

Supporting information for

**Towards Active Macro-mesoporous Hydrotreating Catalyst: Synthesis and
Assembly of Mesoporous Alumina Microspheres**

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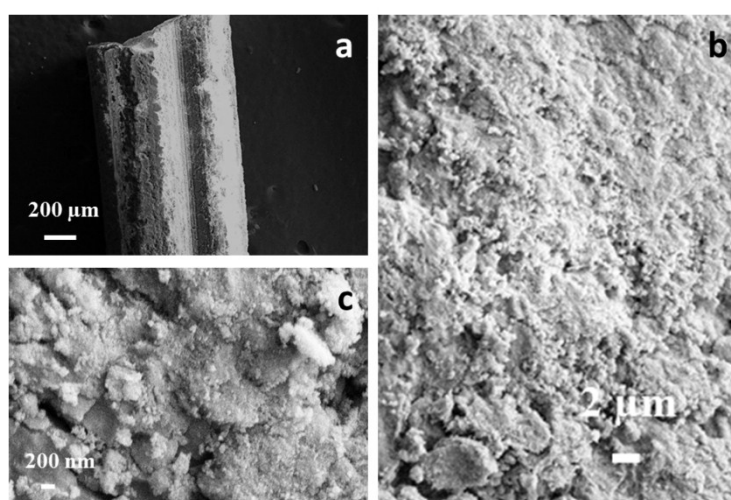


Fig. S1 SEM images of commercial alumina.

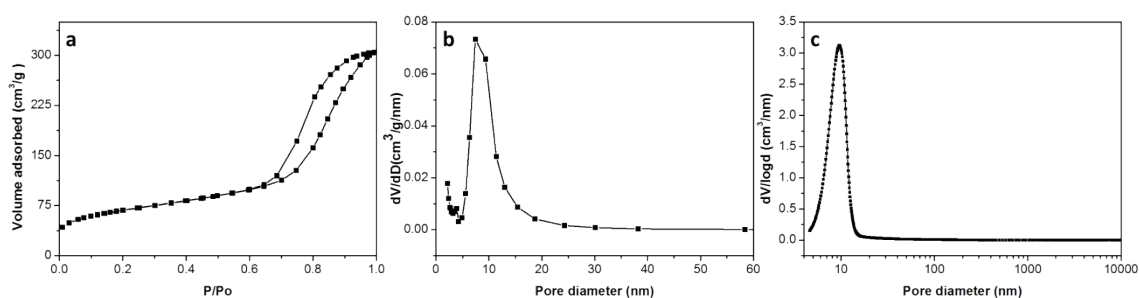


Fig. S2 (a) N₂ adsorption isotherms, (b)BJH pore size distribution and (c) mercury porosimetry data of the commercial alumina.

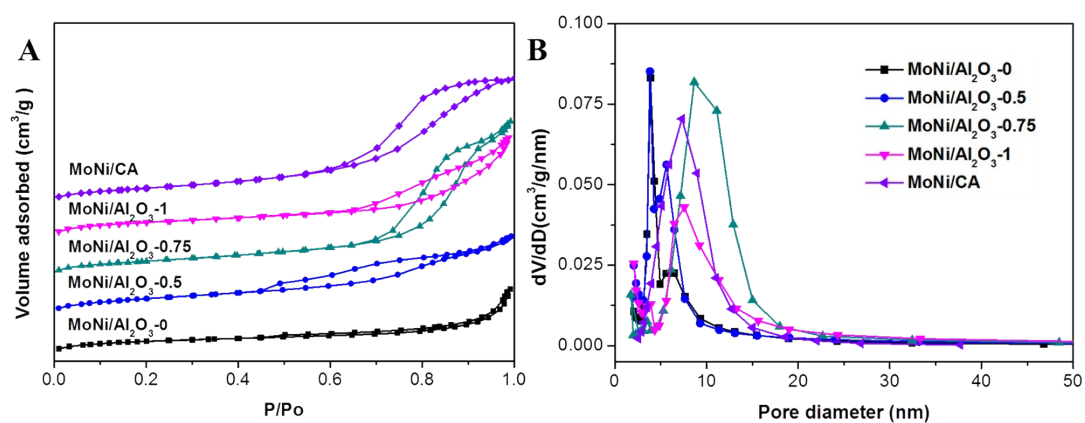


Fig. S3 (A) N₂ adsorption isotherms and (B) BJH pore size distribution of the catalysts.

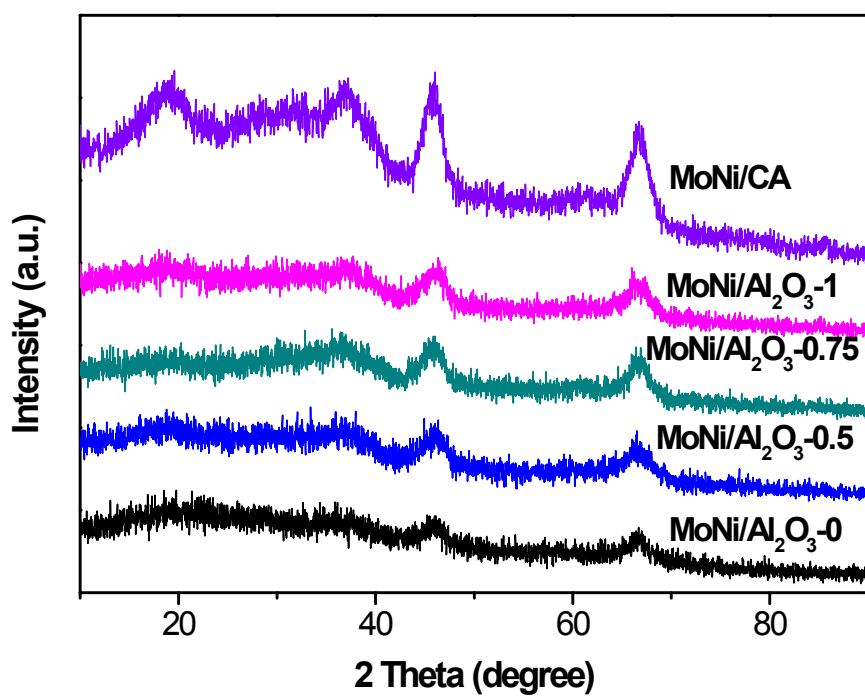


Fig. S4 XRD patterns of MoNi/Al₂O₃-0, MoNi/Al₂O₃-0.5, MoNi/Al₂O₃-0.75, MoNi/Al₂O₃-1 and MoNi/AC.

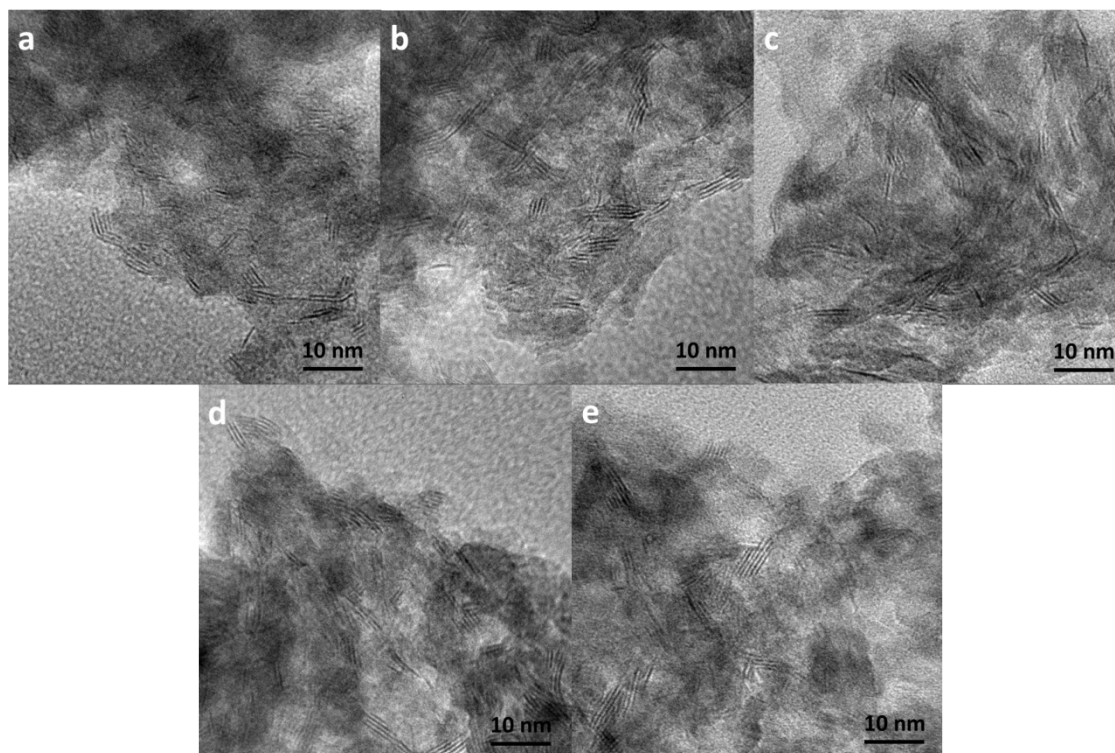
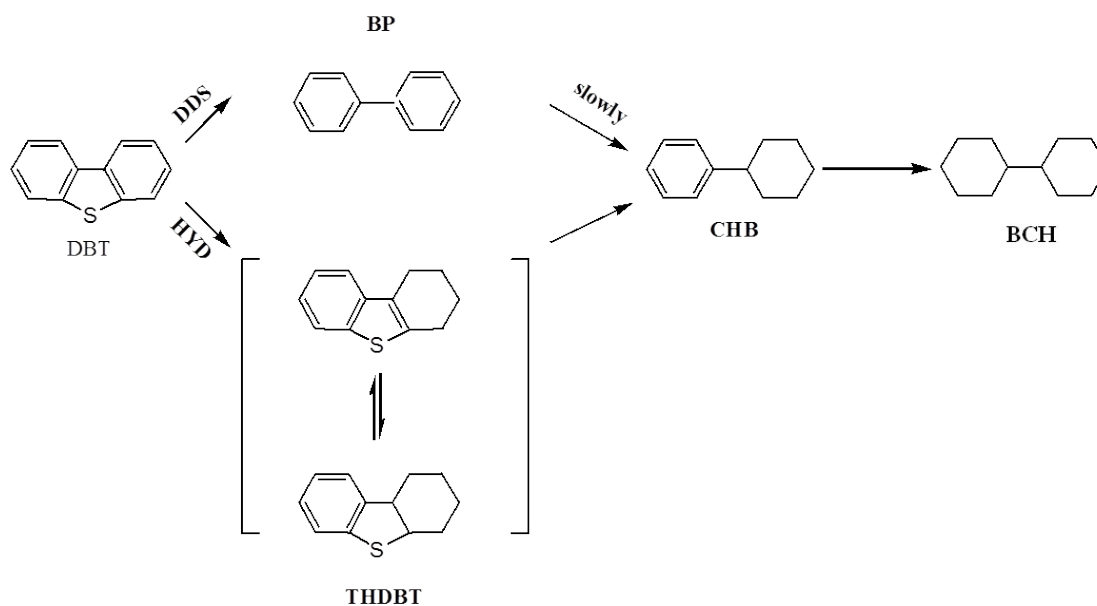


Fig. S5 HRTEM images of the sulfided MoNi/ Al_2O_3 -0(a), MoNi/ Al_2O_3 -0.5 (b), MoNi/ Al_2O_3 -0.75(c), MoNi/ Al_2O_3 -1 (d) and MoNi/AC (e).



Schem

e 1S Reaction network of HDS of DBT.

Table S1 Initial activities and product distributions for the transformation of
dibenzothiophene over sulfided catalysts^a

Sample	R ^b (10 ⁻⁷ mol·g ⁻¹ ·s ⁻¹)	Product selectivity (%)			HYD/DDS
					ratio
		BP ^c	CHB ^c	BCH ^c	
NiMo/Al ₂ O ₃ -0	5.07	87.8	9.9	2.3	0.14
NiMo/Al ₂ O ₃ -0.5	6.09	85.3	12.6	2.1	0.17
NiMo/Al ₂ O ₃ -0.75	6.90	83.9	13.6	2.5	0.19
NiMo/Al ₂ O ₃ -1	5.54	86.3	11.5	2.2	0.16
NiMo/AC	5.28	85.5	12.2	2.3	0.17

a: Reaction conditions: temperature, 270 °C; H₂ pressure, 1.5 Mpa; LHSV, 10 h⁻¹;

H₂/liquid (v/v), 600; feed, 2wt.% DBT in cyclohexane

b: Average specific rate (moles of DBT transformed per second and per gram of catalyst)

c: BP: biphenyl; CHB: cyclohexylbenzene; BCH: bicyclohexyl.

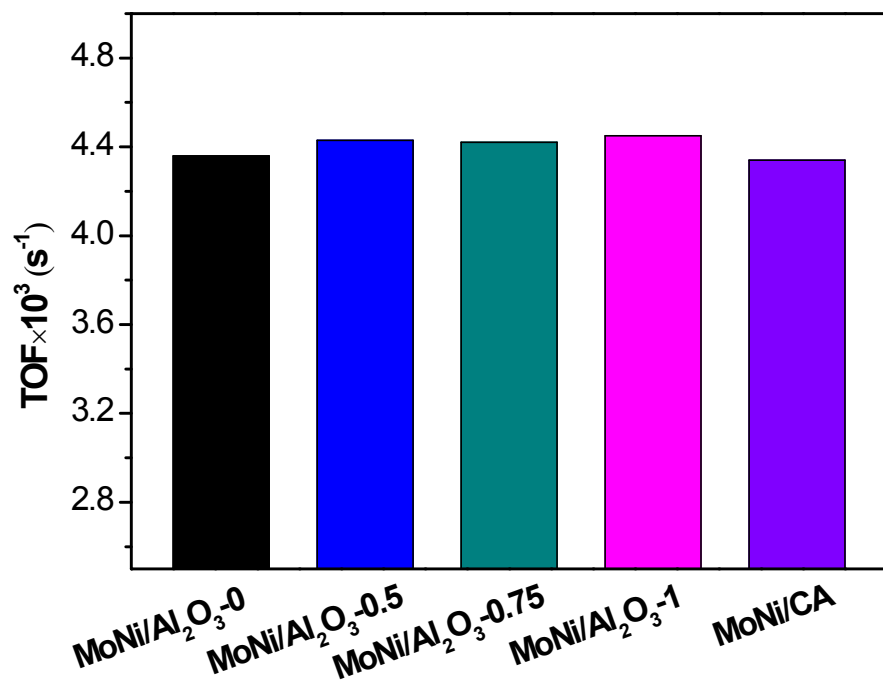


Fig. S6 Turnover frequency (TOF, s⁻¹) of HDS for the sulfided catalysts

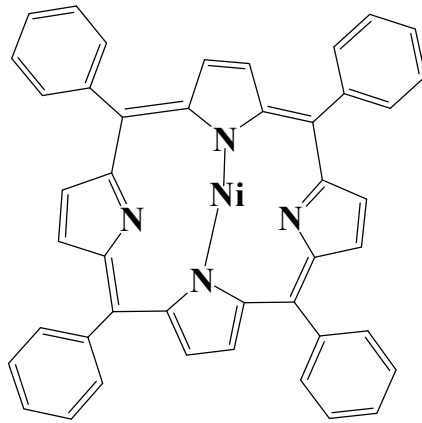


Fig. S7 Molecular formula of Nickel-5, 10, 15, 20-tetraphenylporphyrin (Ni-TPP)