Supporting Information

Noble-metal-free MoS₂ nanosheets modified-InVO₄ heterostructures for

enhanced visible-light-driven photocatalytic H₂ production

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Experimental section

General Procedures: Indium chloride (InCl₃), ethanolamine, ammonium vanadate (NH₄VO₃), sodium molybdate (Na₂MoO₄•2H₂O), thiourea, ethanol and methanol were obtained from Mike Chemical Reagent Co., Ltd Cd. All reagents were used as received without further purification.

Characterization: Powder X-ray diffraction (XRD) patterns were performed on a Rigaku D/Max-2550pc powder diffractometer equipped with Cu-Ka ($\lambda = 0.15406$ nm) radiation in the 20 ranging from 10° to 80° with a scan rate of 10° per minute. The optical properties of the products were characterized by a Varian Cary 500 UV-vis spectrophotometer, in which BaSO₄ was used as the internal reflectance standard. Xray photoelectron spectroscopy (XPS) measurements were performed by using a VG ESCALAB MKII XPS system with Al Ka X-ray source and a charge neutralizer. The morphology of the photocatalysts was determined by using a field-emission scanning electron microscopy (FE-SEM, vltra55), high-resolution transmission electron microscopy (HR-TEM, JEM 2010) with an accelerating voltage of 200 kV. The SEM and TEM samples were obtained by depositing a drop of diluted suspensions in ethanol on a silicon chip and carbon-film-coated copper grid, respectively. Raman spectra of products were recorded at room temperature using Raman spectrometer (J-Y T64000) in the backscattering geometry with 514.5 nm wavelength incident laser light.

Preparation of InVO₄.

In a typical synthesis of $InVO_4$, 1 mmol of $InCl_3$ and 1 mmol NH_4VO_3 were dissolved in 60 mL of distilled water containing 10 mM ethanolamine, which was then transferred into a 100 mL Teflon-lined autoclave and kept at 150 °C for 16 h. After that,

the green precipitate was collected by centrifugation, washed two times with distilled water and ethanol, and then dried in an oven at 60 °C for 12 h.

Preparation of MoS₂/InVO₄ photocatalysts.

In a typical synthesis of $MoS_2/InVO_4$ composite photocatalyst, 500 mg of the prepared InVO₄ was sonicated thoroughly in 50 ml aqueous solution containing 7.5 mg Na2MoO₄-2H₂O and 15 mg thiourea. Next, the heterogeneous solution was transferred into a 100 mL Teflon-lined stainless steel autoclave and kept at 200 °C for 24 h. After the reaction solution was cooled to room temperature, the grey precipitate was separated by centrifugation, washed three times with absolute ethanol, and then dried in an oven at 60 °C to obtain 1 wt% MoS₂/InVO₄ sample. In order to investigate the effect of MoS₂ content in the MoS₂/InVO₄ composite photocatalysts on the photocatalytic H₂ evolution activity, the weight percentages of MoS₂ to InVO₄ were varied from 0 to 5 (0, 1, 2, 3, 4 and 5 wt%) by varying the weight of Na2MoO4+2H₂O and thiourea (2 equiv of Na2MoO4+2H₂O), the resulting samples were annealed at 400 °C for 3 h under nitrogen atmosphere.

Photocatalytic H₂ production.

The photocatalytic H₂-evolution experiments were performed in a 350 mL Pyrex flask, which was connected to a closed gas-circulation lation glass system. H₂ production experiments experiment was performed by dispersing 100 mg powder photocatalyst in 250 ml aqueous solution containing methanol (20%, V/V) as sacrificial electron donor. After the solution was degassed to remove air completely, the mixture solution was irradiated by a 300 W Xe lamp equipped with a cut-off filter ($\lambda > 420$ nm) with gentle magnetic stirring. The evolving H₂ were periodically detected in situ on a gas chromatograph (Jiedao GC1690, TCD, argon as a carrier gas and MS-5A molecular sieve column). The standard errors in the photocatalytic H_2 evolution performance for all photocatalytic experiments were less than 2.0%.



Fig. S1 SEM image of pure InVO₄ sample.



Fig. S2 TEM image of pure InVO₄ sample.



Fig. S3 TEM image of pure InVO₄ sample.



Fig. S4 X-ray diffraction patterns for 3 wt% $MoS_2/InVO_4$ photocatalyst before and after visible light irradiation.