

Supporting Information

Synthesis, structure, and magnetic properties of new layered phosphate halides $\text{Sr}_2\text{Cu}_5(\text{PO}_4)_4\text{X}_2 \cdot 8\text{H}_2\text{O}$ ($\text{X} = \text{Cl}, \text{Br}$) with a crown-like building unit

Chaoqun Qiu^{†,§}, Zhangzhen He^{†,*}, Meiyang Cui^{†,§}, Yingying Tang[†], and Sihuai Chen[†]

[†]State Key Laboratory of Structural Chemistry, Fujian Institute of Research on the Structure of Matter, Chinese Academy of Sciences, Fuzhou, Fujian 350002, China.

[§]University of the Chinese Academy of Sciences, Beijing, 100039, China.

*E-mail: hc1988@hotmail.com; hezz@fjirsm.ac.cn

Figure S1. Experimental (Black line) and Simulated (Red line) powder X-ray diffraction patterns of $\text{Sr}_2\text{Cu}_5(\text{PO}_4)_4\text{X}_2 \cdot 8\text{H}_2\text{O}$ [$\text{X} =$ (a) Cl and (b) Br].

Figure S2. The χ^{-1} versus T curves of $\text{Sr}_2\text{Cu}_5(\text{PO}_4)_4\text{X}_2 \cdot 8\text{H}_2\text{O}$ [$\text{X} =$ (a) Cl and (b) Br].

Figure S3. The fit of square-lattice Heisenberg antiferromagnetic spin model for magnetic susceptibility data of $\text{Sr}_2\text{Cu}_5(\text{PO}_4)_4\text{X}_2 \cdot 8\text{H}_2\text{O}$ [$\text{X} =$ (a) Cl and (b) Br].

Table S1. The Wyckoff positions, atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for $\text{Sr}_2\text{Cu}_5(\text{PO}_4)_4\text{X}_2 \cdot 8\text{H}_2\text{O}$ ($\text{X} = \text{Cl}$ and Br). $U(\text{eq})$ is defined as one third of the trace of the orthogonalized U_{ij} tensor.

Table S2. Anisotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for $\text{Sr}_2\text{Cu}_5(\text{PO}_4)_4\text{X}_2 \cdot 8\text{H}_2\text{O}$ ($\text{X} = \text{Cl}$ and Br). The anisotropic displacement factor exponent takes the form: $-2 \pi^2 [h^2 a^{*2} U_{11} + \dots + 2 h k a^* b^* U_{12}]$.

Table S3. Bond lengths [\AA] and angles [deg] for $\text{Sr}_2\text{Cu}_5(\text{PO}_4)_4\text{Cl}_2 \cdot 8\text{H}_2\text{O}$.

Table S4. Bond lengths [\AA] and angles [deg] for $\text{Sr}_2\text{Cu}_5(\text{PO}_4)_4\text{Br}_2 \cdot 8\text{H}_2\text{O}$.

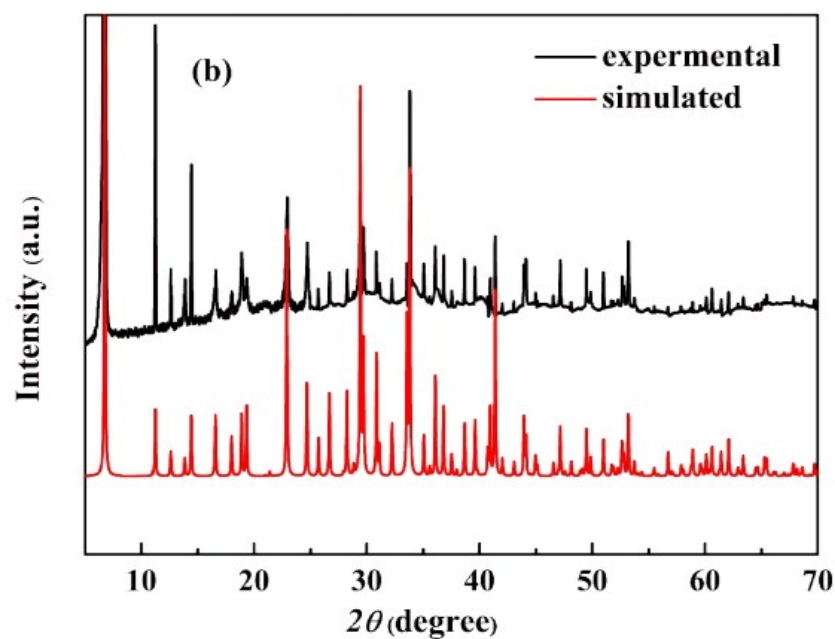
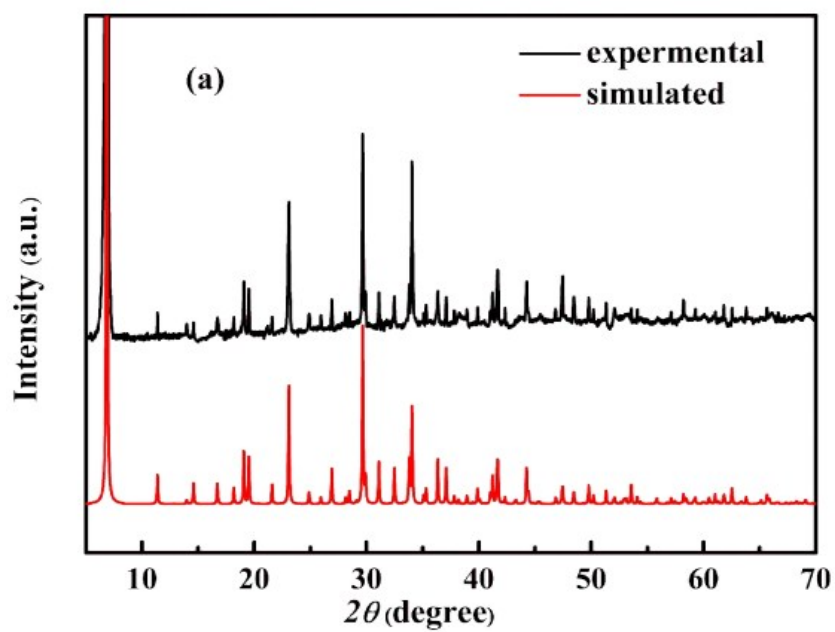


Figure S1. Experimental (Black line) and Simulated (Red line) powder X-ray diffraction patterns of $\text{Sr}_2\text{Cu}_5(\text{PO}_4)_4\text{X}_2 \cdot 8\text{H}_2\text{O}$ [X = (a) Cl and (b) Br].

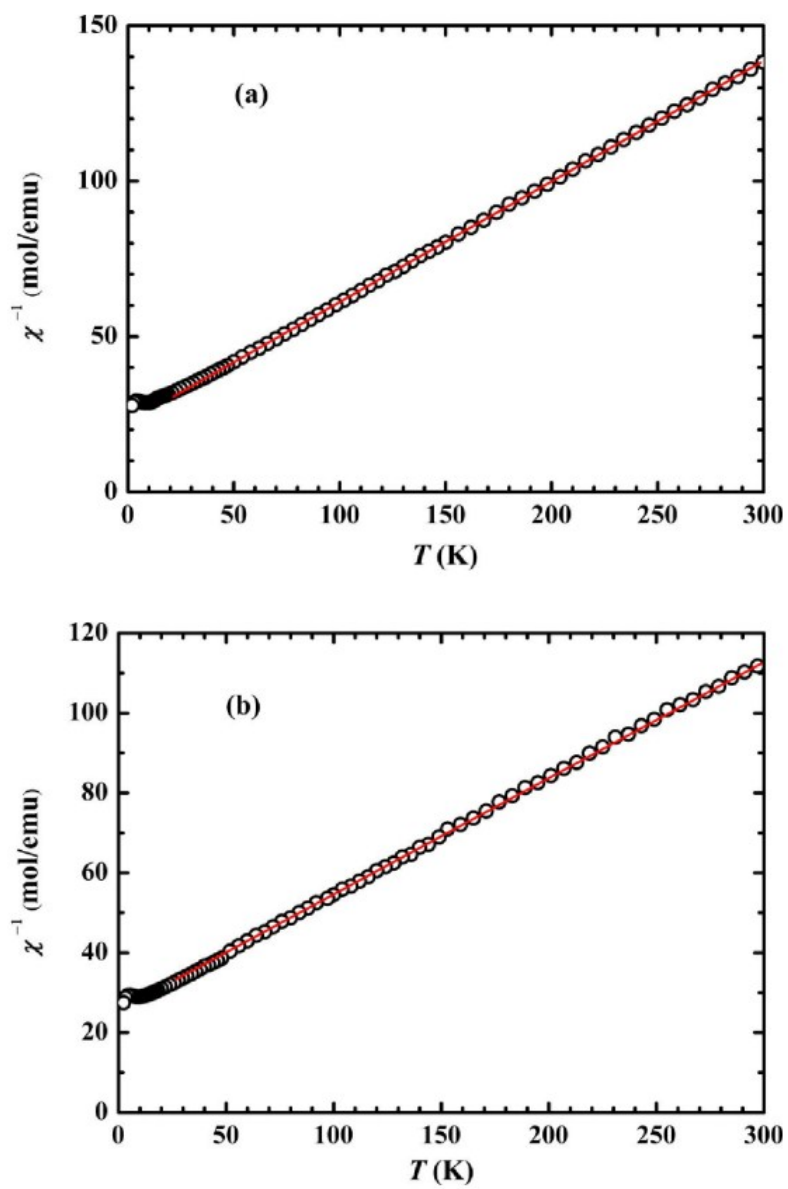


Figure S2. The χ^{-1} versus T curves of $\text{Sr}_2\text{Cu}_5(\text{PO}_4)_4\text{X}_2 \cdot 8\text{H}_2\text{O}$ [$X =$ (a) Cl and (b) Br].

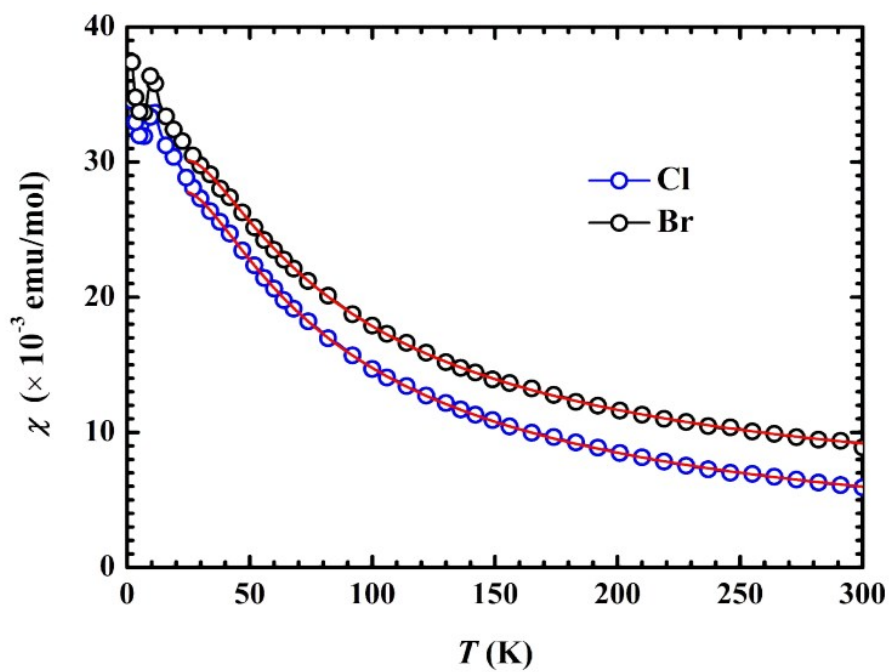


Figure S3. The fit of square-lattice Heisenberg antiferromagnetic spin model for magnetic susceptibility data of $\text{Sr}_2\text{Cu}_5(\text{PO}_4)_4\text{X}_2 \cdot 8\text{H}_2\text{O}$ [X = (a) Cl and (b) Br].

Table S1. The Wyckoff positions, atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for $\text{Sr}_2\text{Cu}_5(\text{PO}_4)_4\text{X}_2 \cdot 8\text{H}_2\text{O}$ ($\text{X} = \text{Cl}, \text{Br}$). $U(\text{eq})$ is defined as one third of the trace of the orthogonalized U_{ij} tensor.

atom	Wyck.	Site	x	y	z	U(eq)
Cu(1)	8g	1	7690(1)	4862(1)	10542(1)	13(1)
Cu(2)	2c	4..	10000	5000	7594(1)	12(1)
Cl(1)	2c	4..	10000	5000	11778(2)	19(1)
Cl(2)	8g	1	10000	5000	5334(2)	39(1)
P(1)	8g	1	7791(1)	6903(1)	8613(1)	11(1)
O(1)	8g	1	8475(3)	6342(3)	9656(2)	12(1)
O(2)	8g	1	6841(3)	3461(3)	11450(3)	14(1)
O(3)	8g	1	6628(3)	6228(3)	11326(3)	17(1)
O(4)	8g	1	8315(3)	6110(3)	7644(2)	17(1)
O(5)	8g	1	9112(6)	8176(5)	5994(3)	62(1)
O(6)	8g	1	5893(6)	7305(6)	6287(4)	93(2)
Sr(1)	4d	2..	10000	10000	7358(1)	24(1)
Cu(1)	8i	.m.	5195(1)	7500	552(1)	15(1)
Cu(2)	2c	4mm	2500	2500	2375(1)	9(1)
Br(1)	2c	4mm	7500	7500	1947(1)	15(1)
Br(2)	2c	4mm	2500	2500	4745(1)	26(1)
P(1)	8j	..m	4552(1)	4552(1)	1372(1)	23(1)
O(1)	8j	..m	3929(2)	3929(2)	328(2)	14(1)
O(2)	16k	1	4231(2)	6105(2)	1375(2)	26(1)
O(3)	16k	1	4168(3)	3622(3)	2343(3)	11(1)
Sr(1)	4f	2mm.	2500	7500	2612(1)	16(1)
O(4A)	16k	1	4604(5)	6474(5)	3786(4)	33(1)
O(4B)	16k	1	1463(5)	5723(4)	3932(4)	37(1)

Table S2. Anisotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for $\text{Sr}_2\text{Cu}_5(\text{PO}_4)_4\text{X}_2 \cdot 8\text{H}_2\text{O}$ (X = Cl, Br). The anisotropic displacement factor exponent takes the form: $-2 \pi^2 [h^2 a^{*2} U11 + \dots + 2 h k a^* b^* U12]$.

atom	U11	U22	U33	U23	U13	U12
Cu(1)	12(1)	8(1)	19(1)	1(1)	3(1)	0(1)
Cu(2)	9(1)	9(1)	18(1)	0	0	0
Cl(1)	18(1)	18(1)	22(1)	0	0	0
Cl(2)	43(1)	43(1)	29(1)	0	0	0
P(1)	8(1)	9(1)	17(1)	-2(1)	-1(1)	1(1)
O(1)	10(1)	10(1)	16(1)	0(1)	-1(1)	0(1)
O(2)	17(2)	9(2)	18(2)	-1(1)	4(1)	2(1)
O(3)	15(2)	10(2)	25(2)	1(1)	6(2)	2(1)
O(4)	13(2)	17(2)	20(2)	-6(1)	-1(1)	4(1)
O(5)	119(4)	34(2)	34(3)	0(2)	-19(3)	7(3)
O(6)	100(5)	100(5)	79(4)	62(4)	-41(3)	-56(4)
Sr(1)	18(1)	38(1)	18(1)	0	0	4(1)
Cu(1)	11(1)	26(1)	9(1)	0	3(1)	0
Cu(2)	8(1)	8(1)	13(1)	0	0	0
Br(1)	15(1)	15(1)	13(1)	0	0	0
Br(2)	29(1)	29(1)	20(1)	0	0	0
P(1)	31(1)	31(1)	6(1)	1(1)	1(1)	-26(1)
O(1)	17(1)	17(1)	9(1)	0(1)	0(1)	-7(1)
O(2)	25(1)	33(1)	20(1)	-15(1)	13(1)	-18(1)
O(3)	10(2)	15(2)	7(2)	3(1)	-1(1)	-3(2)
Sr(1)	14(1)	25(1)	10(1)	0	0	0
O(4A)	40(3)	31(3)	27(3)	-3(2)	-8(2)	7(2)
O(4B)	62(3)	25(3)	23(3)	-1(2)	11(2)	0(2)

Table S3. Bond lengths [Å] and angles [deg] for Sr₂Cu₅(PO₄)₄Cl₂·8H₂O.

Cu(1)-O(3)	1.932(3)	Sr(1)-O(3)#8	2.560(3)
Cu(1)-O(2)	1.941(3)	Sr(1)-O(3)#6	2.560(3)
Cu(1)-O(1)	1.956(3)	Sr(1)-O(5)#9	2.591(5)
Cu(1)-O(1)#1	1.968(3)	Sr(1)-O(6)#10	2.730(5)
Cu(1)-Cl(1)	2.709(2)	Sr(1)-O(6)#3	2.730(5)
P(1)-O(4)	1.515(3)	Sr(1)-O(2)#8	2.748(3)
P(1)-O(3)#5	1.531(3)	Sr(1)-O(2)#6	2.748(3)
P(1)-O(2)#6	1.544(3)	Sr(1)-O(5)	2.591(4)
P(1)-O(1)	1.553(3)	Cu(2)-O(4)#4	1.945(3)
Cu(2)-O(4)#3	1.945(3)	Cu(2)-O(4)	1.945(3)
Cu(2)-O(4)#1	1.945(3)	Cu(2)-Cl(2)	2.817(3)
O(3)-Cu(1)-O(2)	87.47(12)	O(4)#3-Cu(2)-O(4)#1	176.29(19)
O(3)-Cu(1)-O(1)	89.61(13)	O(4)#3-Cu(2)-O(4)#4	89.94(1)
O(2)-Cu(1)-O(1)	177.02(12)	O(4)#1-Cu(2)-O(4)#4	89.94(1)
O(3)-Cu(1)-O(1)#1	175.44(14)	O(4)#3-Cu(2)-O(4)	89.94(1)
O(2)-Cu(1)-O(1)#1	93.16(12)	O(4)#1-Cu(2)-O(4)	89.94(1)
O(1)-Cu(1)-O(1)#1	89.69(15)	O(4)#4-Cu(2)-O(4)	176.29(19)
O(3)-Cu(1)-Cl(1)	96.56(11)	O(4)#3-Cu(2)-Cl(2)	91.85(10)
O(2)-Cu(1)-Cl(1)	92.80(11)	O(4)#1-Cu(2)-Cl(2)	91.85(10)
O(1)-Cu(1)-Cl(1)	88.16(9)	O(4)#4-Cu(2)-Cl(2)	91.85(10)
O(1)#1-Cu(1)-Cl(1)	87.92(9)	O(4)-Cu(2)-Cl(2)	91.85(10)
O(4)-P(1)-O(3)#5	106.24(18)	O(5)#9-Sr(1)-O(6)#10	74.50(2)
O(4)-P(1)-O(2)#6	111.91(18)	O(5)#9-Sr(1)-O(6)#3	67.91(15)
O(3)#5-P(1)-O(2)#6	113.34(16)	O(5)#9-Sr(1)-O(2)#8	77.08(13)
O(4)-P(1)-O(1)	110.50(16)	O(5)#9-Sr(1)-O(2)#6	159.04(14)
O(3)#5-P(1)-O(1)	108.37(19)	O(6)#10-Sr(1)-O(6)#3	121.40(2)
O(2)#6-P(1)-O(1)	106.47(18)	O(6)#10-Sr(1)-O(2)#8	128.30(13)
O(3)#8-Sr(1)-O(3)#6	100.26(16)	O(6)#10-Sr(1)-O(2)#6	84.80(16)
O(3)#8-Sr(1)-O(5)#9	122.17(14)	O(6)#3-Sr(1)-O(2)#8	84.80(16)
O(3)#8-Sr(1)-O(6)#10	163.23(17)	O(6)#3-Sr(1)-O(2)#6	128.30(13)
O(3)#8-Sr(1)-O(6)#3	70.72(13)	O(2)#8-Sr(1)-O(2)#6	114.53(14)
O(3)#8-Sr(1)-O(2)#8	60.47(8)	O(5)-Sr(1)-O(3)#8	108.04(12)
O(3)#8-Sr(1)-O(2)#6	78.44(9)	O(5)-Sr(1)-O(3)#6	122.17(14)
O(3)#6-Sr(1)-O(5)#9	108.04(12)	O(5)-Sr(1)-O(5)#9	97.90(2)
O(3)#6-Sr(1)-O(6)#10	70.72(13)	O(5)-Sr(1)-O(6)#10	67.91(15)
O(3)#6-Sr(1)-O(2)#8	78.44(9)	O(5)-Sr(1)-O(2)#8	159.04(14)
O(3)#6-Sr(1)-O(2)#6	60.47(8)	O(5)-Sr(1)-O(2)#6	77.08(13)
O(3)#6-Sr(1)-O(6)#3	163.23(17)	O(5)-Sr(1)-O(6)#3	74.50(2)

Symmetry transformations used to generate equivalent atoms:

#1 -y+3/2,x-1/2,z	#2 x-1/2,-y+3/2,-z+2	#3 y+1/2,-x+3/2,z
#4 -x+2,-y+1,z	#5 y,x,-z+2	#6 -x+3/2,y+1/2,-z+2
#7 -x+3/2,y-1/2,-z+2	#8 x+1/2,-y+3/2,-z+2	#9 -x+2,-y+2,z
#10 -y+3/2,x+1/2,z		

Table S4. Bond lengths [Å] and angles [deg] for Sr₂Cu₅(PO₄)₄Br₂·8H₂O.

Cu(1)-O(2)#1	1.938(2)	Sr(1)-O(4B)#15	2.587(5)
Cu(1)-O(2)	1.938(2)	Sr(1)-O(4B)#1	2.587(5)
Cu(1)-O(1)#2	1.961(2)	Sr(1)-O(4B)#9	2.587(5)
Cu(1)-O(1)#3	1.961(2)	Sr(1)-O(4B)	2.587(5)
Cu(1)-Br(1)	2.832(1)	Sr(1)-O(2)#9	2.649(2)
P(1)-O(2)#5	1.535(2)	Sr(1)-O(2)#15	2.649(2)
P(1)-O(2)	1.535(2)	Sr(1)-O(2)#1	2.649(2)
P(1)-O(3)	1.556(3)	Sr(1)-O(2)	2.649(2)
P(1)-O(1)	1.560(3)	Sr(1)-O(4A)#1	2.700(4)
Cu(2)-O(3)	1.946(3)	Sr(1)-O(4A)	2.700(4)
Cu(2)-O(3)#6	1.946(3)	Sr(1)-O(4A)#9	2.700(4)
Cu(2)-O(3)#7	1.946(3)	Sr(1)-O(4A)#15	2.700(4)
Cu(2)-O(3)#8	1.946(3)		
Cu(2)-Br(2)	2.964(2)		
O(2)#1-Cu(1)-O(2)	88.34(13)	O(3)-Cu(2)-O(3)#6	177.60(2)
O(2)#1-Cu(1)-O(1)#2	176.64(10)	O(3)-Cu(2)-O(3)#7	89.98(1)
O(2)-Cu(1)-O(1)#2	90.87(10)	O(3)#6-Cu(2)-O(3)#7	89.98(1)
O(2)#1-Cu(1)-O(1)#3	90.87(10)	O(3)-Cu(2)-O(3)#8	89.98(1)
O(2)-Cu(1)-O(1)#3	176.64(10)	O(3)#6-Cu(2)-O(3)#8	89.98(1)
O(1)#2-Cu(1)-O(1)#3	89.74(15)	O(3)#7-Cu(2)-O(3)#8	177.60(2)
O(2)#1-Cu(1)-Br(1)	93.00(7)	O(3)-Cu(2)-Br(2)	91.19(11)
O(2)-Cu(1)-Br(1)	93.00(7)	O(3)#6-Cu(2)-Br(2)	91.19(11)
O(1)#2-Cu(1)-Br(1)	90.30(7)	O(3)#7-Cu(2)-Br(2)	91.19(11)
O(1)#3-Cu(1)-Br(1)	90.30(7)	O(3)#8-Cu(2)-Br(2)	91.19(11)
O(2)#5-P(1)-O(2)	113.38(16)	O(4B)#1-Sr(1)-O(2)#15	77.78(12)
O(2)#5-P(1)-O(3)	96.58(13)	O(4B)#1-Sr(1)-O(4B)	83.30(2)
O(2)-P(1)-O(3)	121.09(15)	O(4B)#1-Sr(1)-O(4B)#9	100.70(2)
O(2)#5-P(1)-O(1)	107.60(10)	O(4B)#1-Sr(1)-O(2)#9	117.77(13)
O(2)-P(1)-O(1)	107.60(10)	O(4B)#1-Sr(1)-O(2)#1	106.20(12)
O(3)-P(1)-O(1)	109.70(16)	O(4B)#1-Sr(1)-O(4A)#1	72.60(17)
O(4B)#15-Sr(1)-O(4B)#1	45.70(2)	O(4B)#1-Sr(1)-O(4A)	100.95(15)
O(4B)#15-Sr(1)-O(4B)#9	83.30(2)	O(4B)#1-Sr(1)-O(4A)#9	66.70(15)
O(4B)#15-Sr(1)-O(2)#9	163.07(12)	O(4B)#1-Sr(1)-O(4A)#15	27.81(15)
O(4B)#15-Sr(1)-O(4B)	100.70(2)	O(4B)#9-Sr(1)-O(2)#9	106.20(12)
O(4B)#15-Sr(1)-O(2)#15	106.20(11)	O(4B)#9-Sr(1)-O(4B)	45.70(2)
O(4B)#15-Sr(1)-O(2)#1	77.78(12)	O(4B)#9-Sr(1)-O(2)#15	163.07(12)
O(4B)#15-Sr(1)-O(4A)#15	72.60(17)	O(4B)#9-Sr(1)-O(2)#1	117.77(13)
O(4B)#15-Sr(1)-O(4A)#1	27.81(15)	O(4B)#9-Sr(1)-O(4A)#1	66.70(15)
O(4B)#15-Sr(1)-O(4A)	66.70(15)	O(4B)#9-Sr(1)-O(4A)	27.81(15)
O(4B)#15-Sr(1)-O(4A)#9	100.95(15)	O(4B)#9-Sr(1)-O(4A)#9	72.60(17)
O(4B)-Sr(1)-O(2)#9	77.78(12)	O(4B)#9-Sr(1)-O(4A)#15	100.95(15)
O(4B)-Sr(1)-O(2)#1	163.07(12)	O(2)#9-Sr(1)-O(2)#15	61.27(8)

O(4B)-Sr(1)-O(4A)#1	100.95(15)	O(2)#9-Sr(1)-O(2)#1	108.60(10)
O(4B)-Sr(1)-O(2)#15	117.77(13)	O(2)#9-Sr(1)-O(4A)	127.37(11)
O(4B)-Sr(1)-O(4A)	72.60(17)	O(2)#9-Sr(1)-O(4A)#1	169.03(11)
O(4B)-Sr(1)-O(4A)#9	27.81(15)	O(2)#9-Sr(1)-O(4A)#9	69.72(12)
O(4B)-Sr(1)-O(4A)#15	66.70(15)	O(2)#9-Sr(1)-O(4A)#15	91.61(12)
O(2)#15-Sr(1)-O(2)#1	78.44(8)	O(2)#1-Sr(1)-O(4A)#1	69.72(12)
O(2)#15-Sr(1)-O(4A)#1	127.37(11)	O(2)#1-Sr(1)-O(4A)	91.61(12)
O(2)#15-Sr(1)-O(4A)	169.04(11)	O(2)#1-Sr(1)-O(4A)#9	169.04(11)
O(2)#15-Sr(1)-O(4A)#9	91.61(12)	O(2)#1-Sr(1)-O(4A)#15	127.37(11)
O(2)#15-Sr(1)-O(4A)#15	69.72(12)	O(4A)-Sr(1)-O(4A)#9	97.90(2)
O(4A)#1-Sr(1)-O(4A)	43.20(2)	O(4A)-Sr(1)-O(4A)#15	114.10(2)
O(4A)#1-Sr(1)-O(4A)#9	114.10(2)	O(4A)#9-Sr(1)-O(4A)#15	43.20(2)
O(4A)#1-Sr(1)-O(4A)#15	97.90(2)	O(2)-Sr(1)-O(2)#1	61.27(8)
O(2)-Sr(1)-O(4B)#15	117.77(13)	O(2)-Sr(1)-O(4A)#1	91.61(12)
O(2)-Sr(1)-O(4B)#1	163.07(12)	O(2)-Sr(1)-O(4A)	69.72(12)
O(2)-Sr(1)-O(4B)#9	77.78(12)	O(2)-Sr(1)-O(4A)#9	127.37(11)
O(2)-Sr(1)-O(4B)	106.20(11)	O(2)-Sr(1)-O(4A)#15	169.03(11)
O(2)-Sr(1)-O(2)#9	78.44(8)	O(2)-Sr(1)-O(2)#15	108.60(10)

Symmetry transformations used to generate equivalent atoms:

#1 $x, -y+3/2, z$	#2 $-x+1, -y+1, -z$	#3 $-y+1, x+1/2, -z$
#4 $-y+1/2, -x+1/2, z$	#5 y, x, z	#6 $-x+1/2, -y+1/2, z$
#7 $y, -x+1/2, z$	#8 $-y+1/2, x, z$	#9 $-x+1/2, y, z$
#10 $x, -y+1/2, z$	#11 $-y+3/2, x, z$	#12 $y, -x+3/2, z$
#13 $-x+3/2, -y+3/2, z$	#14 $y-1/2, -x+1, -z$	#15 $-x+1/2, -y+3/2, z$