

## Supplementary Information

### **KFeCuTe<sub>2</sub>: A new compound to study the removal of interstitial Fe in layered telluride**

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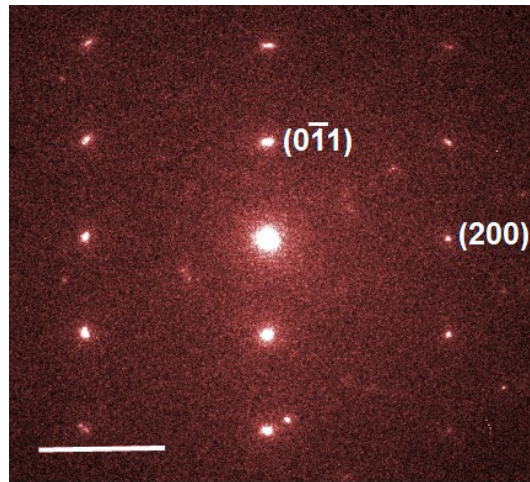
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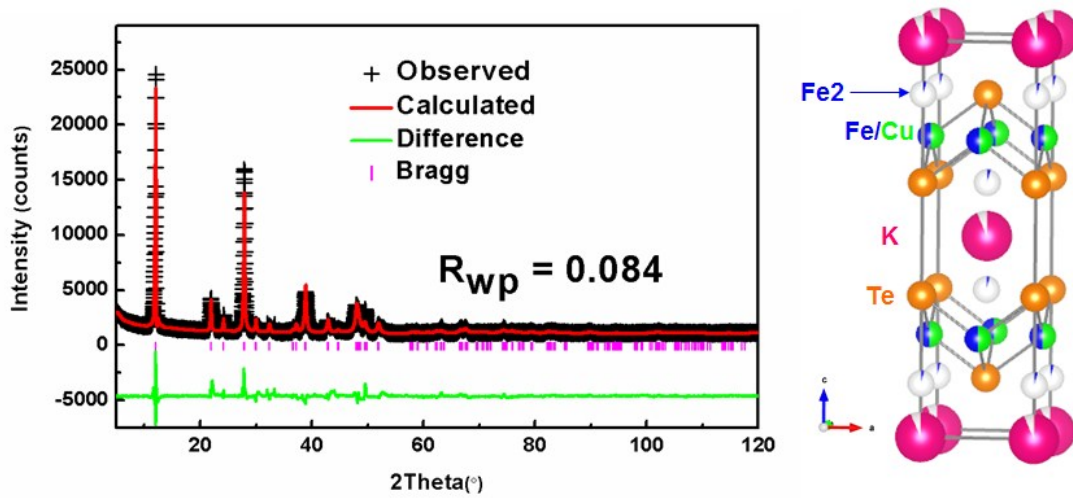
*‡ Fan Sun and Zhongnan Guo contribute equally to this work.*

**Table S1. Structural parameters for KFeCuTe<sub>2</sub> by PXRD Rietveld refinement.**

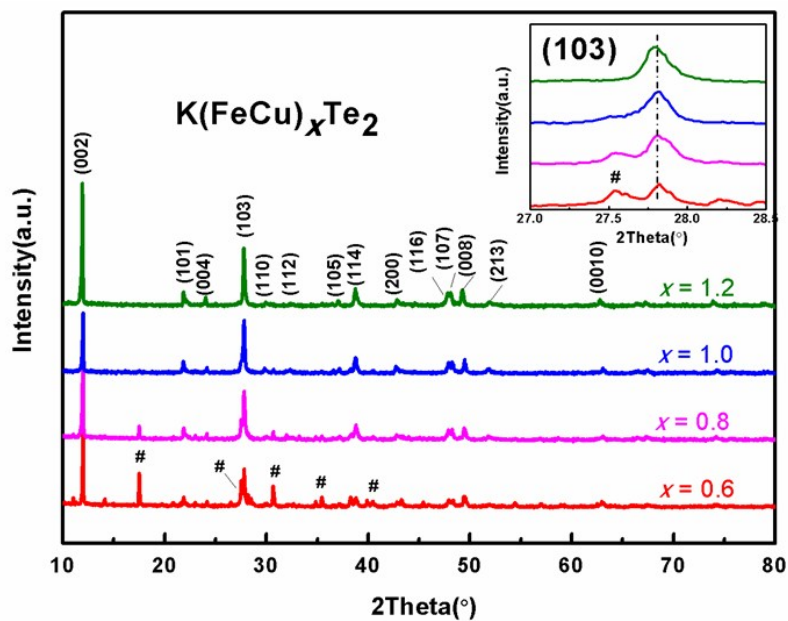
Chemical formula	K <sub>0.94</sub> FeCuTe <sub>2</sub>		
Space group	<i>I4/mmm</i>		
<i>a</i> (Å)	4.2148(4)		
<i>c</i> (Å)	14.7286(11)		
<i>V</i> (Å <sup>3</sup> )	261.65(4)		
<i>Z</i>	2		
$\chi^2$	8.79		
<i>R</i> <sub>wp</sub> (%)	7.80		
<i>d</i> <sub>Fe(Cu)-Te</sub> (Å)	2.6344(8)		
Atom	<i>x</i>	<i>y</i>	<i>z</i>
K	0	0	0
Fe(Cu)	0	0.5	0.25
Te	0	0	0.3573(1)



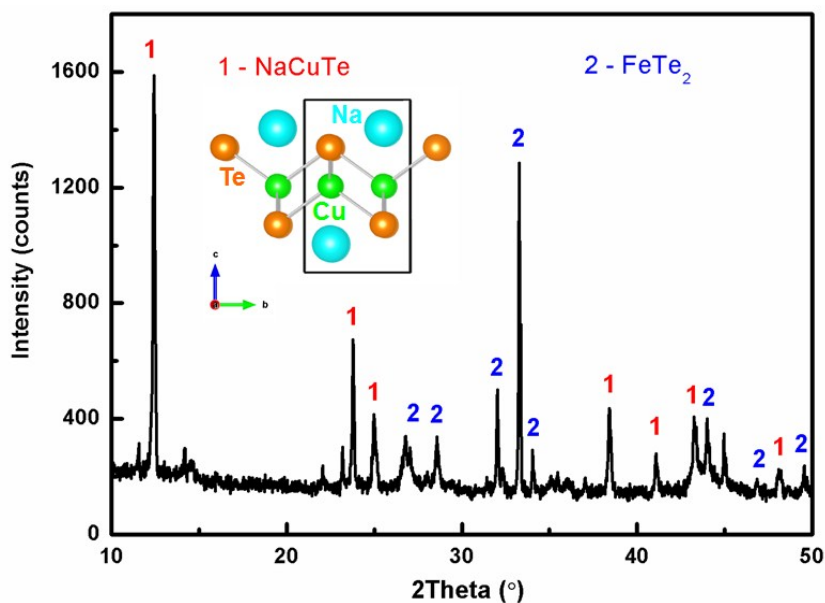
**Fig. S1** SAED pattern of  $\text{KFeCuTe}_2$  crystal along the  $[011]$  zone axes, showing the tetragonal unit cell. The scale bar in the left bottom is 5  $1/\text{nm}$ .



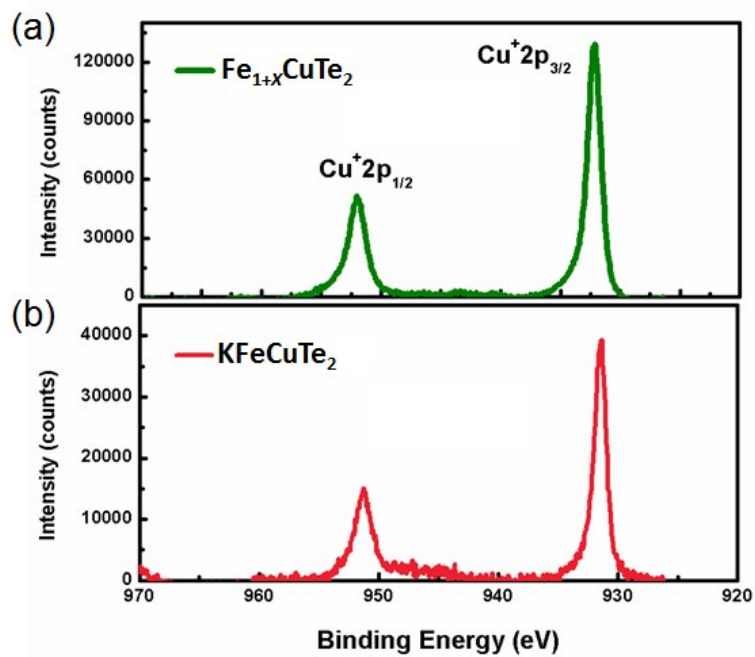
**Fig. S2** Rietveld refinement with the structure model containing 4% interstitial Fe atoms  $(0.5, 0.5, z)$ , resulting the  $R_{wp} = 0.084$ , which is larger than that without the interstitial Fe (0.078).



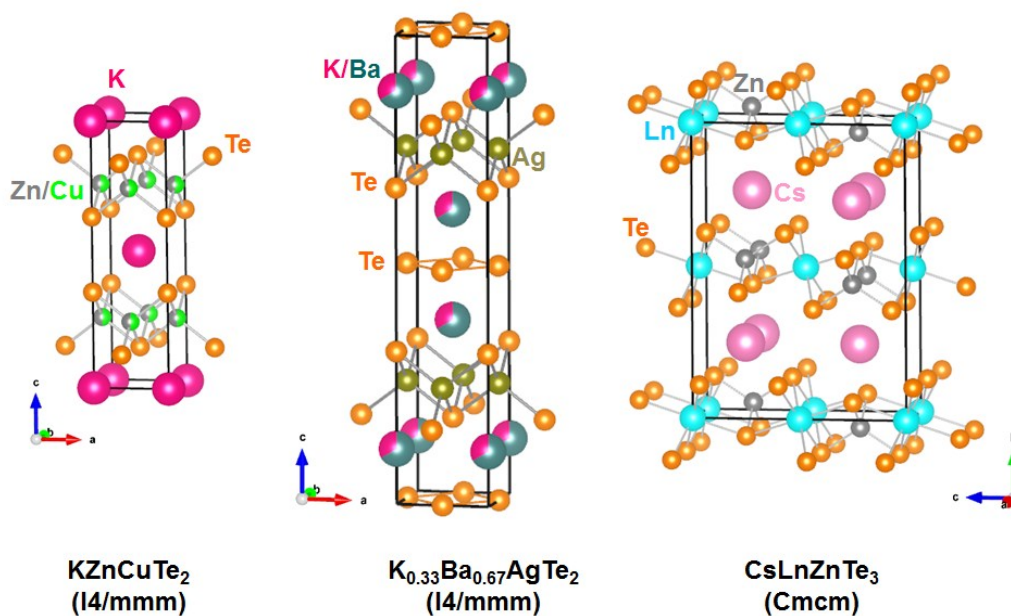
**Fig. S3** PXR D patterns of the crystals grown from different nominal compositions. Inset shows the enlarged (103) diffraction peak with the same location, suggesting that the obtained main phases in different patterns are the same. # represents the impurity of  $\text{K}_2\text{Te}_3$  (ICDD: 71-0490).



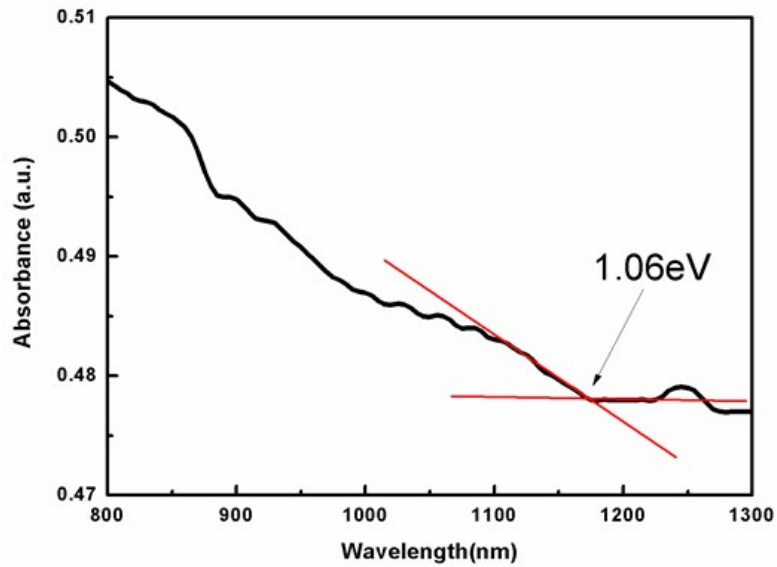
**Fig. S4** PXR D pattern of the nominal composition  $\text{NFeCuTe}_2$ , showing the products of  $\text{NaCuTe}$  and  $\text{FeTe}_2$ . Inset shows the crystal structure of  $\text{NaCuTe}$ .



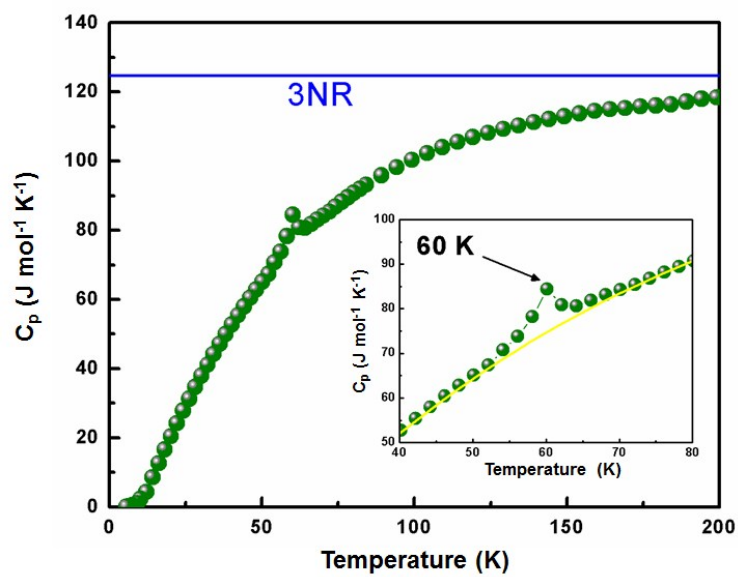
**Fig. S5** XPS Cu spectra of (a)  $\text{Fe}_{1+x}\text{CuTe}_2$  and (b)  $\text{KFeCuTe}_2$ , showing +1 valence.<sup>1</sup>



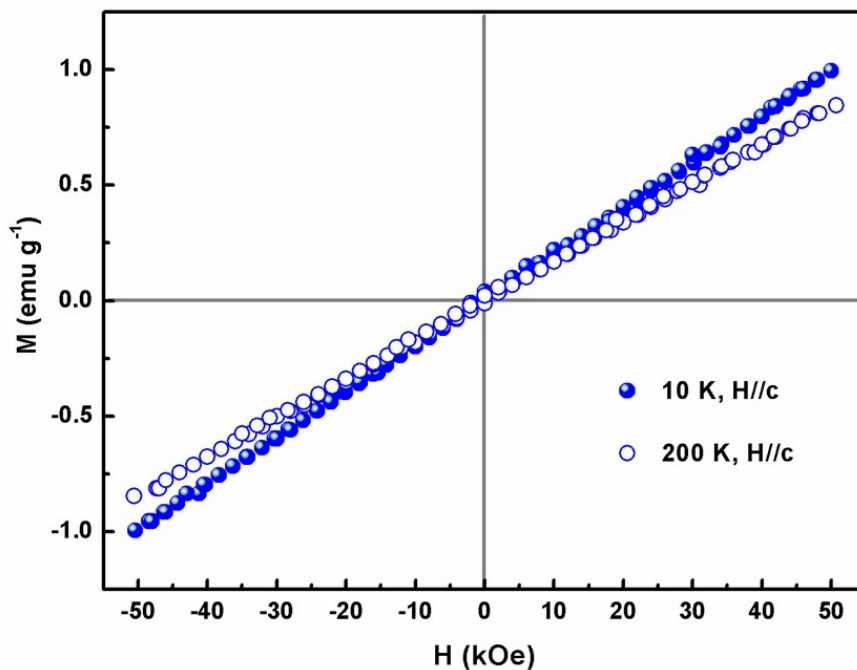
**Fig. S6** The crystal structures of some typical layered transition metal tellurides:  $\text{KZnCuTe}_2$ ,<sup>2</sup>  $\text{K}_{0.33}\text{Ba}_{0.67}\text{AgTe}_2$ <sup>3</sup> and  $\text{CsLnZnTe}_3$  ( $\text{Ln}$ = rare earth)<sup>4</sup>.



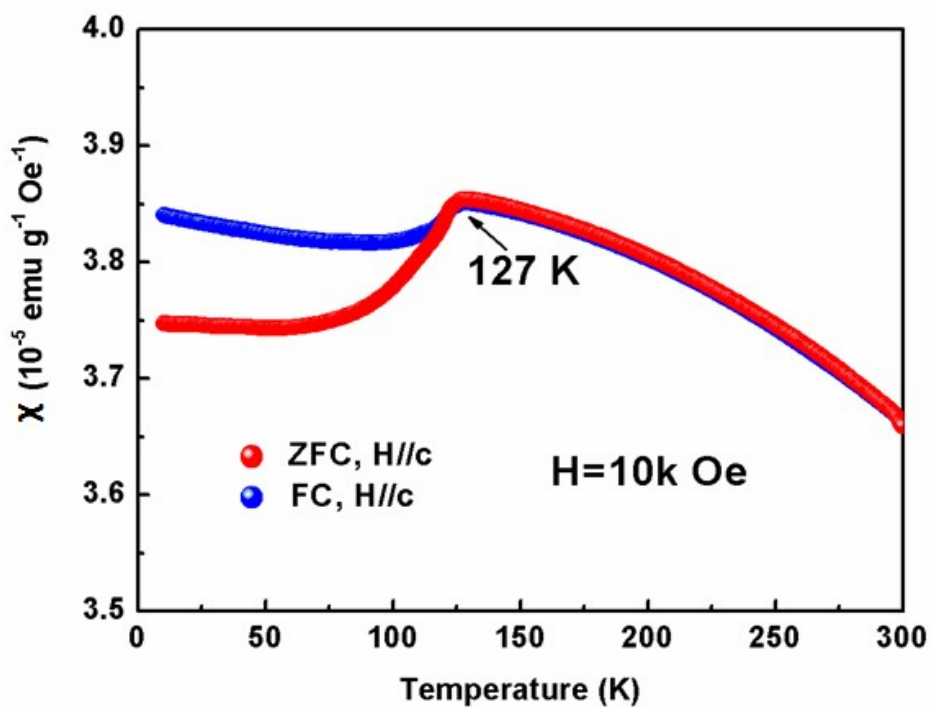
**Fig. S7** UV–vis–NIR diffuse-reflectance spectra of KFeCuTe<sub>2</sub>, showing a band gap  $E_g = 1.06$  eV.



**Fig. S8** Temperature dependence of specific heat for KFeCuTe<sub>2</sub>. The inset shows the enlarged area near the magnetic transition at 60 K, with the yellow solid curve representing the lattice contribution by a polynomial. The blue line is the Dulong-Petit value of  $3NR$ , which approaches the high temperature heat capacity.

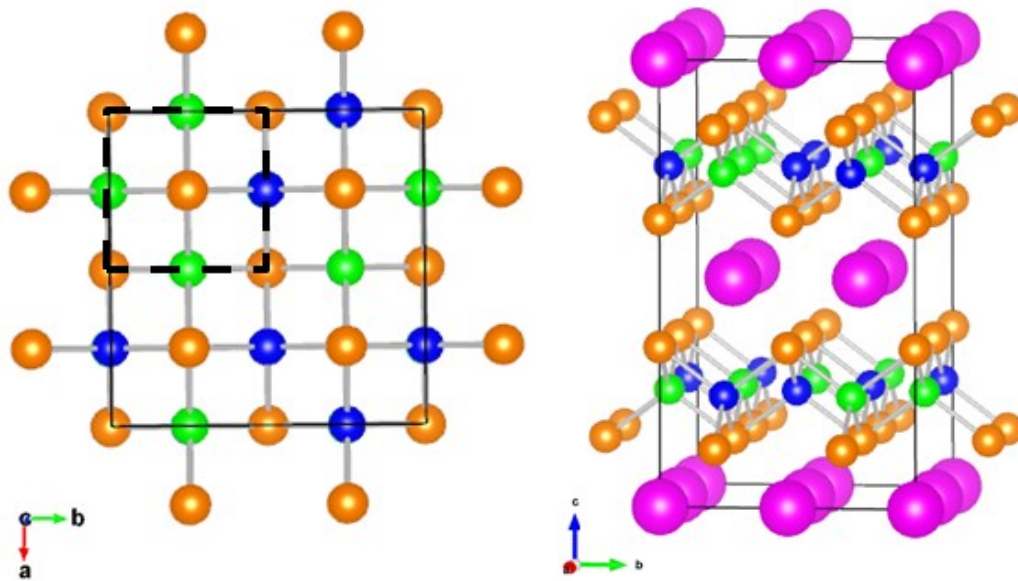


**Fig. S9** Magnetization loops  $M(H)$  with  $H//c$  at 10 K and 200 K.

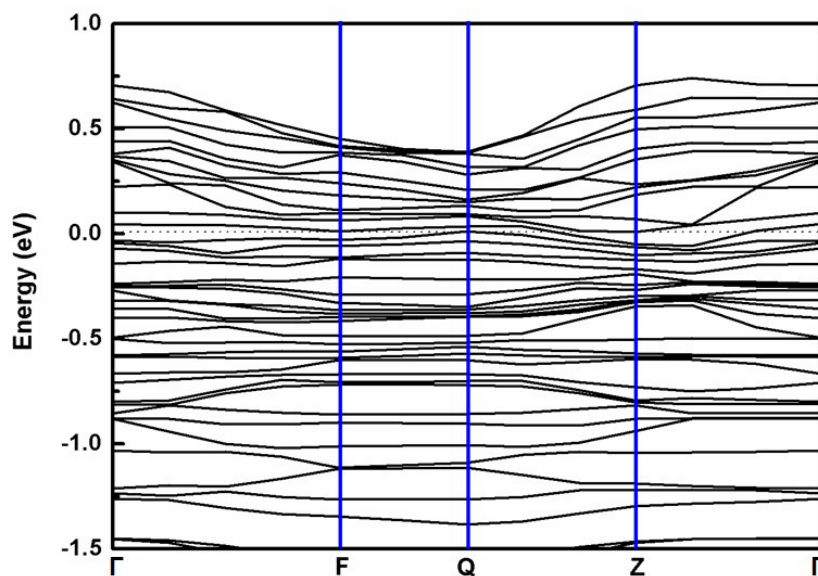


**Fig. S10** Temperature dependence of magnetic susceptibility of the sample after air exposure for a few days, showing an anomaly at 127 K and the rising susceptibility of FC curve at low temperature, which is attributed to the formation of  $\text{Fe}_3\text{O}_4$  impurity.





**Fig. S11** The structural model used in the first principle calculation: a  $2 \times 2$  supercell with the randomly distributed Fe/Cu atoms.



**Fig. S12** The band structure of non-magnetic  $\text{KFeCuTe}_2$ , showing a metallic state.

### References

1. D. Cahen, P. J. Ireland, L. L. Kazmerski and F. A. Thiel, *Journal of Applied Physics*, 1985, **57**, 4761-4771.



2. J. Li, H.-Y. Guo, R. A. Yglesias and T. J. Emge, *Chemistry of Materials*, 1995, **7**, 599-601.
3. X. Zhang, X. Li, B. Foran, S. Lee, S. Guo, S. Hogan, S. Kannewurf and M. G. Kanatzidis, *Journal of the American Chemical Society*, 1995, **117**, 10513-10520.
4. J. Yao, B. Deng, L. J. Sherry, A. D. McFarland, D. E. Ellis, R. P. Van Duyne and J. A. Ibers, *Inorganic Chemistry*, 2004, **43**, 7735-7740.