

Electronic Supplementary Information (ESI)

Highly Shape- and Regio-selective Peroxy-trifluoromethylation of Styrene by Metal-Organic Framework Cu₃(BTC)₂

Bo Qi,^a Tiexin Zhang,^{a*} Mochen Li,^a Cheng He^a and Chunying Duana,^{b*}

^aState Key Laboratory of Fine Chemicals, Dalian University of Technology, Dalian 116024, China.

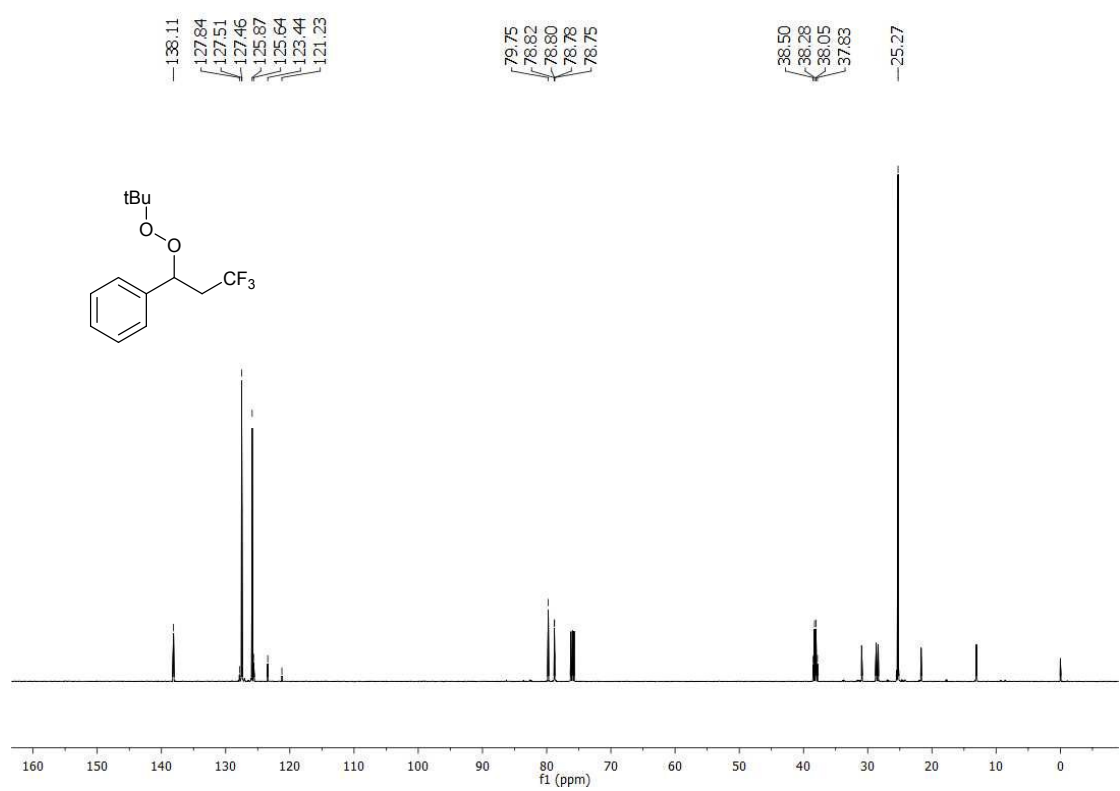
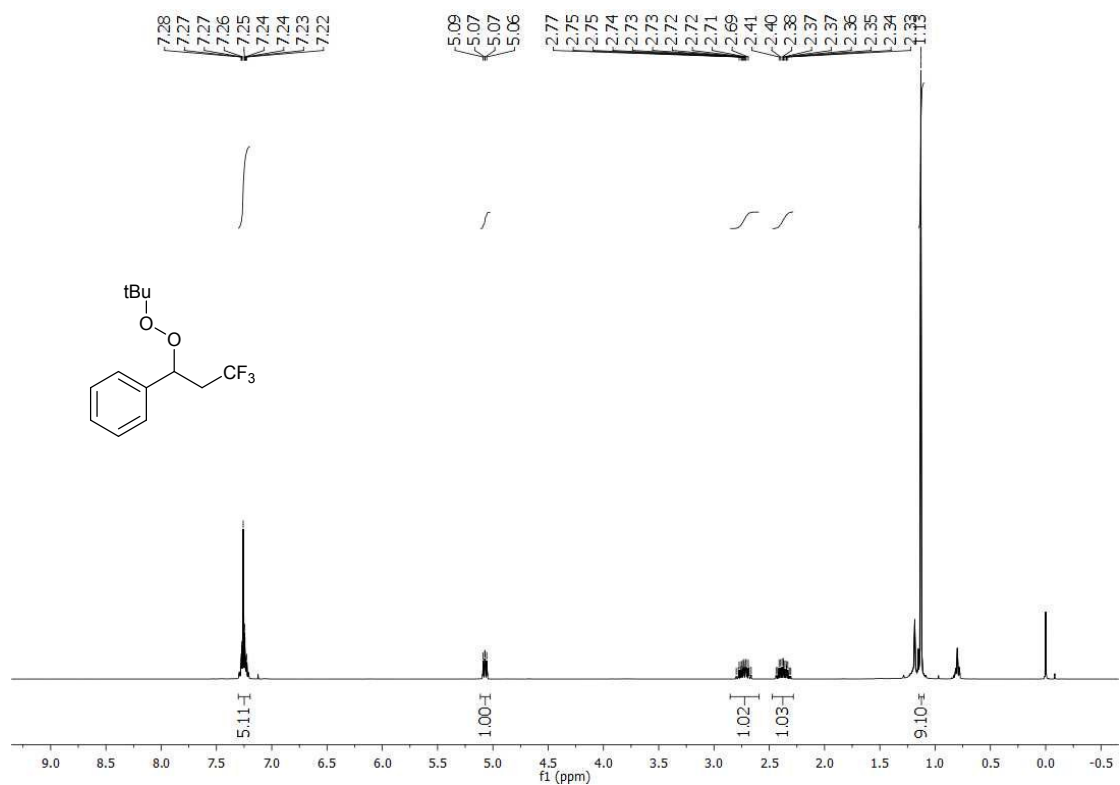
^bCollaborative Innovation Center of Chemical Science and Engineering, Tianjin 300071, China.

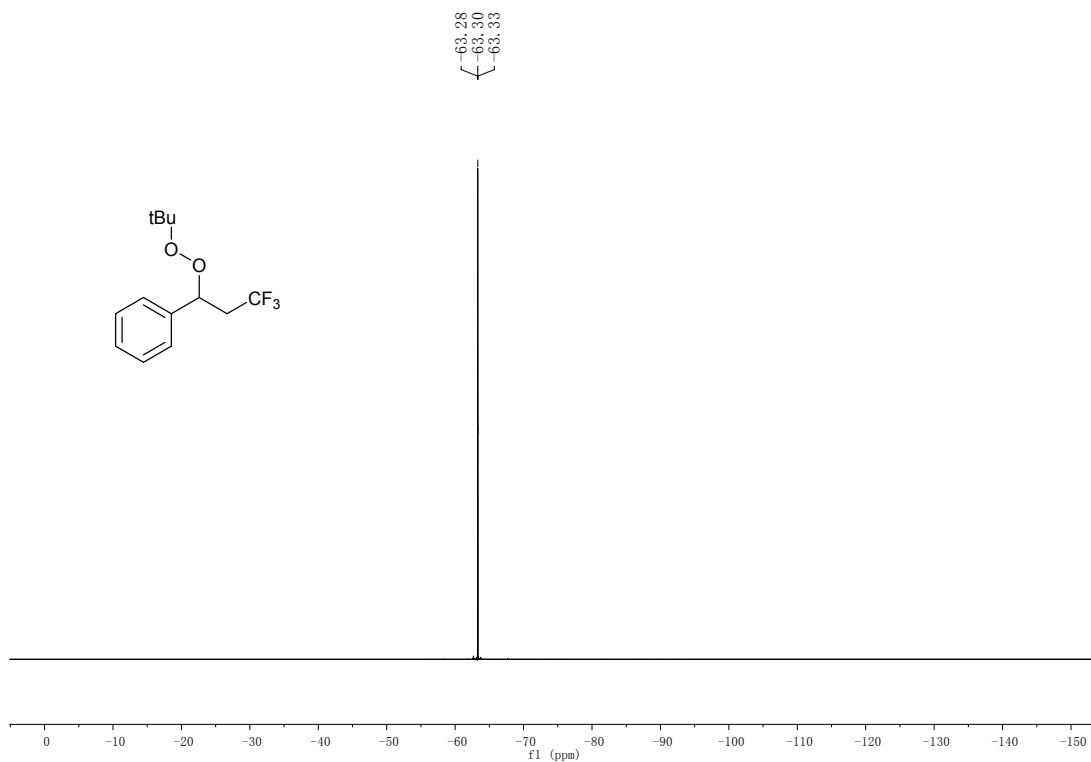
*Corresponding author: Phone and Fax: +86-411-84986476;

E-mail address: zhangtiexin@dlut.edu.cn; cyduan@dlut.edu.cn

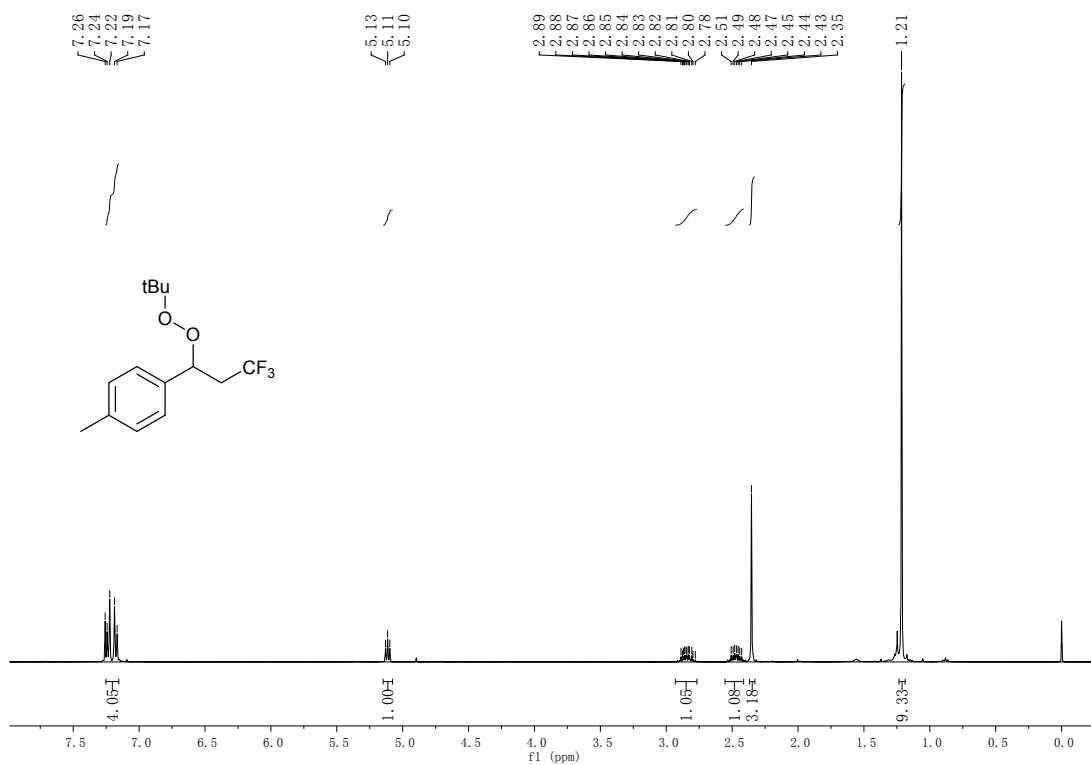
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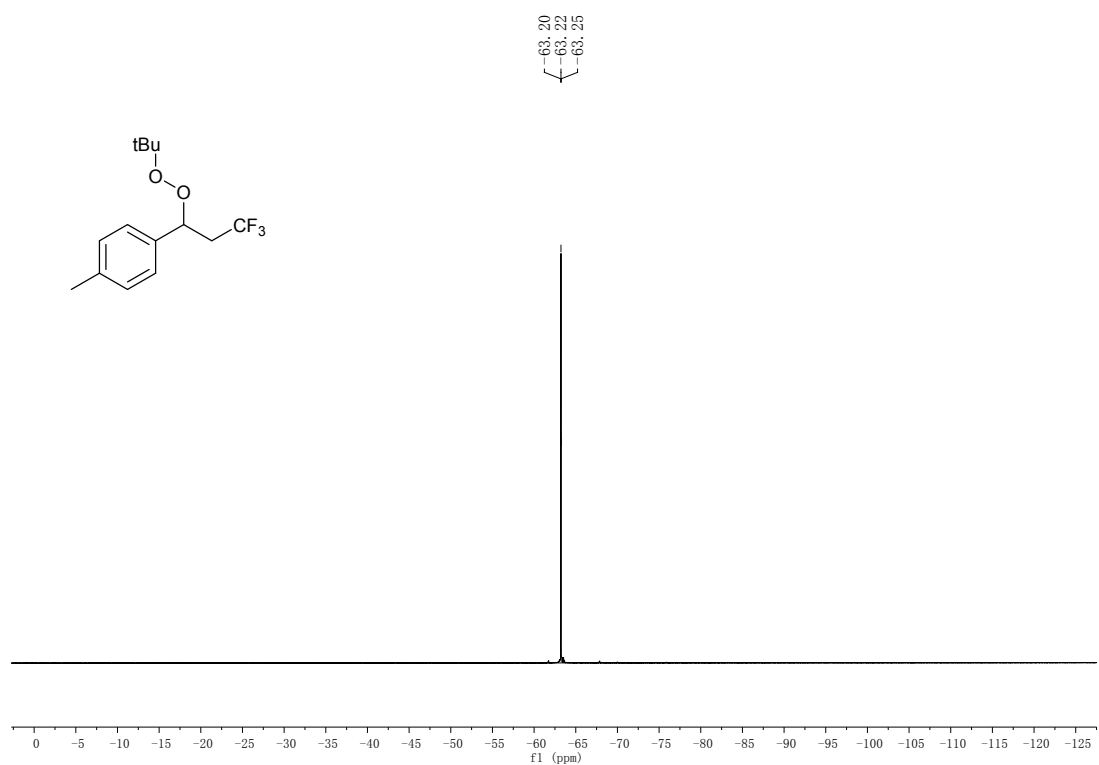
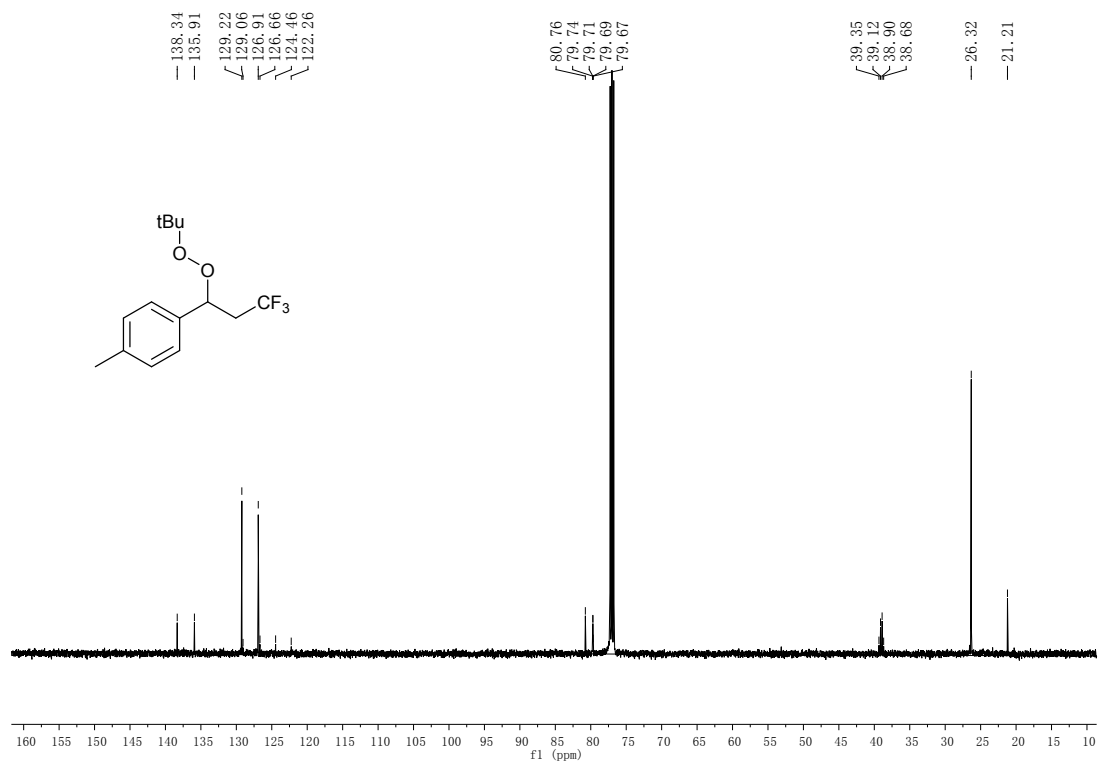
(1-(*tert*-butylperoxy)-3,3,3-trifluoropropyl)benzene (2a)



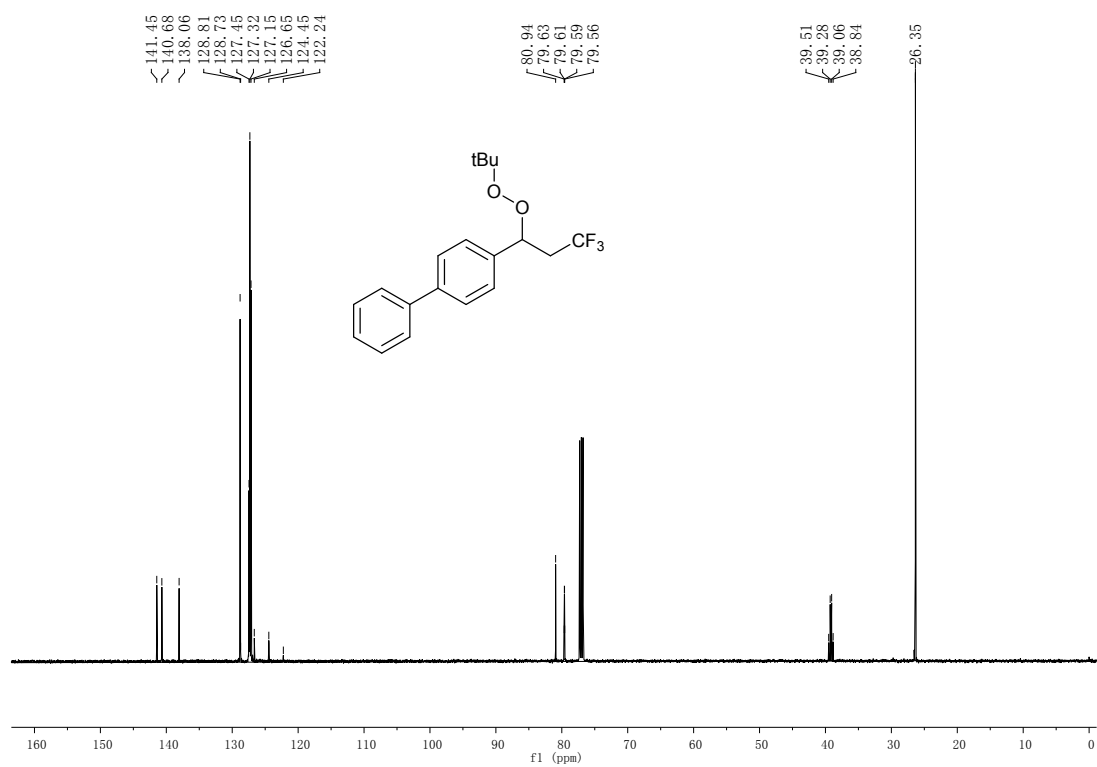
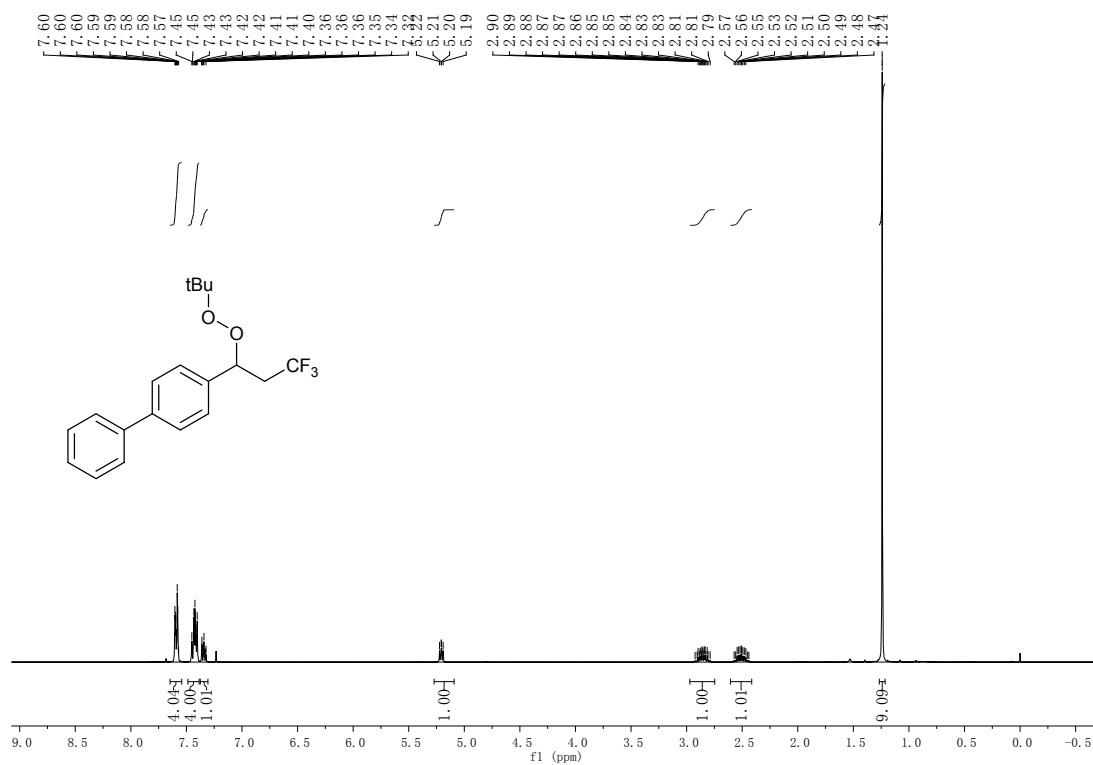


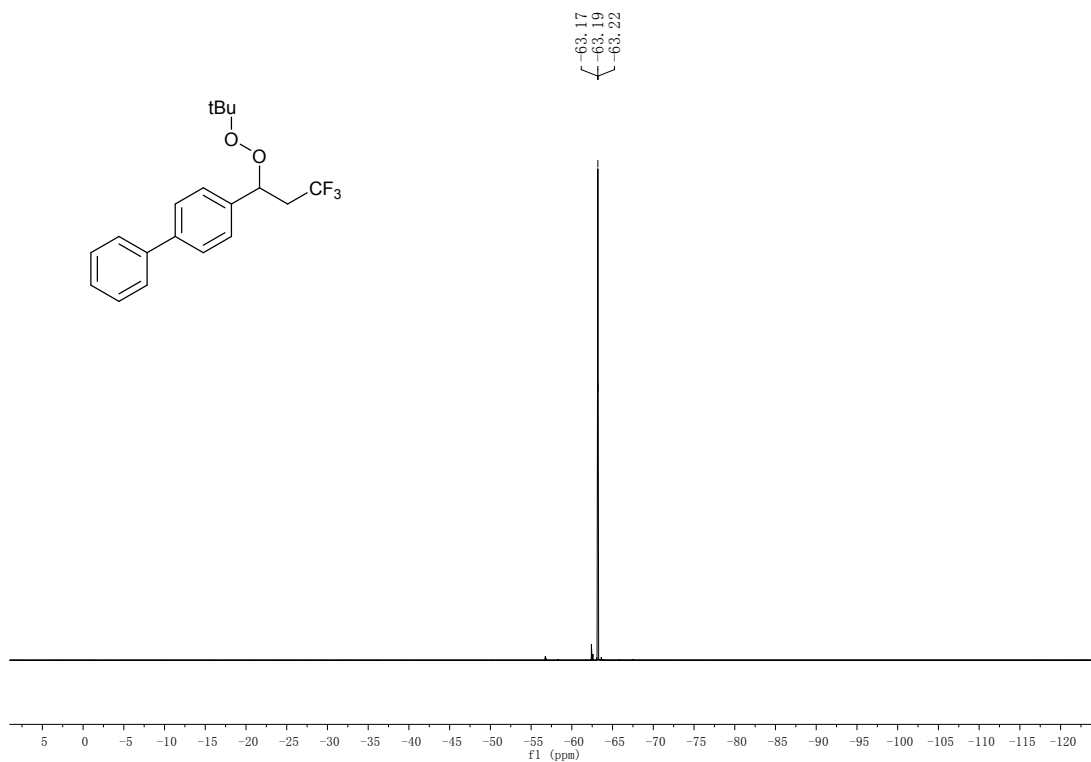
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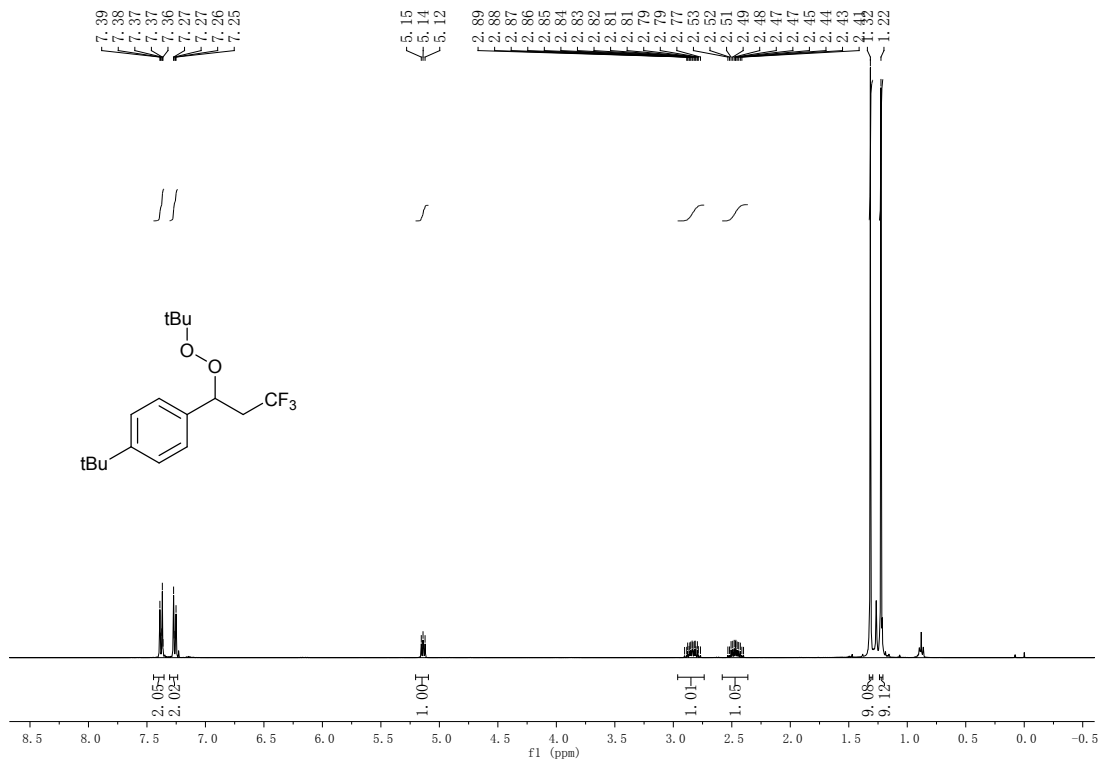


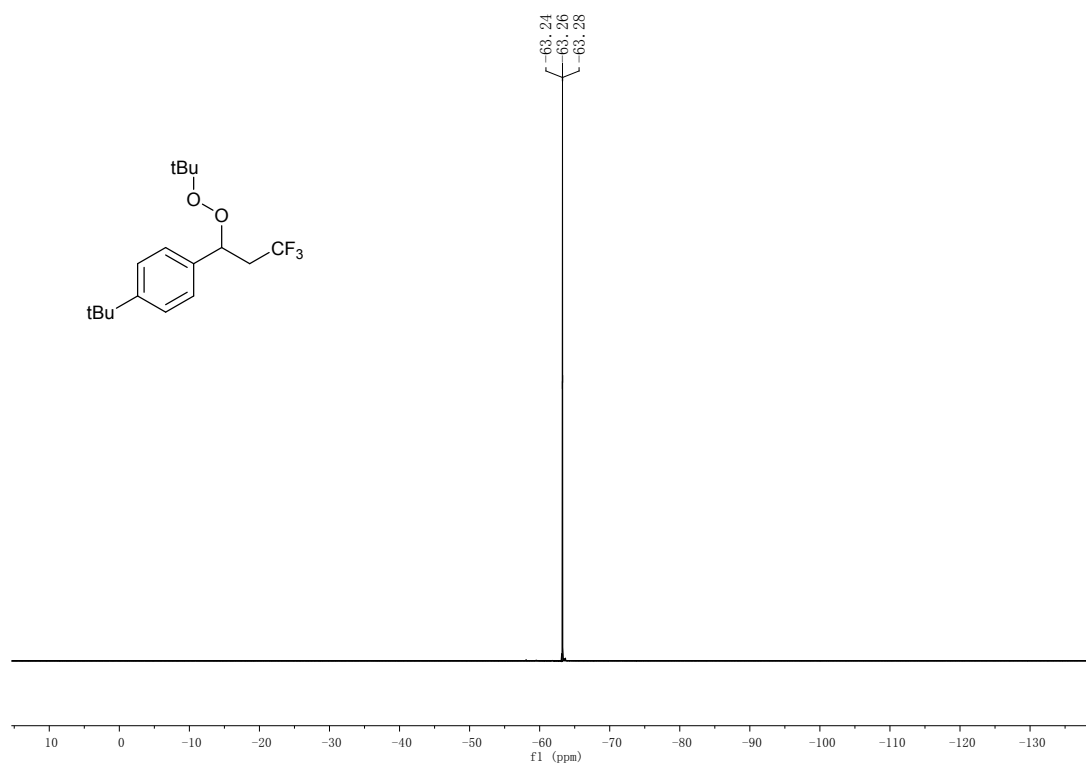
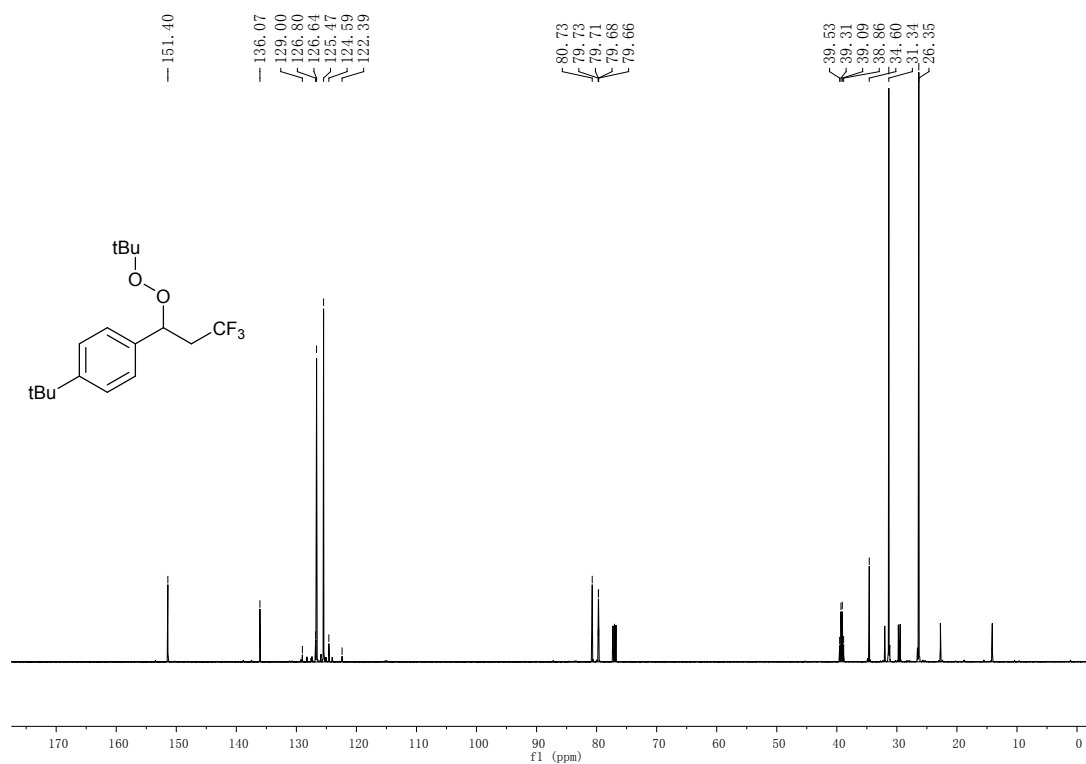
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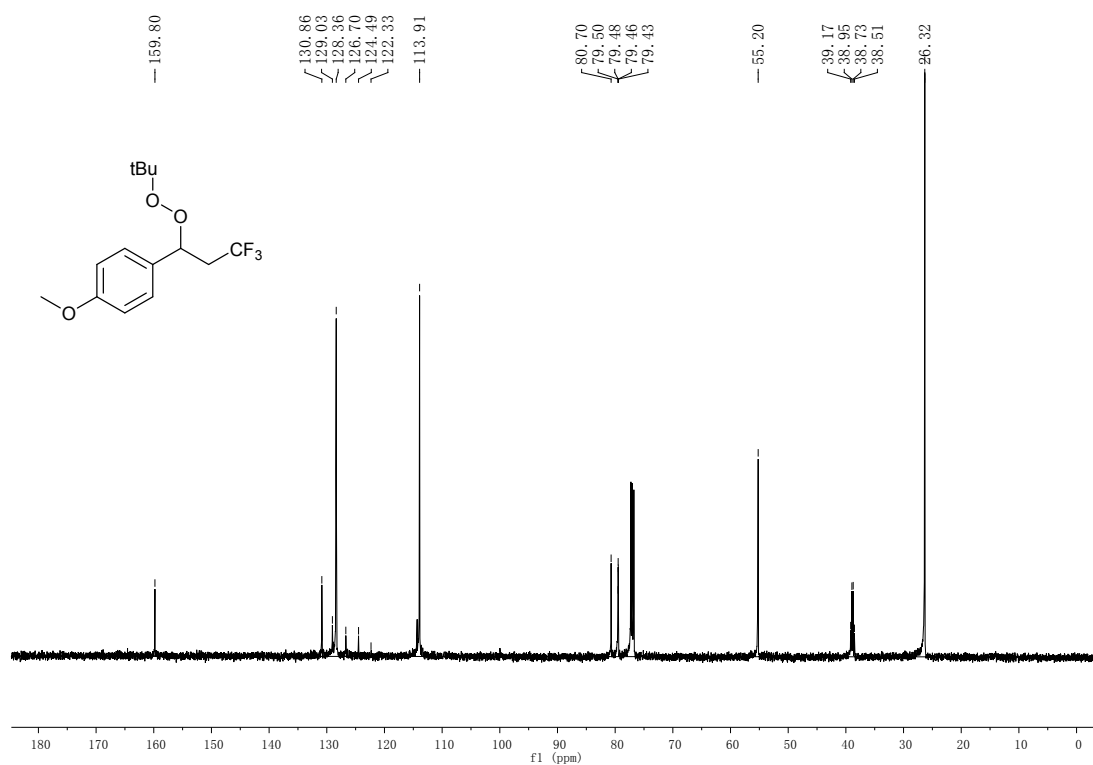


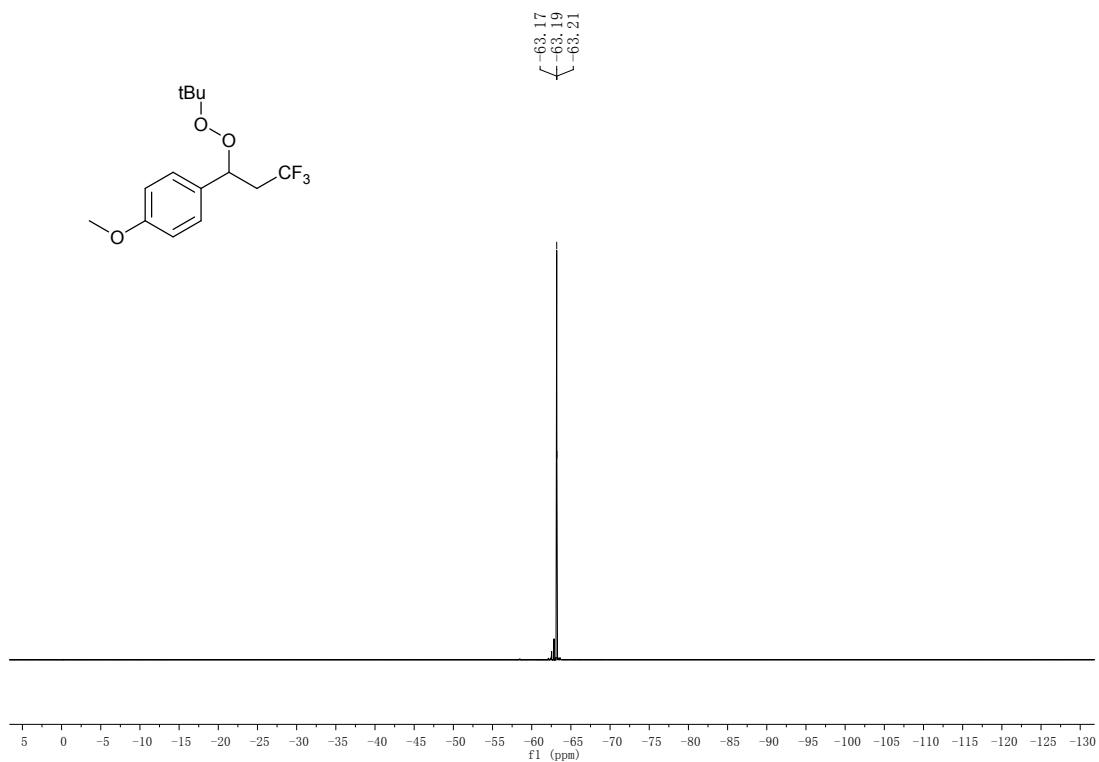
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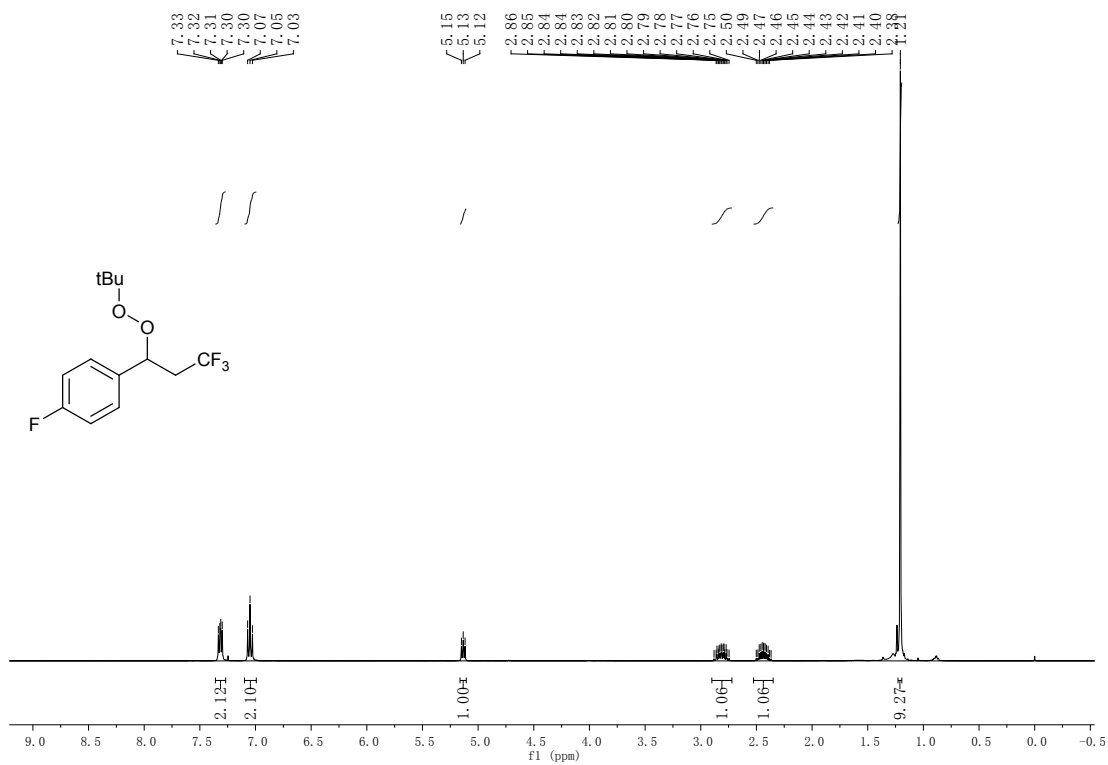


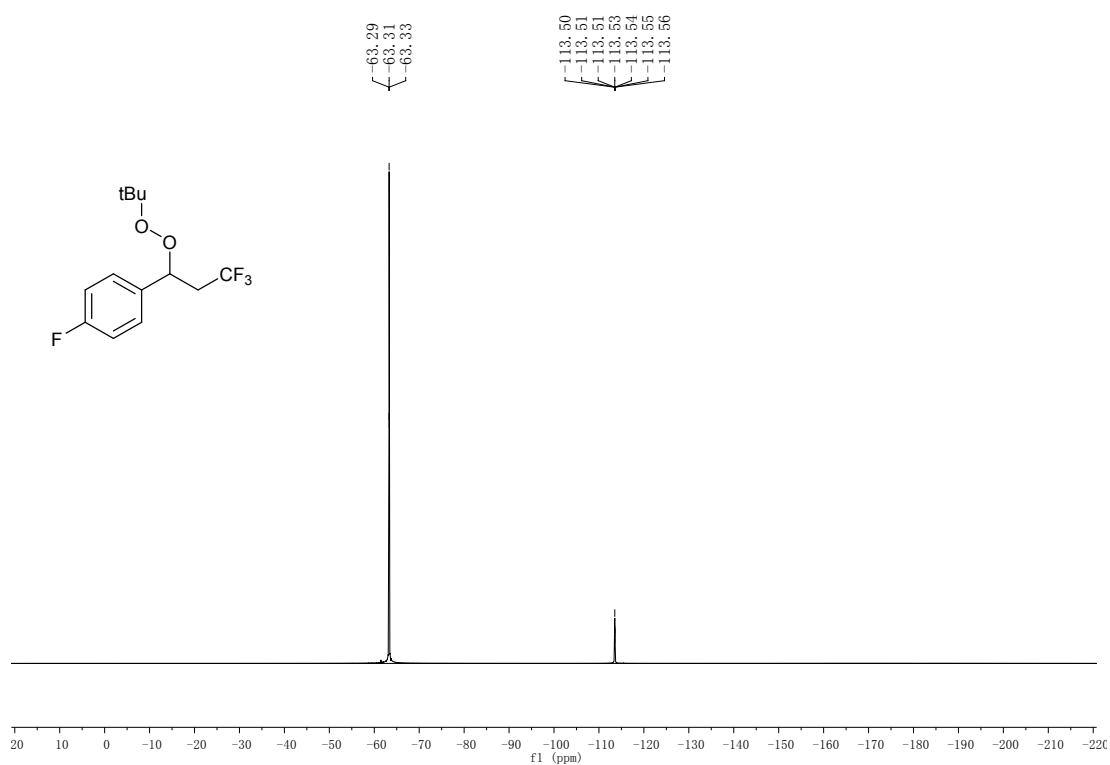
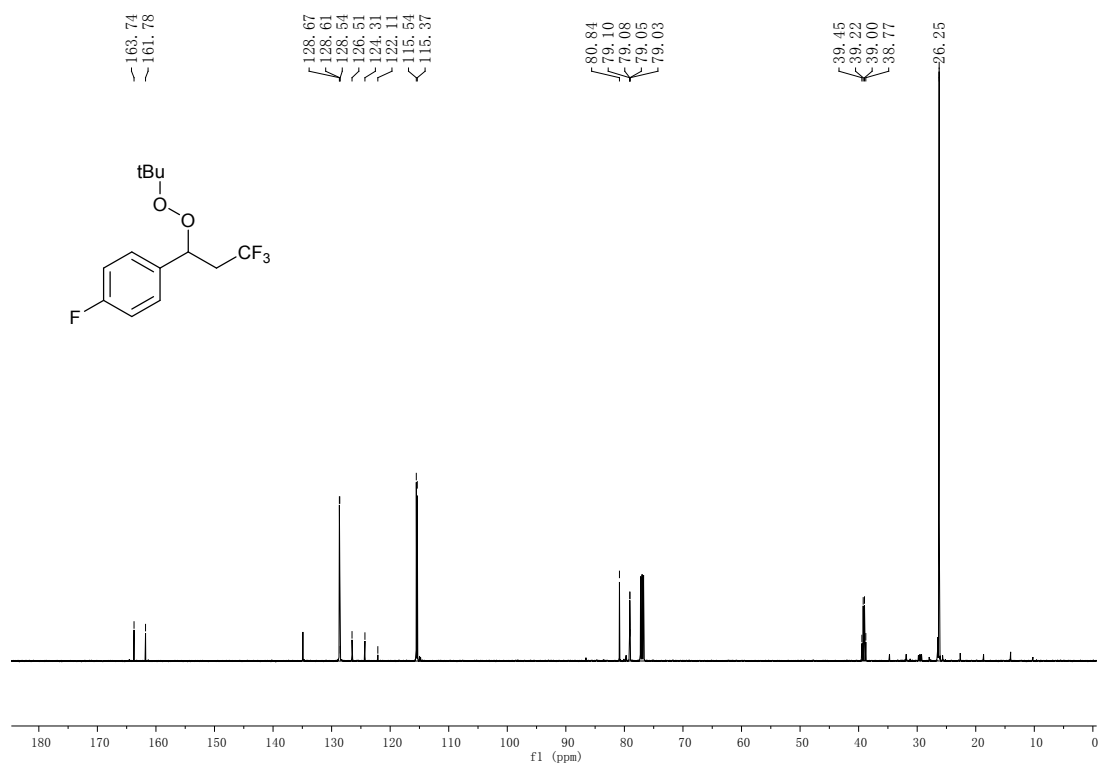
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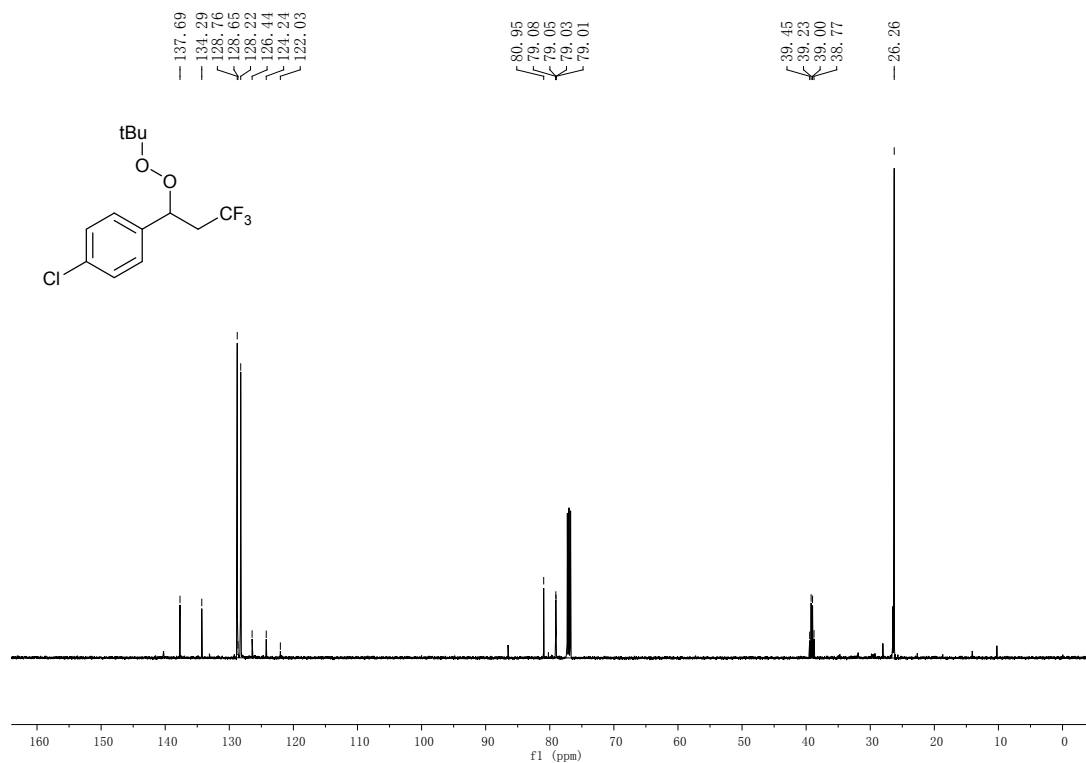
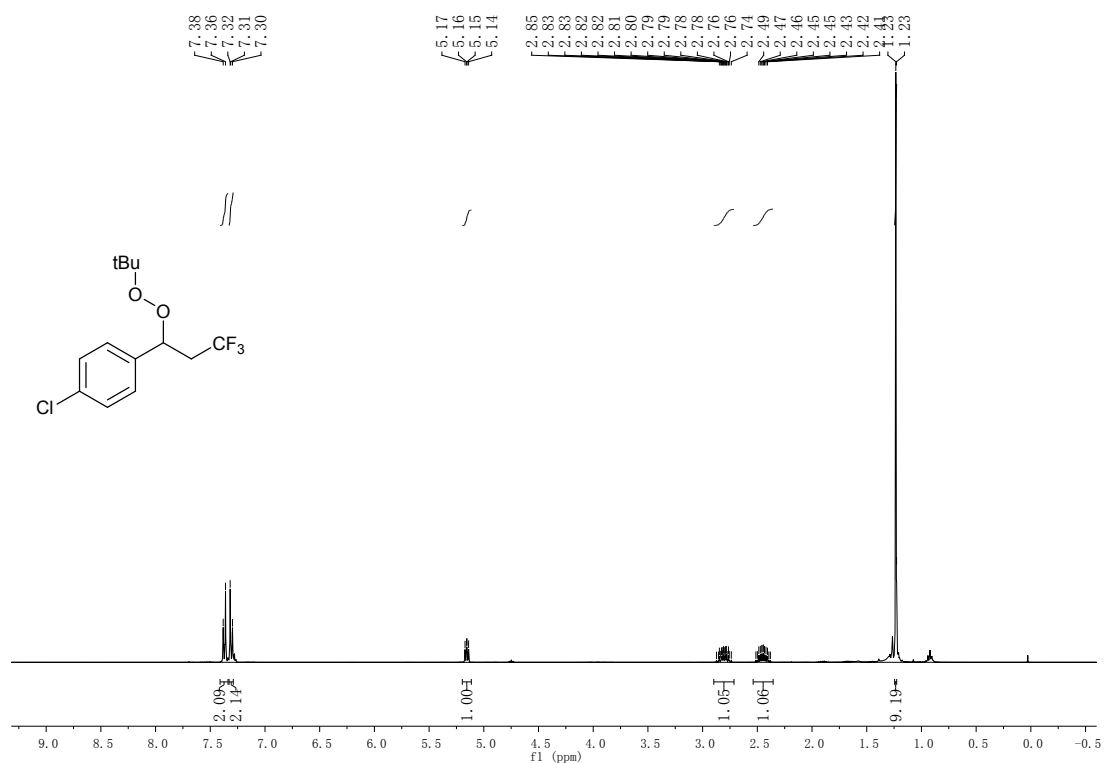


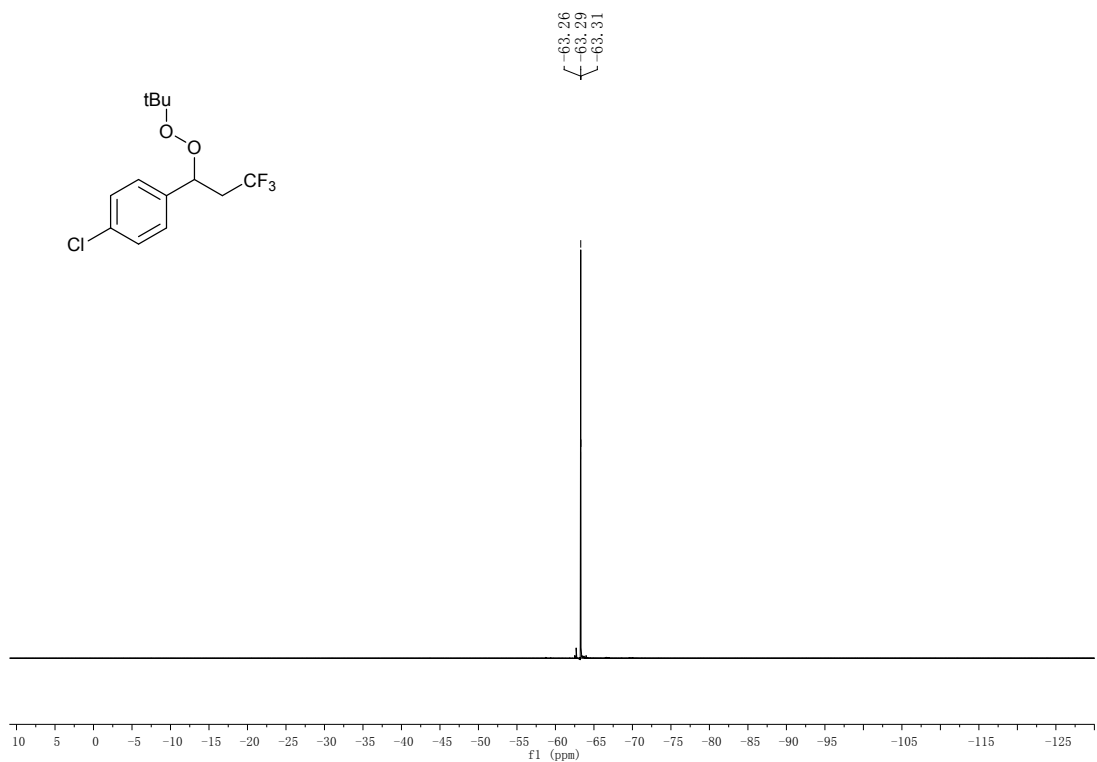
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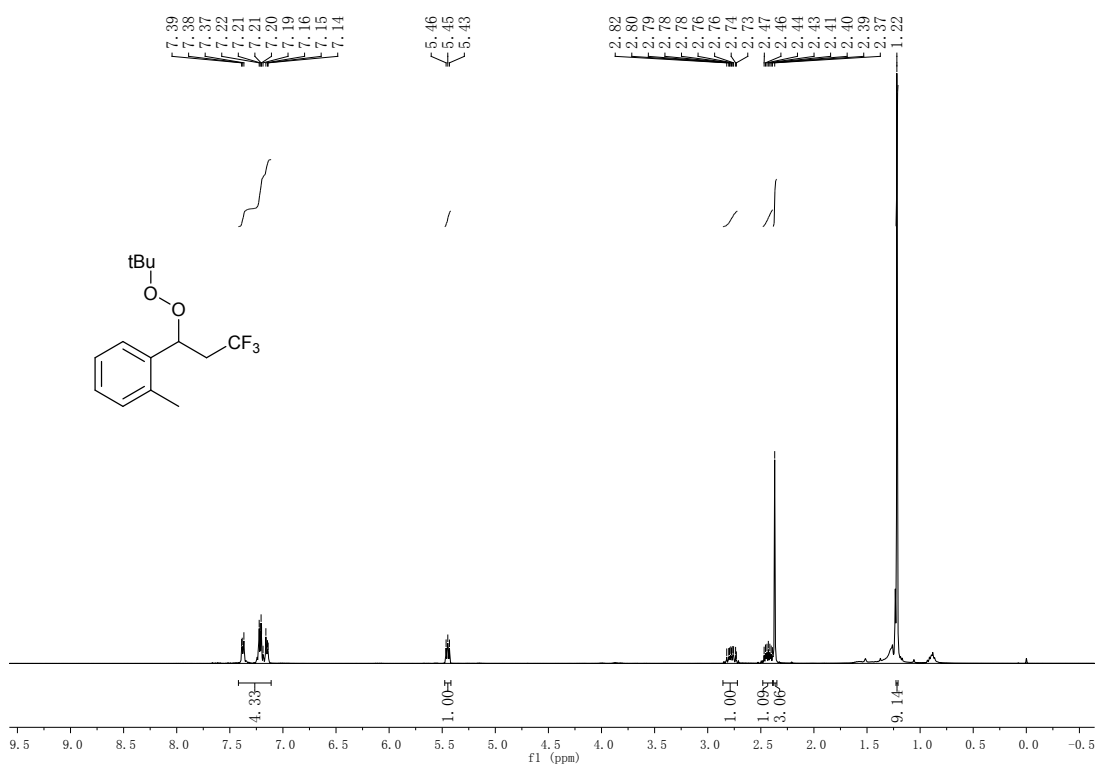


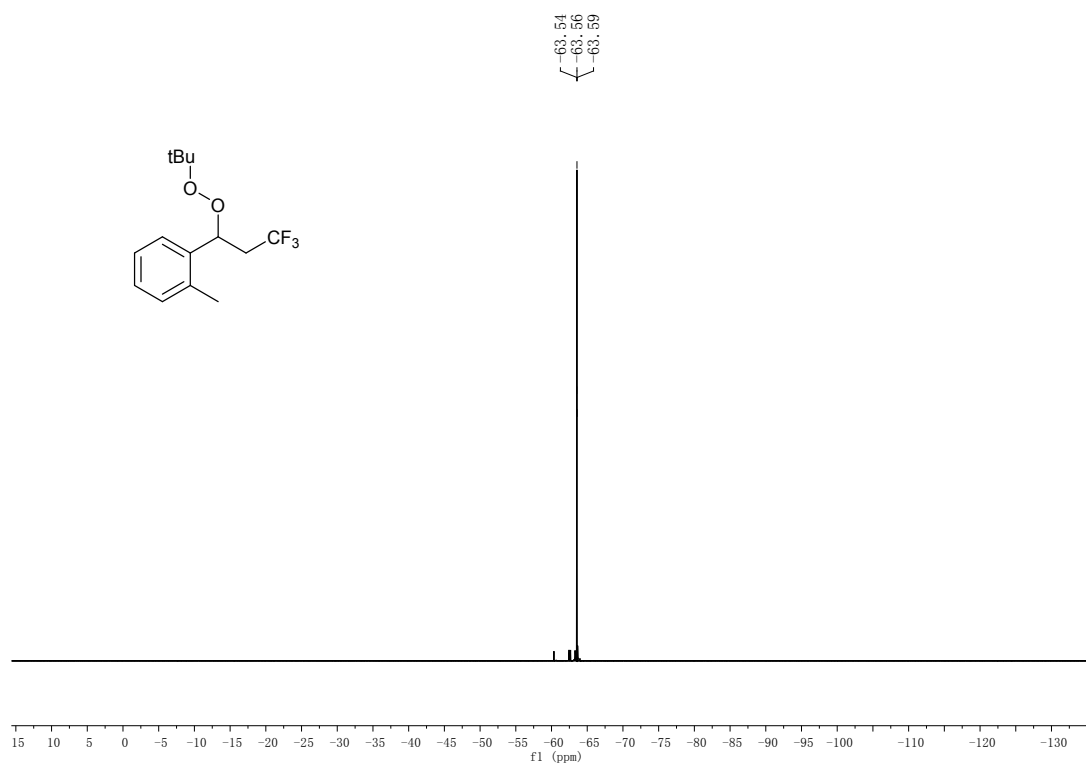
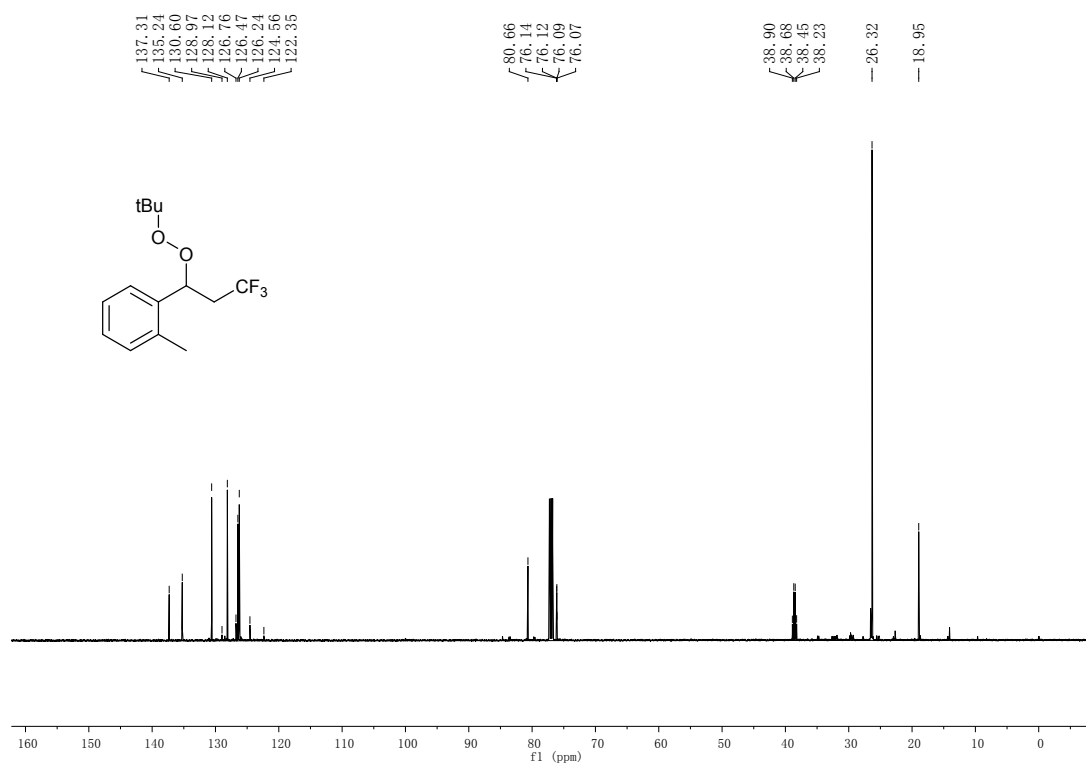
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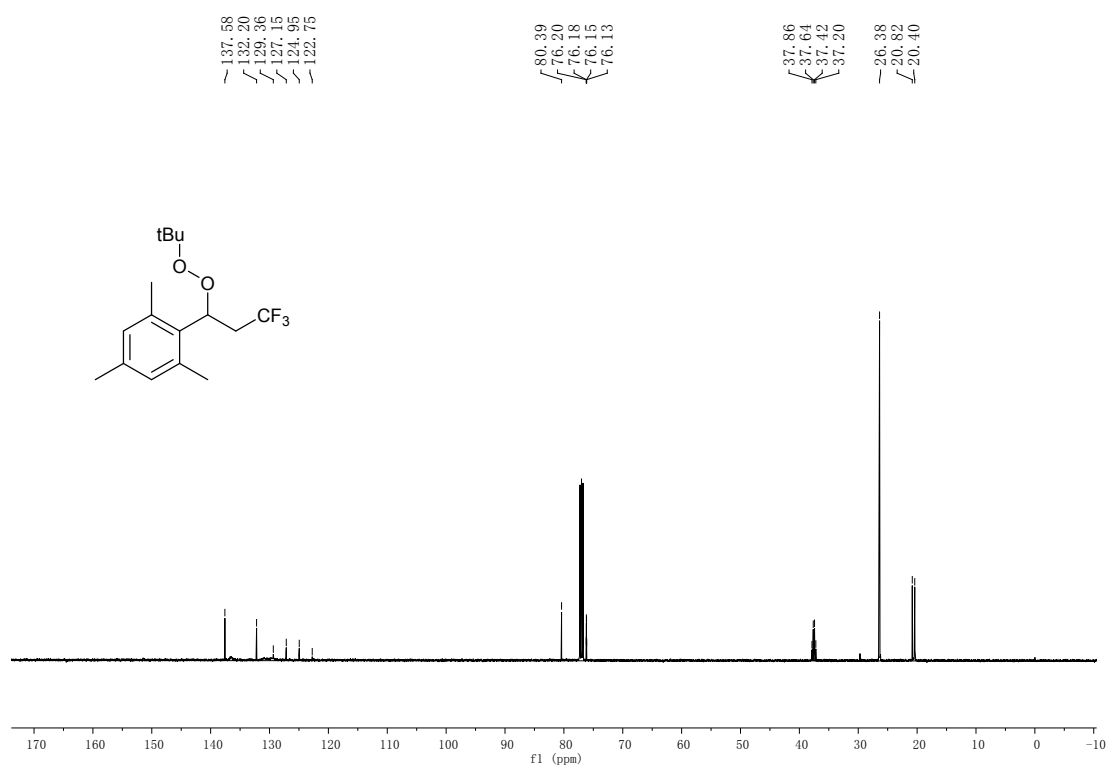
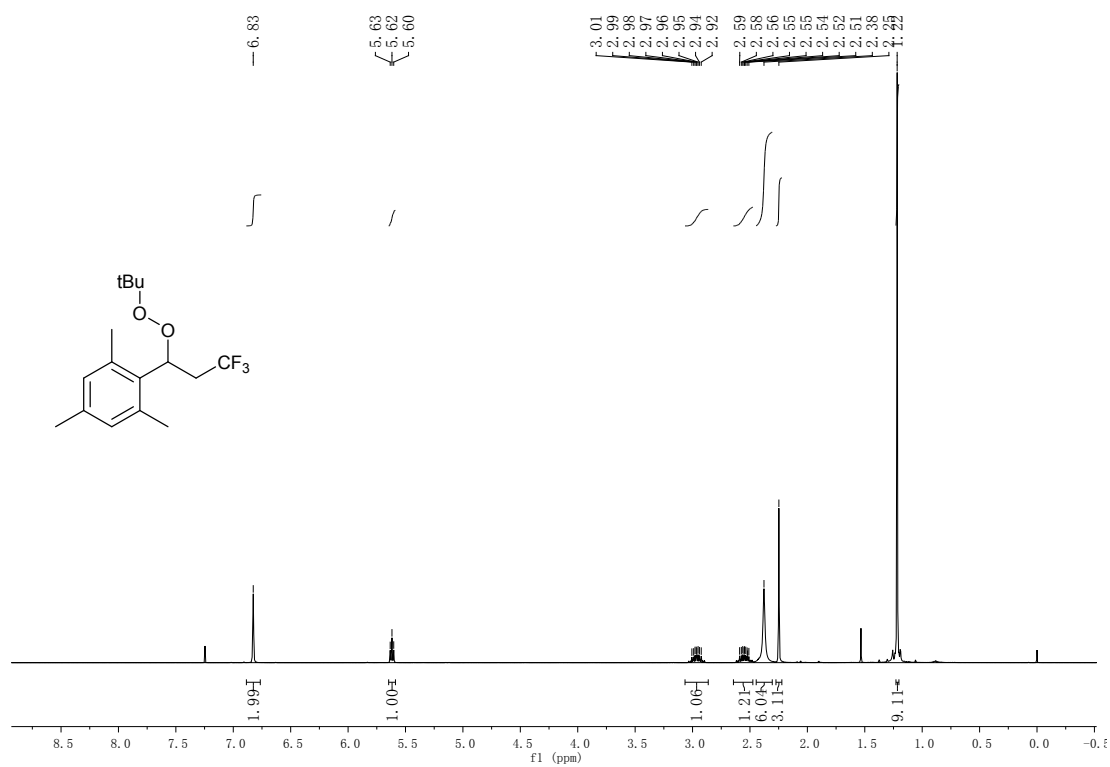


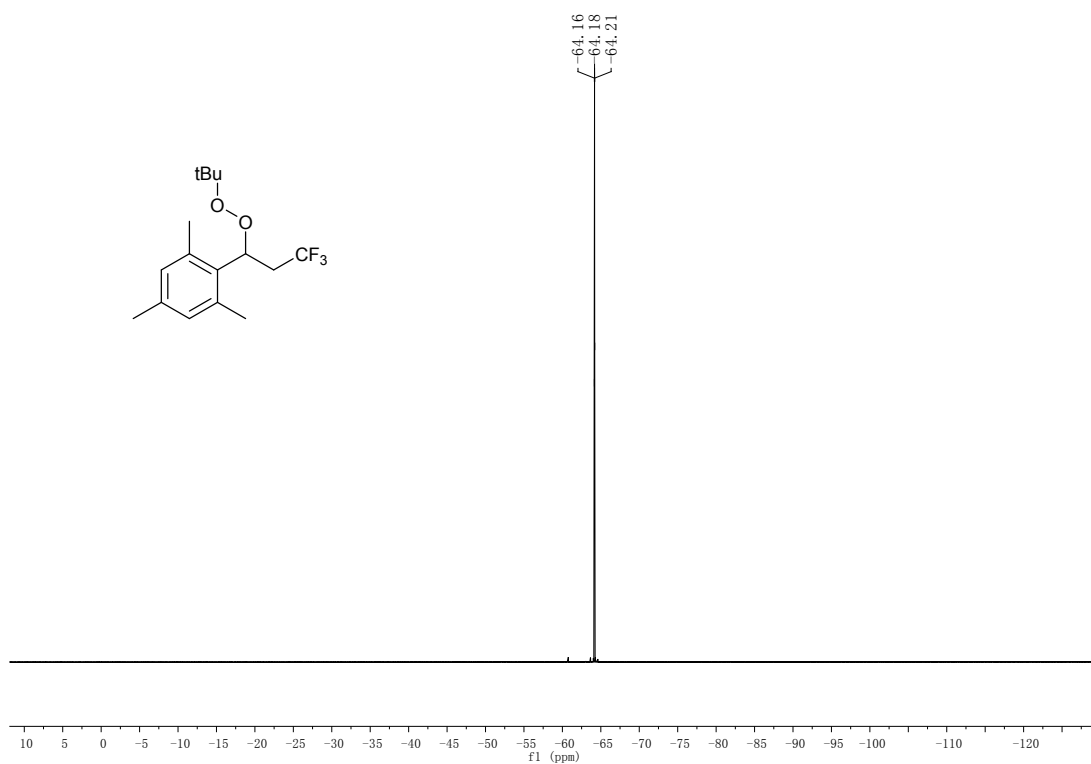
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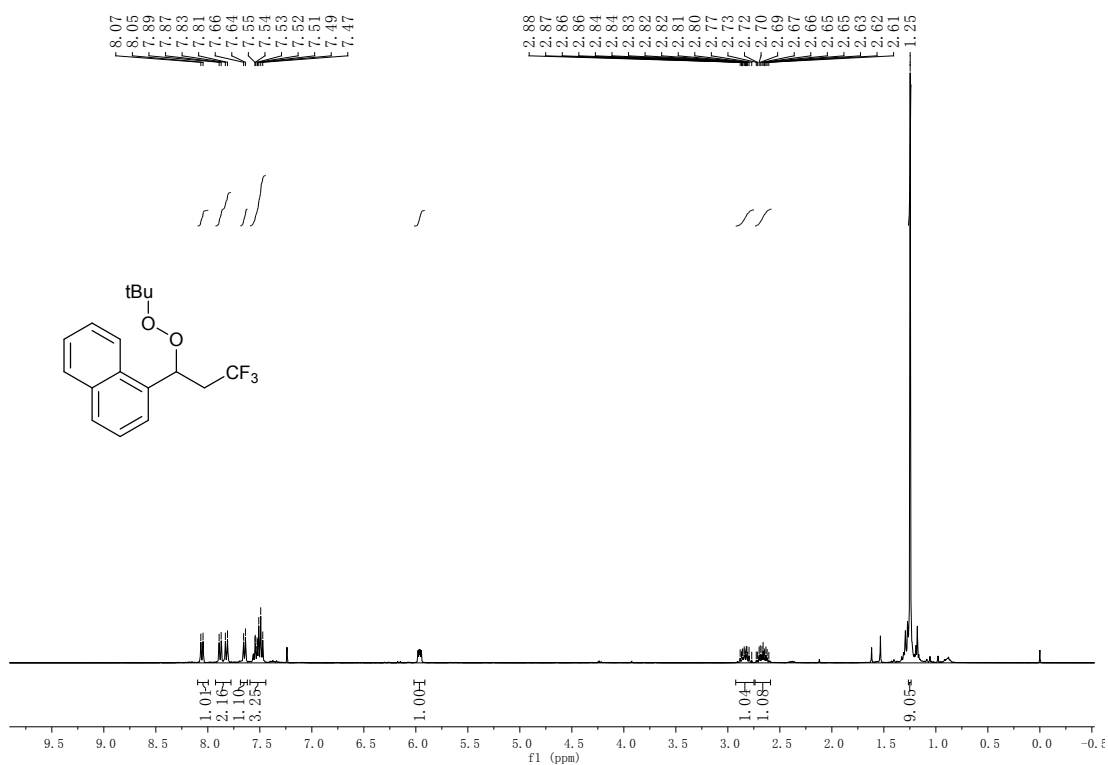


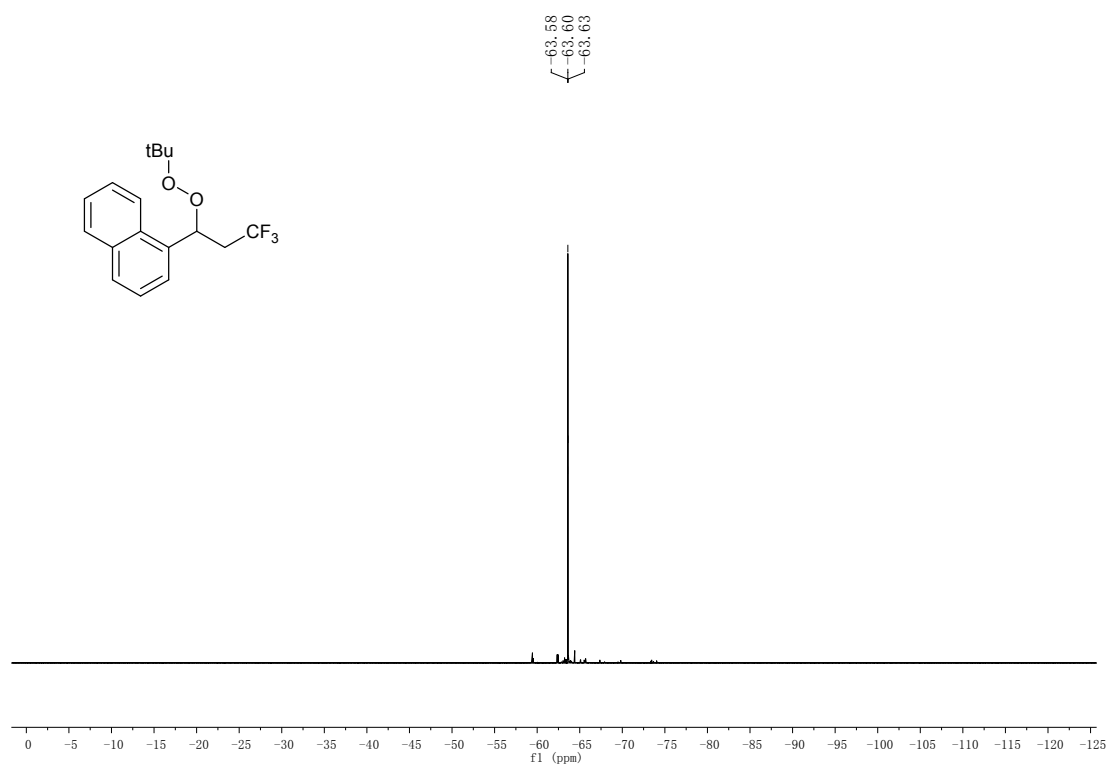
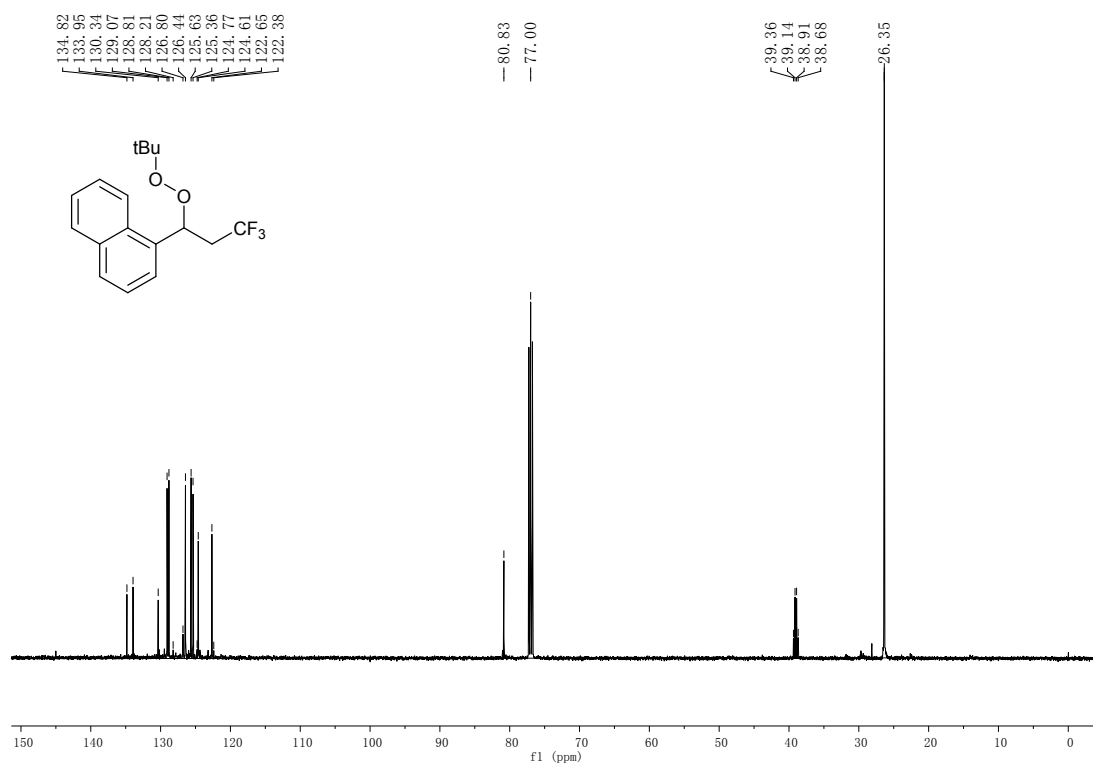
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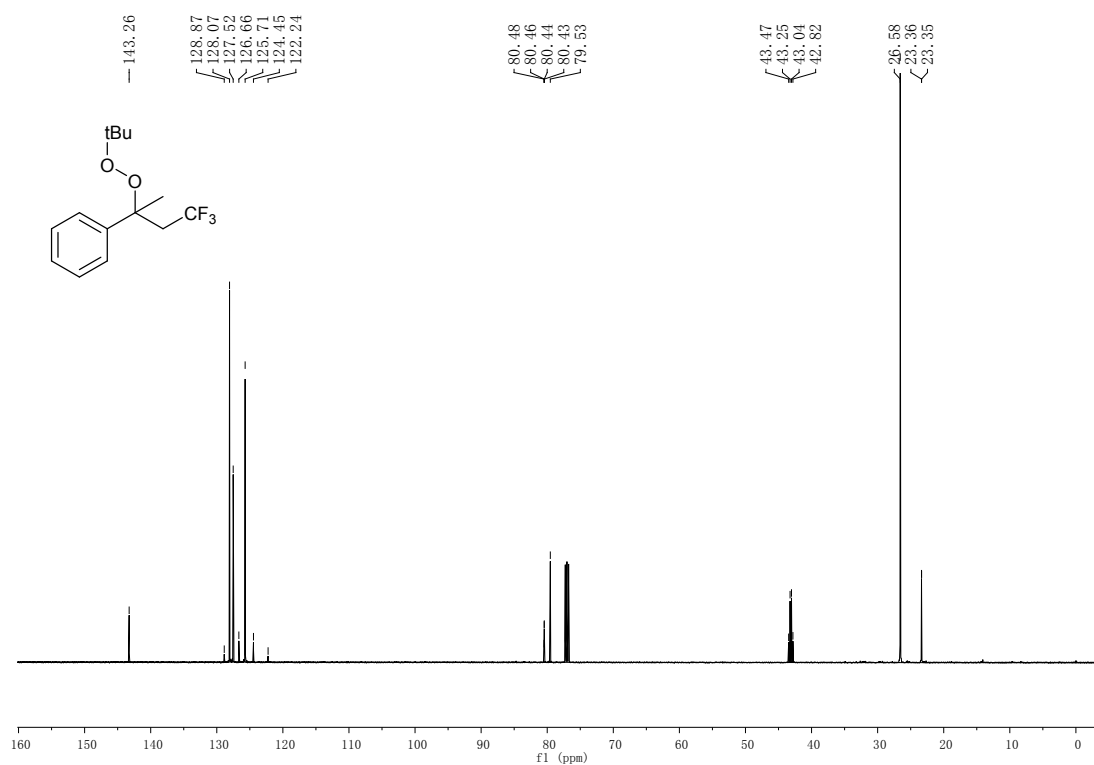
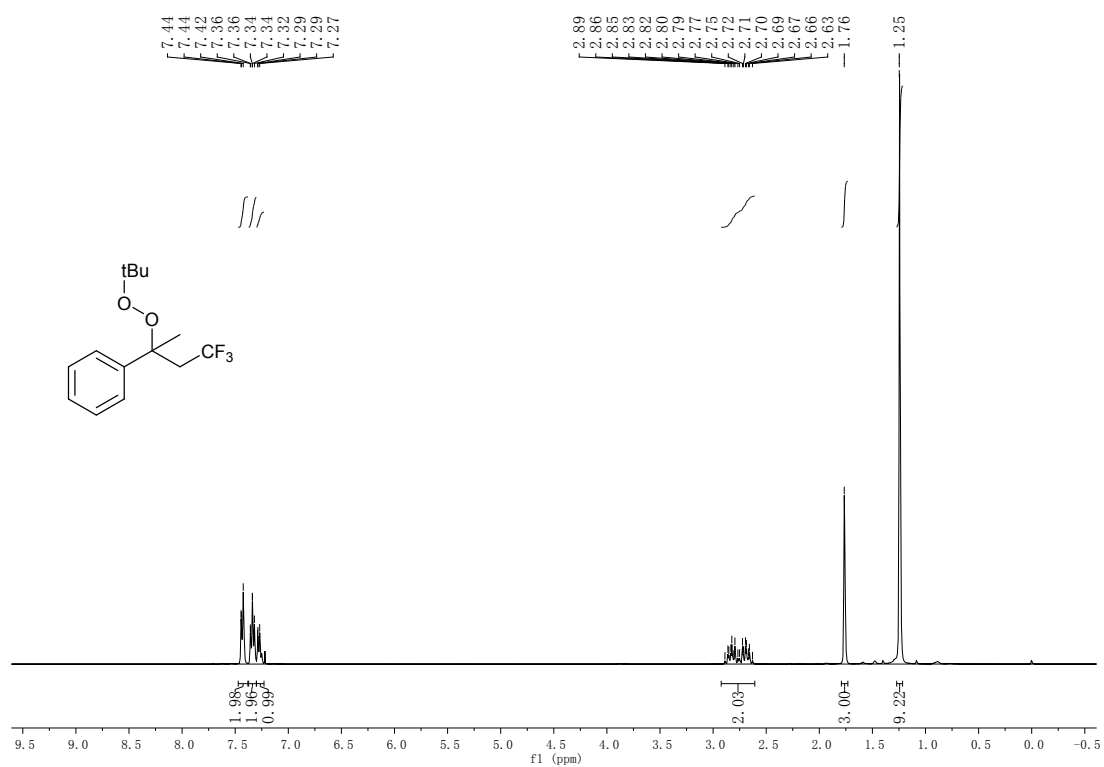


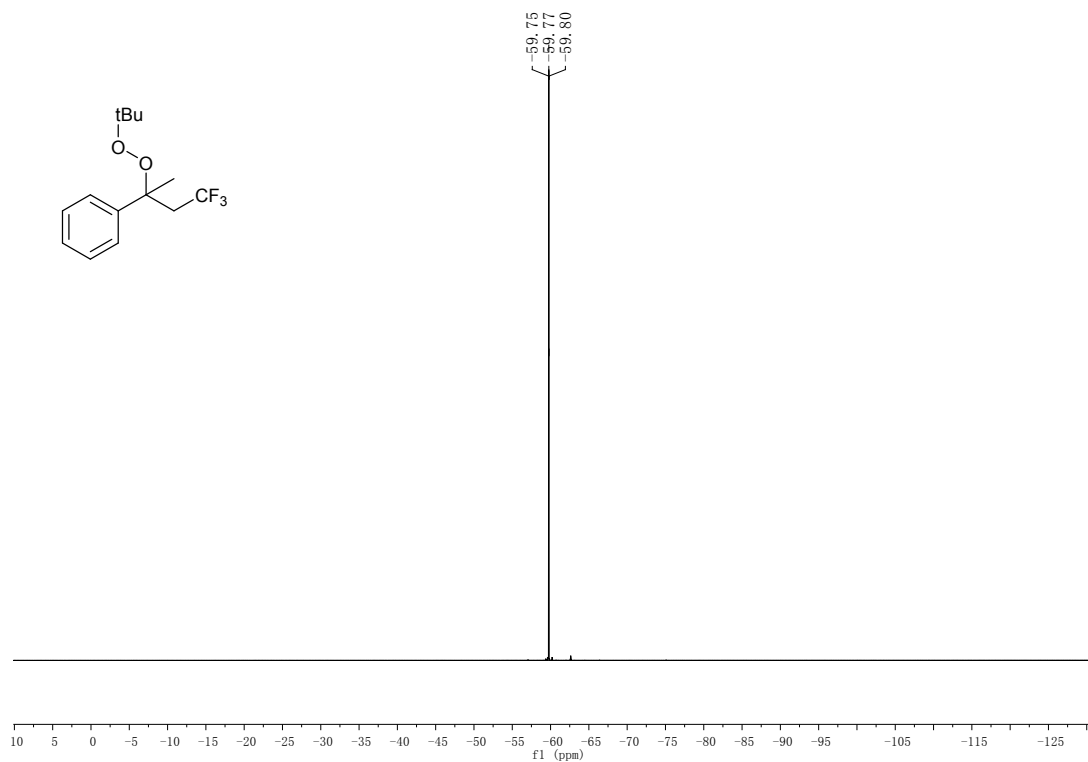
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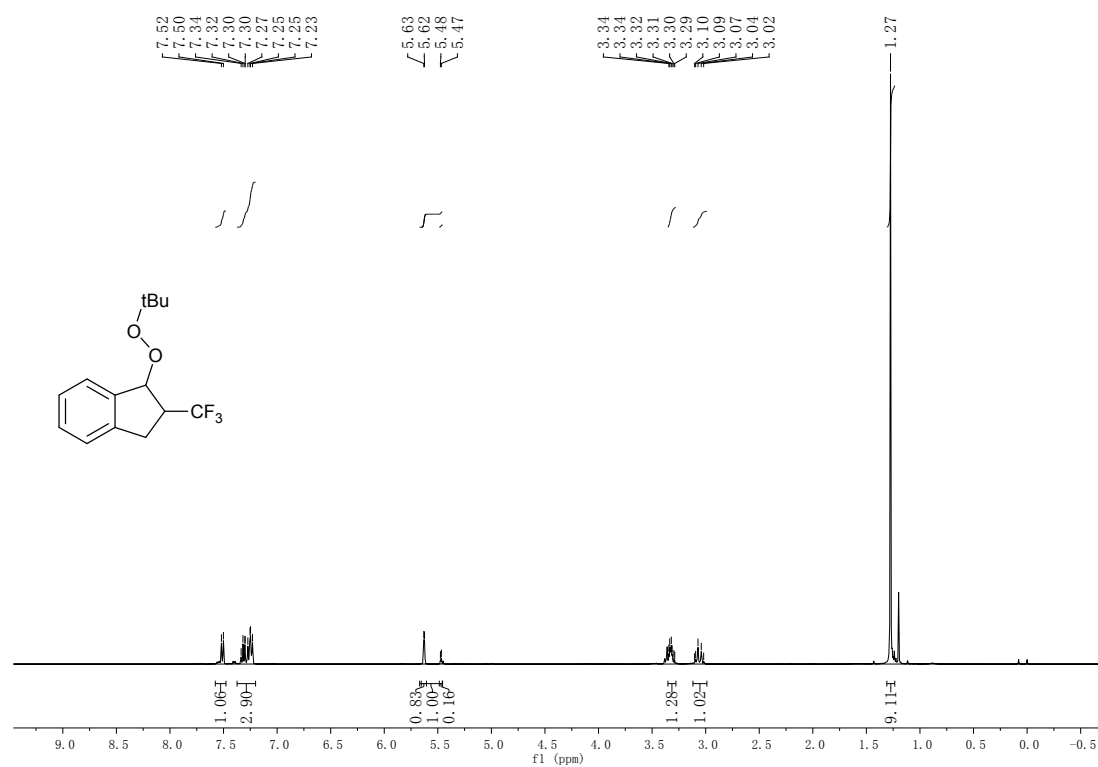


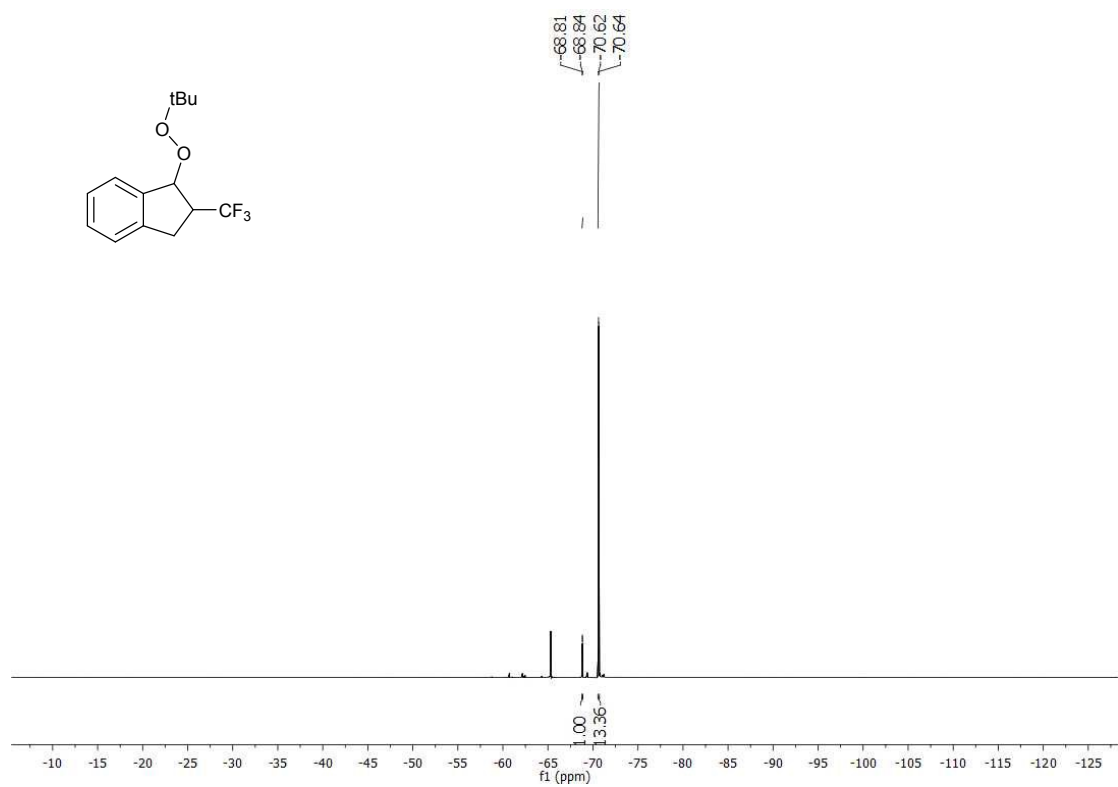
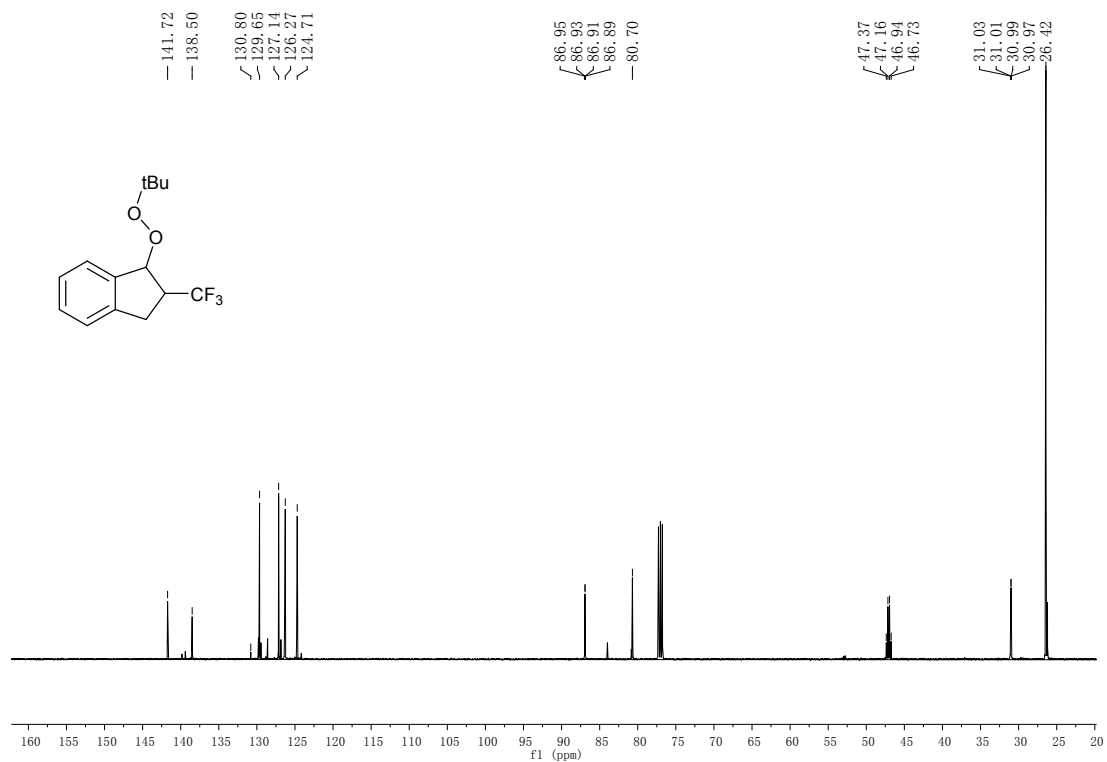
(2-(*tert*-butylperoxy)-4,4,4-trifluorobutan-2-yl)benzene (2k)



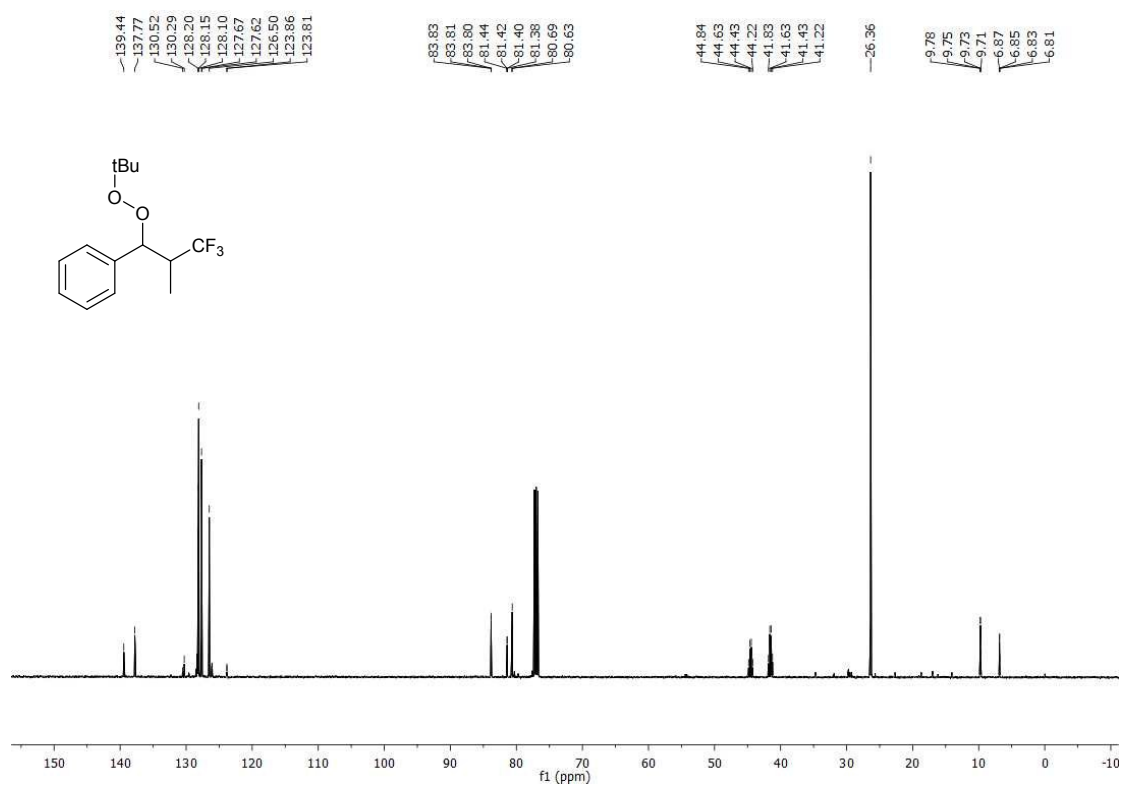
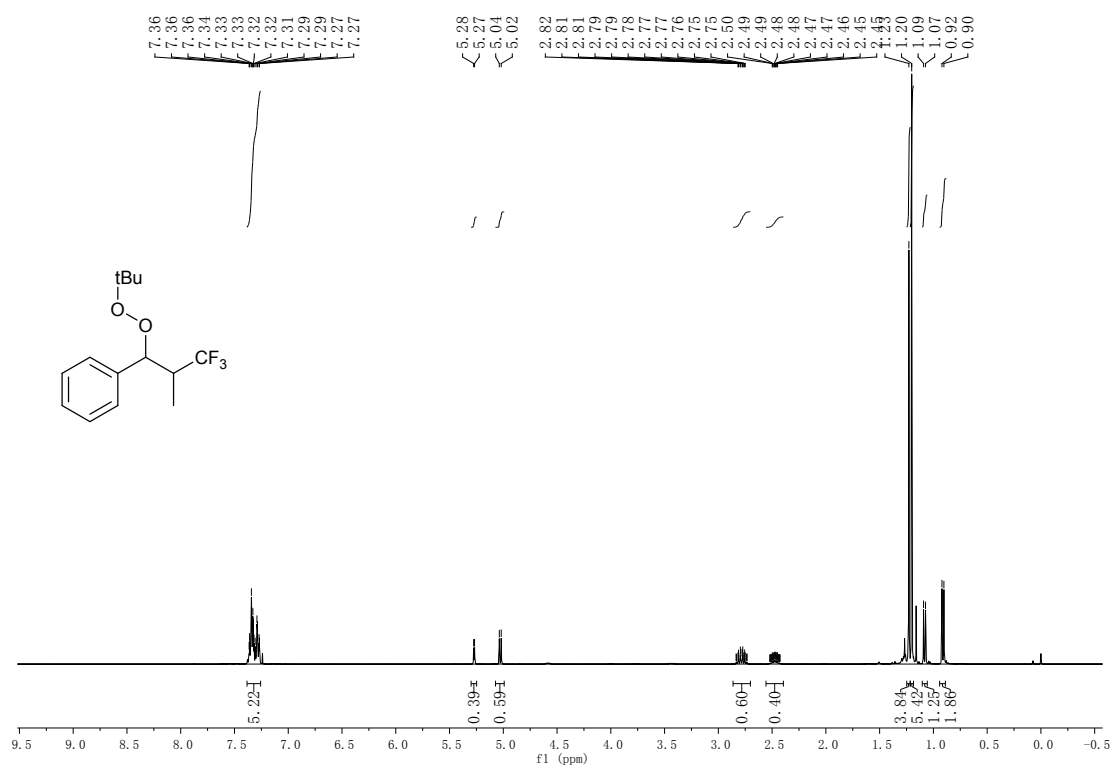


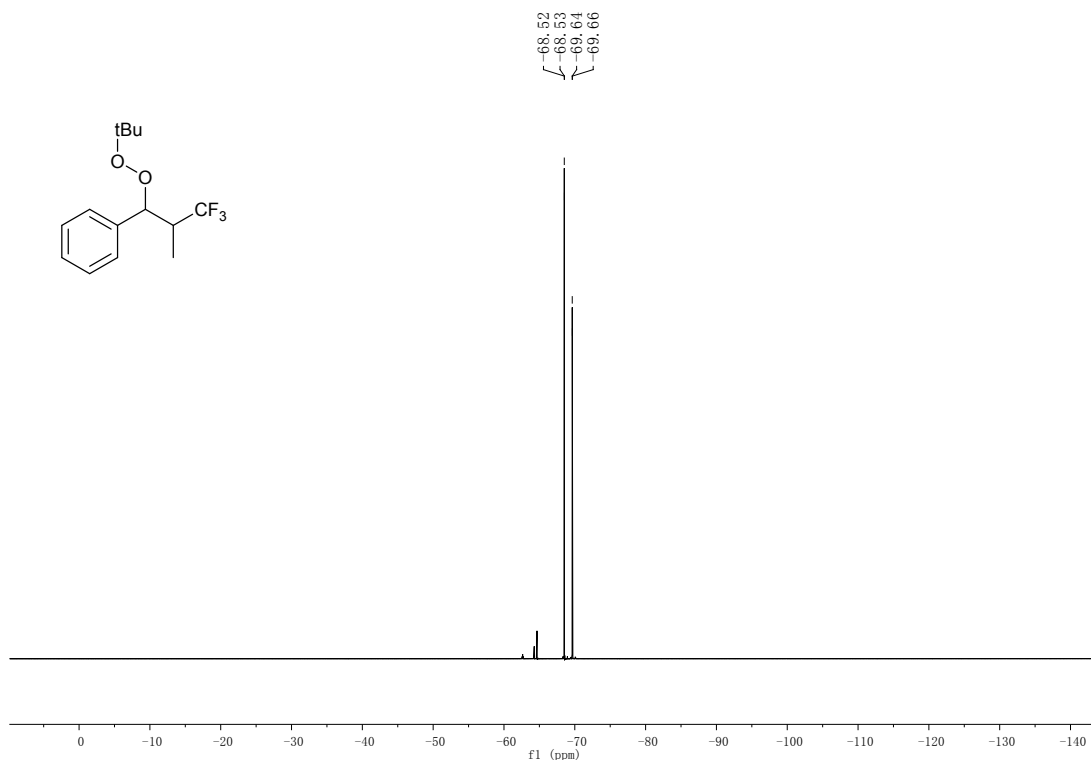
1-(*tert*-butylperoxy)-2-(trifluoromethyl)-2,3-dihydro-1*H*-indene (2l)



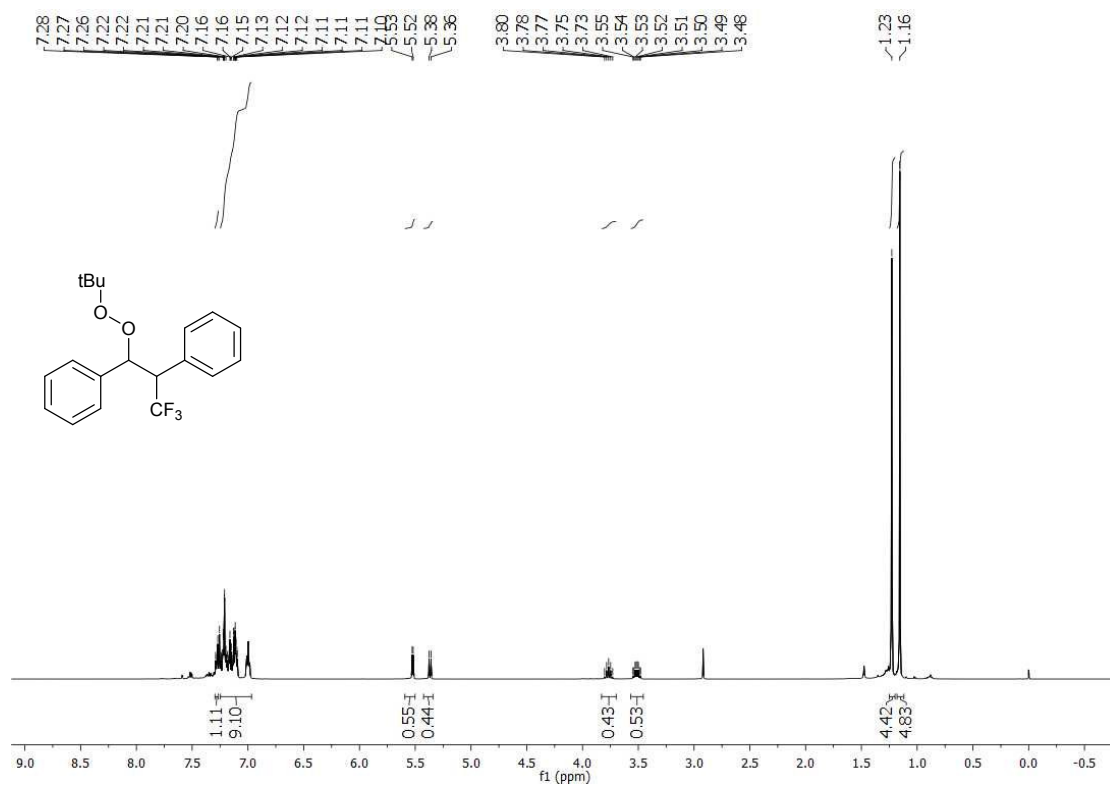


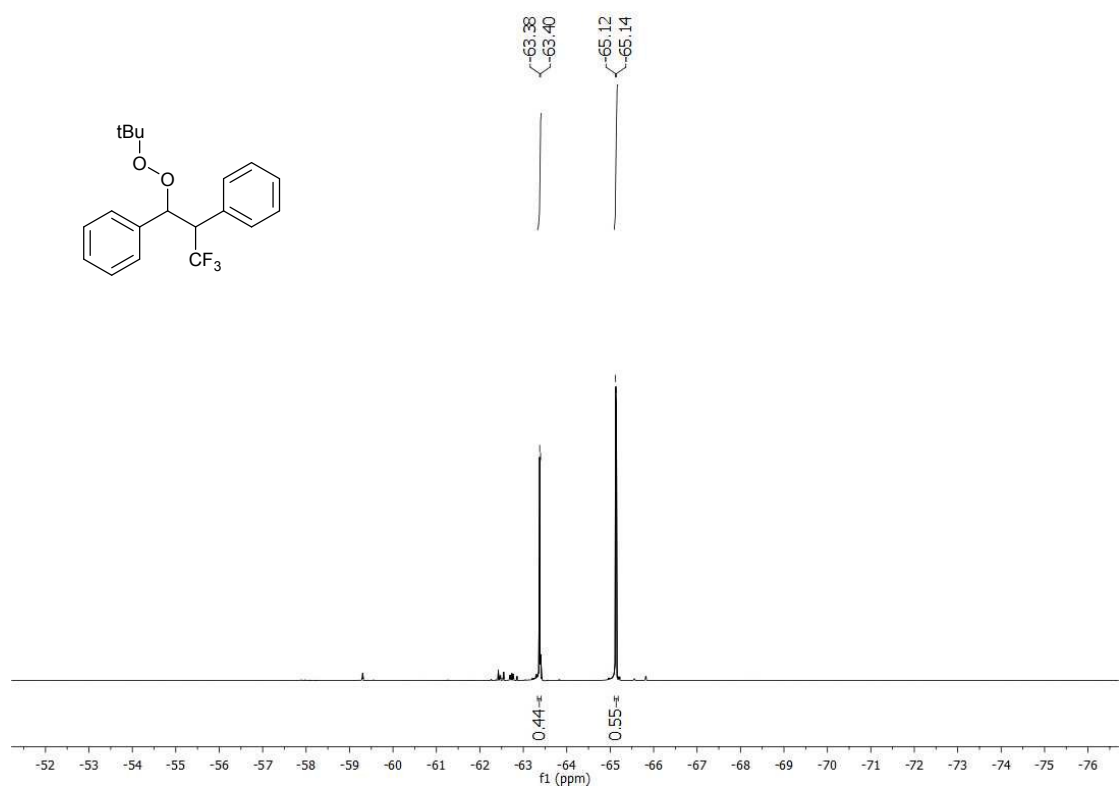
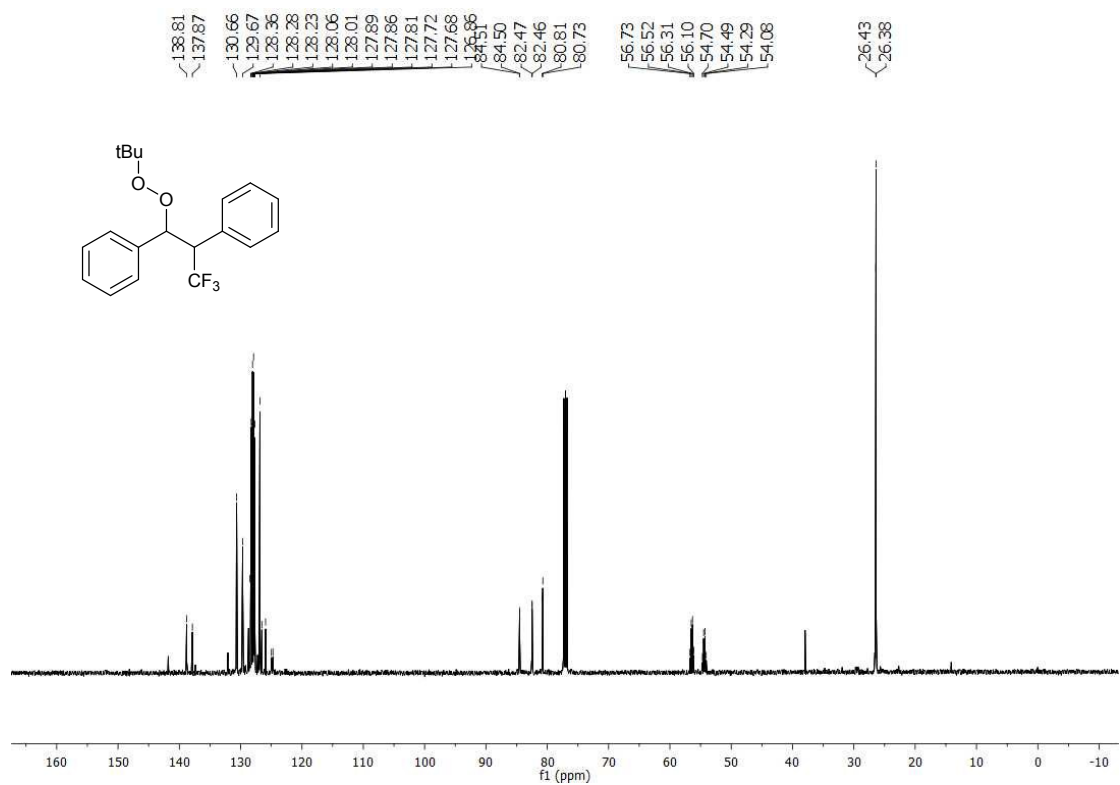
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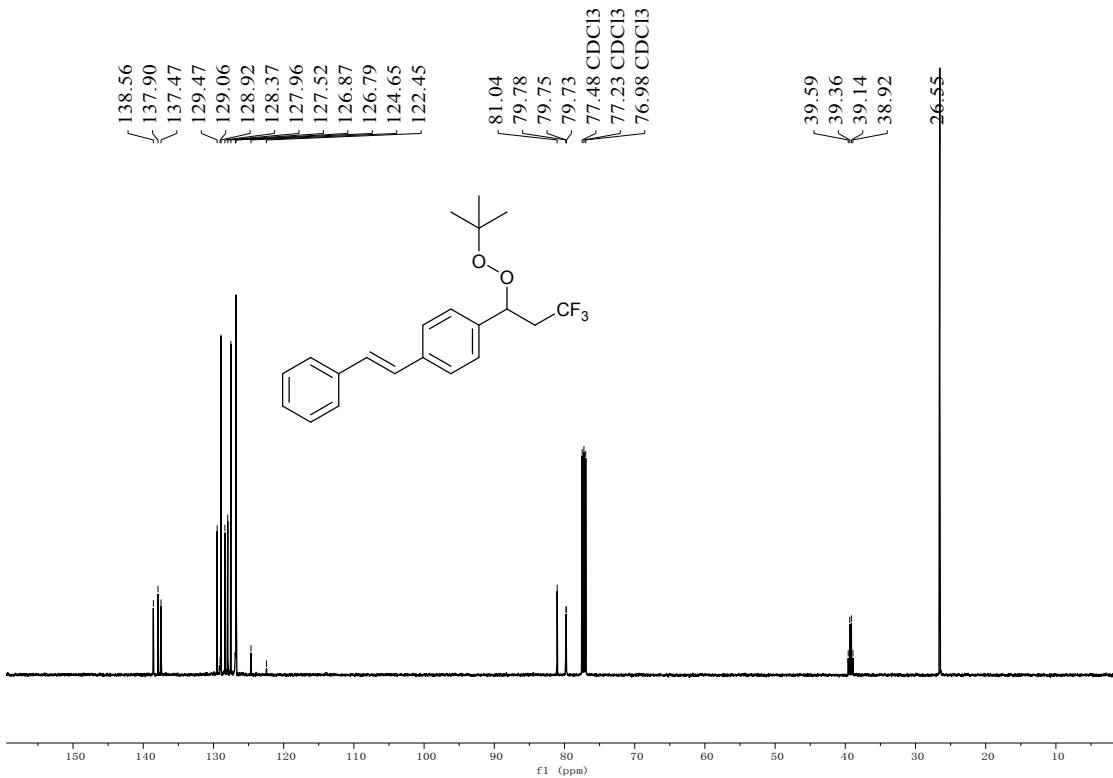
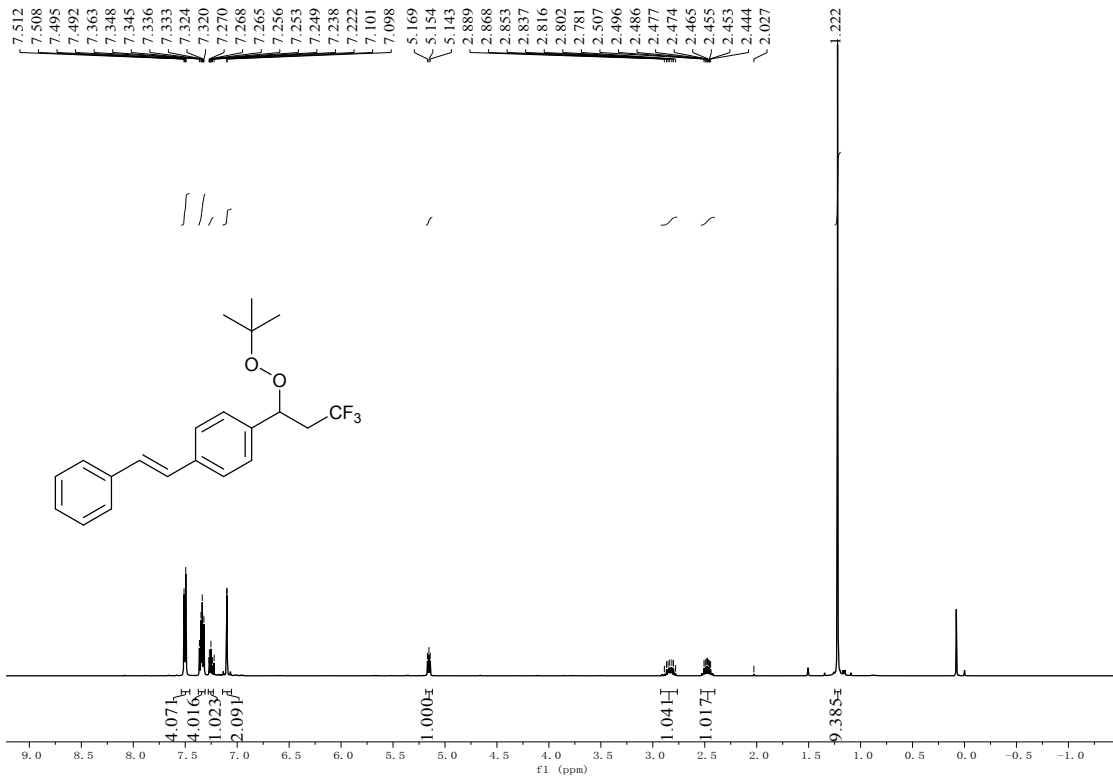


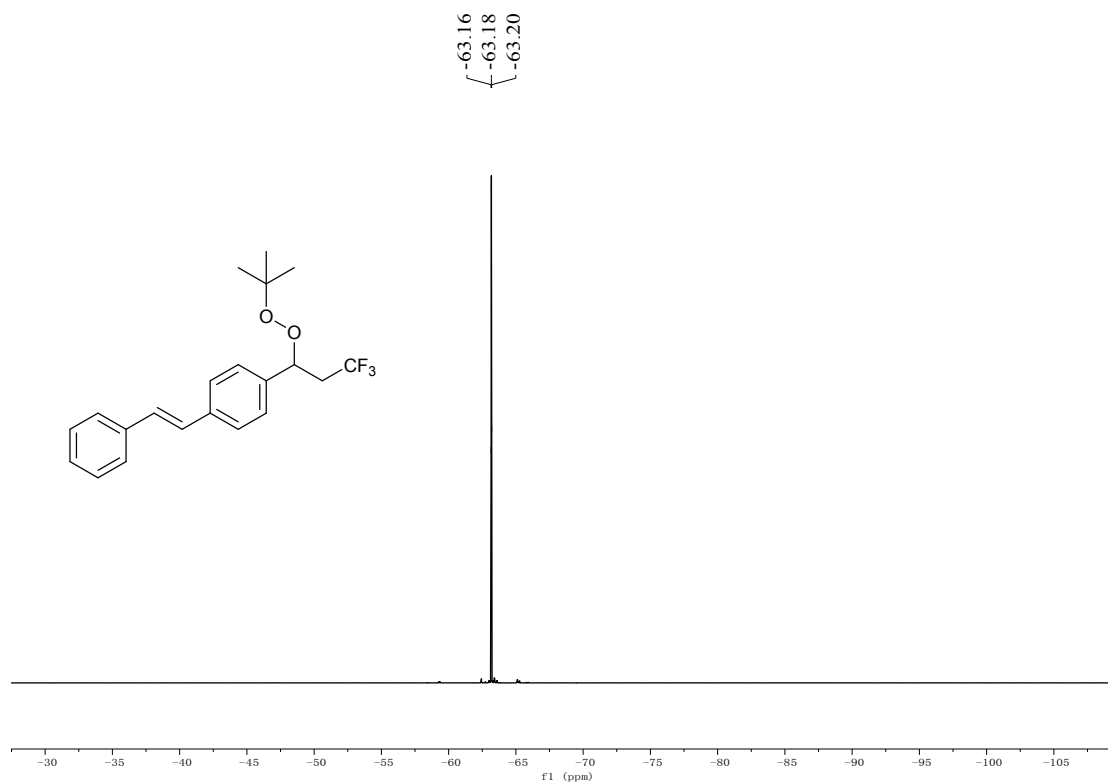
(1-(*tert*-butylperoxy)-3,3,3-trifluoropropane-1,2-diyl)dibenzene (2n)



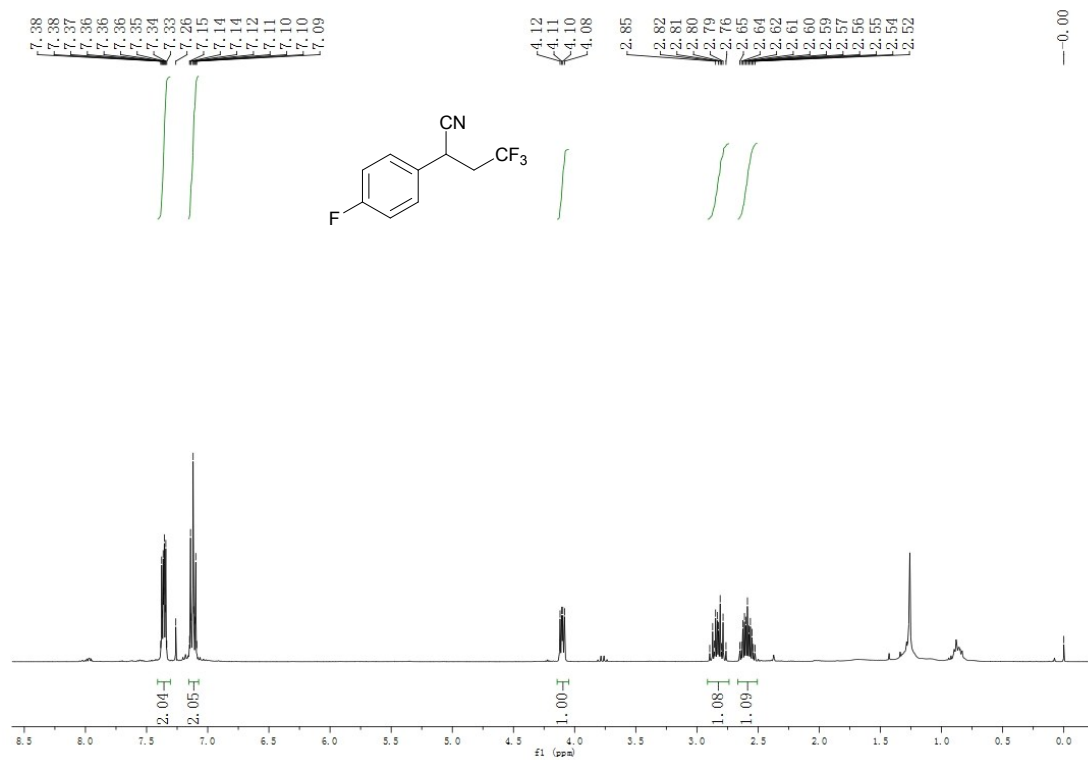


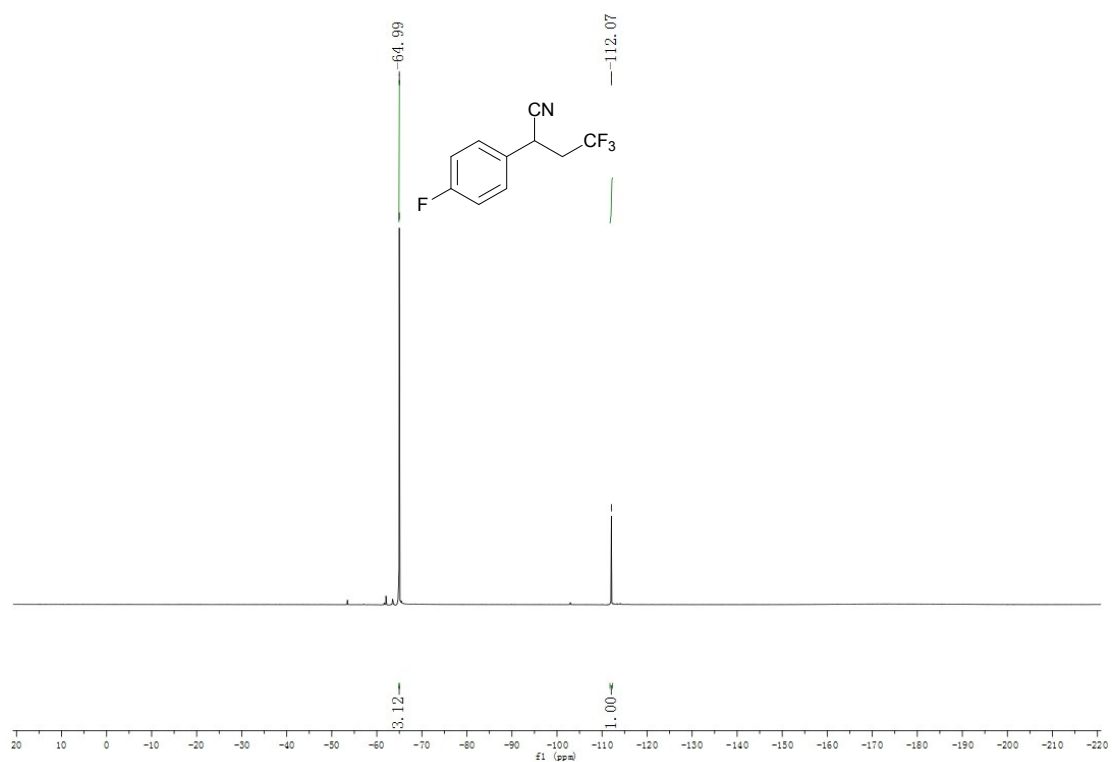
(E)-1-(1-(*tert*-butylperoxy)-3,3,3-trifluoropropyl)-4-styrylbenzene (2o)



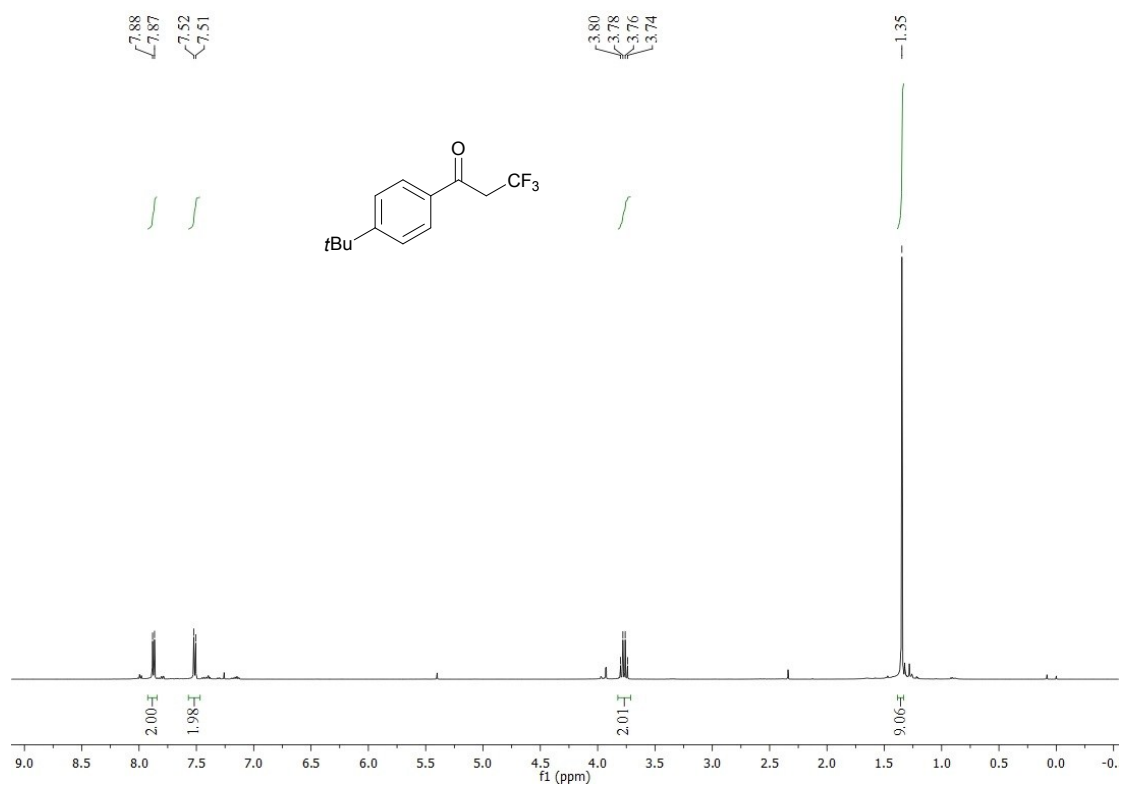


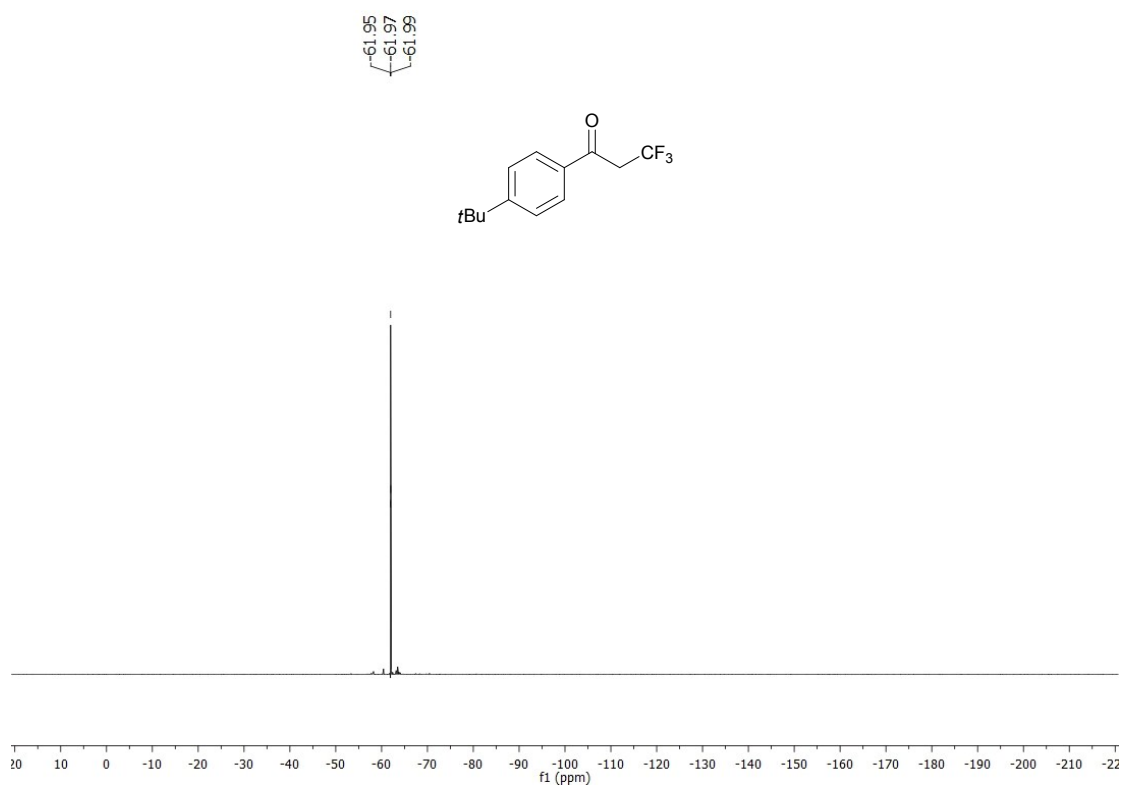
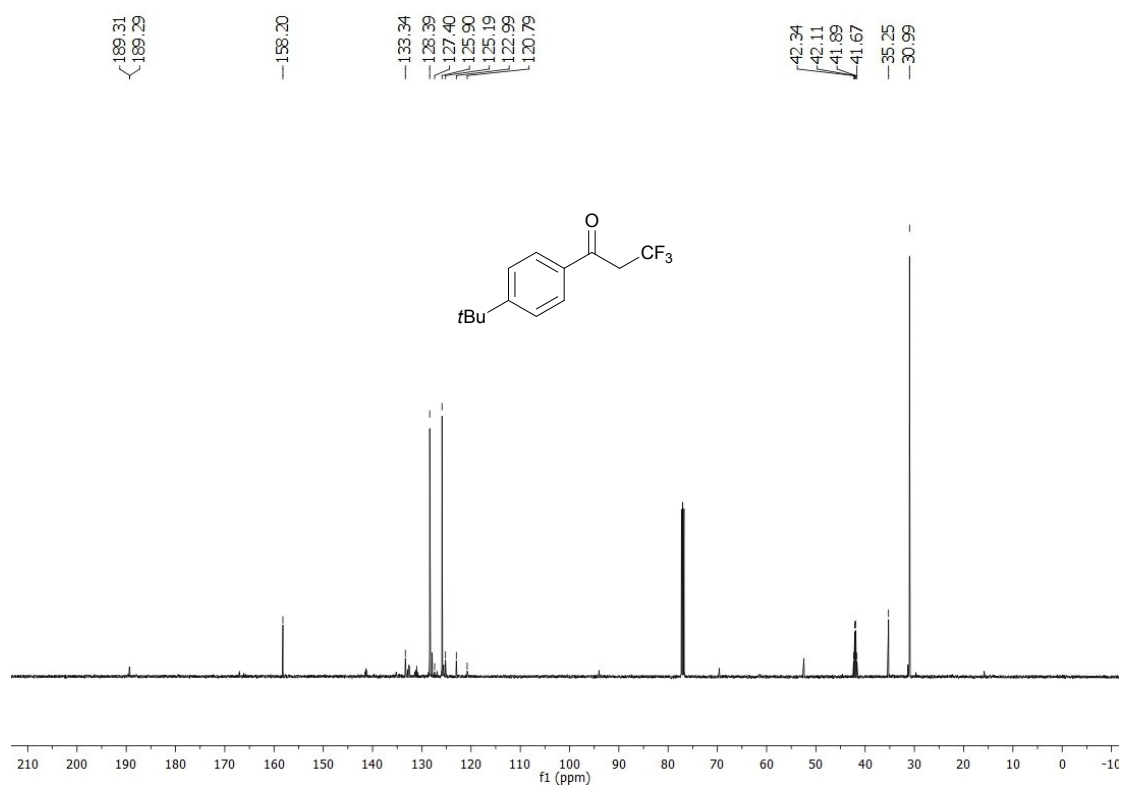
4,4,4-trifluoro-2-(4-fluorophenyl)butanenitrile (3)



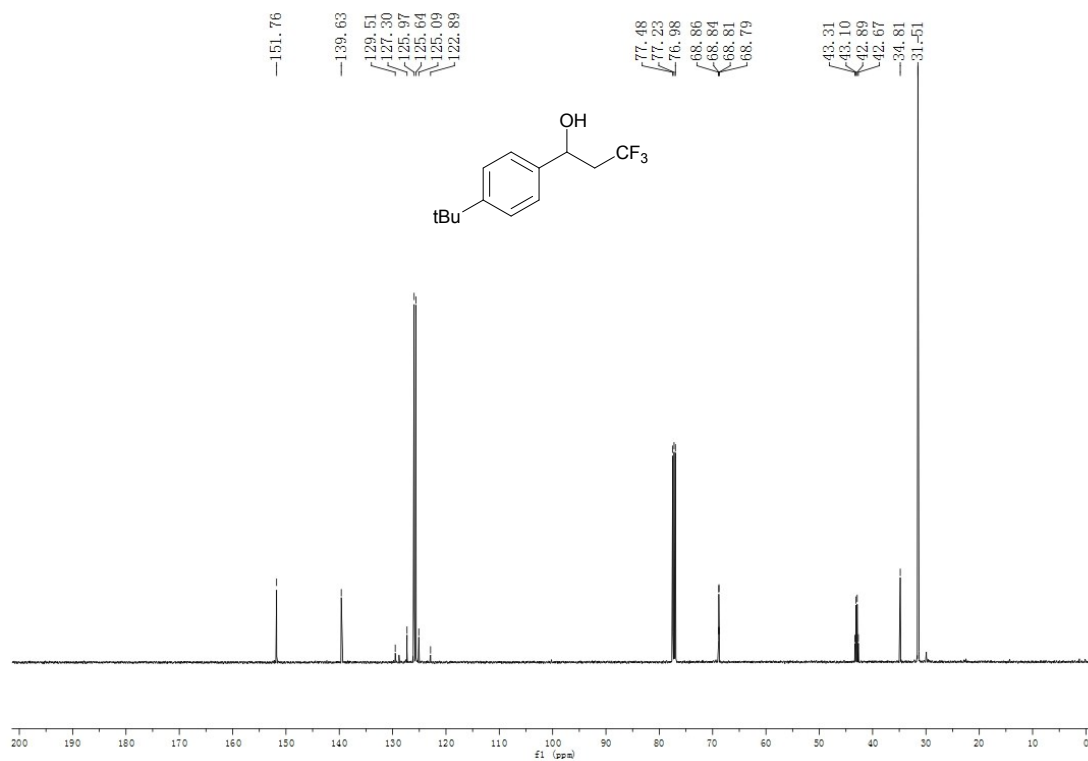
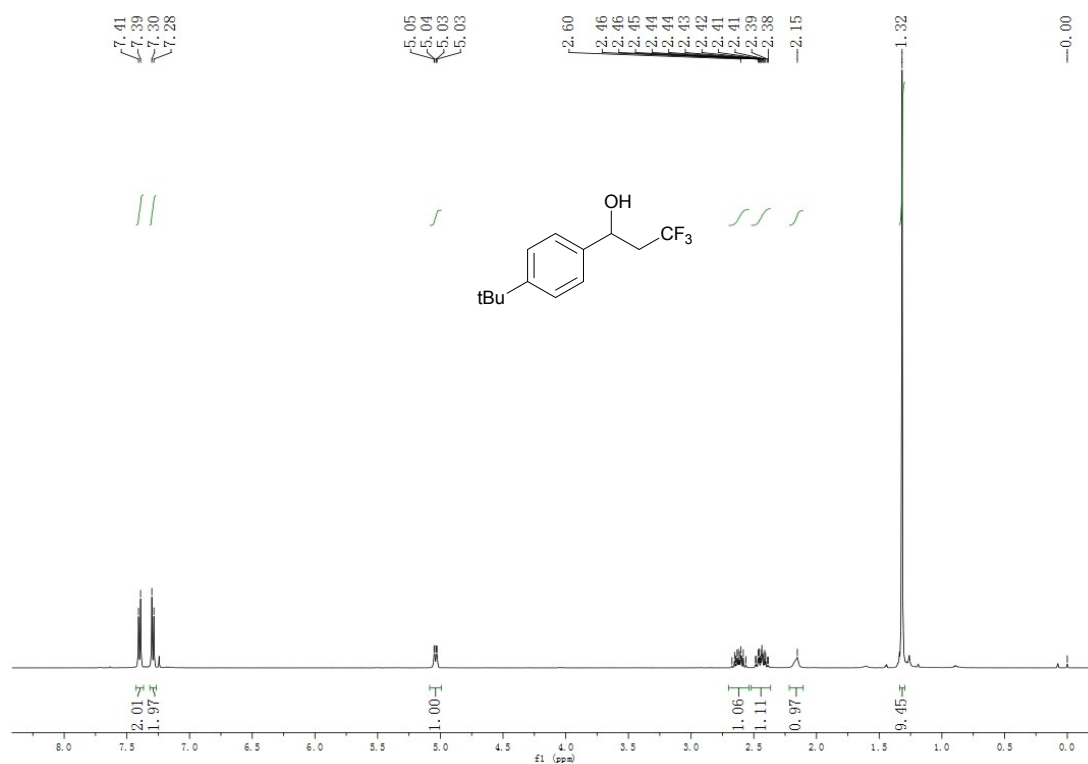


1-(4-(*tert*-butyl)phenyl)-3,3,3-trifluoropropan-1-one (4)





1-(4-(*tert*-butyl)phenyl)-3,3,3-trifluoropropan-1-ol (5)



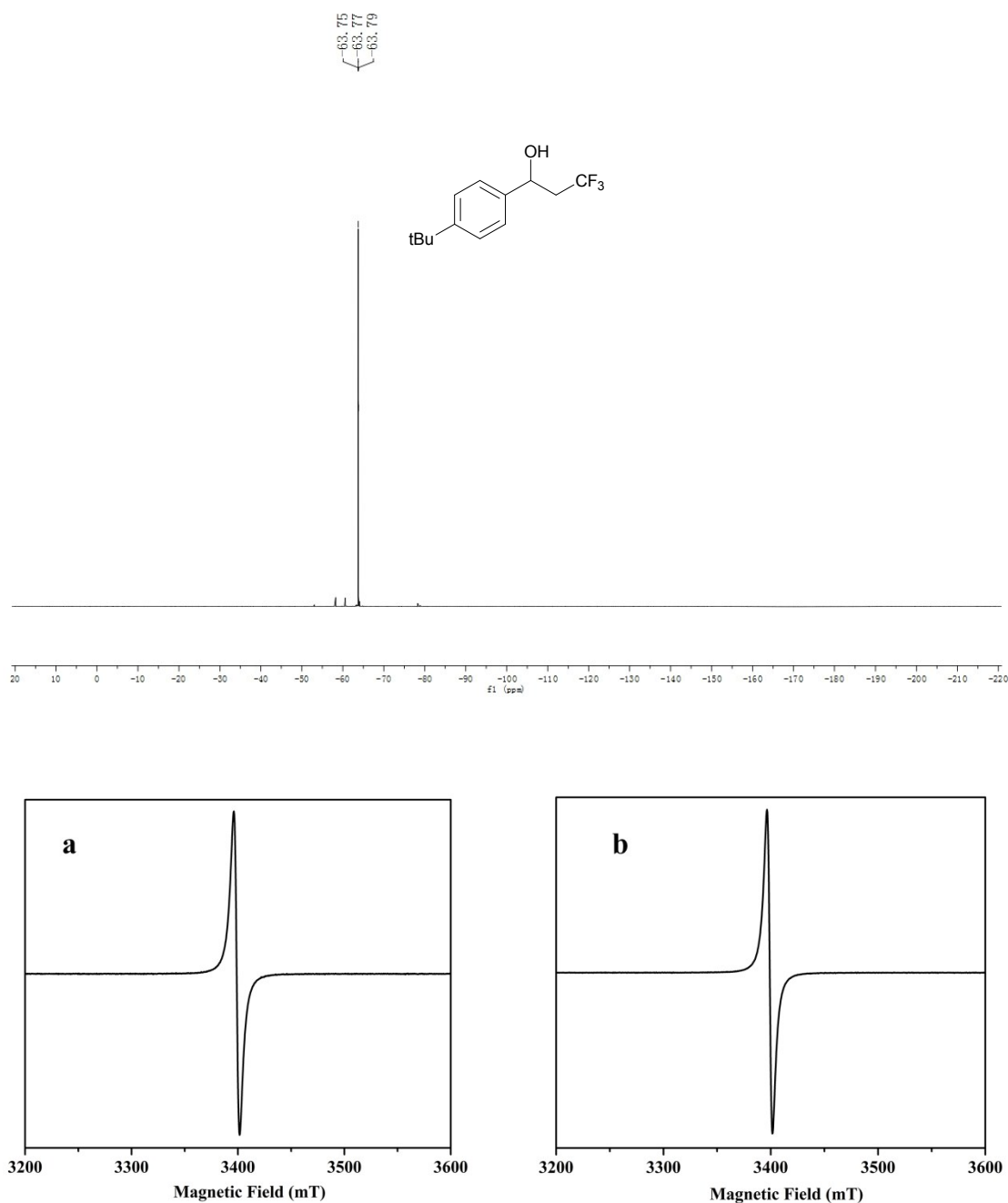


Figure S1. EPR Spectra of TEMPO (a) before reaction ($g = 2.0544$) and (b) a recovery after reaction ($g = 2.0547$). Reaction condition: 2.4 mmol TEMPO was added to the suspension of 0.02 mmol $\text{Cu}_3(\text{BTC})_2$ in 2 mL of anhydrous CH_3CN and the reaction mixture was stirred under 60 °C. After 24 hrs, the $\text{Cu}_3(\text{BTC})_2$ was removed out through centrifugation and TEMPO was recovered by evaporating the solvent. EPR spectra were applied to TEMPO before reaction and after reaction. The same g value suggested that the TEMPO was functionally compatible with the $\text{Cu}_3(\text{BTC})_2$ during this procedure.

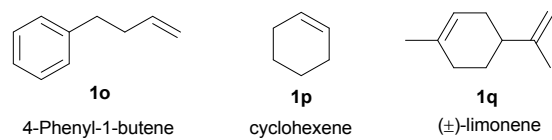


Figure S2. Substrates with negligible conversions.