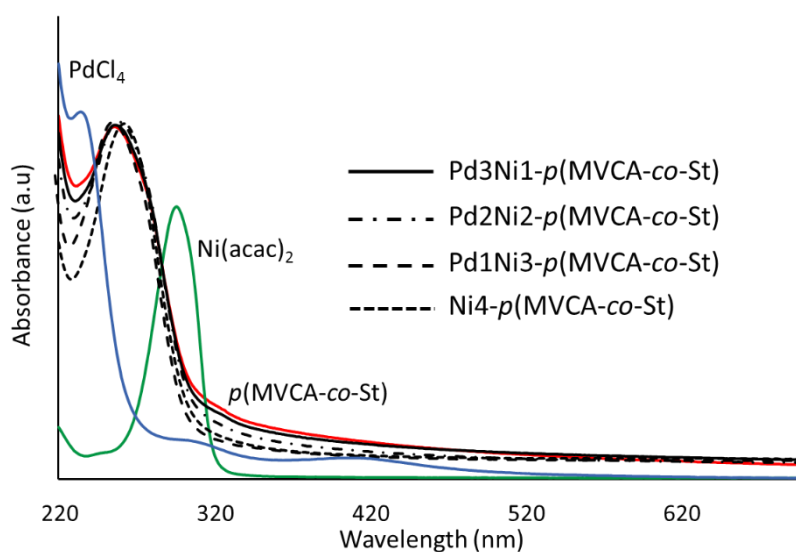


Fig. S1. UV spectra of Ni(acac)₂ (green); Na₂[PdCl₄] (blue); p(MVCA-co-St) (red); and Pd3Ni1-p(MVCA-co-St), Pd2Ni2-p(MVCA-co-St), Pd1Ni3-p(MVCA-co-St) and Ni4-p(MVCA-co-St) (H₂O, 20 °C, l = 0.5 cm).

Table S1. Amount of Pd and Ni in the compositions measured by inductively coupled plasma atomic emission spectrometry (ICP-AES).^a

	Theoretical yield		ICP-AES data, yield		Molar ratio Pd-Ni measured by ICP-AES	
	Pd, C (mg/l)	Ni, C (mg/l)	Pd, C (mg/l), $\pm 10\%$	Ni, C (mg/l), $\pm 10\%$	Pd	Ni
Pd3Ni1-p(MVCA-co-St)	2.3	0.42	1.78 (77%)	0.37 (88 %)	0.73	0.27
Pd2Ni2-p(MVCA-co-St)	1.52	0.84	1.31 (86 %)	0.62 (74 %)	0.54	0.46
Pd1Ni3-p(MVCA-co-St)	0.76	1.25	0.66 (87 %)	0.98 (78 %)	0.27	0.73
Ni4-p(MVCA-co-St)	-	1.68	-	1.31 (78 %)	-	1

^aThe samples solutions were diluted 70-fold before measurement

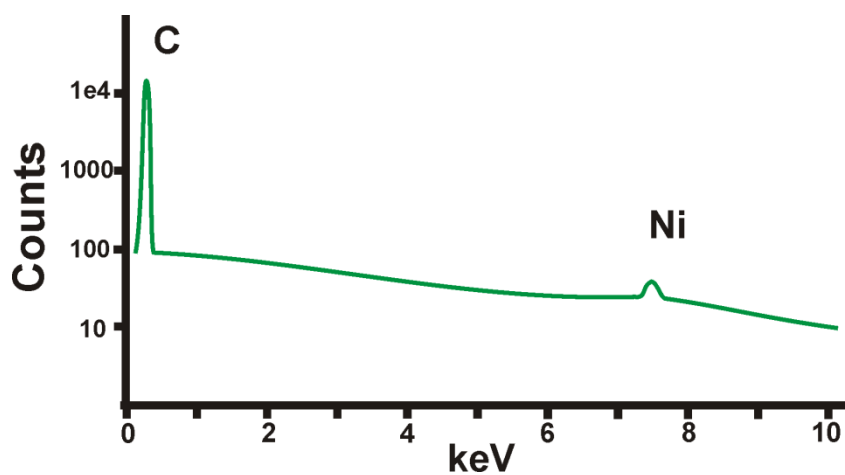


Fig. S2. EDX spectrum of Ni4-p(MVCA-co-St) after background subtraction.

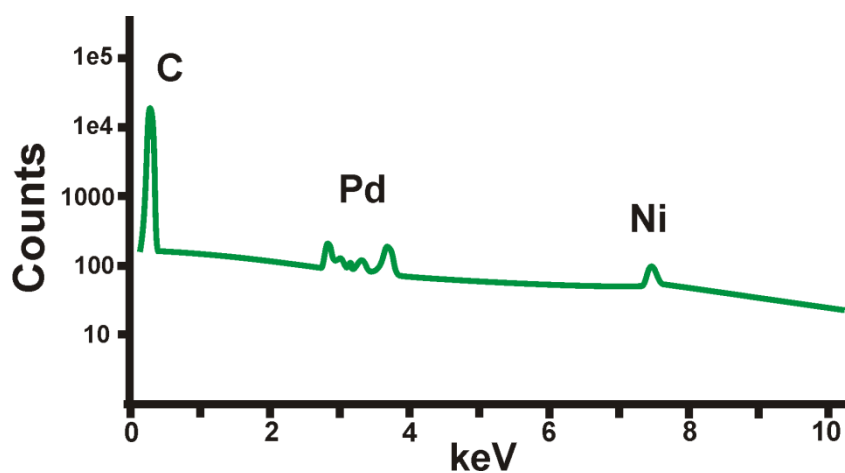


Fig. S3. EDX spectrum of Pd1Ni3-p(MVCA-co-St) after background subtraction.

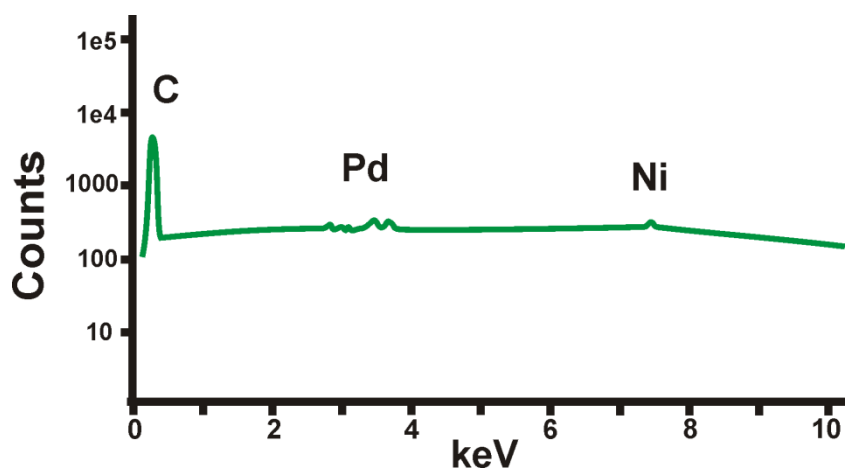


Fig. S4. EDX spectrum of Pd₂Ni₂-*p*(MVCA-*co*-St) after background subtraction.

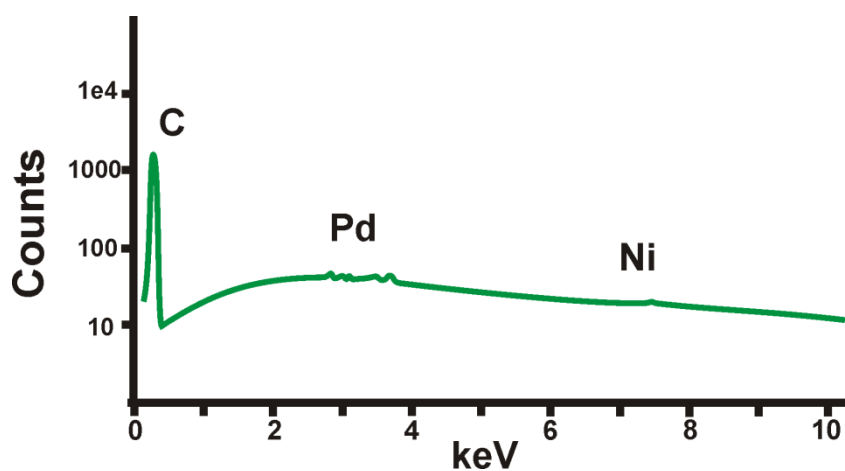


Fig. S5. EDX spectrum of Pd₃Ni₁-*p*(MVCA-*co*-St) after background subtraction.

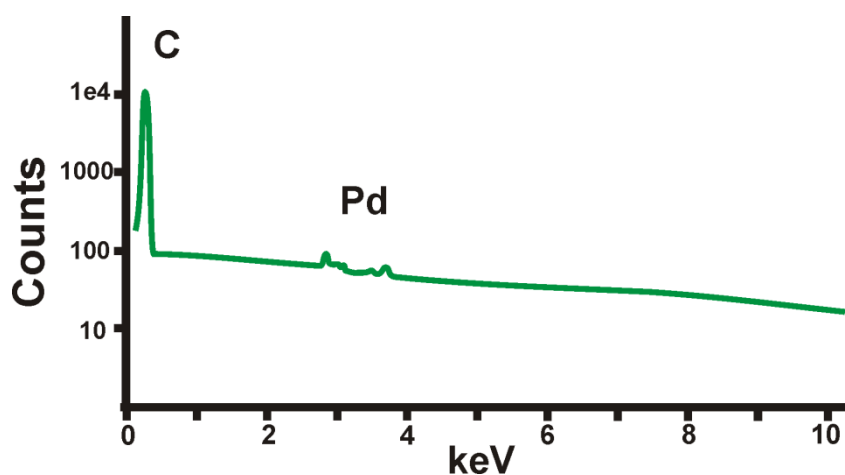


Fig. S6. EDX spectrum of Pd₄-*p*(MVCA-*co*-St) after background subtraction.

Characteristic lines in the EDX spectra:

Ni lines: $K\alpha_1 = 7.478$ keV, $K\alpha_2 = 7.461$ keV (one averaged line is observed);

Pd lines: $L\alpha_1 = 2.84$ keV, $L\beta_1 = 2.99$ keV, $L\beta_3 = 3.07$ keV, $L\gamma_1 = 3.32$ keV, $L\gamma_3 = 3.56$ keV