

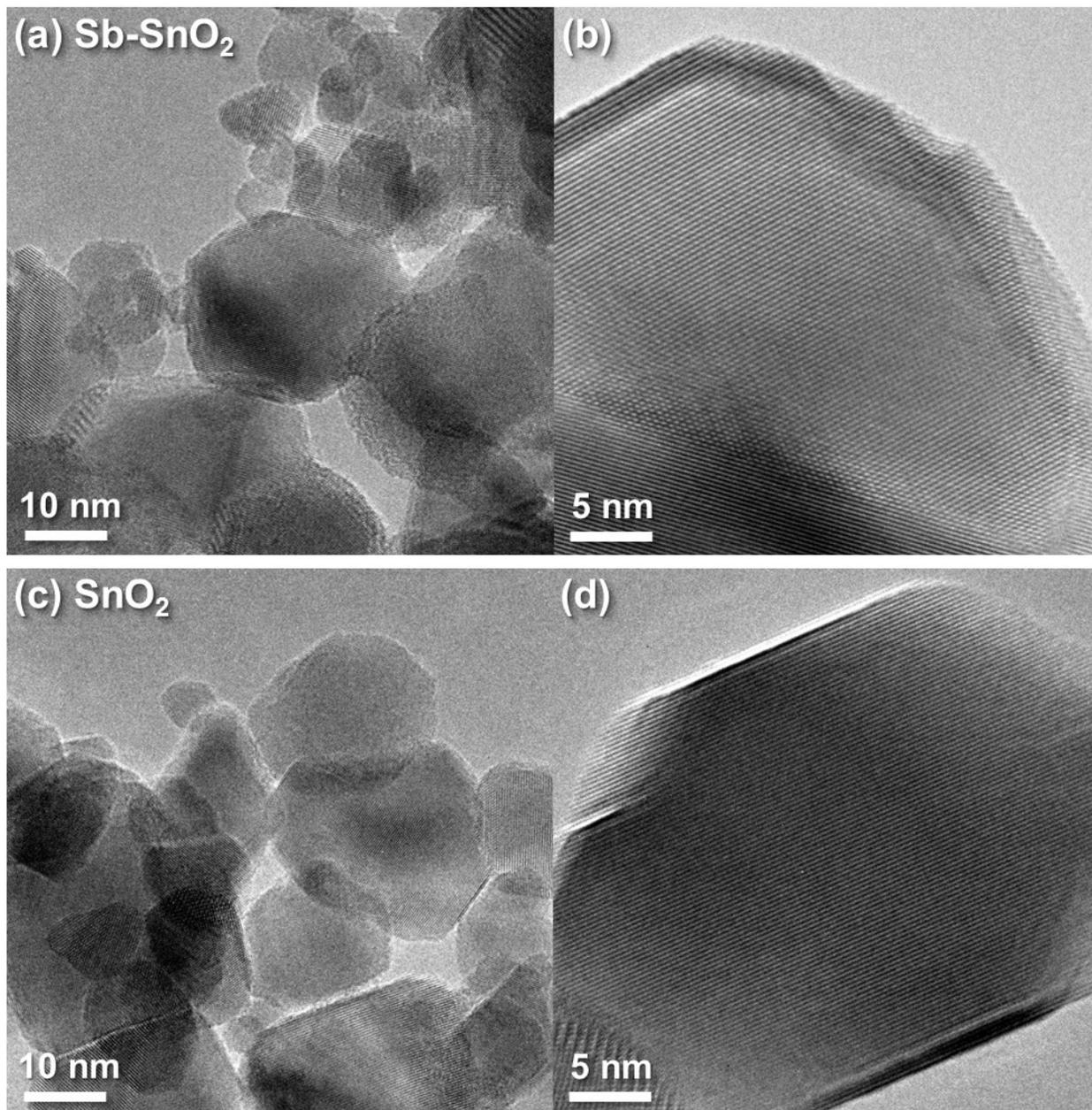
**Supporting Information**

# CO oxidation on SnO<sub>2</sub> surface enhanced by metal doping

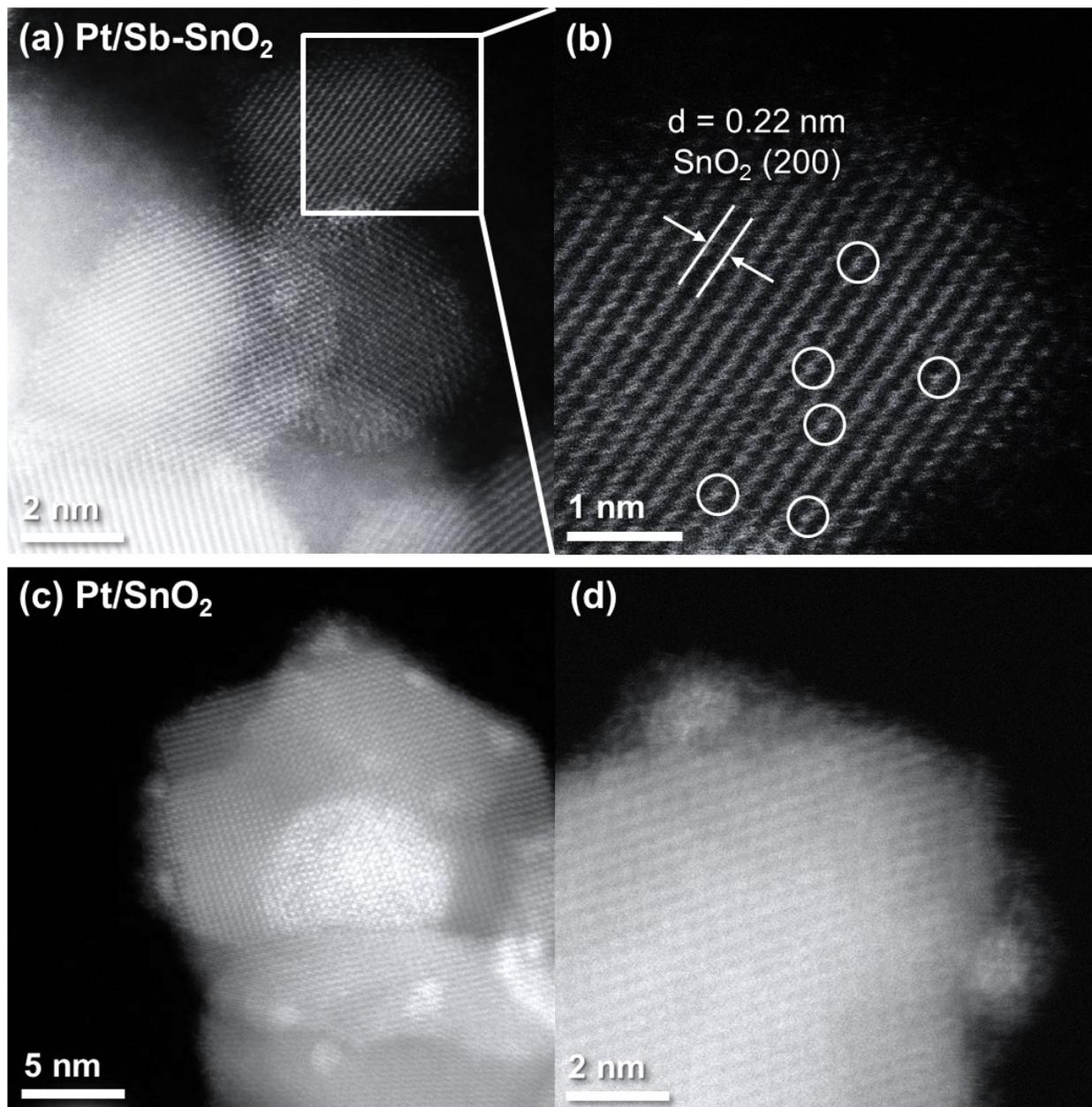
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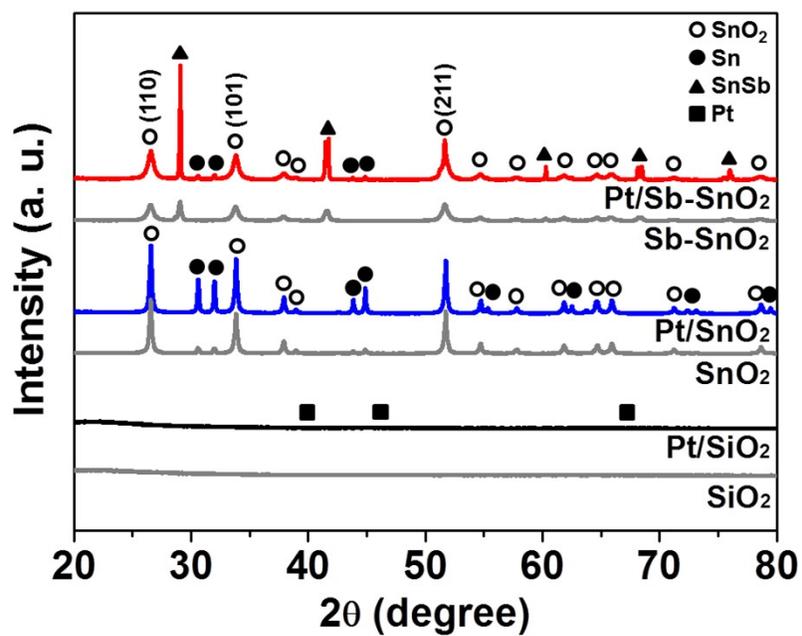
**Additional data; Figure S1-S8**



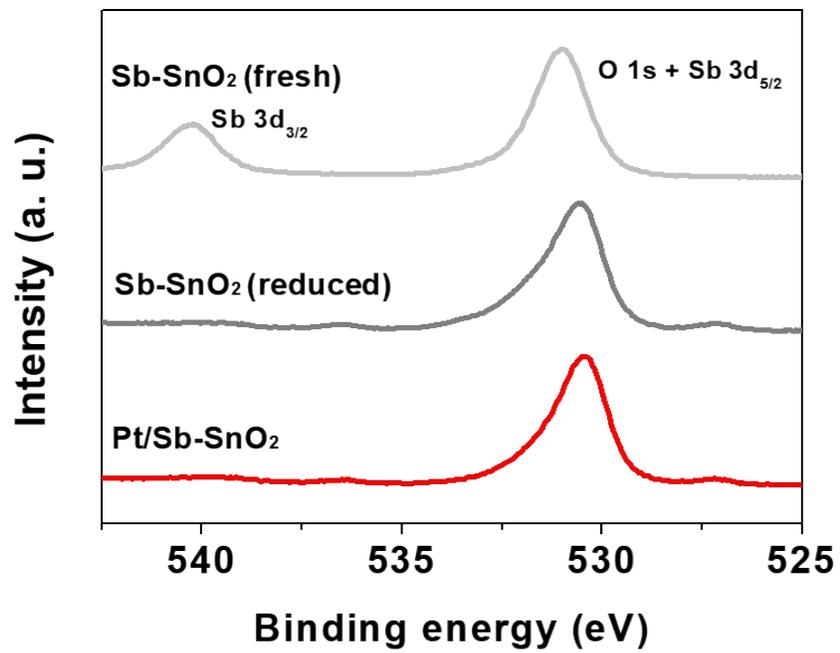
**Figure S1.** HR-TEM images of bare metal oxide supports reduced at 400 °C; (a, b) Sb-SnO<sub>2</sub> and (c, d) SnO<sub>2</sub>.



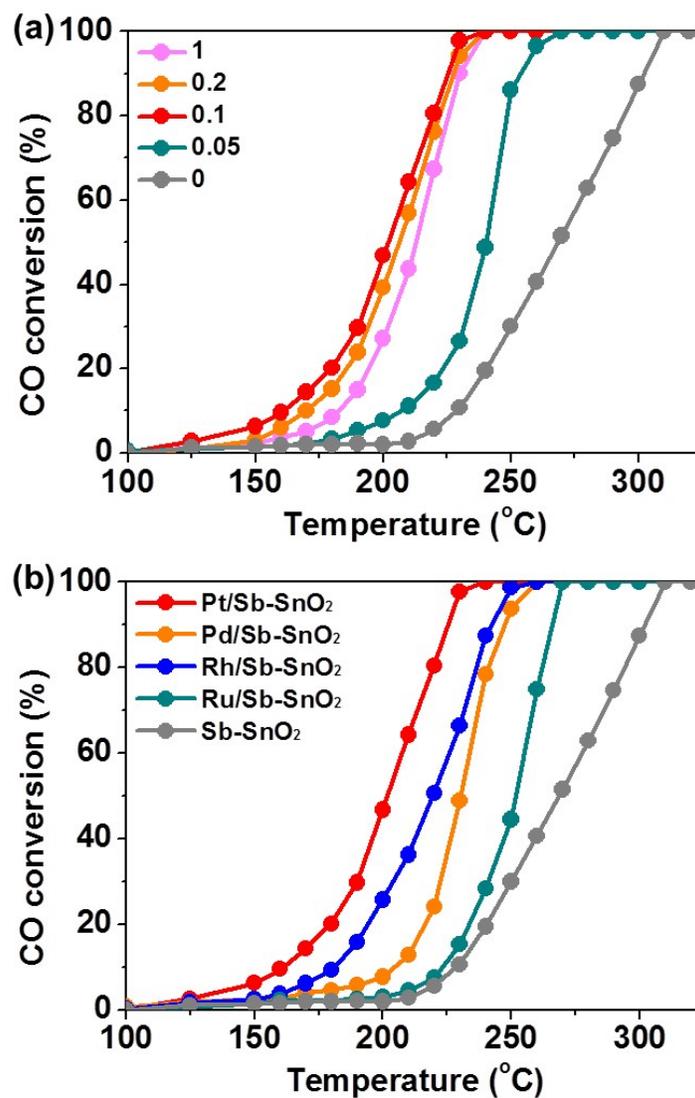
**Figure S2.** HAADF-STEM images of (a) 0.1 wt% Pt/Sb-SnO<sub>2</sub>, (b) magnified image of (a), and (c, d) 0.1 wt% Pt/SnO<sub>2</sub> at different locations. White circles in (b) indicate single Pt atoms at SnO<sub>2</sub> (200) phase.



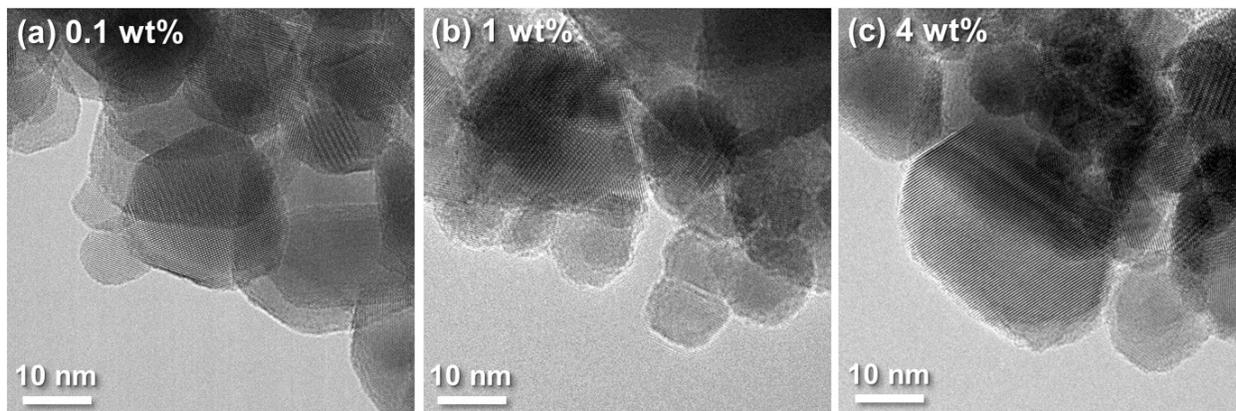
**Figure S3.** XRD patterns of the 0.1 wt% Pt/Sb-SnO<sub>2</sub>, Pt/SnO<sub>2</sub>, Pt/SiO<sub>2</sub>, and their corresponding bare supports of Sb-SnO<sub>2</sub>, SnO<sub>2</sub>, SiO<sub>2</sub>.



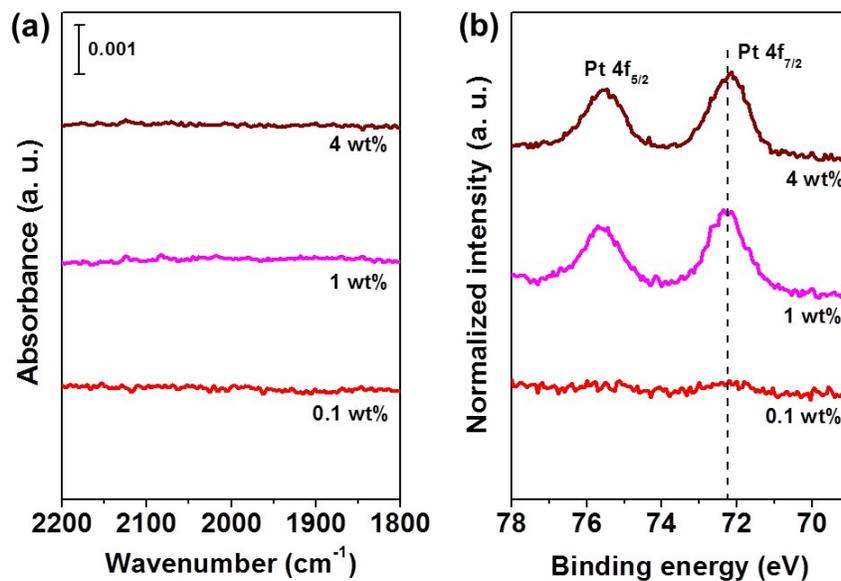
**Figure S4.** XPS Sb 3d spectra of fresh Sb-SnO<sub>2</sub>, Sb-SnO<sub>2</sub> reduced at 400 °C, and 0.1 wt% Pt/Sb-SnO<sub>2</sub>.



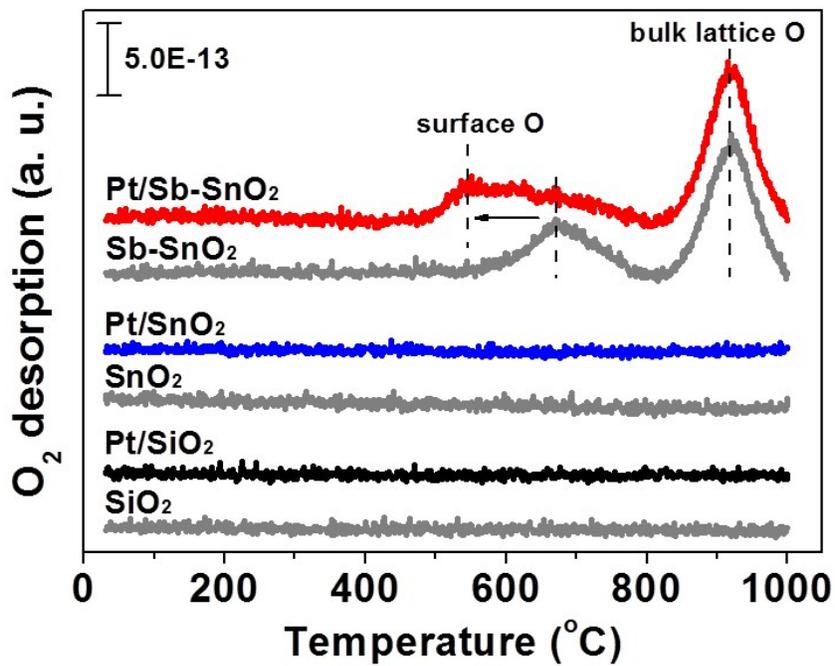
**Figure S5.** CO oxidation results for (a) Pt/Sb-SnO<sub>2</sub> with various Pt weight percentages and (b) 0.1 wt% M/Sb-SnO<sub>2</sub> (M = Pt, Pd, Rh, and Ru). CO oxidation was performed at 120,000 ml·g<sup>-1</sup>·h<sup>-1</sup> with an inlet gas flow of 1% CO and 1% O<sub>2</sub> balanced with He.



**Figure S6.** HR-TEM images of (a) 0.1, (b) 1, and (c) 4 wt% Pt/Sb-SnO<sub>2</sub> catalysts.



**Figure S7.** (a) DRIFT spectra for CO chemisorption and (b) XPS Pt 4f spectra of the 0.1, 1, and 4 wt% Pt/Sb-SnO<sub>2</sub> catalysts.



**Figure S8.** O<sub>2</sub>-TPD results by monitoring O<sub>2</sub> (m/z 32) desorption with a mass spectrometer on the 0.1 wt% Pt/Sb-SnO<sub>2</sub>, Pt/SnO<sub>2</sub>, Pt/SiO<sub>2</sub>, and their corresponding bare supports of Sb-SnO<sub>2</sub>, SnO<sub>2</sub>, SiO<sub>2</sub>.