Reaction Pathway for Partial Hydrogenation of 1,3-Butadiene over Pt/SiO₂

Chaoquan Hu ^{a,*}, Jiahan Sun ^a, Yafeng Yang ^a, Qingshan Zhu ^{a,*}, Bin Yu ^b

^a State Key Laboratory of Multiphase Complex Systems, Institute of Process Engineering, Chinese Academy of Sciences, Beijing 100190, PR China.

^b Shangdong CISRI-CHALCO Rare Earth Technology Co., Ltd. Jining 277600, Shandong, PR China.

E-mail: cqhu@ipe.ac.cn, qszhu@ipe.ac.cn;

Fax: +86 10 62536108;

Tel.: +86 10 62558393.

Supporting information



Figure S1. Light-off curves for 1,3-butadiene (0.25%) hydrogenation in 15% H_2 /Ar over the Pt/SiO₂ catalyst with different particle size.



Figure S2. Light-off curves for hydrogenation of 1,3-butadiene (0.25%) in 15% H_2/Ar over the Pt/SiO₂ catalyst with different amounts of catalyst and gas flow rates (STP).



Figure S3.DRIFT spectra for uptake of 1-butene and *trans*-2-butene on the Pt/SiO_2 catalyst at room temperature.



Figure S4. Top view of the carbonaceous species with labeling the number of the carbon atoms.

Table S1. Geometrical parameters (expressed in Å) of C_4H_x and transition states for partial hydrogenation of 1,3-butadiene on the Pt(111) surface. Pt-C represents the distance between the carbon and its coordinating Pt.

| Spacios | Dt C | Dt C | Dt C | Dt C | C ₁ - | C ₂ - | C ₃ - | С-Н |
|--|---------|-------------------|-------------------|-------|------------------|------------------|------------------|---------|
| species | | rt-C ₂ | rt-C ₃ | rt-C4 | C ₂ | C ₃ | C ₄ | forming |
| $C_4H_6(\text{tetra-}\sigma)$ | 2.123 | 2.204 | 2.201 | 2.126 | 1.475 | 1.477 | 1.475 | |
| $C_4H_6(di-\pi)$ | 2.204 | 2.228 | 2.249 | 2.185 | 1.413 | 1.458 | 1.414 | |
| $C_4H_6(di-\sigma)$ | 2.129 | 2.179 | - | - | 1.485 | 1.463 | 1.348 | |
| $C_4H_7(1B3R)(tri-\sigma)$ | 2.120 | 2.247 | 2.150 | - | 1.472 | 1.478 | 1.507 | |
| $C_4H_7(1B4R)(tri-\sigma)$ | 2.143 | 2.145 | - | 2.108 | 1.486 | 1.520 | 1.516 | |
| 1-butene(di-σ) | 2.122 | 2.172 | - | - | 1.486 | 1.517 | 1.517 | |
| <i>tran</i> -2-butene(di-σ) | - | 2.144 | 2.147 | - | 1.515 | 1.500 | 1.515 | |
| <i>cis</i> -2-butene(di- σ) | - | 2.160 | 2.161 | - | 1.510 | 1.499 | 1.512 | - |
| C_4H_6 (tetra- σ) + H(top) | 2.125 | 2.203 | 2.218 | 2.121 | 1.473 | 1.474 | 1.475 | |
| C_4H_6 (di- π)+ H(top) | 2.204 | 2.229 | 2.244 | 2.188 | 1.414 | 1.458 | 1.414 | |
| C_4H_6 (di- σ)+ H(top) | 2.124 | 2.189 | - | - | 1.485 | 1.464 | 1.345 | |
| C_4H_7 (1B3R) (tri- σ) + | 2 1 2 2 | 2 244 | 2 1 4 0 | | 1 472 | 1 479 | 1 507 | |
| H(top) | 2.125 | 2.244 | 2.149 | - | 1.4/5 | 1.4/0 | 1.307 | |
| cis -C ₄ H ₇ (1B3R) (tri- σ) | 2 1 2 8 | 2 218 | 2 1/18 | | 1 472 | 1 / 87 | 1 5 1 5 | |
| + H(top) | 2.120 | 2.210 | 2.140 | - | 1.472 | 1.407 | 1.515 | |
| TS ₁ | 2.113 | 2.240 | - | - | 1.481 | 1.431 | 1.489 | 1.929 |
| TS ₂ | - | 2.222 | 2.143 | - | 1.470 | 1.488 | 1.503 | 2.064 |
| TS ₃ | - | 2.266 | 2.144 | - | 1.450 | 1.488 | 1.511 | 2.069 |
| TS ₄ | 2.123 | 2.215 | 2.213 | - | 1.476 | 1.484 | 1.457 | 1.494 |
| TS ₅ | 2.132 | 2.152 | - | 2.130 | 1.484 | 1.474 | 1.458 | 1.786 |
| TS ₆ | 2.122 | 2.251 | - | - | 1.474 | 1.439 | 1.382 | 1.643 |
| TS ₇ | 2.137 | 2.160 | - | - | 1.489 | 1.480 | 1.381 | 1.776 |
| C_4H_7 (1B3R) ($\eta_1 \sigma$) + | 2.141 | - | - | - | 1.457 | 1.353 | 1.486 | - |
| 2H(top) | | | | | | | | |
| TS ₈ | 2.178 | - | - | - | 1.431 | 1.414 | 1.511 | 1.469 |
| 1-butene (π)+ H(top) | 2.211 | 2.269 | - | - | 1.407 | 1.503 | 1.534 | - |
| TS ₉ | 2.178 | 2.292 | 2.235 | - | 1.428 | 1.460 | 1.438 | 1.699 |
| TS ₁₀ | 2.236 | 2.213 | - | 2.146 | 1.392 | 1.473 | 1.441 | 1.652 |
| C ₄ H ₇ (1B3R)(π-σ) | 2.179 | 2.348 | 2.145 | - | 1.415 | 1.466 | 1.514 | - |
| $C_4H_7(1B4R)(\pi-\sigma)$ | 2.235 | 2.233 | - | 2.096 | 1.404 | 1.502 | 1.515 | - |



Figure S5. Optimized structures (side and top views) of reactant, transition states, and the products for hydrogenation of C_4H_6 with the di- π configuration to C_4H_7 on the Pt(111) surface.