

Electronic Supplementary Information (ESI)

**Au-Pd NPs immobilised on nanostructured ceria and titania:
impact of support morphology on the catalytic activity for
selective oxidation**

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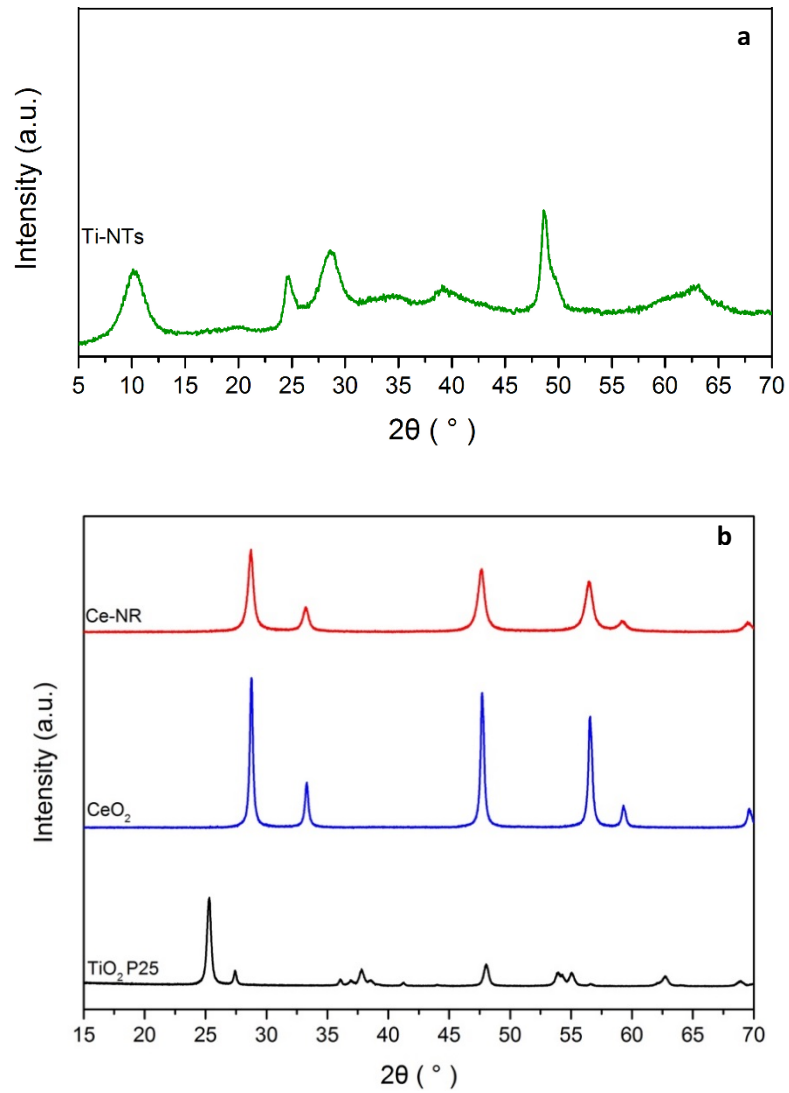


Figure S1. X-ray powder diffraction patterns of Ti-NTs (a) and Ce-NRs, CeO₂ and TiO₂ P25 nanopowders (b).

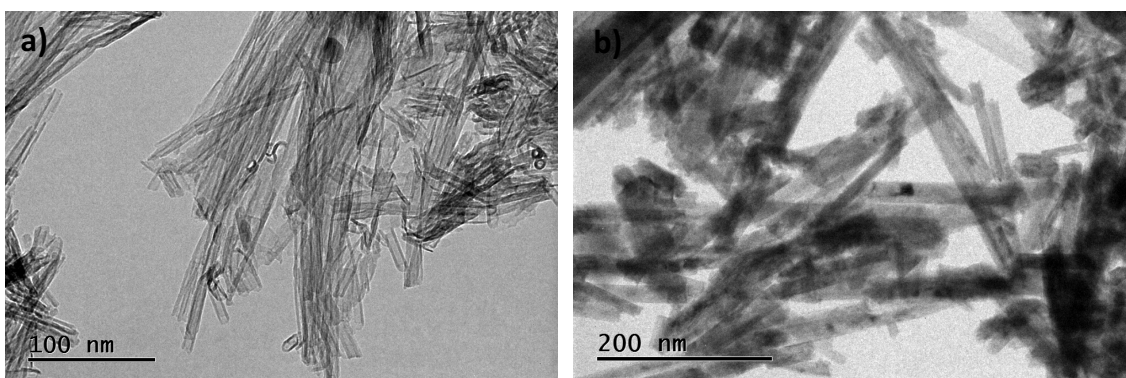


Figure S2. TEM images of a) Ti-NTs, and b) Ce-NRs.

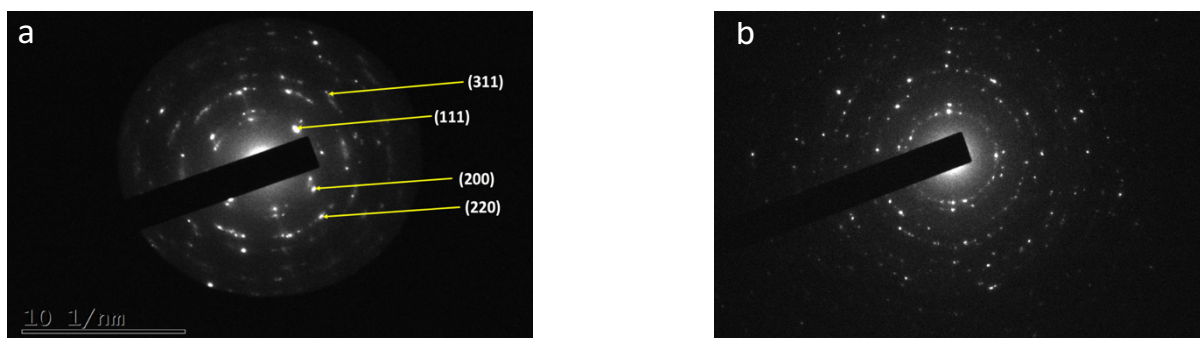


Figure S3. Selected area electron diffraction (SAED) patterns for (a) Ce-NR and (b) CeO₂ obtained with the JEOL JEM-2100F. The ring patterns confirm the polycrystalline nature of the as-Ce-NR and CeO₂ nanopowder. The rings are indexed against the CeO₂ fluorite structure.

Table S1. Textural properties determined from N₂ adsorption-desorption measurements.

Support	SBET (m ² /g)	Pore volume (cm ³ /g)	Pore diameter (nm)
Ti-NT	236	0.51	7.7
Ce-NR	61.7	0.34	17.3

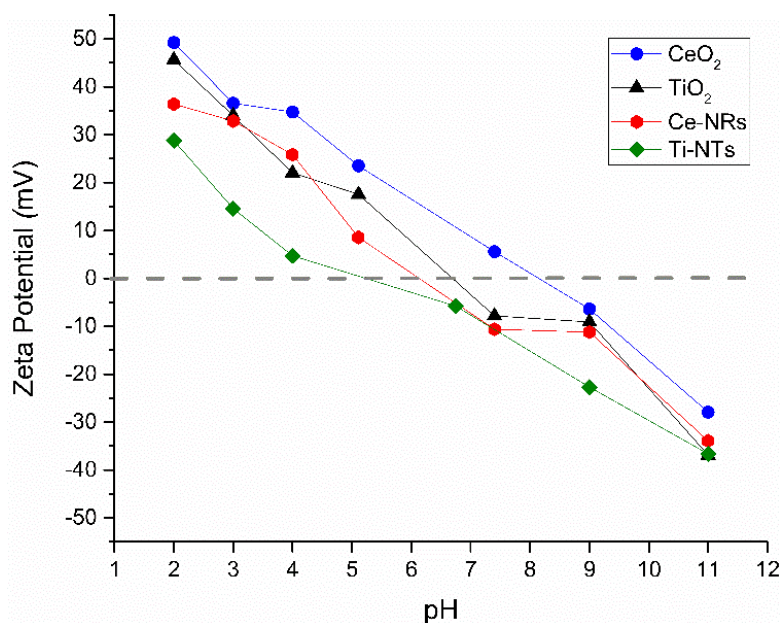


Figure S4. Zeta potential as a function of pH for Ce-NRs, CeO₂ nanopowders, Ti-NTs, and TiO₂ P25.

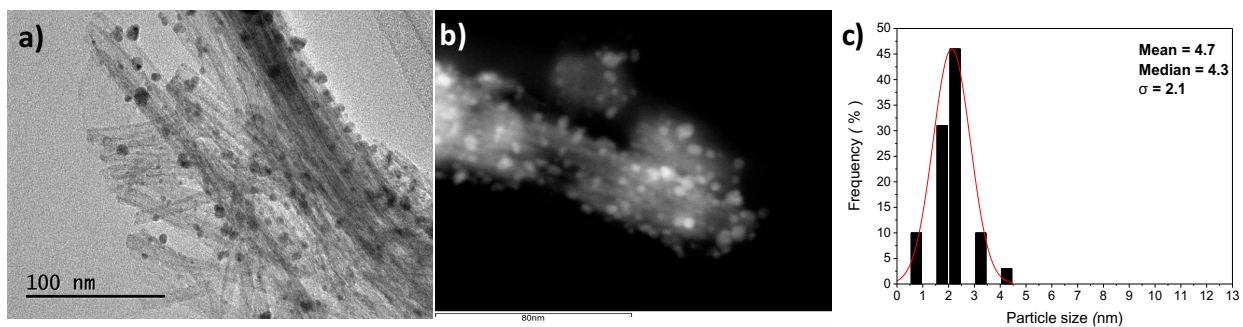


Figure S5. a) TEM image, b) STEM image for Au-Pd/Ti-NT, and c) corresponding particle size distribution.

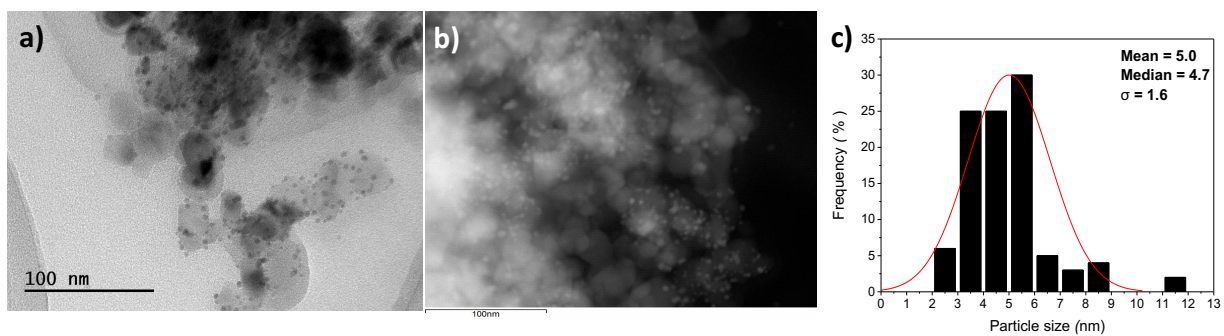


Figure S6. a) TEM image, b) STEM image for Au-Pd/TiO₂, and c) corresponding particle size distribution.

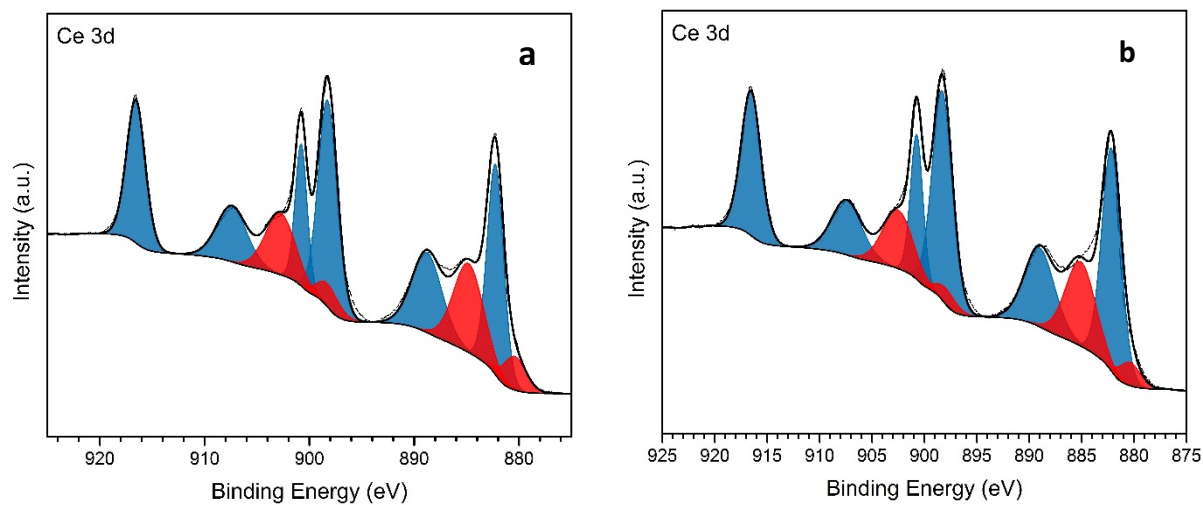


Figure S7. XPS spectra of Ce 3d for (a) Ce-NR and (b) CeO₂ nanopowder.

Table S2. Concentration of Ce³⁺ and O species for the ceria supports from XPS.

Support	Ce ³⁺ (at.%) ^a	O _α (at.%) ^b	O _β (at.%) ^b	O _γ (at.%) ^b
Ce-NR	31.0	61.7	35.2	3.0
CeO ₂	27.3	77.5	20.1	2.2

^a Determined from the Ce 3d XPS spectra.

^b Determined from the O1s XPS spectra.

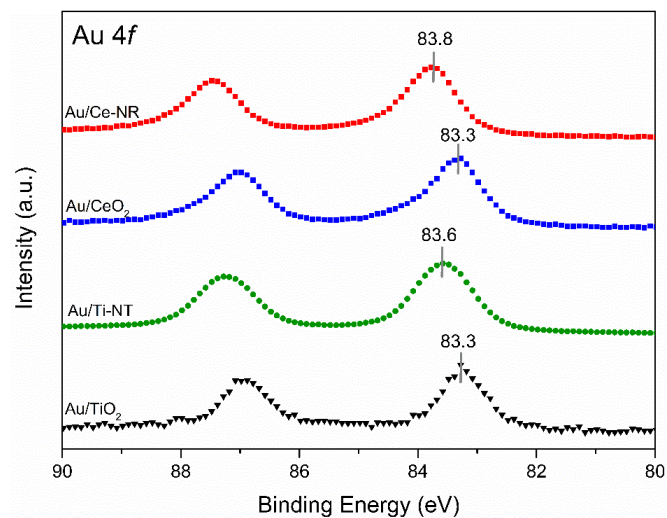


Figure S8. XPS spectra of Au 4f for monometallic catalyst samples Au/Ce-NR, Au/CeO₂, Au /Ti-NT and Au-Pd/TiO₂

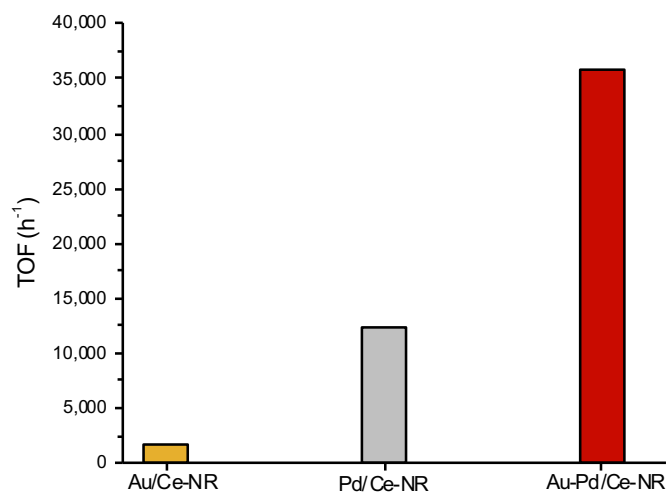


Figure S9. TOF (h⁻¹) for monometallic and bimetallic catalysts prepared by sol-immobilisation. Reaction conditions: T= 120°C, pO₂= 2 bar, stirring rate=1,000 rpm, molar ratio of benzyl alcohol/metal = 50,000. TOF calculated after 0.5 hour.