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Supplementary Materials Section

Wolfram W. Rudolph and Gert Irmer: "Raman spectroscopic characterization of light rare earth ions: La³⁺, Ce³⁺, Pr³⁺, Nd³⁺ and Sm³⁺ - Hydration and ion pair formation." Dalton Trans., 02/2017



Figure S1A. UV absorption spectrum of an aqueous 0.0025 molL⁻¹ Ce(ClO₄)₃ solution at 22 °C. Shown are the five characteristic UV bands of Ce³⁺(aq). Note the small contribution at 296.4 nm.



Figure S1B. UV/UV-vis absorption spectrum of an aqueous $0.317 \text{ molL}^{-1} Pr(ClO_4)_3$ solution at 22 °C.



Figure S1C. UV/UV-vis absorption spectrum of an aqueous 0.256 molL⁻¹ Nd(ClO₄)₃ solution at 22 °C.



Figure S2. Raman spectrum of crystalline NH_4ClO_4 at 22 °C Note, that not all band positions are given. The inset shows the $v_1 ClO_4^-$ band in full detail. Note the band split into two components at 922 cm⁻¹ and 937.5 cm⁻¹. The band at 922 cm⁻¹ is due to Fermi resonance.



Figure S3. Raman scattering profiles (from top to bottom: R_{VV} , R_{VH} and R_{ios}) of a 0.317 molL⁻¹ Pr(ClO₄)₃ solution at 22 °C. Left inset: Isotropic profile of the low frequency region shows the v₁ Pr-O breathing mode at 347 cm⁻¹ in detail. Right inset: The isotropic band profile of the v₁ ClO₄⁻¹ mode at 932.15 cm⁻¹ at full scale



Figure S4. Raman scattering profiles (from top to bottom: R_{VV} , R_{VH} and R_{ios}) of a 0.256 molL⁻¹ Nd(ClO₄)₃ solution at 22 °C. Inset: Isotropic profile of the low frequency region showing the v₁ Nd-O breathing mode at 352 cm⁻¹.



Figure S5. Raman scattering profiles (from top to bottom: R_{VV} , R_{VH} and R_{ios}) of a 0.274 molL⁻¹ Sm(ClO₄)₃ solution at 22 °C. Inset: Isotropic profile of the low frequency region showing the v₁ Sm-O breathing mode at 363 cm⁻¹.



Figure S6. Concentration plot of the integrated band intensity of the breathing mode v_1 Ce-O in Ce(ClO₄)₃(aq). Concentrations of Ce(ClO₄)₃: at 0.202, 0.404, 0.942, and 1.889 molL⁻¹ (R² = 0.998).



Figure S7. <u>Upper panel:</u> Raman scattering profiles (R_{VV} , R_{VH} and R_{iso} , with slightly thicker line) of a 1.281 molL⁻¹ Ce(ClO₄)₃ solution in heavy water at 22 °C. Inset: Isotropic profile (R_{iso}) of the low frequency region showing the v_1 Ce-O breathing mode at 326 cm⁻¹ of the [Ce(OD₂)₉]³⁺ species. <u>Lower panel</u>: The same scattering profiles of a 2.562 molL⁻¹ Ce(ClO₄)₃ solution in heavy water. Inset: Isotropic profile (R_{iso}) of the terahertz frequency region showing the v_1 Ce-O breathing mode at 328 cm⁻¹ of the [Ce(OD₂)₉]³⁺ species.



Figure S8. Raman scattering profiles (From top to bottom: I_{VV} , I_{iso} , and underneath I_{VH} ; not that I_{iso} with slightly thicker line) of aqueous Ce(ClO₄)₃ solutions in the water deformation and O-H stretch region from $1500 - 4200 \text{ cm}^{-1}$. Note the evolution of the mode at 3548 cm⁻¹ with increasing Ce(ClO₄)₃ concentration. This mode is due to weak, decoupled O-H stretching modes (weak H-bonds: H-OH····OClO₃⁻). The broad water bands at 3235 and 3400 cm⁻¹ diminishes with increasing Ce(ClO₄)₃ concentration.



Figure S9. Raman scattering profiles of a 3.118 molL⁻¹ Ce(ClO₄)₃ solution in wavenumber region from $1500 - 4200 \text{ cm}^{-1}$. Upper panel shows the I_{VV} and I_{VH} scattering profiles. The weak band at 1627 cm⁻¹ is due to the deformation mode of water and the mode at 3230 cm⁻¹ the Fermi resonance component of the weak and very broad O-H stretching profile. The dominant band at 3538 cm⁻¹ is due to the stretching band of water forming very weak H-bonds with ClO₄⁻ (O-H····OClO₃⁻). The lower panel shows the isotropic scattering component.



Figure S10. Isotropic Raman spectrum, R_{iso} , of a 3.118 molL⁻¹ Ce(ClO₄)₃ in terahertz region at 22°C. The v₁ Ce-O breathing mode is broad and peaks at 340 cm⁻¹. Given are the measured Raman spectrum, the sum curve of the fit and the two band components at 327 cm⁻¹ and 344 cm⁻¹.



Figure S11A. Raman scattering profiles in R-format (R_{VV} and R_{VH}) of the Ce(ClO₄)₃ hydrate melt (Ce(ClO₄)₃ plus 6 H₂O) at 161°C in the low frequency region (60-800 cm⁻¹). The deformation modes of ClO₄⁻ at 458 cm⁻¹ and 628 cm⁻¹ are the prominent bands in the spectrum. (The polarized band profiles, R_{VV} , for both deformation modes including the sum curve of the fit and the components are given in greater detail in Figs. S11B and C further below.) The inset shows the R_{iso} scattering profile in the terahertz frequency region in larger detail (v_1 Ce-O breathing mode at 326 cm⁻¹).



Figure S11B. Polarized Raman scattering profile in R-format (R_{VV}) of the Ce(ClO₄)₃ hydrate melt (Ce(ClO₄)₃ plus 6 H₂O) at 161°C from 370 – 575 cm⁻¹. Shown is the profile of the deformation mode, $v_2(e)$ ClO₄⁻ at 458 cm⁻¹ which contains clearly two band components. Given is the measured Raman band profile, the sum curve of the band fit as well as the two band components. Underneath the band profiles, the residual curve of the fit is given (R_{VV} -profile – sum curve; $R^2 = 0.9999$).



Figure S11C. Polarized Raman scattering profile in R-format (R_{VV}) of the Ce(ClO₄)₃ hydrate melt (Ce(ClO₄)₃ plus 6 H₂O) at 161°C from 530- 775 cm⁻¹. Shown is the band complex of the deformation mode, the $v_4(f_2)$ ClO₄⁻ at 628 cm⁻¹. Given is the measured Raman profile, the sum curve of the band fit as well as the three band components. Underneath the band profiles, the residual curve of the fit is given (R_{VV} -profile – sum curve; $R^2 = 0.9999$).



Figure S12. Isotropic Raman scattering profiles in R format (R_{iso}) of the v_1 ClO₄⁻ band at 934 cm⁻¹ of the Ce(ClO₄)₃ hydrate melt (Ce(ClO₄)₃ + 6 H₂O) at 161°C. Given is the measured curve as well as the sum curve of the band. The component bands are located at 914 cm⁻¹, 933 cm⁻¹ and 941 cm⁻¹. Underneath the profiles, the residual curve of the fit is given (R_{iso} -profile – sum curve; $R^2 = 0.9999$).



Figure S13. Raman scattering profiles of a Ce(ClO₄)₃ hydrate melt (Ce(ClO₄)₃ plus 6 H₂O) at 161°C. The upper panel depicts the I_{VV} and I_{VH} scattering profiles in the symmetric and antisymmetric stretching region (v_1 and v_3) region of ClO₄⁻. The inset shows the weak v_3 region of ClO₄⁻ in larger detail. The three components in the v_3 region of ClO₄⁻ are clearly seen. The lower panel shows the isotropic scattering profile with the v_1 ClO₄⁻ band region with its peak position at 934 cm⁻¹ and the very weak band at 1033 cm⁻¹.