

Electronic Supplementary Information (ESI)

Dinuclear Zirconium Complex Bearing 1,5-Bridged-Calix[8]arene Ligand as Effective Catalyst for the Synthesis of Macrolactones

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1. NMR CHARACTERIZATION

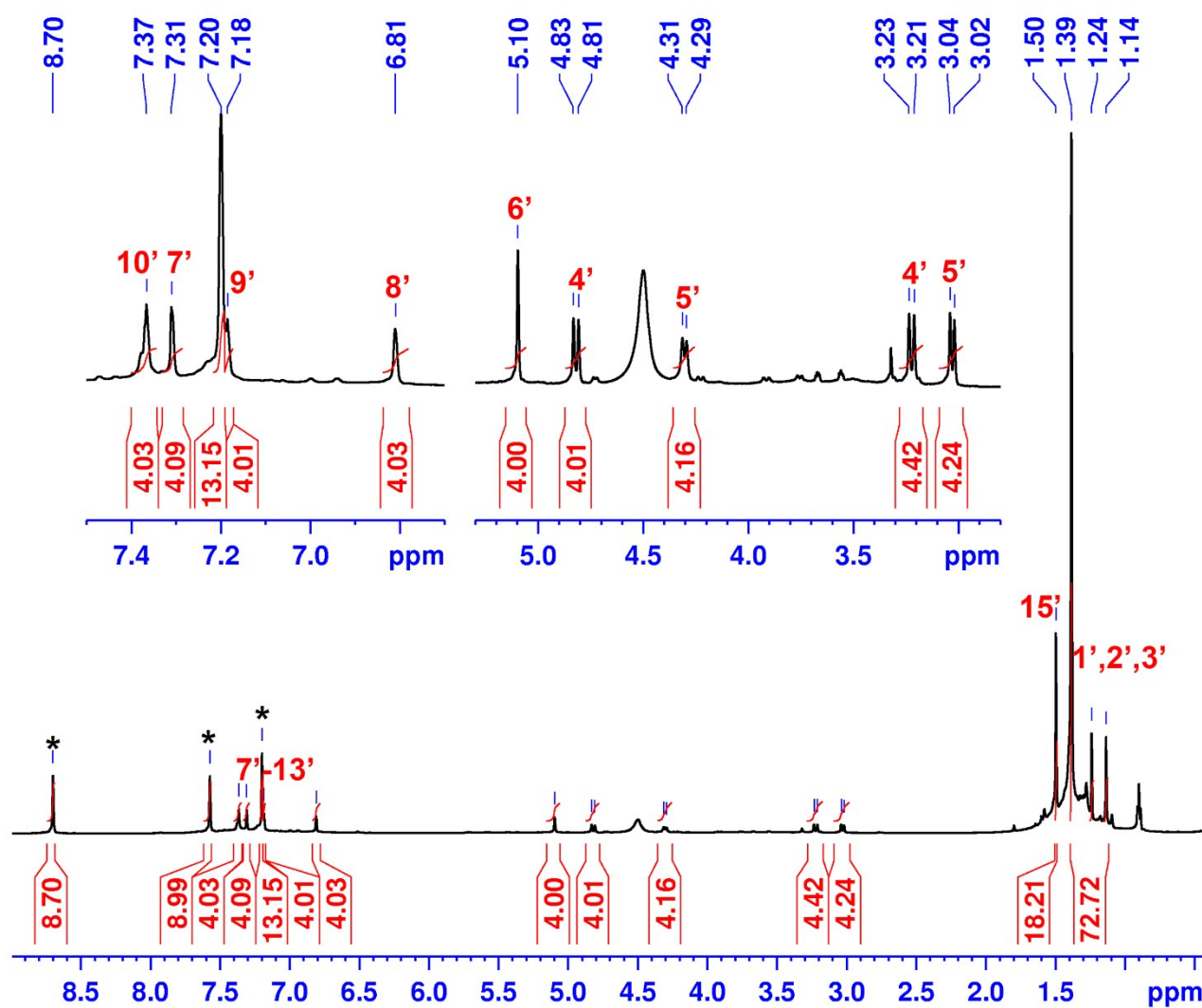


Figure S1. ^1H NMR spectrum of C (600 MHz, $\text{pyridine-}d_5$, 25 °C).

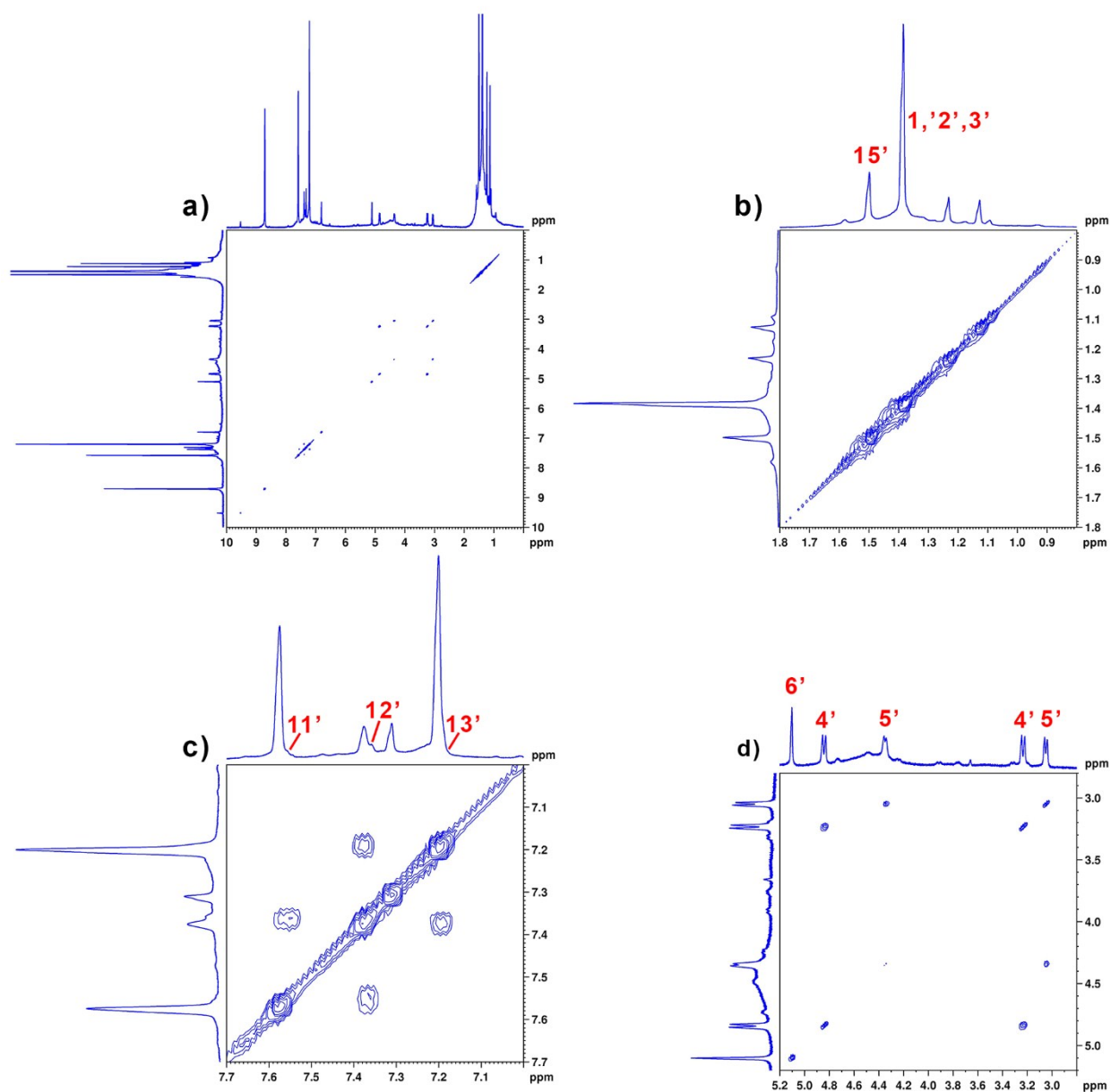


Figure S2. ^1H - ^1H COSY spectrum (600 MHz, pyridine- d_5 , 90°C) of C (a) with magnifications of the diagnostic regions (b-d).

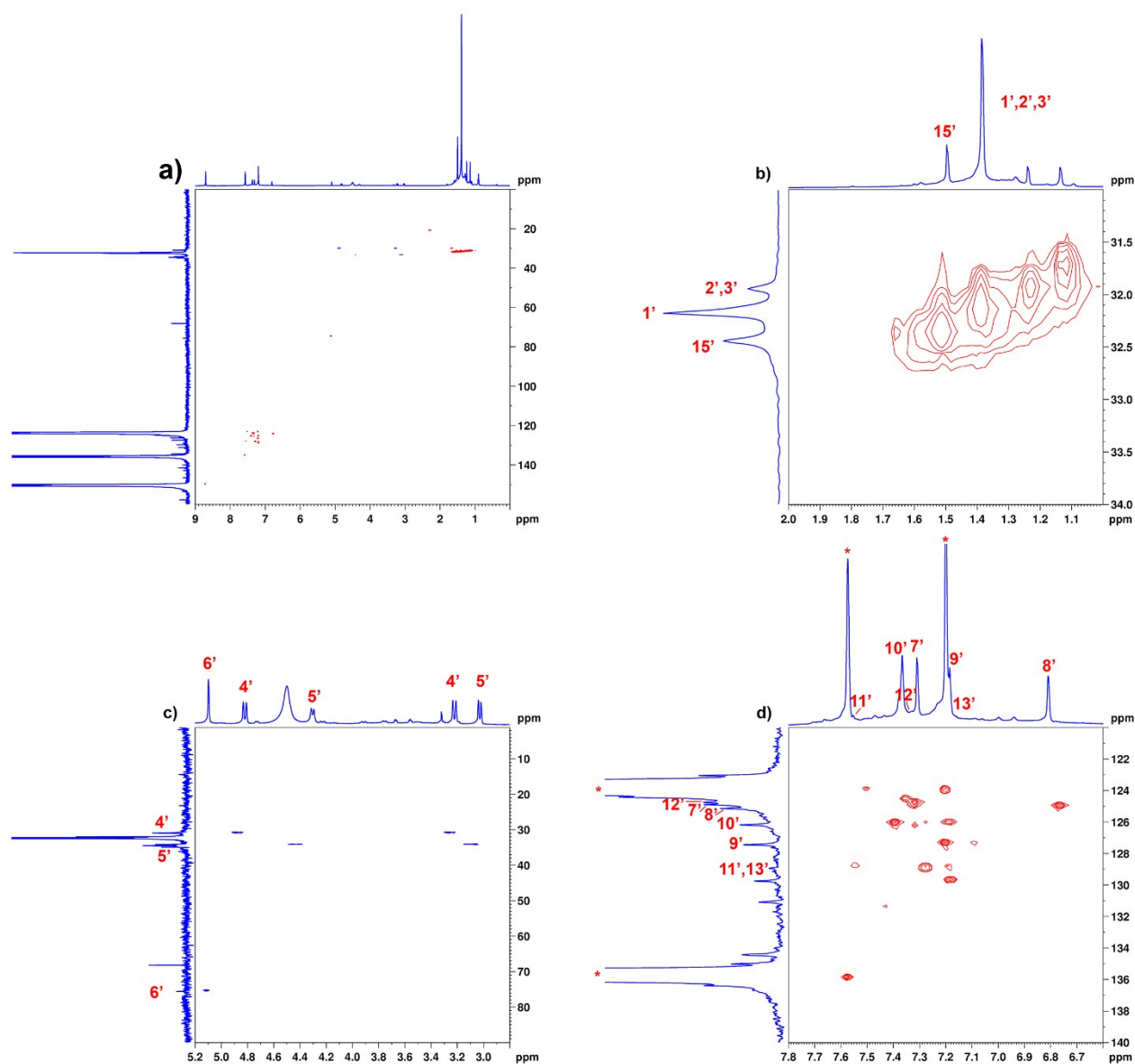


Figure S3. ^1H - ^{13}C HSQC spectrum (600 MHz, pyridine- d_5 , 25°C) of the complex C (a) with magnifications of the diagnostic regions (b-d).

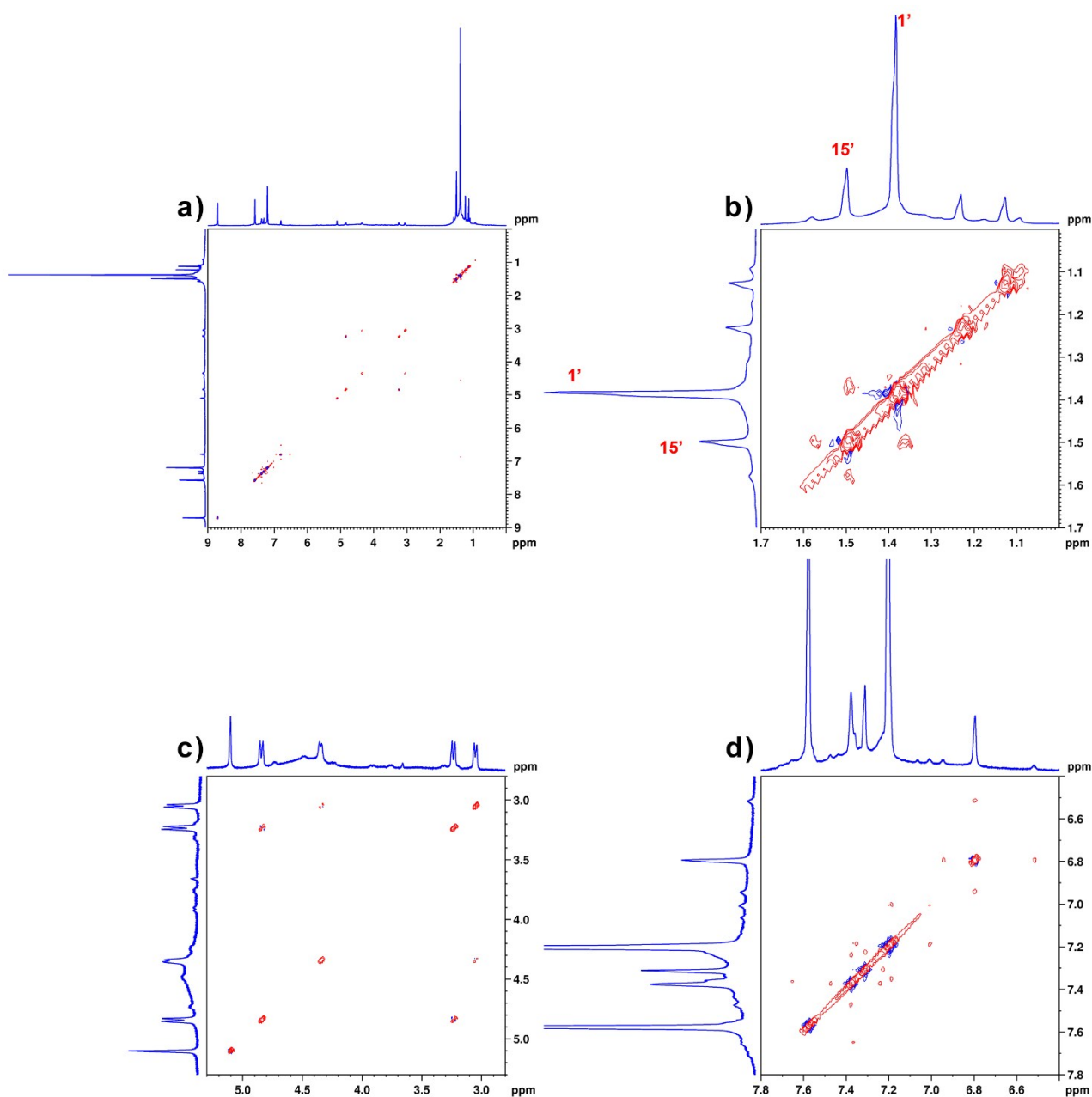


Figure S4. ^1H - ^1H NOESY spectrum (600 MHz, pyridine- d_5 , 90°C) of the complex **C** (a) with magnifications of the diagnostic regions (b-d).

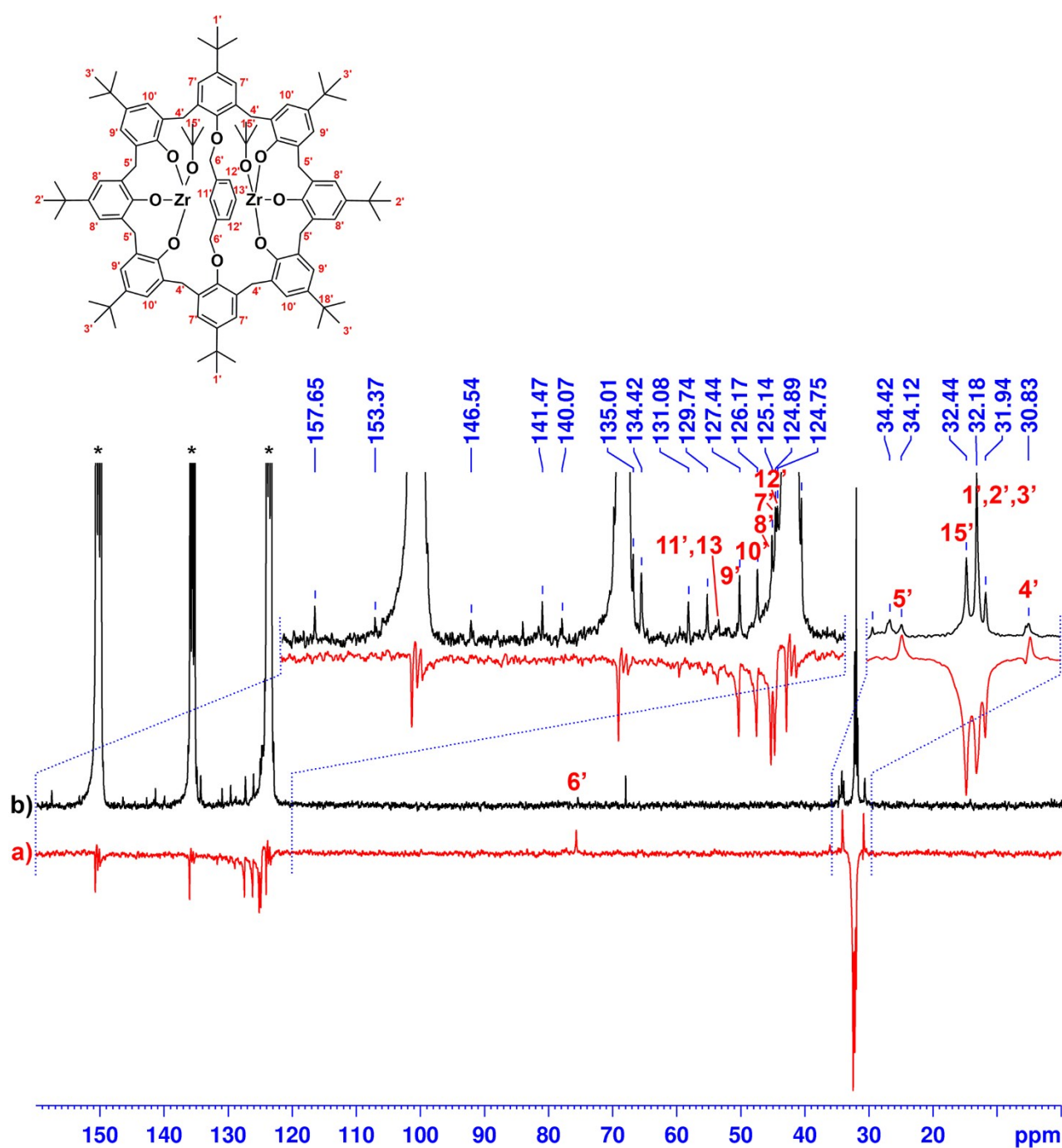


Figure S5. DEPT135 (a) and ^{13}C NMR (b) spectra (600 MHz, pyridine- d_5 , 90°C) of the complex **C** (Scheme 1).

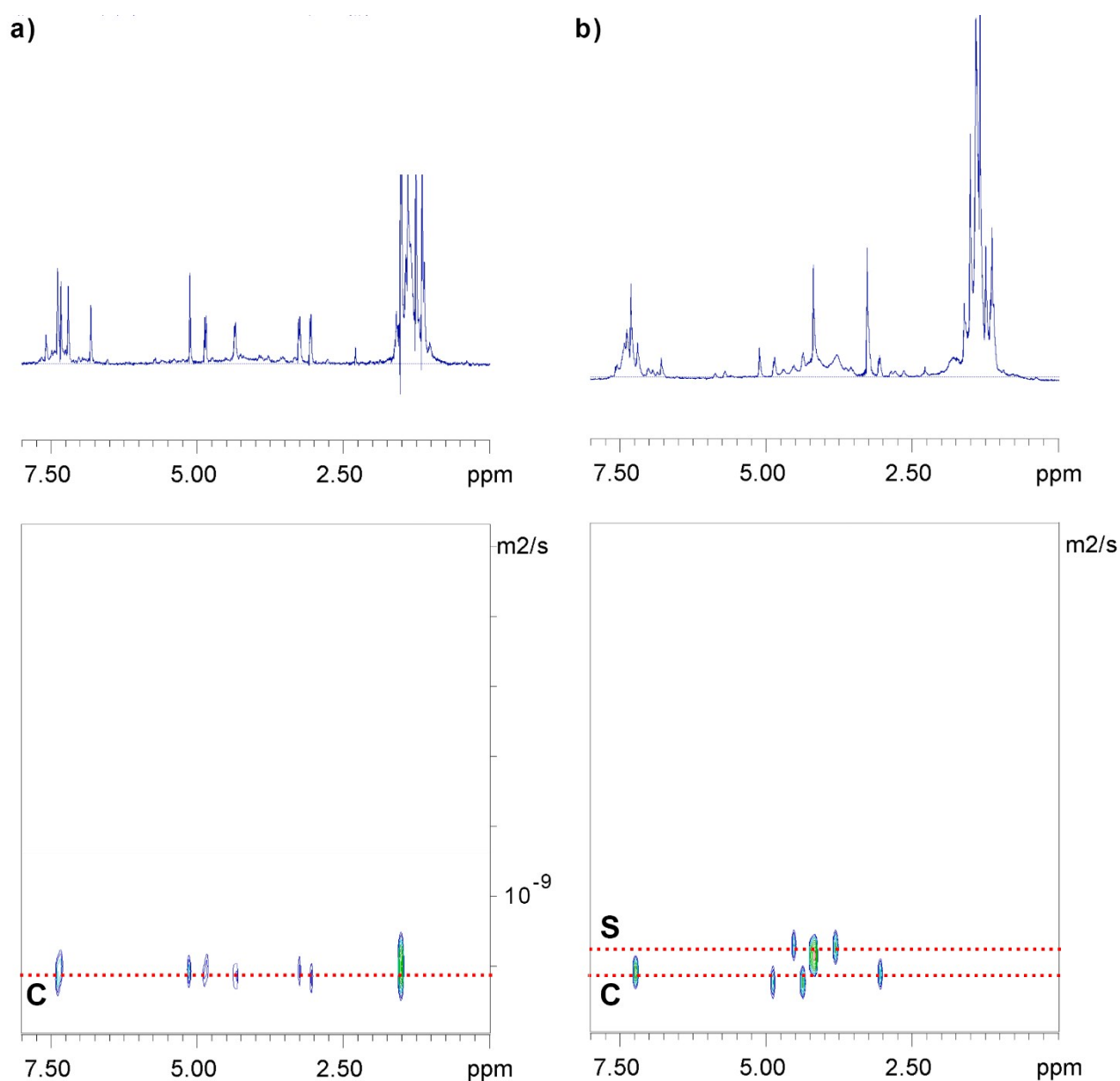


Figure S6. DOSY NMR spectra of the complex **C** (600 MHz, 25°C, benzene-*d*₆) (a) and of **C** (600 MHz, 25°C, benzene-*d*₆) in presence of the internal standard **S** as reference (b). Diffusion coefficients: **C** = $5.6 \cdot 10^{-8} \pm 4.6 \cdot 10^{-10} \text{ m}^2 \text{ s}^{-1}$; **S** = $6.4 \cdot 10^{-8} \pm 3.9 \cdot 10^{-10} \text{ m}^2 \text{ s}^{-1}$.

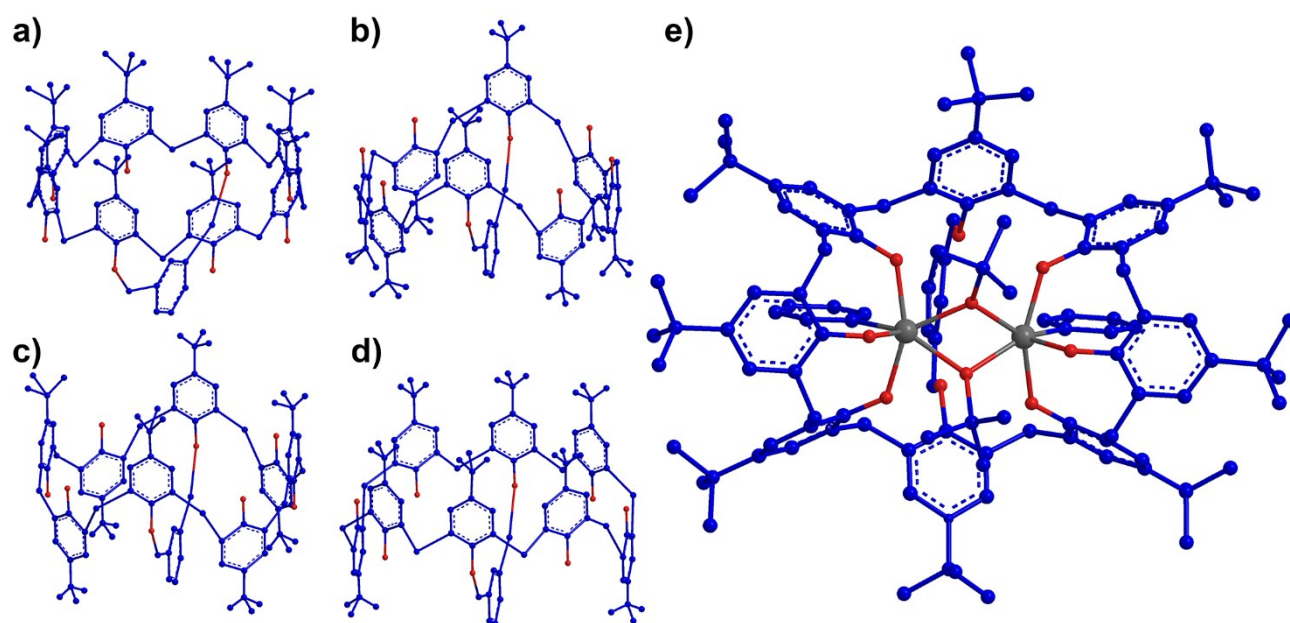


Figure S7. Possible configurations of the ligand **L** with C_{2v} symmetry (a-d) and the structure of the complex **C**, compatible with NMR information, resulting from the configuration in b).

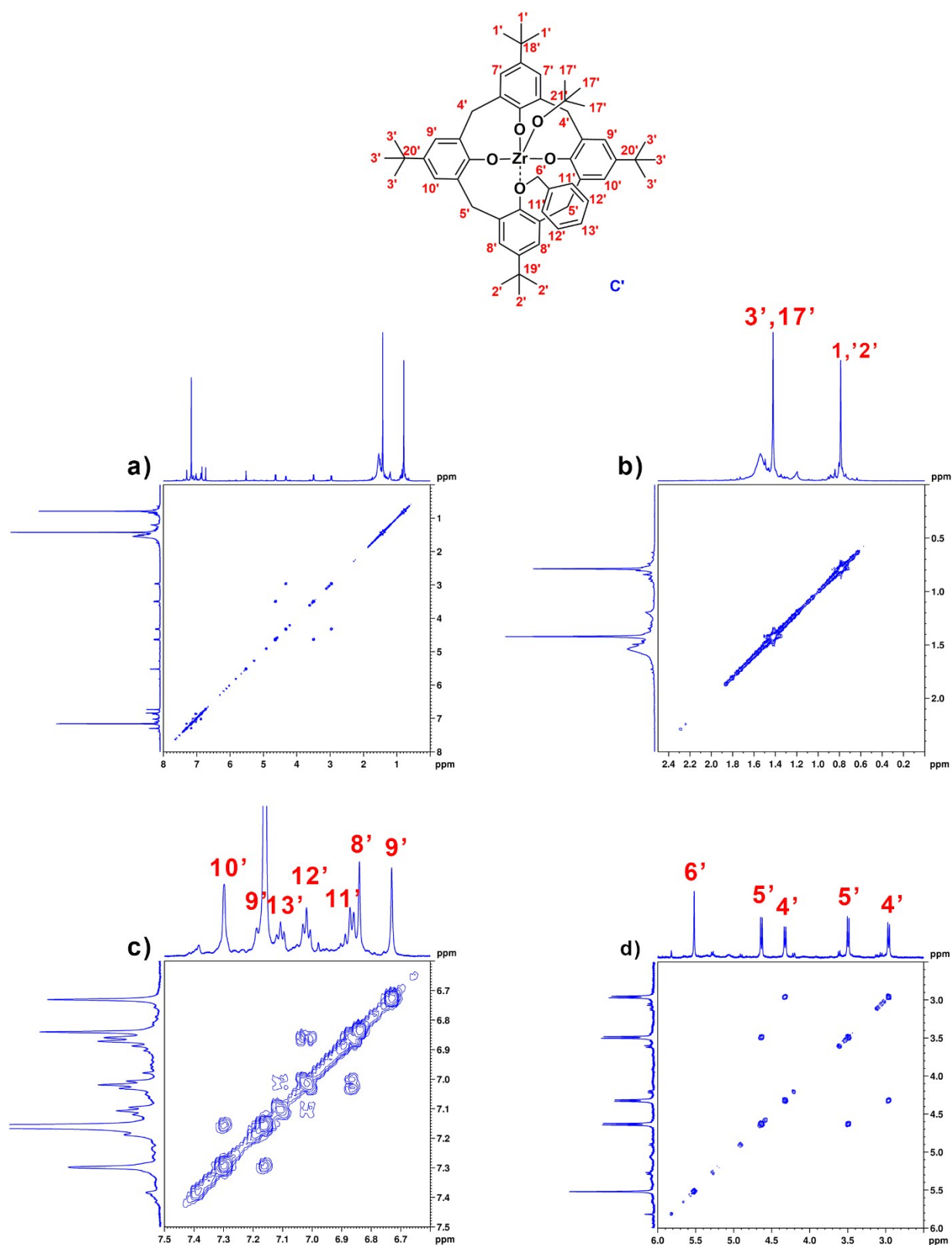


Figure S8. ^1H - ^1H COSY spectrum (600 MHz, benzene- d_6 , 25°C) of **C'** (a) with magnifications of the diagnostic regions (b-d).

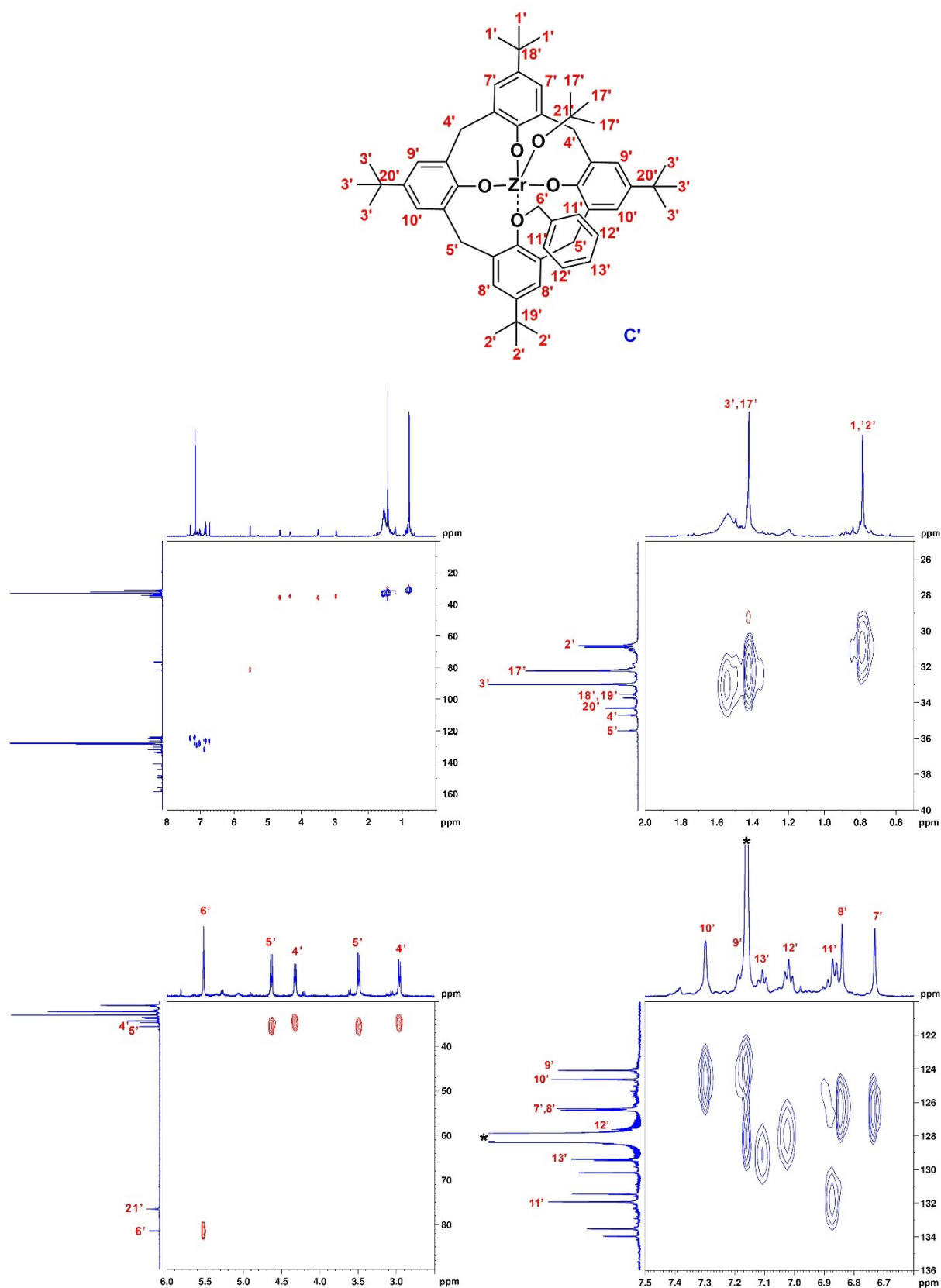


Figure S9. ^1H - ^{13}C HSQC spectrum (600 MHz, benzene- d_6 , 25°C) of the complex **C'** (a) with magnifications of the diagnostic regions (b-d).

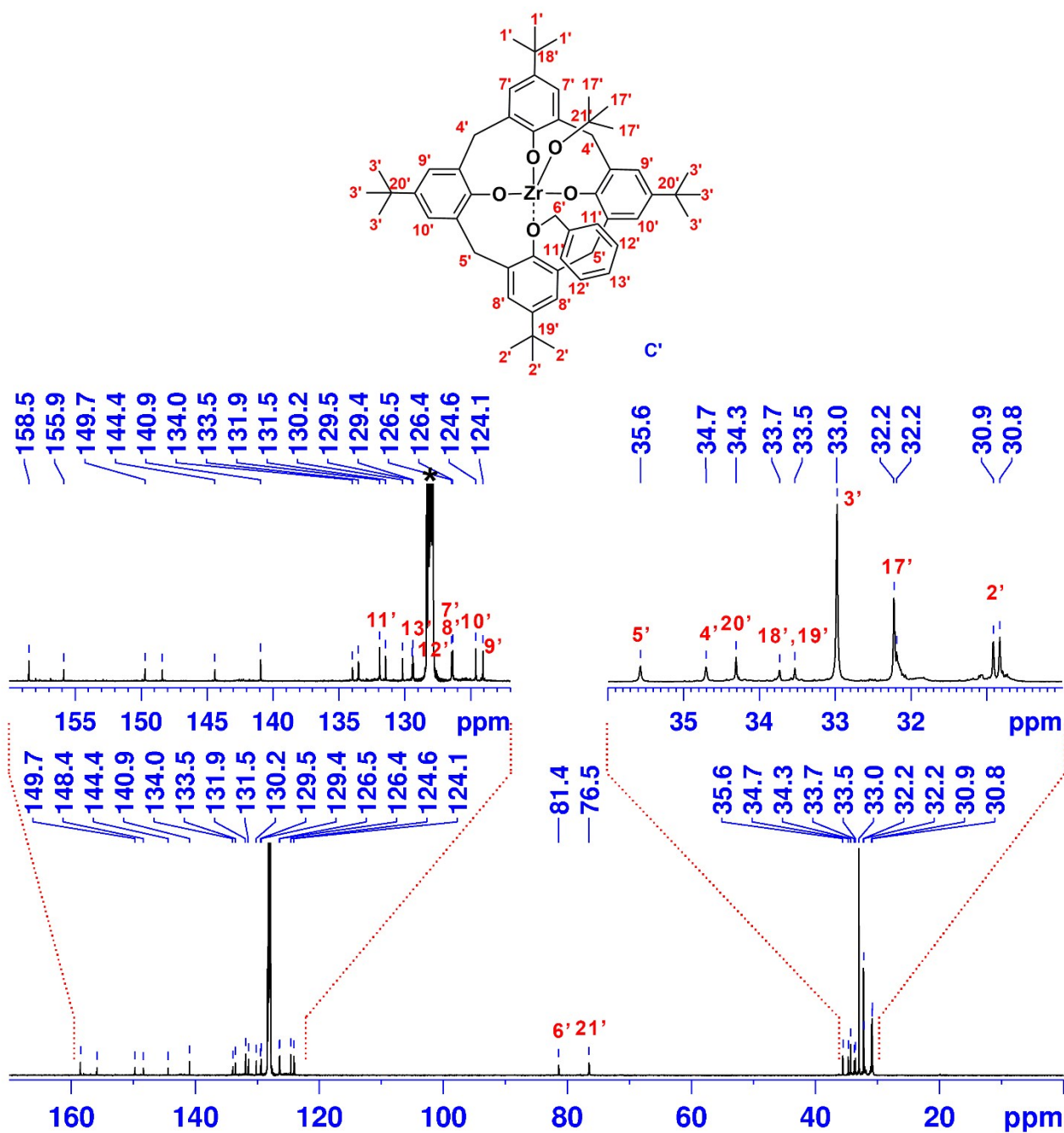


Figure S10. ^{13}C NMR spectrum of the complex **C'** with diagnostic signal labelled (600 MHz, benzene- d_6 , 25°C).

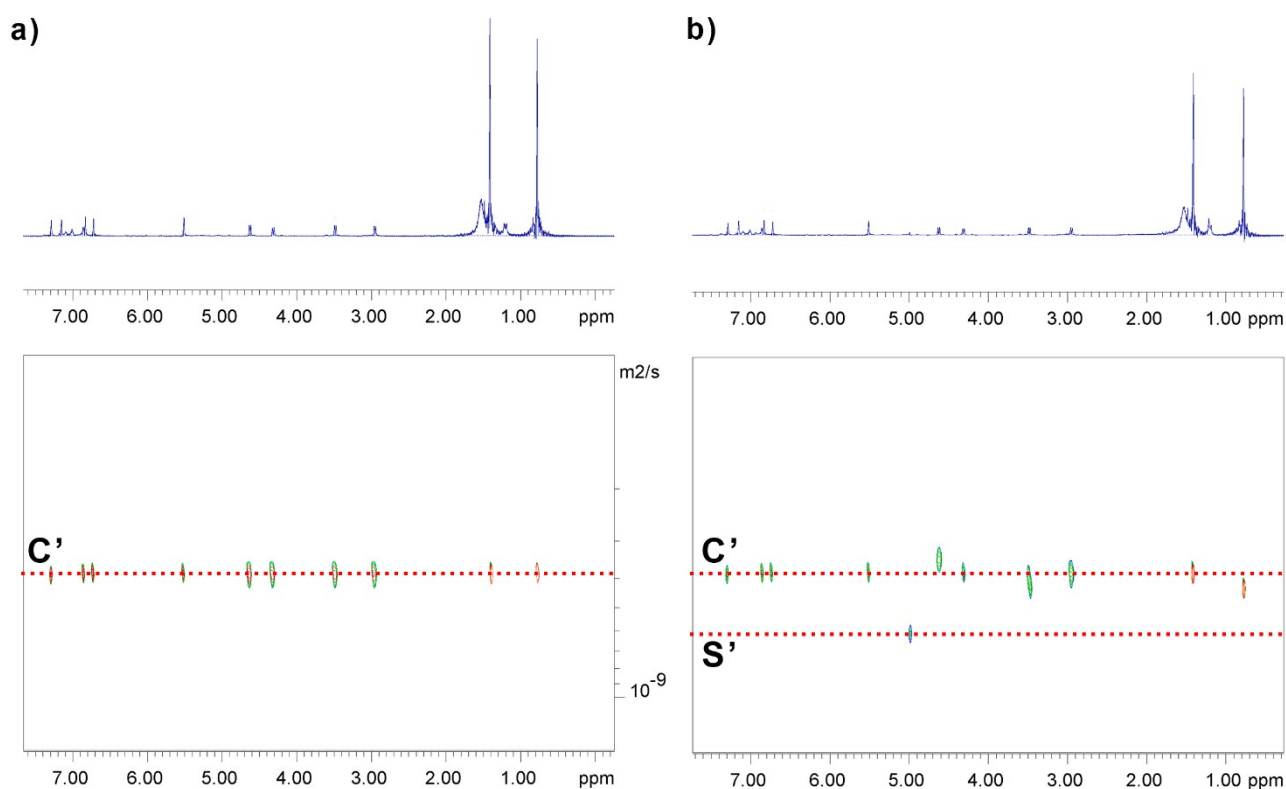


Figure S11. DOSY NMR spectra of the complex **C'** (a) and of **C'** in presence of the internal standard **S'** as reference (b). Diffusion coefficients: $C = 4.30 \cdot 10^{-10} \pm 8.56 \cdot 10^{-13} \text{ m}^2\text{s}^{-1}$; $S = 4.35 \cdot 10^{-10} \pm 5.68 \cdot 10^{-13} \text{ m}^2\text{s}^{-1}$.

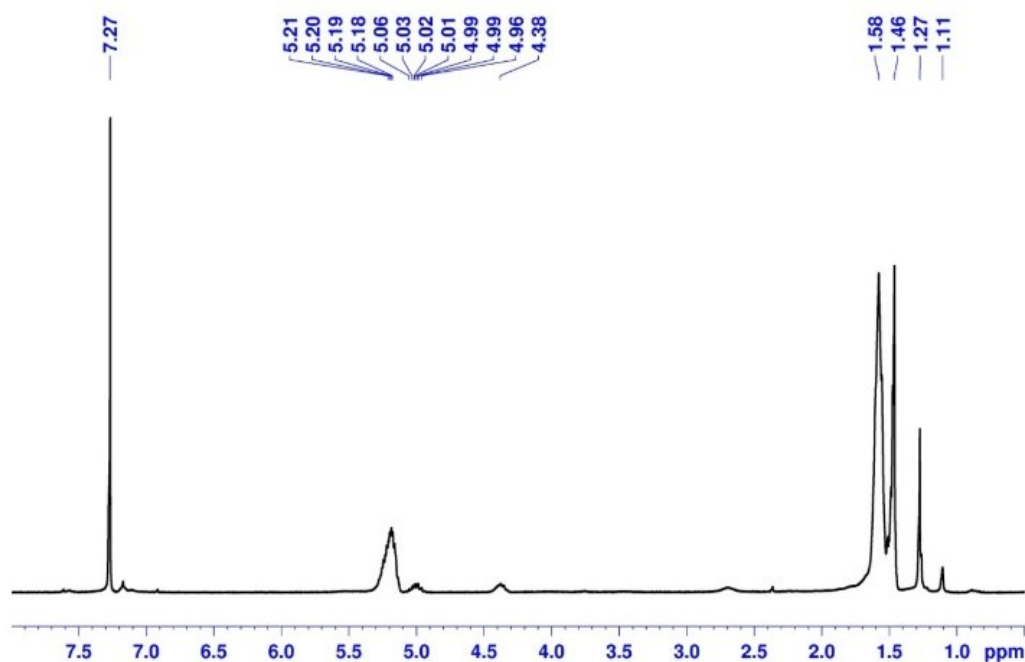


Figure S12. ^1H NMR spectrum of cyclic PLA synthesized by C (entry 4 of Table 1; CDCl_3 , 300 MHz, 25°C).

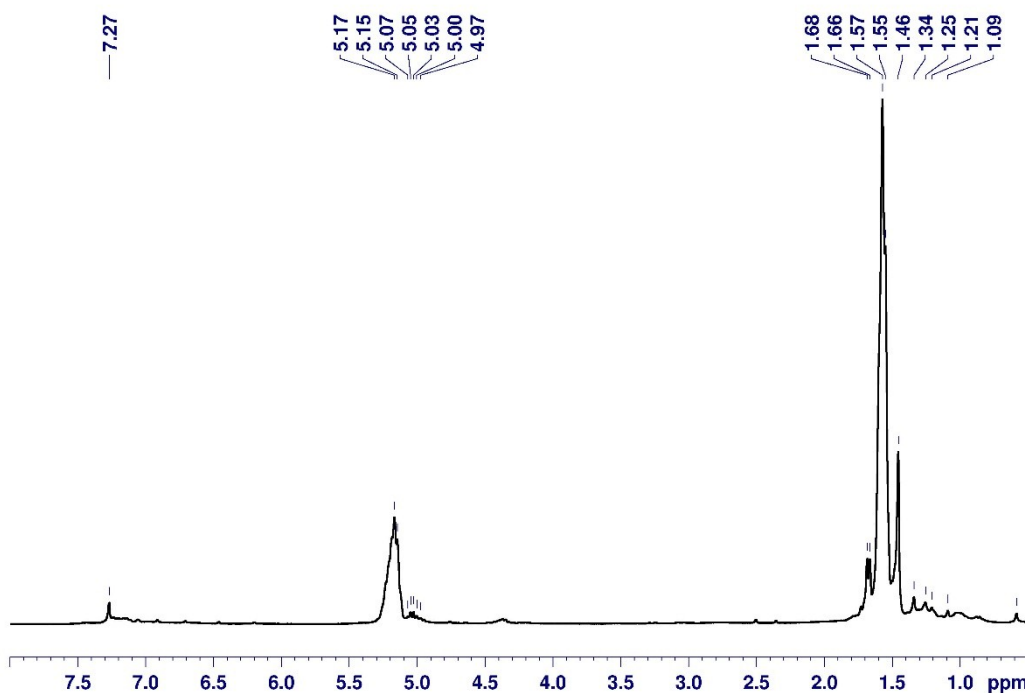


Figure S13. ^1H NMR spectrum of linear PLA synthesized by C' (entry 9 of Table 1; CDCl_3 , 300 MHz, 25°C).

2. UV-VIS ANALYSIS

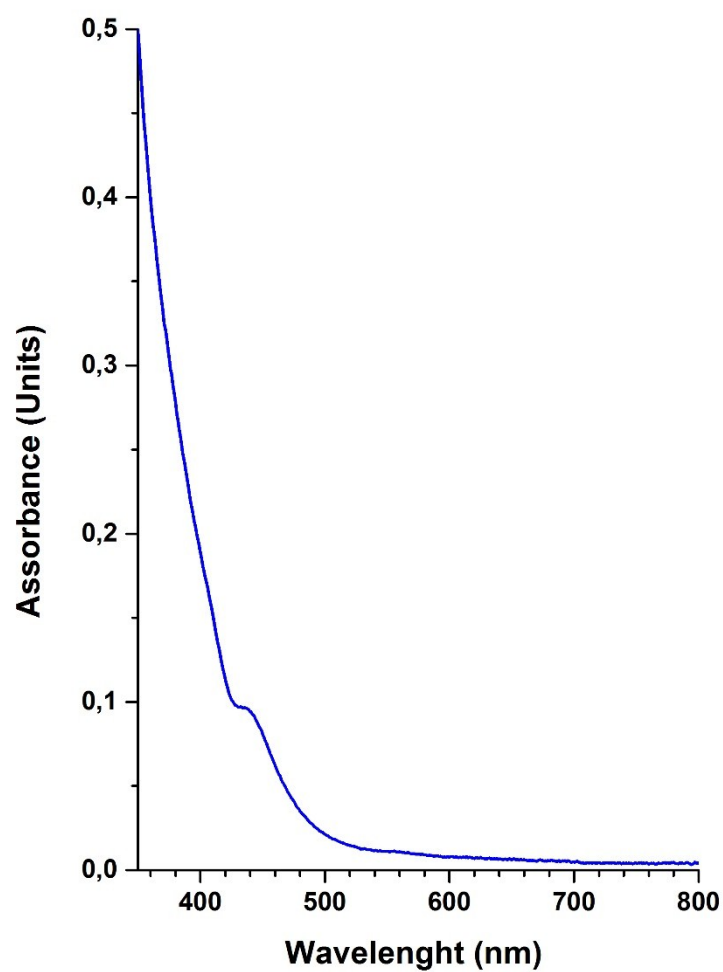


Figure S14. UV-Vis spectrum of C ($5.8 \cdot 10^{-3}$ M; pyridine; 25 °C; $\epsilon_{436} = 166 \text{ Lmol}^{-1}\text{cm}^{-1}$).

3. ESI-MS AND MALDI-MS ANALYSES

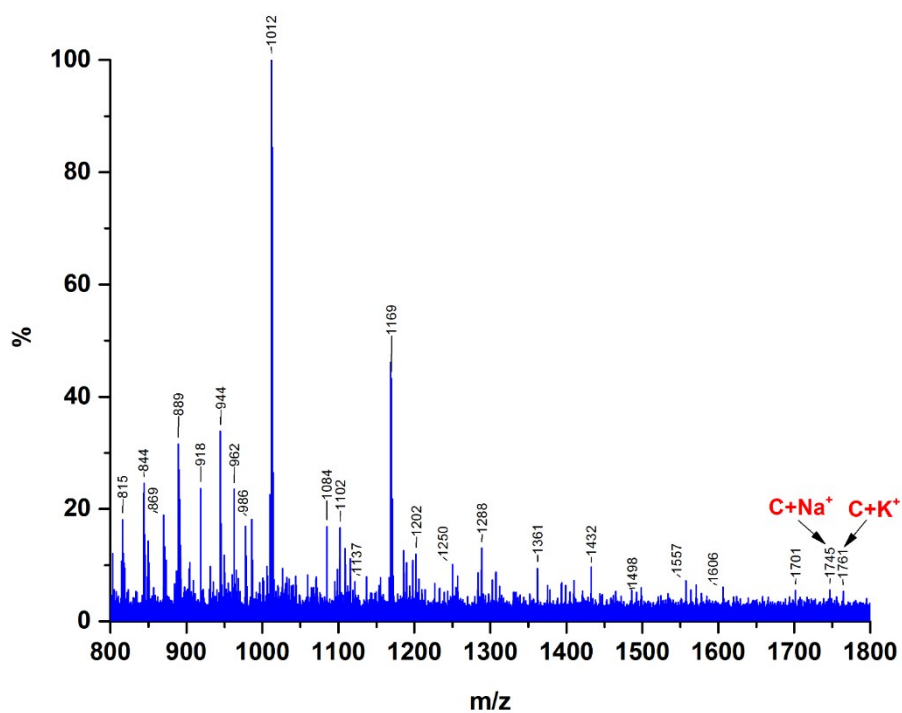


Figure S15. ESI-MS spectrum of the complex C (toluene/methanol solvents mixture).

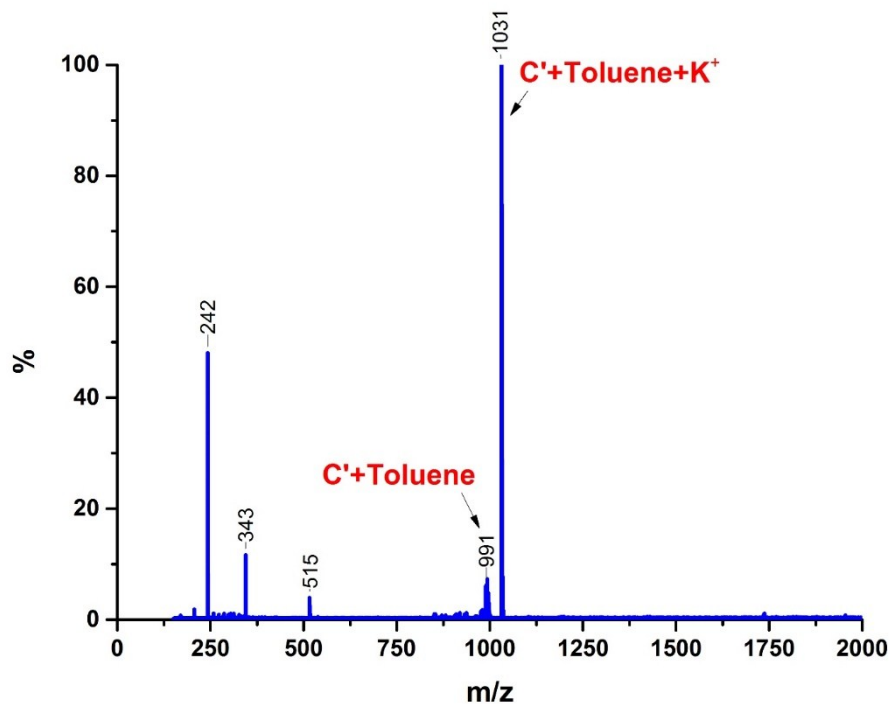


Figure S16. ESI-MS spectrum of the complex C' (toluene/methanol solvents mixture).

4. Kinetic Investigations

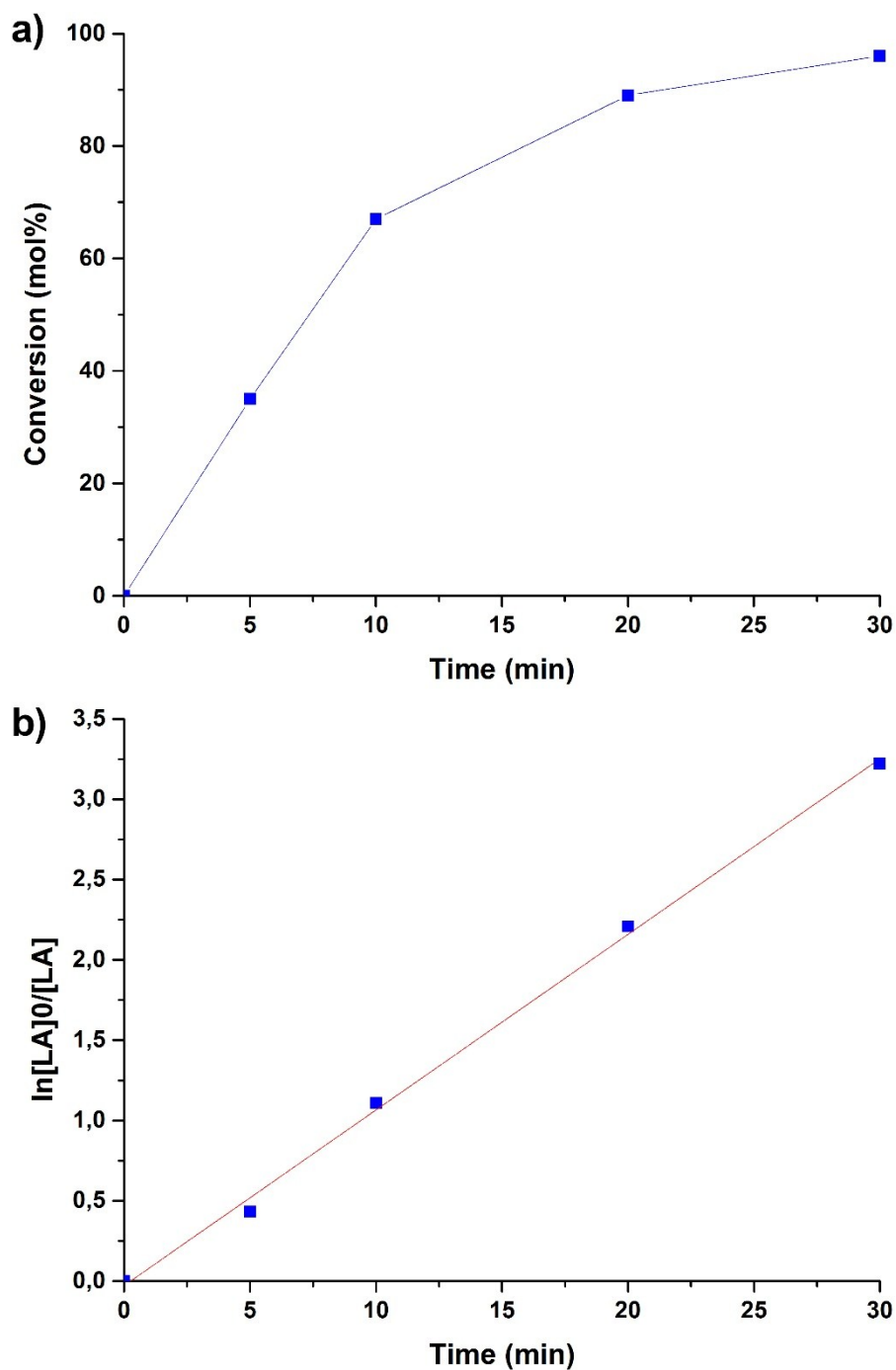


Figure S17. Polymerization of *rac*-LA catalyzed by C (a) with the corresponding plot of $\ln([LA]_0/[LA])$ versus time.

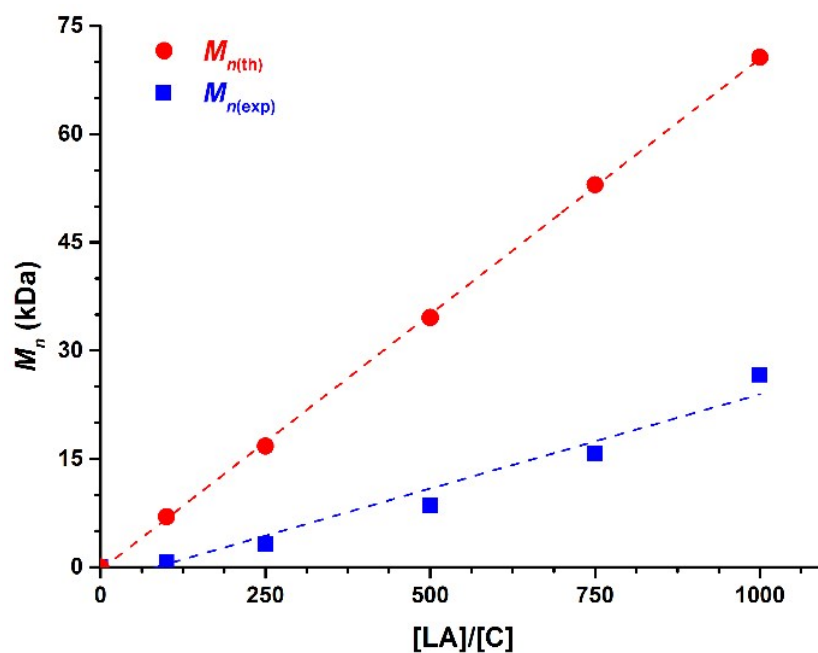


Figure S18. Plot of number-averaged molecular weights $M_{n(exp)}$ (square) vs monomer to initiator ratio with theoretical molecular weights $M_{n(th)}$ (dots) for LA polymerization catalyzed by **C** (reaction conditions in Table 1).

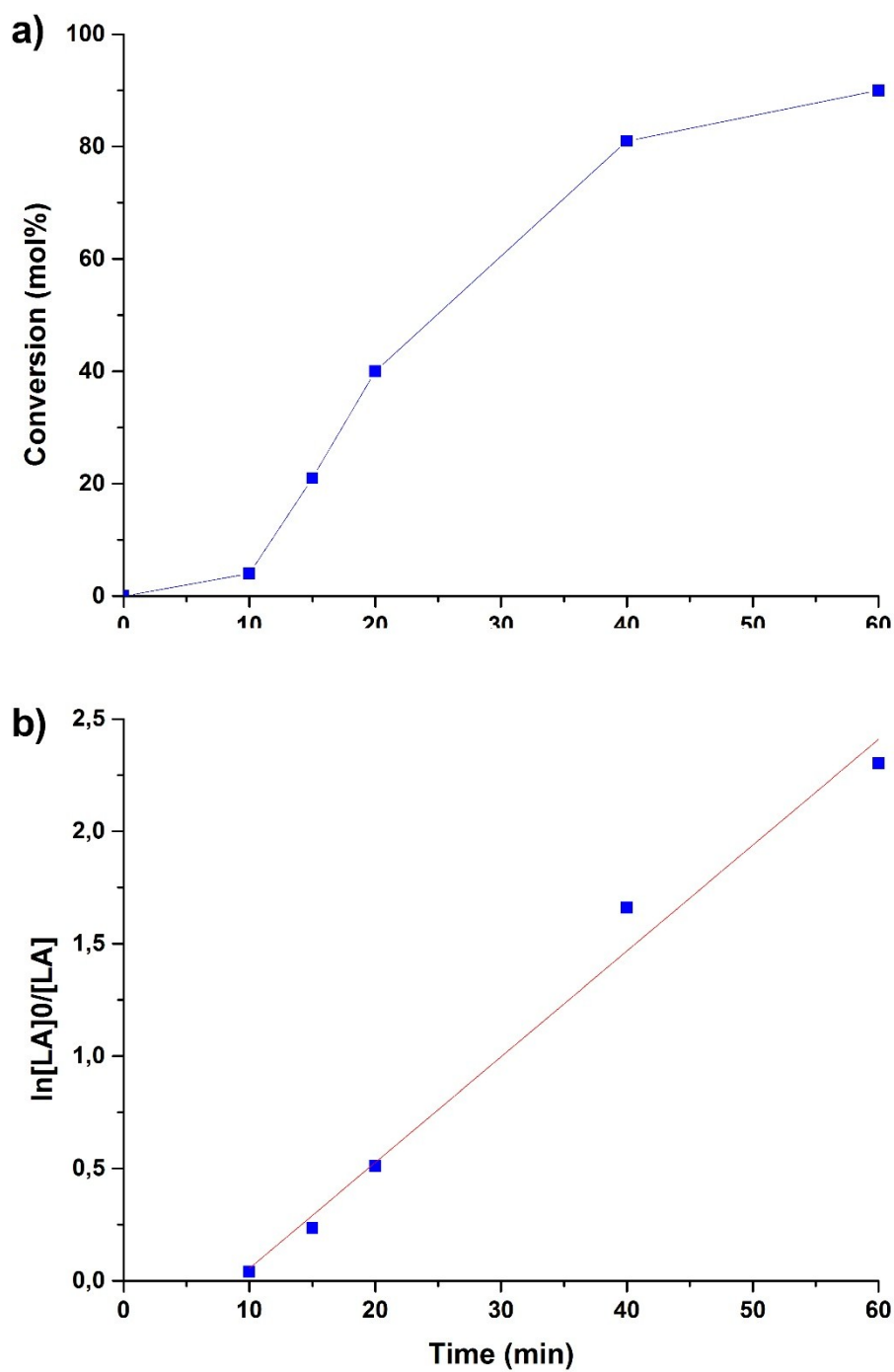


Figure S19. Polymerization of *rac*-LA catalyzed by C' (a) with the corresponding plot of $\ln([LA]_0/[LA])$ versus time.

Table S1. Reaction rate as a function of catalyst concentration for ROP of LA with C.

Entry	[C] (M)	[<i>rac</i> -LA]/[C] (molar ratio)	r (M s ⁻¹)
S1	$1.42 \cdot 10^{-2}$	50	$2.7 \cdot 10^{-3} \pm 1.5 \cdot 10^{-4}$
S2	$7.07 \cdot 10^{-3}$	100	$1.7 \cdot 10^{-3} \pm 1.3 \cdot 10^{-4}$
S3	$4.72 \cdot 10^{-3}$	150	$1.2 \cdot 10^{-3} \pm 6.3 \cdot 10^{-5}$
S4	$3.53 \cdot 10^{-3}$	200	$7.3 \cdot 10^{-4} \pm 3.7 \cdot 10^{-5}$

Reaction conditions: [*rac*-LA] = 0.706 M, TCE-*d*₂ = 0.6 mL, T = 80 °C; reactions carried out inside NMR tubes and monitored with interval of one minute.

5. Gel Permeation Chromatography.

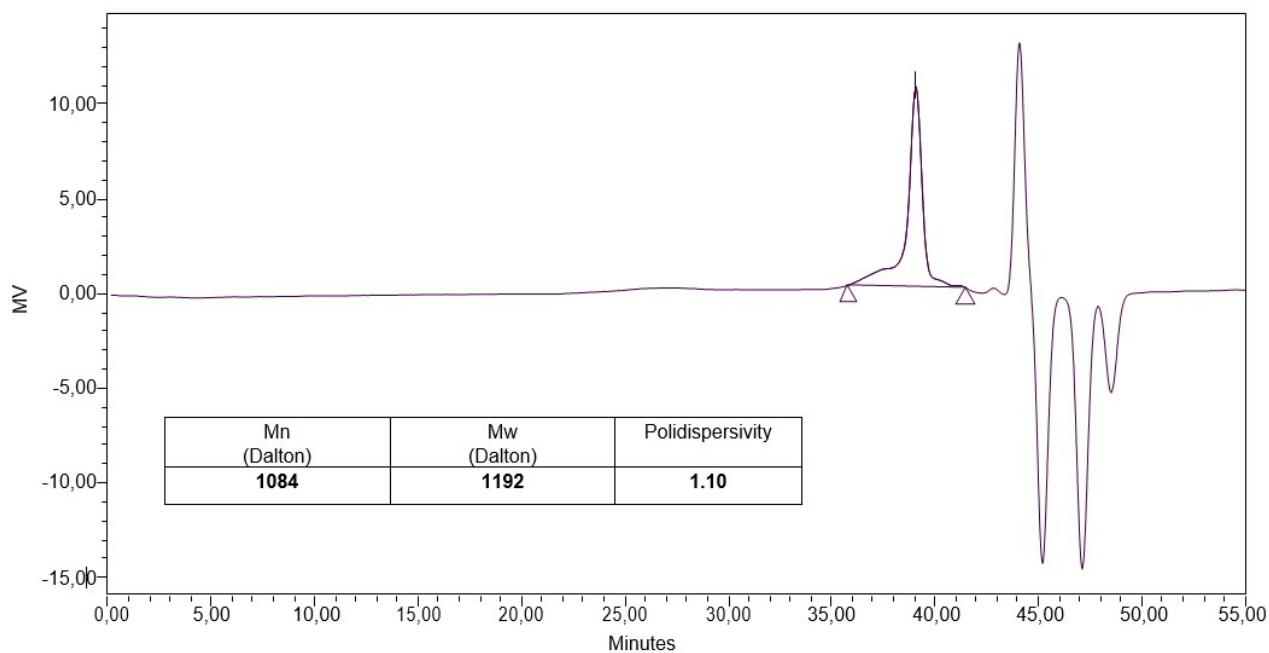


Figure S20. Gel permeation chromatogram of the polymer sample from entry **1** of Table 1.

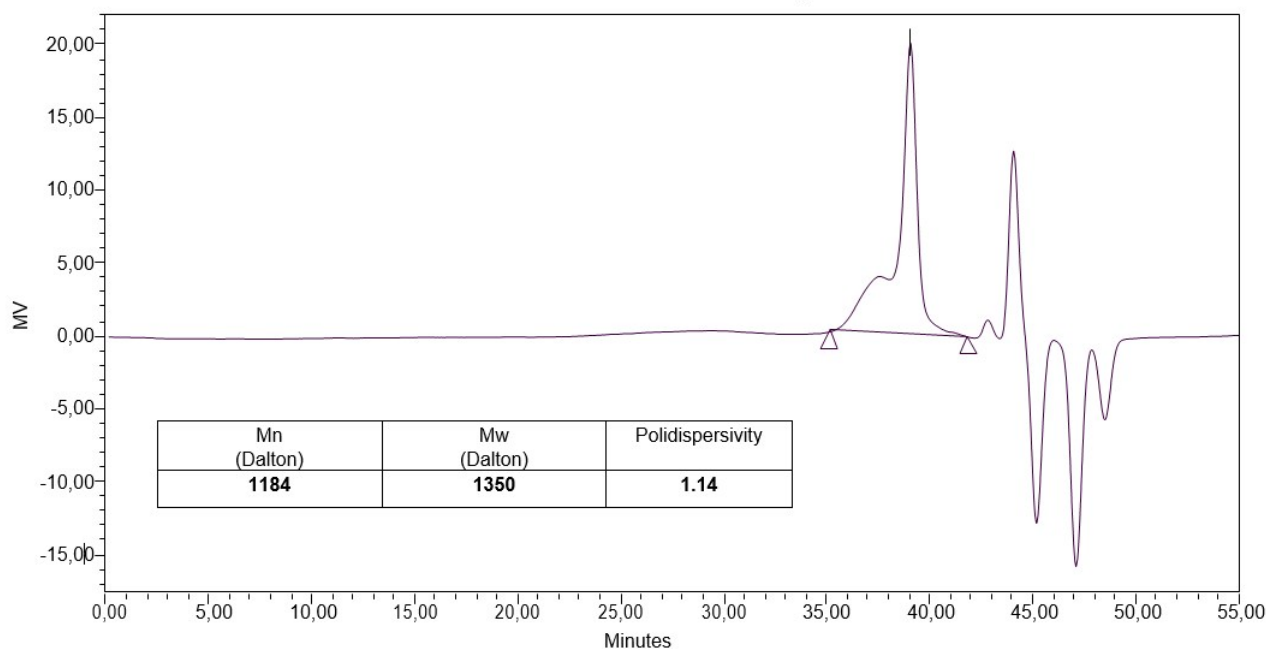


Figure S21. Gel permeation chromatogram of the polymer sample from entry **2** of Table 1.

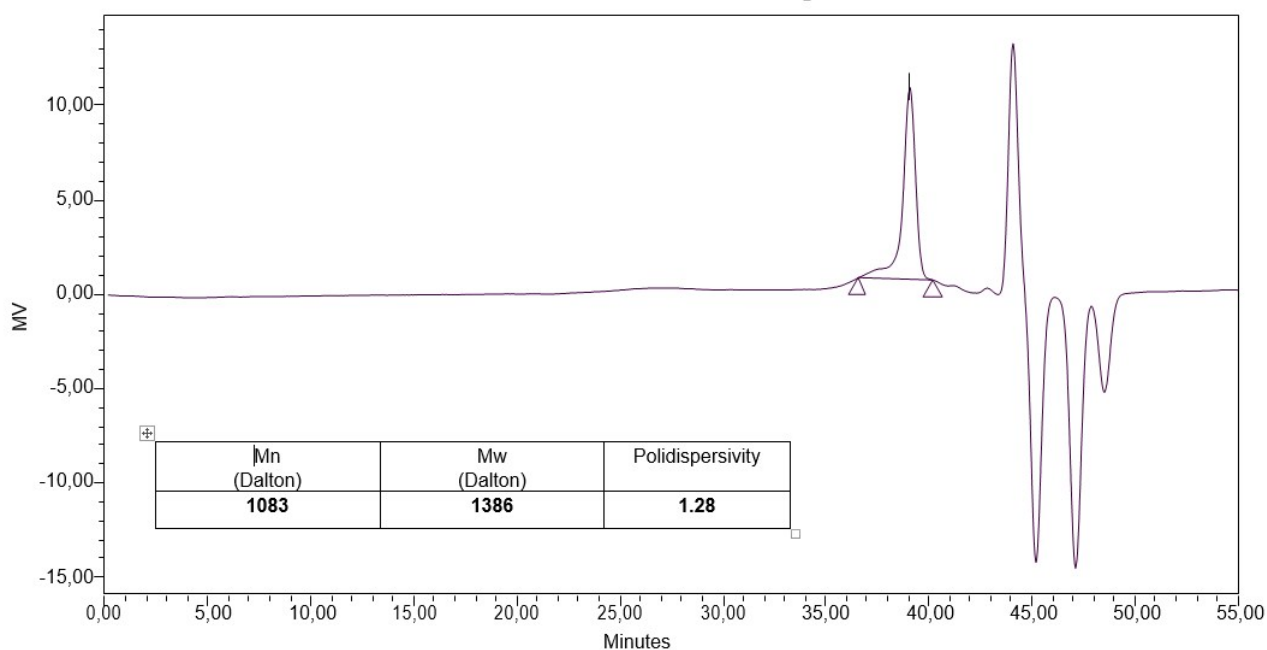


Figure S22. Gel permeation chromatogram of the polymer sample from entry **3** of Table 1.

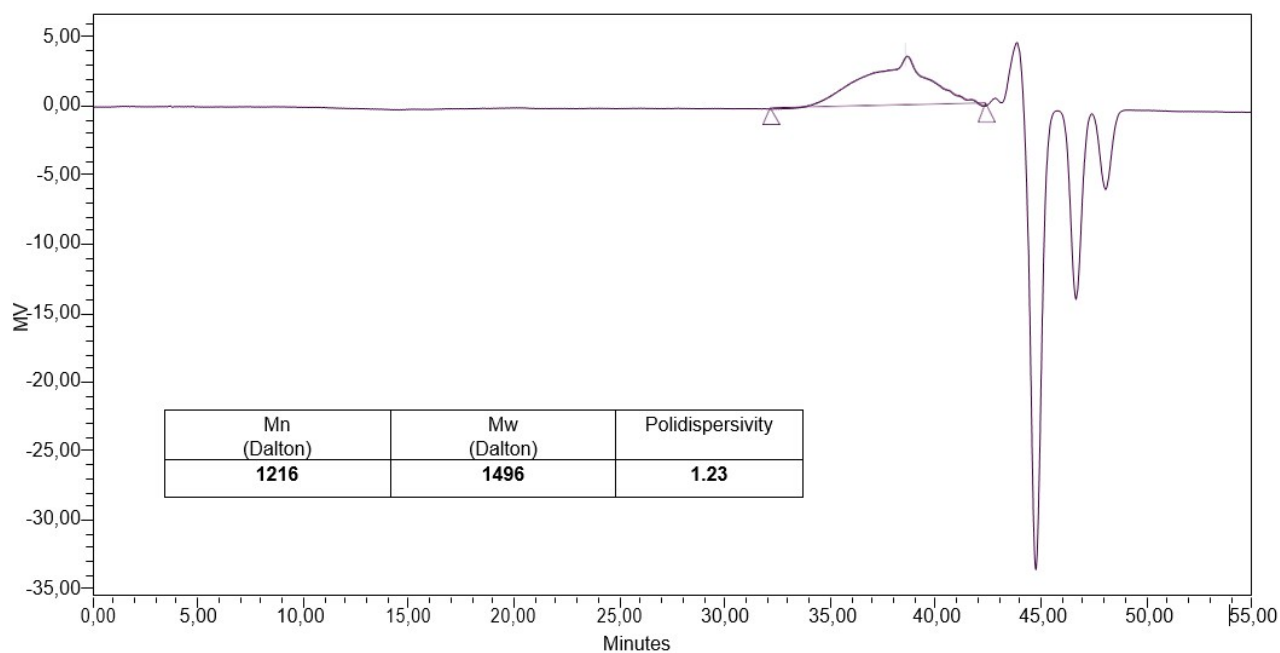


Figure S23. Gel permeation chromatogram of the polymer sample from entry **4** of Table 1.

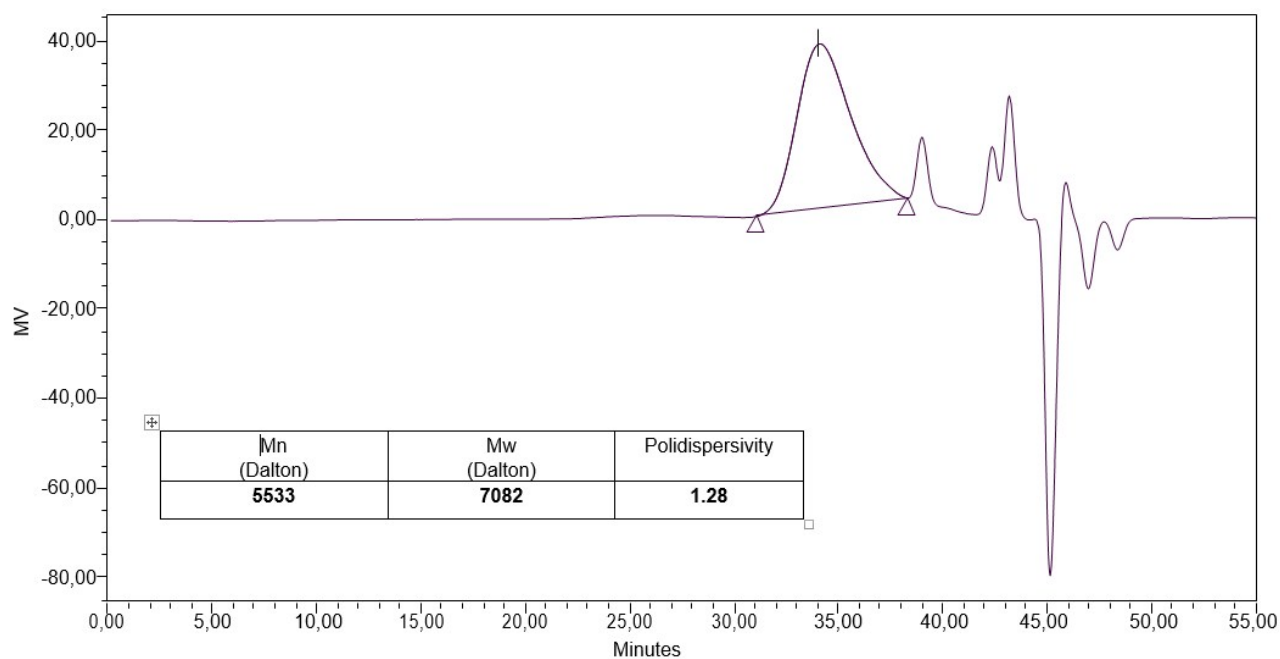


Figure S24. Gel permeation chromatogram of the polymer sample from entry **5** of Table 1.

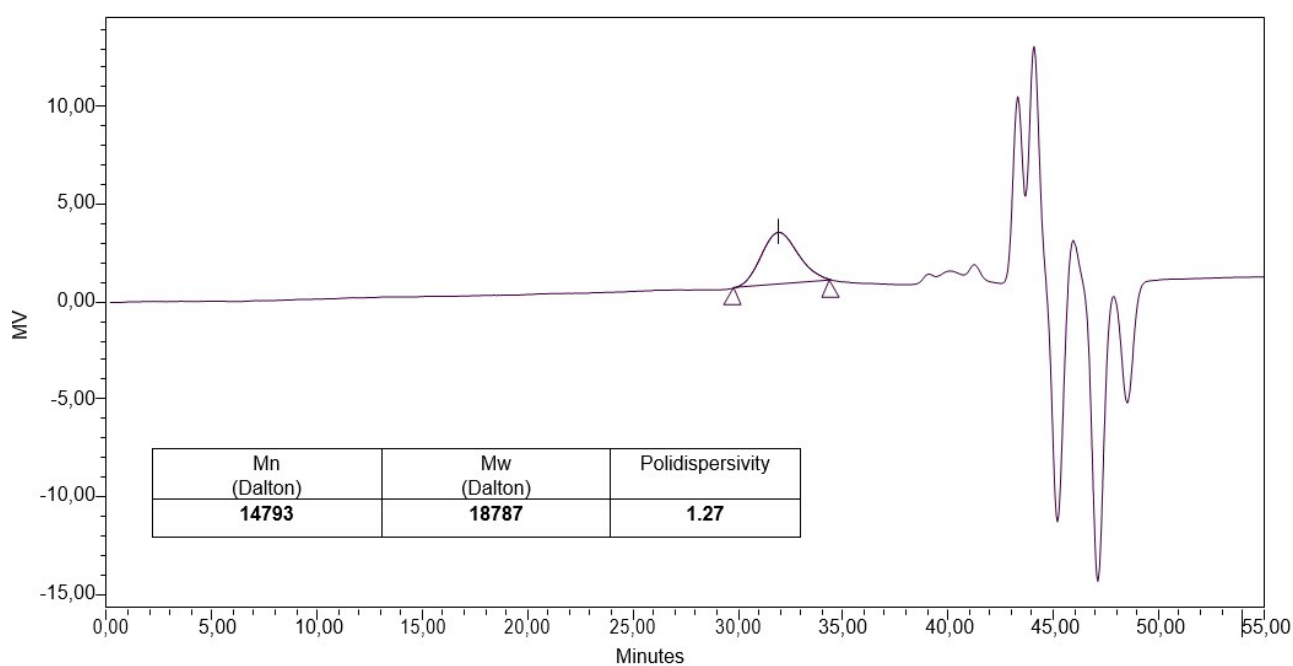


Figure S25. Gel permeation chromatogram of the polymer sample from entry **6** of Table 1.

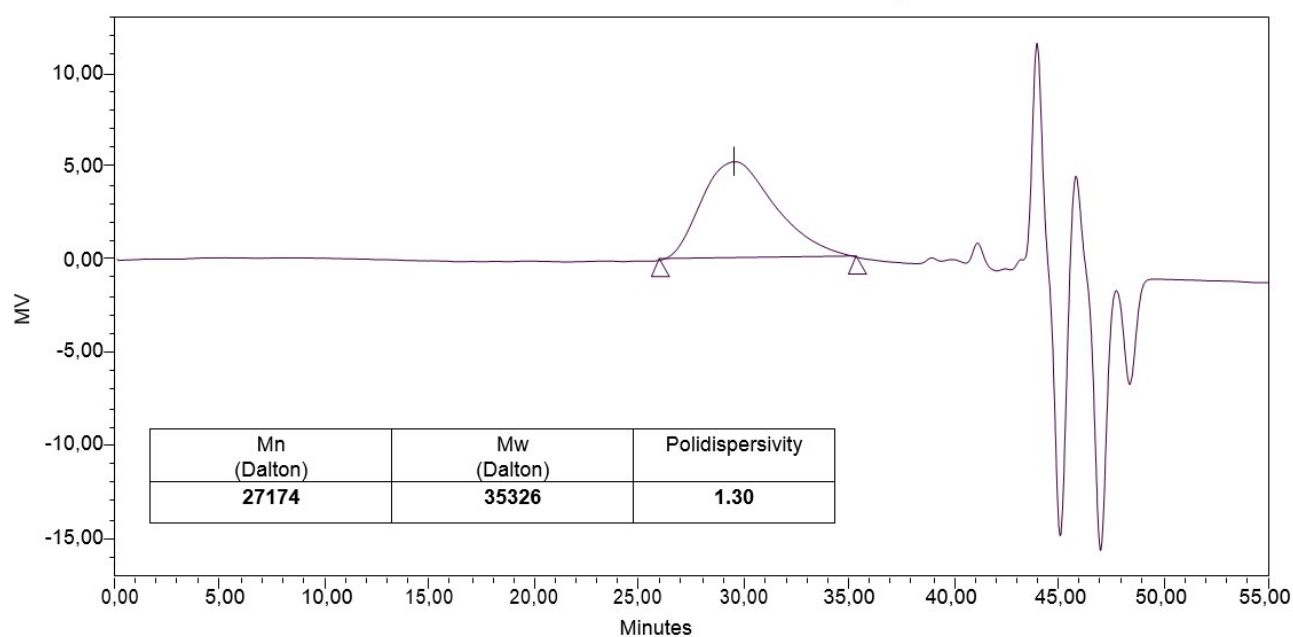


Figure S26. Gel permeation chromatogram of the polymer sample from entry 7 of Table 1.

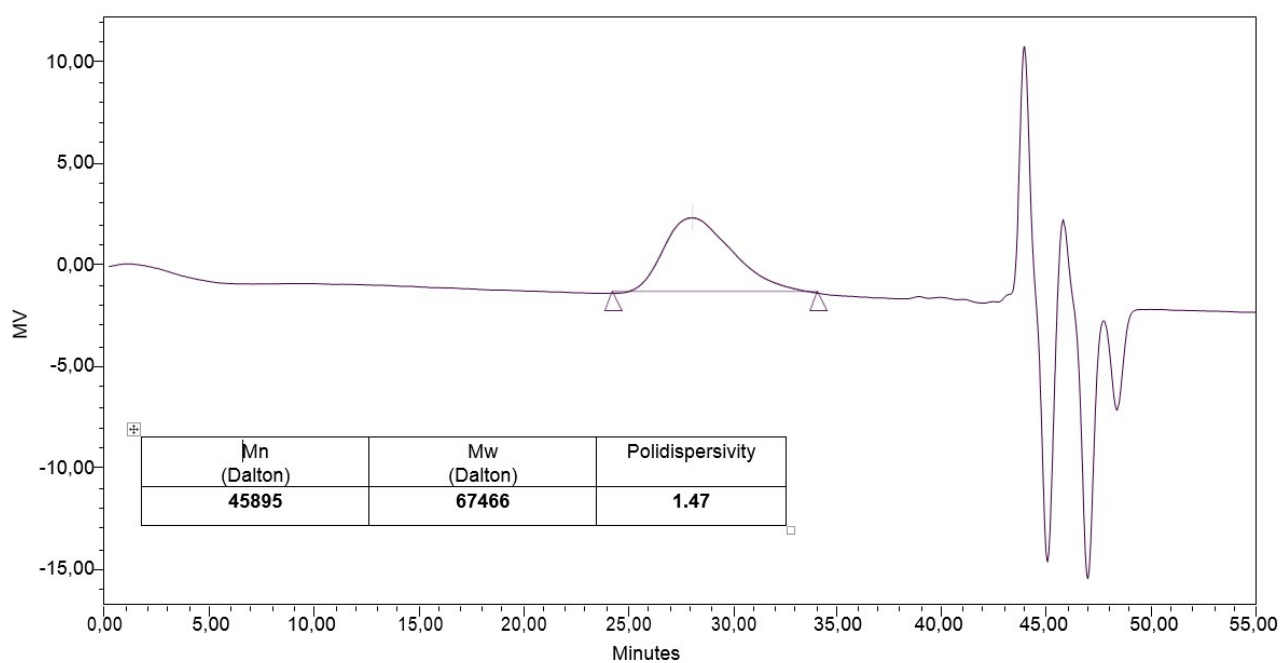


Figure S27. Gel permeation chromatogram of the polymer sample from entry 8 of Table 1.

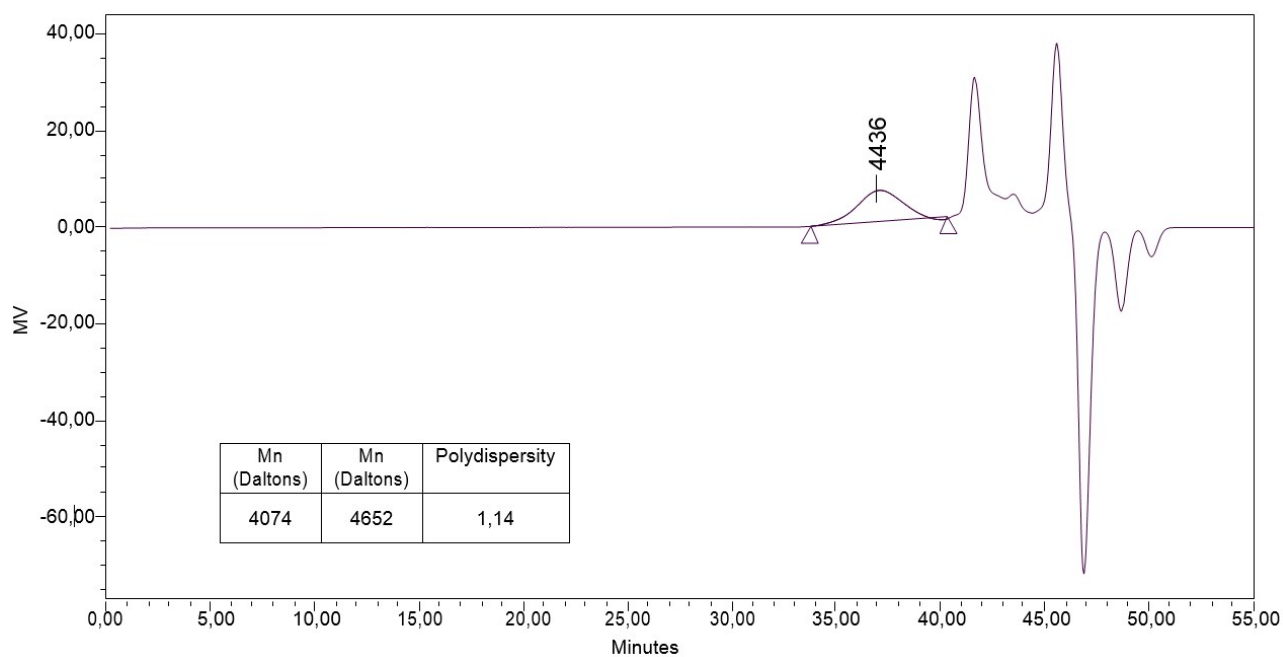


Figure S28. Gel permeation chromatogram of the polymer sample from entry **9** of Table 1.

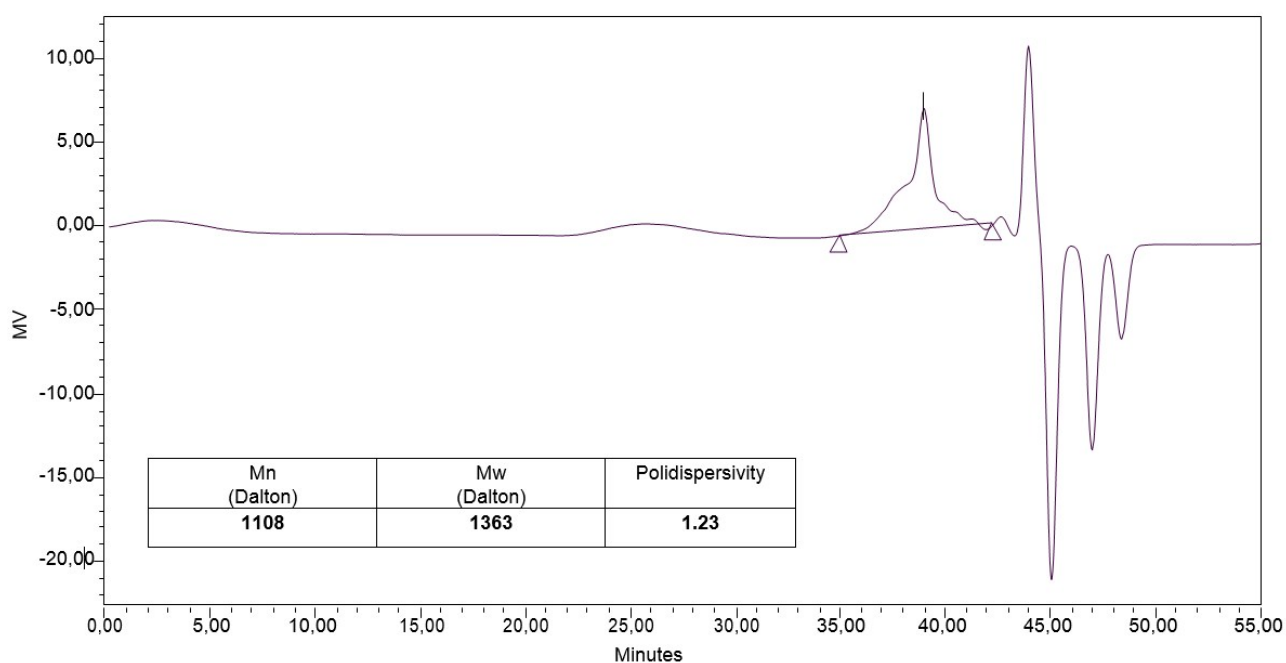


Figure S29. Gel permeation chromatogram of the polymer sample from entry **10** of Table 1.

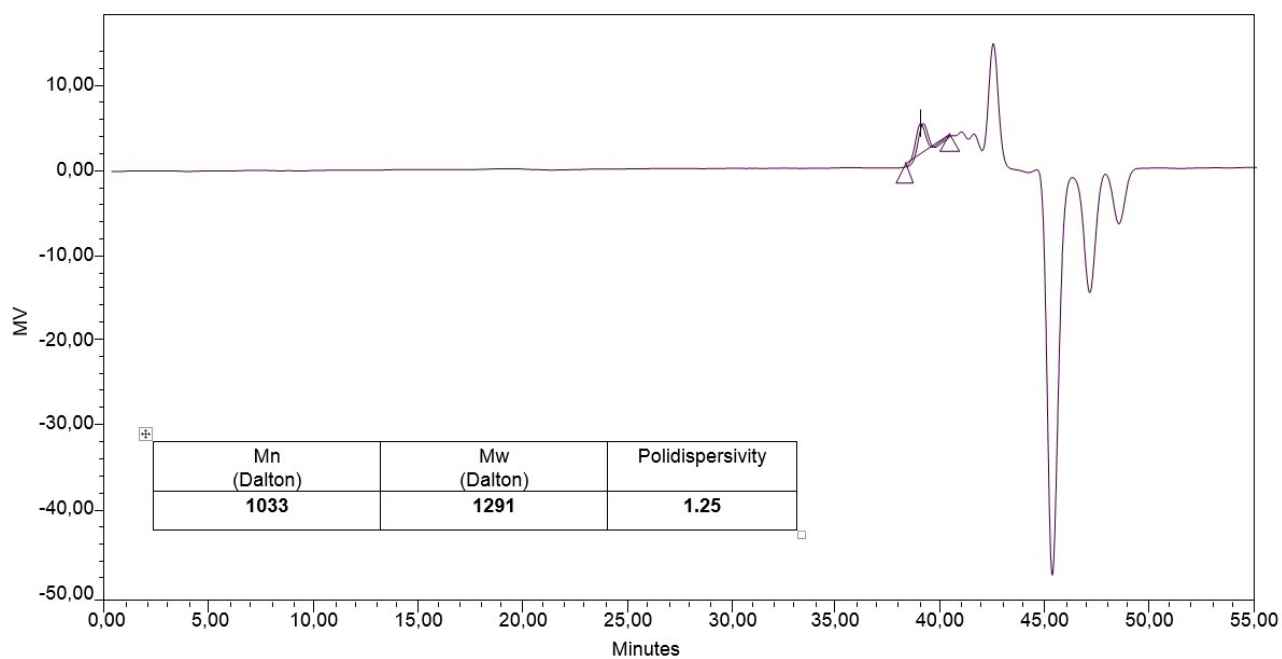


Figure S30. Gel permeation chromatogram of the polymer sample from entry **11** of Table 1.