## Reagents

All experiments on transuranium elements were conducted in the Atalante facility (CEA Marcoule, France), in a laboratory dedicated to these highly radioactive elements. All experiments were carried out with approved safety operating procedures in negative pressure glove box. <sup>239, 240</sup> Pu(IV) solution was prepared by dissolving the corresponding oxide, PuO<sub>2</sub>, with concentrated HNO<sub>3</sub> in a glove box. It was then purified by fixation on a DOWEX anion exchange resin at 7 mol.L<sup>-1</sup> HNO<sub>3</sub> and elution with 0.5 mol.L<sup>-1</sup> HNO<sub>3</sub>. Oxidation state and concentration of the actinide solutions were checked by Vis-NIR spectrophotometry. **DEHBA** *N*,*N*-dialkyl amides were synthesized by Pharmasynthese (France). DODECANE is purchased from Aldrich.

## Sample preparation

The **DEHBA** samples were prepared by liquid-liquid extraction at 25°C. Nitric acid aqueous phases ([HNO<sub>3</sub>] = 3 mol.L<sup>-1</sup> and 11 mol.L<sup>-1</sup>), containing the Pu(IV were contacted for 15 minutes by means of an automatic vortex shaker with organic solutions (aqueous phase volume over organic phase volume ratio: Vaq/Vorg = 1) containing the ligand. Before the extraction, the organic phase was pre-equilibrated with the targeted nitric acid (Vaq/Vorg = 3) for 30 min at 25°C. The phases were separated after centrifugation. Plutonium concentrations in the organic phase were determined by alpha-spectroscopy. The evolution of Pu(IV) was monitored with Vis-NIR spectrophotometry during several days corresponding to the delay between the samples preparation and their analysis at the Synchrotron.

Table 1. Compositions of <b>DEHB</b>	solutions used in EXAFS	and vis-NIR measurements.
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	C <sup>aq</sup> hno3	$C^{\text{org}}_{\text{HNO3}}$ <sup>b</sup>	[Ligand] <sub>org</sub> (mol.L <sup>-1</sup> ) <sup>a</sup>	[Pu] <sub>org</sub> (mmol.L <sup>-1</sup> )	Measurements
Pu-DEHBA 2.8M	2.8 M	0.7	1.2	2.2 <sup>c</sup>	vis-NIR
Pu-DEHBA 3.3M	3.3 M	0.8	1.2	2.2 <sup>d</sup>	EXAFS, vis-NIR
Pu-DEHBA 5M	5 M	1.9	2.7	2.2 <sup>c</sup>	vis-NIR
Pu-DEHBA 8M	8 M	3.4	2.7	2.2 <sup>c</sup>	vis-NIR
Pu-DEHBA 11M	11M	4.2	2.7	2.2 <sup>d</sup>	EXAFS, vis-NIR

<sup>a</sup> Amide concentration in the organic phase. <sup>b</sup> HNO<sub>3</sub> concentration in the organic phase estimated values from ref [20] and [24] of the main manuscript.<sup>c</sup> Pu(IV) concentration in the organic phase estimated values from ref [20] and [24] of the main manuscript. <sup>d</sup>Alfa spectroscopy measurements.

## EXAFS data acquisition and treatment

EXAFS spectra were recorded at Synchrotron SOLEIL (Saint-Aubin, France) (2.75 GeV at 450 mA), at the MARS beamline. MARS is equipped with a water-cooled Si(220) double crystal horizontal focusing monochromator (DCM). Beam vertical collimation/focusing and rejection of higher-order harmonics were achieved with two Ptcoated mirrors before and after the DCM. A 13-element HPGe solid state detector (ORTEC) was used for data collection in fluorescence mode. Monochromator energy calibration was carried out at zirconium K-edge. All measurements were performed at room temperature in 200  $\mu$ L double-layered cells specifically designed for radioactive samples. All spectra were recorded at the plutonium L<sub>3</sub> edge (18057 eV for Pu). The data represent averages of 4 scans. Data processing was carried out with the Athena code.<sup>1</sup> After energy calibration, the e0 energy was set at the maximum of the absorption edge. For all the samples, the maximum energy of the absorption edge confirms the redox state +IV for plutonium (18068.2 eV).<sup>2,3</sup> The EXAFS signal was extracted by subtracting a linear pre-edge background and a combination of cubic spline functions for atomic absorption background and then normalized by the Lengeler-Eisenberg procedure. Pseudo-radial distribution functions (PRDF) were obtained by Fourier transform in k<sup>3</sup> $\chi$ (k) using the ATHENA code between 1.5 and 12.5 Å<sup>-1</sup>. For EXAFS data analysis, no fitting procedure is applied. Theoretical scattering phases and amplitudes were calculated by the FEFF9.<sup>4,5</sup> from computational methods.

## **Computational details**

The geometry and frequency calculations have been performed with Gaussian  $09^6$  at the DFT level of theory with the PBEO functional.<sup>7</sup> A small core quasi-relativistic effective core potential (RECP – 60 electrons)<sup>8,9</sup> by the Stuttgart-Cologne group and its corresponding TZ-valence basis set have been used for Pu<sup>10</sup> and the def-TZVP<sup>11</sup> basis sets for the other atoms. Solvent effects have been taken into account with a polarizable continuum model (IEFPCM) corresponding to *n*-dodecane.<sup>6</sup> The *ab initio* DW factors have been calculated at 300 K for each scattering path from the dynamical matrix extracted from the DFT frequency calculation with the DMDW module of FEFF9.<sup>4,5</sup> The EXAFS spectra have then been simulated with FEFF9 considering this *ab initio* DW factors, and all multiple-scattering paths up to a half-path length of 6 Å. The scattering potentials were calculated using a self-consistent loop. The amplitude factor S<sub>0</sub><sup>2</sup> was fixed to 1. The shift of the photoelectron energy origin  $\Delta E_0$  is the only parameter adjusted in the EXAFS simulation and has been taken so that the first amplitudes of the simulated spectra are in phase with the experimental ones.

(1) Ravel, B.; Newville, M. Journal of Synchrotron Radiation 2005, 12, 537.

(2) Den Auwer, C.; Simoni, E.; Conradson, S.; Madic, C. *European Journal of Inorganic Chemistry* **2003**, 3843.

(3) Morss, L. R.; Edelstein, N.; Fuger, J.; Klatz, J. J. *The Chemistry of Actinide and Transactinide Elements*; Springer: Dordrecht, 2006; Vol. 5.

(4) Vila, F. D.; Lindahl, V. E.; Rehr, J. J. *Physical Review B* **2012**, *85*.

(5) Rehr, J. J.; Kas, J. J.; Vila, F. D.; Prange, M. P.; Jorissen, K. *Physical Chemistry Chemical Physics* **2010**, *12*, 5503.

(6) Frisch, M. J.; Trucks, G. W.; Schlegel, H. B.; Scuseria, G. E.; Robb, M. A.; Cheeseman, J. R.; Scalmani, G.; Barone, V.; Mennucci, B.; Petersson, G. A.; Nakatsuji, H.; Caricato, M.; Li, X.; Hratchian, H. P.; Izmaylov, A. F.; Bloino, J.; Zheng, G.; Sonnenberg, J. L.; Hada, M.; Ehara, M.; Toyota, K.; Fukuda, R.; Hasegawa, J.; Ishida, M.; Nakajima, T.; Honda, Y.; Kitao, O.; Nakai, H.; Vreven, T.; Montgomery Jr., J. A.; Peralta, J. E.; Ogliaro, F. ß.; Bearpark, M. J.; Heyd, J.; Brothers, E. N.; Kudin, K. N.; Staroverov, V. N.; Kobayashi, R.; Normand, J.; Raghavachari, K.; Rendell, A. P.; Burant, J. C.; Iyengar, S. S.; Tomasi, J.; Cossi, M.; Rega, N.; Millam, N. J.; Klene, M.; Knox, J. E.; Cross, J. B.; Bakken, V.; Adamo, C.; Jaramillo, J.; Gomperts, R.; Stratmann, R. E.; Yazyev, O.; Austin, A. J.; Cammi, R.; Pomelli, C.; Ochterski, J. W.; Martin, R. L.; Morokuma, K.; Zakrzewski, V. G.; Voth, G. A.; Salvador, P.; Dannenberg, J. J.; Dapprich, S.; Daniels, A. D.; Farkas, ñ. n.; Foresman, J. B.; Ortiz, J. V.; Cioslowski, J.; Fox, D. J.; Revision D.01 ed.; Gaussian, Inc.: Wallingford, CT, USA, 2009.

- (7) Adamo, C.; Barone, V. *Journal of Chemical Physics* **1999**, *110*, 6158.
- (8) Kuchle, W.; Dolg, M.; Stoll, H.; Preuss, H. Journal of Chemical Physics 1994, 100, 7535.
- (9) Cao, X. Y.; Dolg, M. Journal of Molecular Structure-Theochem 2004, 673, 203.
- (10) Cao, X. Y.; Dolg, M.; Stoll, H. Journal of Chemical Physics **2003**, 118, 487.
- (11) Schafer, A.; Huber, C.; Ahlrichs, R. *Journal of Chemical Physics* **1994**, *100*, 5829.