

Supporting information for:

**Selective C-C coupling at a Pt(IV) centre: 100% preference for
 sp^2 - sp^3 over sp^3 - sp^3 .**

Paul A Shaw and Jonathan P Rourke*

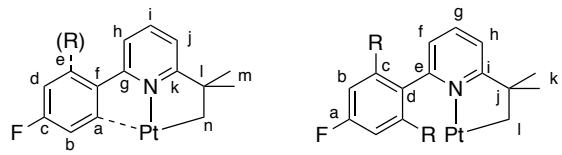
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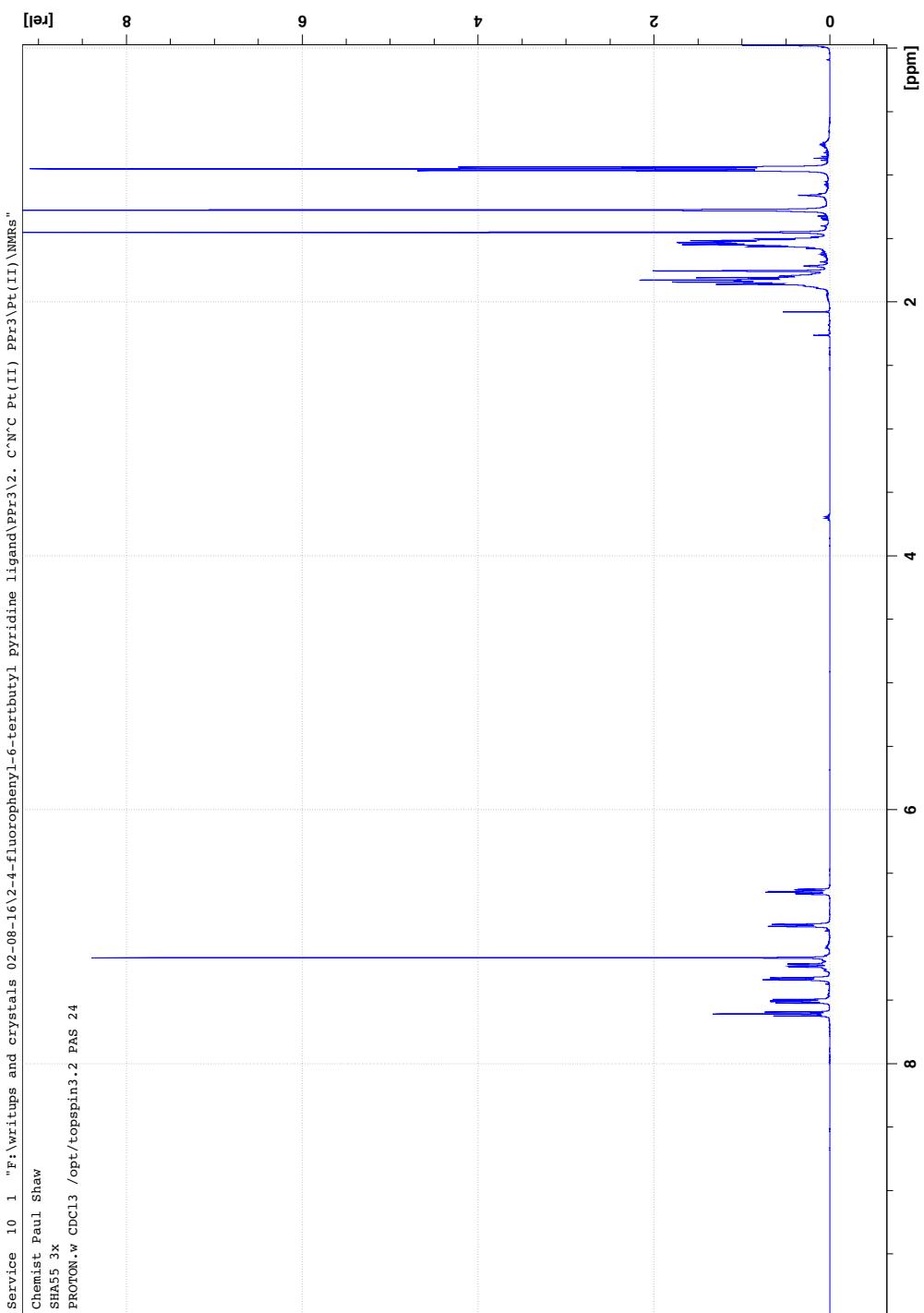
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The leftmost labelling scheme was used for all complexes except **Me₂-3** and **Bn₂-3**, for which the right hand scheme was used:



Complex 1



$\delta_H = 7.63$ (1H, t, $^3J_{H-H} = 8$ Hz, H_i), 7.43 (1H, dd, $^3J_{H-H} = 8$ Hz, $^4J_{H-F} = 5$ Hz, H_e), 7.36 (1H, d, $^3J_{H-H} = 8$ Hz, H_h), 7.25 (1H, dd, $^3J_{H-F} = 9.5$, $^4J_{H-H} = 2.5$ Hz, $^3J_{H-Pt} = 29$ Hz, H_b), 6.93 (1H, d, $^3J_{H-H} = 8$ Hz, H_j), 6.67 (1H, td, $^3J_{H-H} = 3J_{H-F} = 9.5$, $^4J_{H-H} = 2.5$ Hz, H_d), 1.86 (6H, m, PCH₂), 1.78 (2H, s, $^2J_{H-H} = 39$ Hz, H_n), 1.56 (6H, m, PCH₂CH₂), 1.30 (6H, s H_m), 0.97 (9H, t, $^3J_{H-H} = 7.5$, PCH₂CH₂Me) ppm.

**Chemist Paul Shaw
SHA55 3x
C13APT.w CDCl3 /opt/topspin3.2 PAS 24**

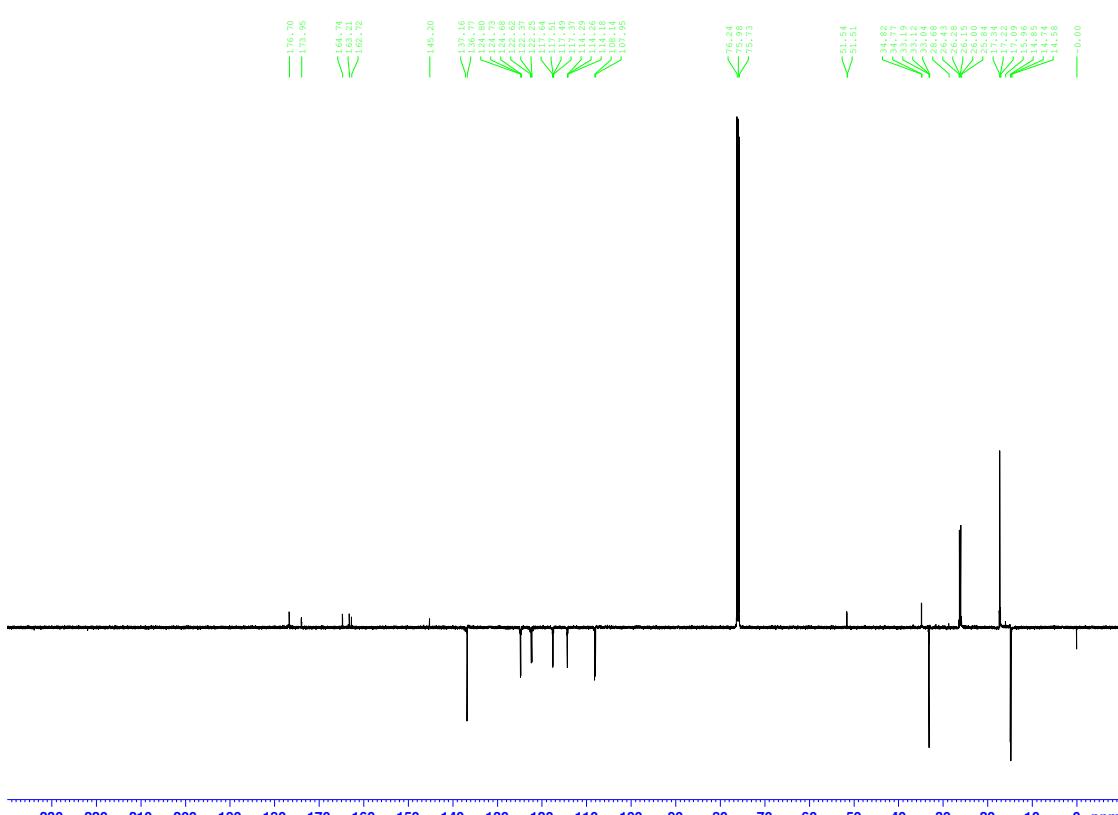


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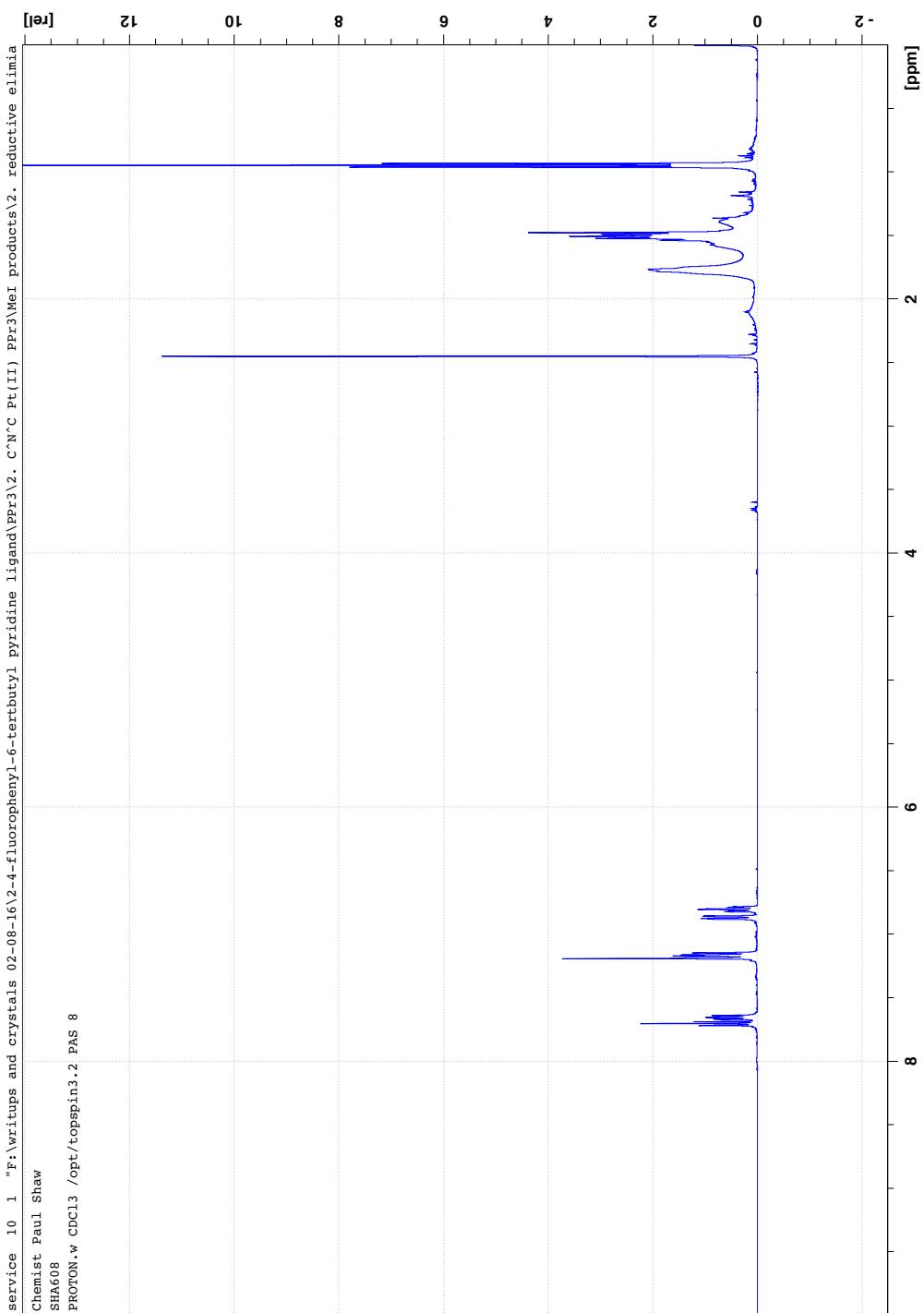
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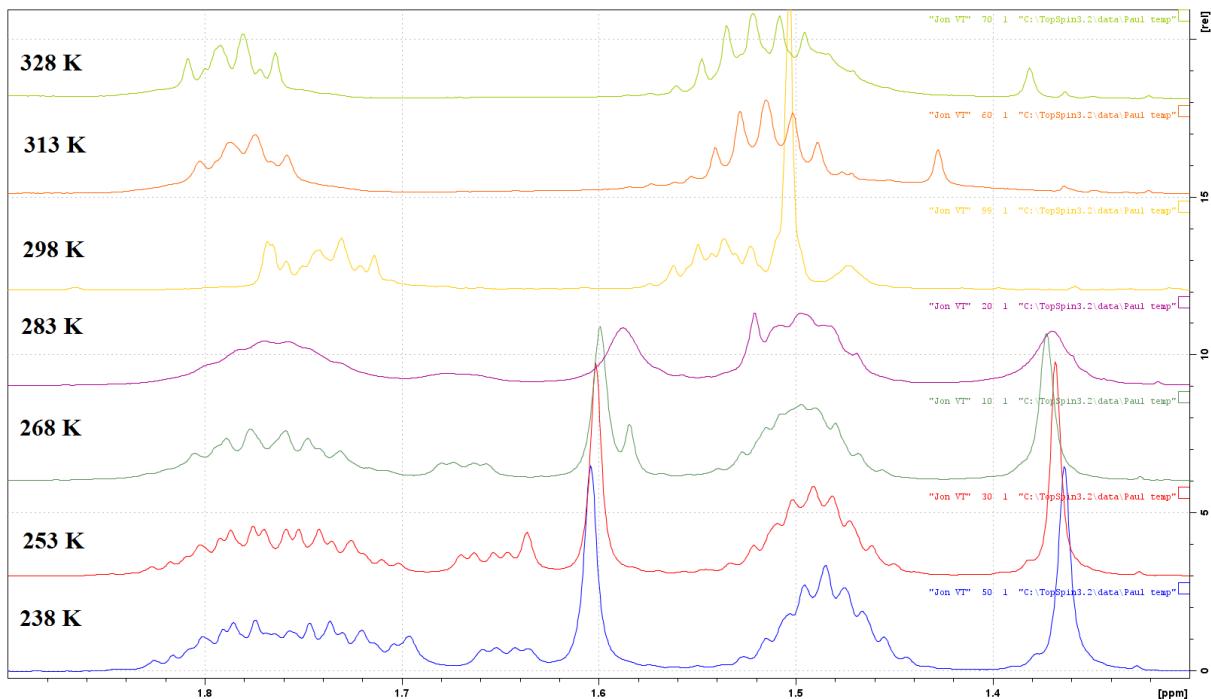
$\delta_{\text{C}} = 14.79$ (d, ${}^3J_{\text{C-P}} = 15.5$ Hz, $\text{PCH}_2\text{CH}_2\text{Me}$), 15.63 (s, ${}^3J_{\text{C-Pt}} = 32.5$ Hz, PCH_2CH_2), 26.14 (d, ${}^1J_{\text{C-P}} = 35.63$, ${}^2J_{\text{C-Pt}} = 37.5$ Hz, PCH_2), 33.11 (s, ${}^3J_{\text{C-Pt}} = 18.5$ Hz, C_m), 34.79 (d, ${}^2J_{\text{C-P}} = 6$ Hz, ${}^1J_{\text{C-Pt}} = 466$ Hz, C_n), 51.52, (d, ${}^3J_{\text{C-P}} = 3.5$ Hz, ${}^2J_{\text{C-Pt}} = 21$ Hz, C_l), 108.05 (d, ${}^2J_{\text{C-F}} = 22$ Hz, C_d), 114.28 (d, ${}^4J_{\text{C-P}} = 3$ Hz, ${}^3J_{\text{C-Pt}} = 25.5$ Hz, C_h), 117.50 (d, ${}^4J_{\text{C-P}} = 2.5$ Hz, ${}^3J_{\text{C-Pt}} = 33$ Hz, C_j), 122.31 (d, ${}^2J_{\text{C-F}} = 15$ Hz, ${}^2J_{\text{C-Pt}} = 64.5$ Hz, C_b), 124.77 (d, ${}^3J_{\text{C-F}} = 9$ Hz, ${}^3J_{\text{C-Pt}} = 30$ Hz, C_e), 136.76 (s, C_i), 145.20 (m, ${}^2J_{\text{C-Pt}} = 26$ Hz, C_f), 163.22 (s, ${}^2J_{\text{C-Pt}} = 52$ Hz, H_g), 163.72 (d, ${}^2J_{\text{C-F}} = 254.5$ Hz, ${}^3J_{\text{C-Pt}} = 52$ Hz, C_c), 173.95 (m, ${}^1J_{\text{C-Pt}} = 686$ Hz, C_a), 176.22 (s, ${}^2J_{\text{C-Pt}} = 19$ Hz, H_k) ppm.

$\delta_F = -111.69$ ($^4J_{F-Pt} = 26.5$ Hz) ppm. $\delta_P = 0.44$ ($^1J_{P-Pt} = 3813$ Hz) ppm. $\delta_{Pt} = -4026$ (d, $^1J_{Pt-P} = \sim 3700$ Hz) ppm.

Complex Me-3



δ_H = 7.70 (1H, t, $^3J_{H-H} = ^3J_{H-F} = 8$ Hz, H_i), 7.65 (1H, dd, $^3J_{H-H} = 8.5$ Hz, $^4J_{H-F} = 6$ Hz, H_a), 6.16 (2H, m, H_{h,j}), 6.86 (1H, dd, $^3J_{H-H} = 10$ Hz, $^4J_{H-H} = 2.5$ Hz, H_d), 6.80 (1H, td, $^3J_{H-H} = ^3J_{H-F} = 8$ Hz, $^4J_{H-H} = 2.5$ Hz, H_b), 2.45 (3H, s, Me), 1.77 (6H, m, PCH₂) 1.49 (14H, m, H_{m,n}, PCH₂CH₂), 0.95 (9H, t, $^3J_{H-H} = 7$ Hz, PCH₂CH₂Me) ppm.



(600 MHz) Note that at 298 K PCH_2 is broader than normal, without its characteristic shape. See ESI. H_m is split into two broad peaks separated by 85 Hz. At 328 K, H_m has coalesced to form one broad lump hidden under PCH_2CH_2 , and PCH_2 has regained its characteristic shape. Lowering the temperature to 268 K, H_m now comprises of two distinct peaks (1.44 and 1.66). Two new peaks also appear which correspond to H_n . These peaks are at 1.74 (1H, dd, $^2J_{\text{H-H}} = 10$ Hz, $^3J_{\text{H-Pt}} = 3.5$ Hz) and 2.21 (1H, d, $^2J_{\text{H-H}} = 10$ Hz, $^2J_{\text{H-Pt}} = \sim 43$ Hz) which were previously hidden under PCH_2CH_2 . Lowering the temperature further to 238 K we see PCH_2 beginning to separate into 2 peaks. Using a separation of 131 Hz and a coalescence temperature of 313 K for H_m we can calculate a barrier of 62.9 kJmol^{-1} ; a coalescence temperature of 298 would equate to a barrier of 59.9 kJmol^{-1} ; variations in the peak separation make a similar difference in calculated value, hence our estimate of an uncertainty of $\pm 5 \text{ kJmol}^{-1}$.

Chemist Paul Shaw
 SHA608
 C13APT.w CDCl₃ /opt/topspin3.2 PAS 8



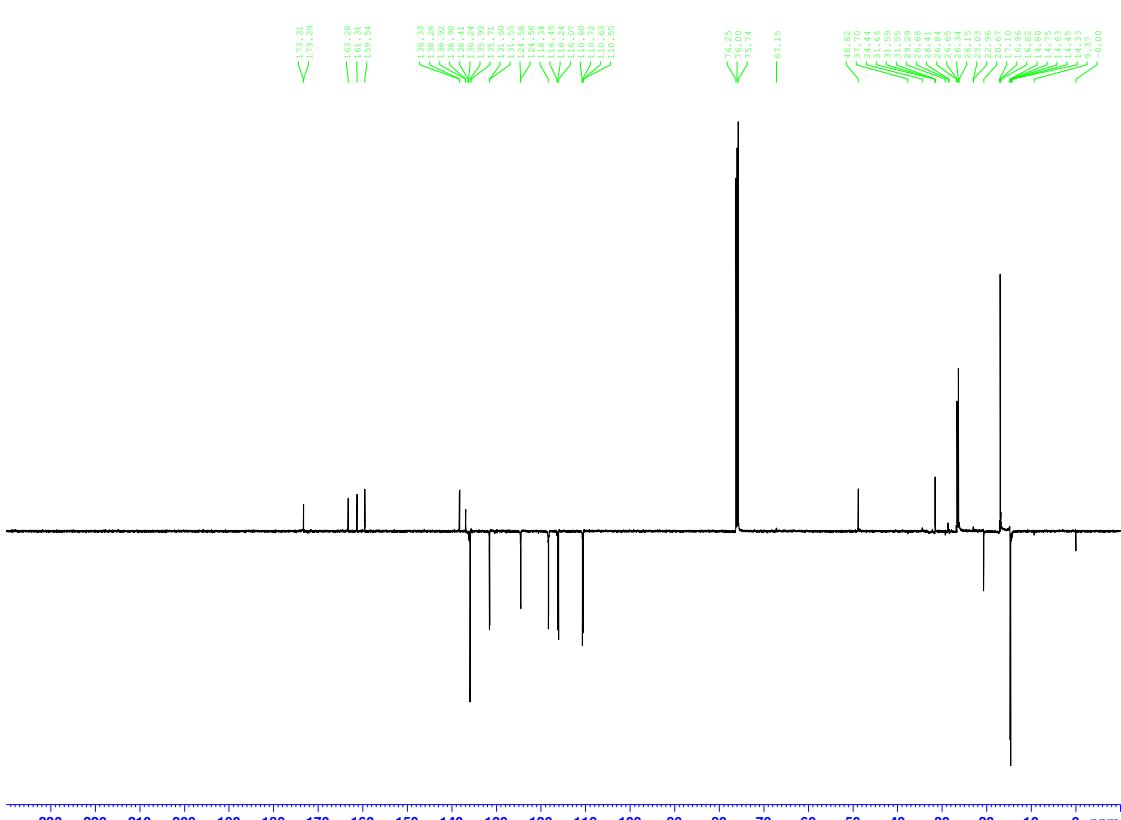
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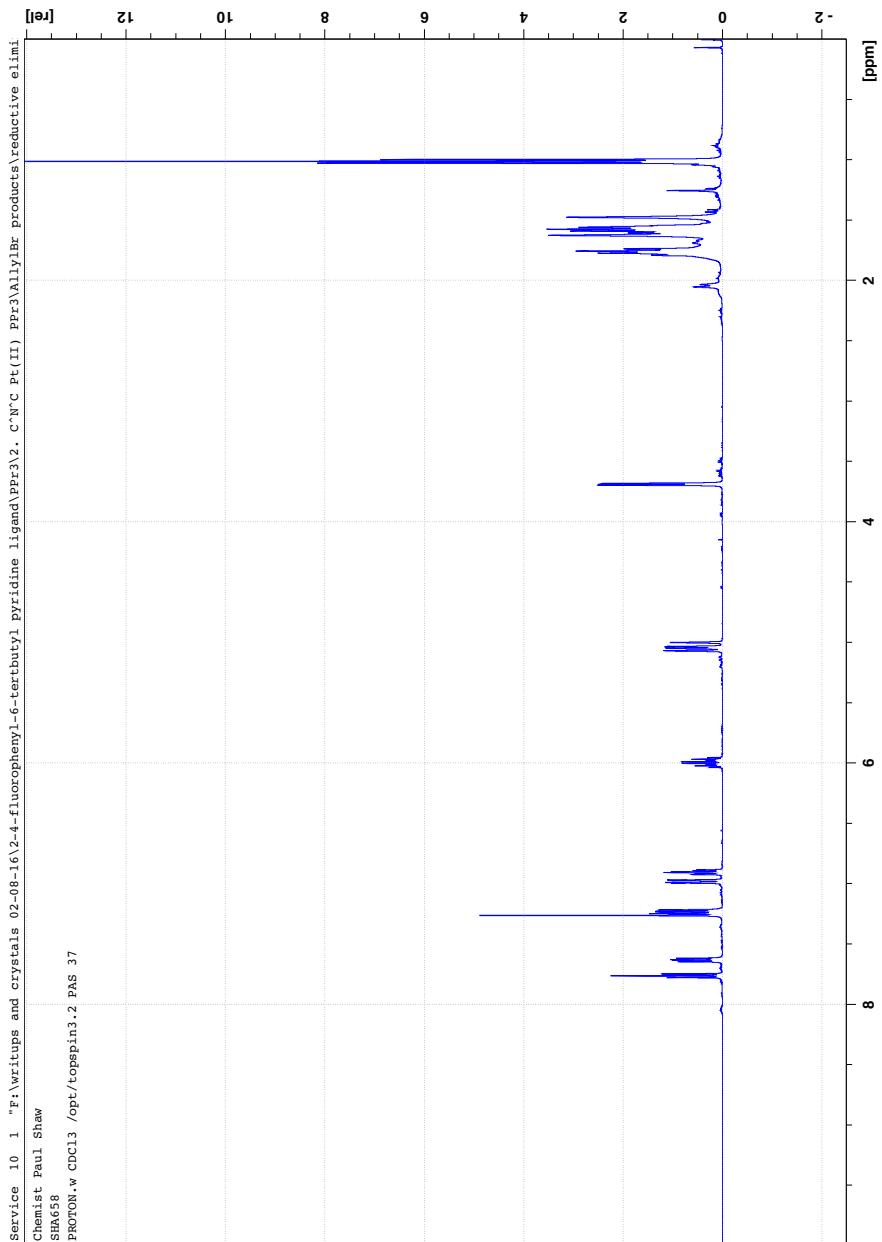
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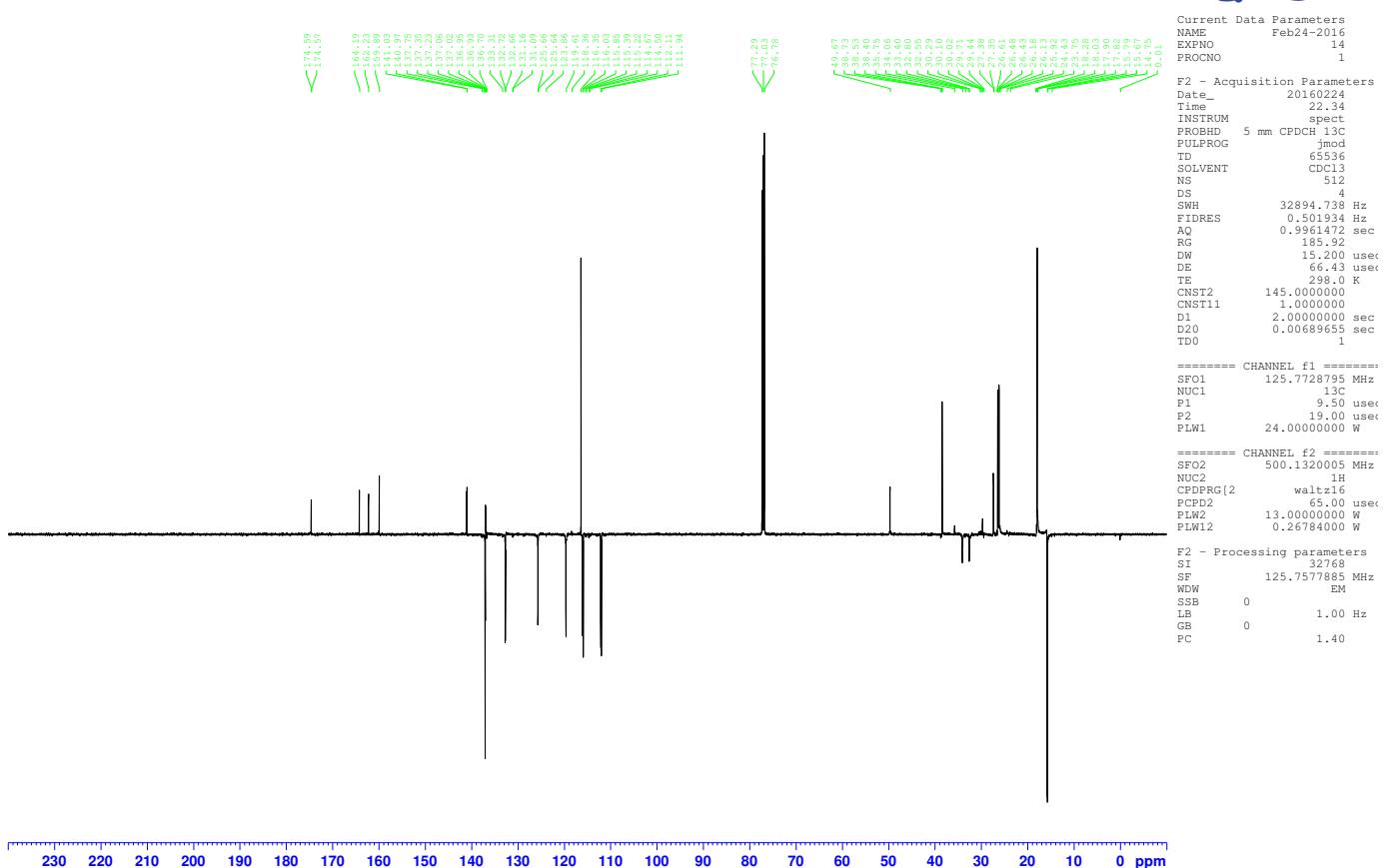
$\delta_F = -113.28$ ppm. $\delta_P = -2.05$ ($^1J_{P-Pt} = 4303$ Hz) ppm. $\delta_{Pt} = -4339$ (d, $^1J_{Pt-P} \sim 4300$ Hz) ppm.

Complex Allyl-3



δ_H = 7.76 (1H, t, ${}^3J_{H-H}$ = 7.5 Hz, H_i), 7.63 (1H, dd, ${}^3J_{H-H}$ = 8 Hz, ${}^4J_{H-F}$ = 6 Hz, H_a), 7.26 (1H, d, ${}^3J_{H-H}$ = 7.5 Hz, H_j), 7.23 (1H, d, ${}^3J_{H-H}$ = 7.5 Hz, H_h), 6.98 (1H, dd, ${}^3J_{H-F}$ = 10 Hz, ${}^4J_{H-H}$ = 2 Hz, H_d), 6.90 (1H, td, ${}^3J_{H-H}$ = ${}^3J_{H-F}$ = 8 Hz, ${}^4J_{H-H}$ = 2 Hz, H_b), 5.99 (1H, ddt, ${}^3J_{H-Htrans}$ = 17 Hz, ${}^3J_{H-Hcis}$ = 10 Hz, ${}^3J_{H-H}$ = 6.5 Hz, H_{allyl}), 5.06 (1H, dd, ${}^3J_{H-H}$ = 10 Hz, ${}^3J_{H-H}$ = 1.5 Hz, H_{allyl}), 5.01 (1H, dd, ${}^3J_{H-H}$ = 17 Hz, ${}^3J_{H-H}$ = 1.5 Hz, H_{allyl}), 3.69 (2H, d, ${}^3J_{H-H}$ = 6.5 Hz, H_{allyl}), 2.05 (1H, d, ${}^2J_{H-H}$ = 9.5 Hz, ${}^2J_{H-Pt}$ = 60 Hz, H_n), 1.77 (6H, m, PCH₂), 1.68 (1H, d, ${}^2J_{H-H}$ = 9.5 Hz, H_n), 1.63 (3H, s, H_m), 1.58 (6H, m, PCH₂CH₂), 1.48 (3H, s, H_m), 1.01 (9H, t, ${}^3J_{H-H}$ = 6 Hz, PCH₂CH₂Me) ppm. Note that at 298K both H_m and H_n have separated into two peaks each, separated by 75 Hz and 183 Hz respectively (600 MHz spectrometer).

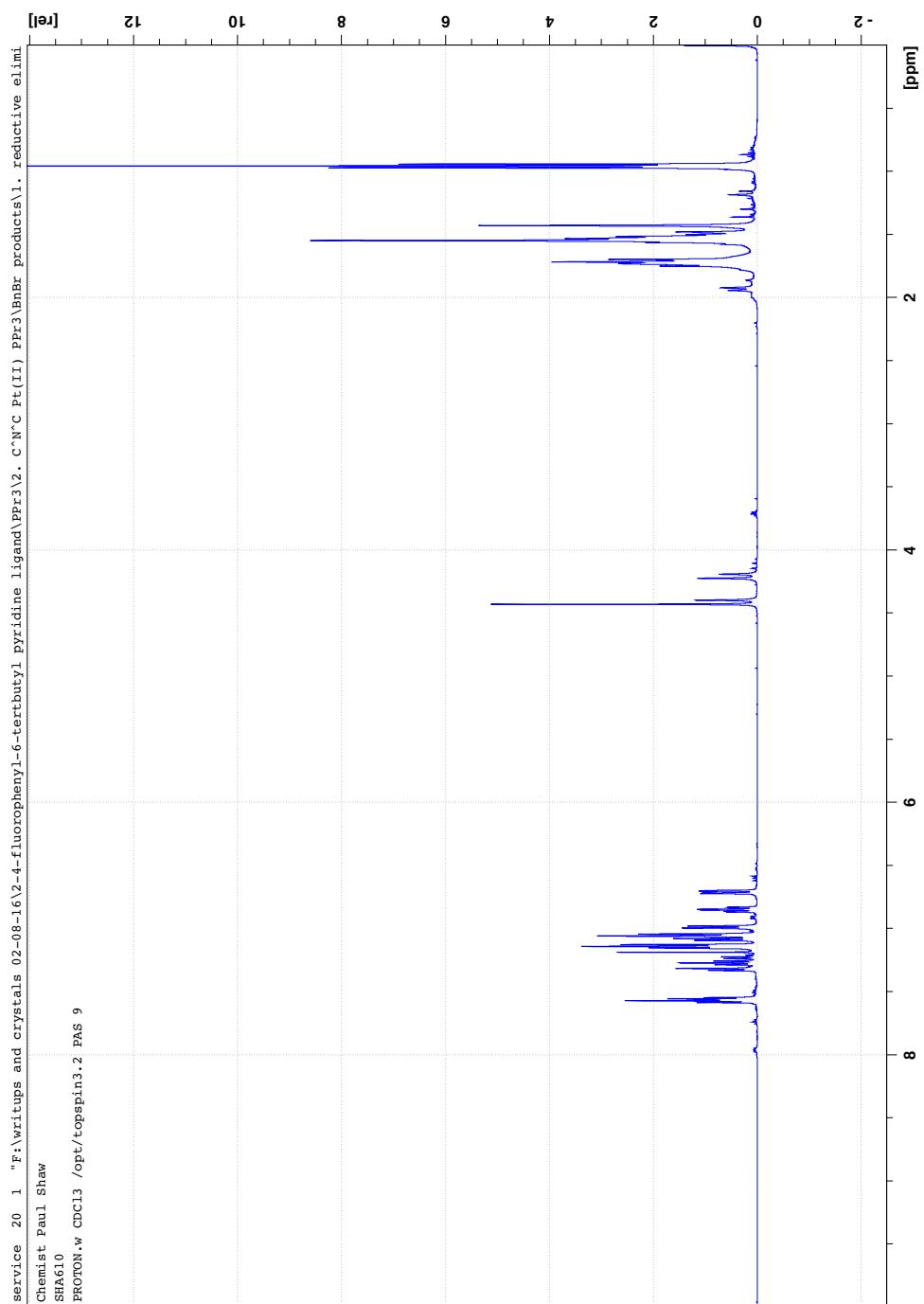
Chemist Paul Shaw
 SHA658
 C13APT.w CDCl₃ /opt/topspin3.2 PAS 37



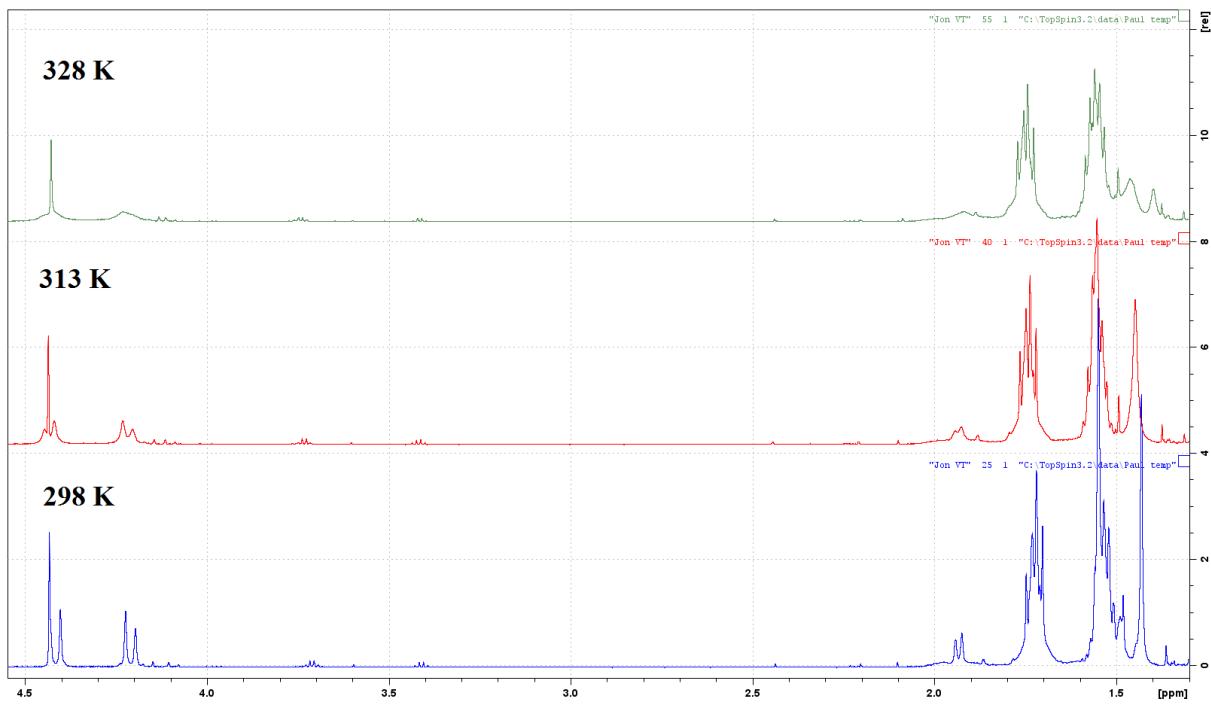
$\delta_{\text{C}} = 15.73$ (d, $^3J_{\text{C-Pt}} = 15$ Hz, PCH₂CH₂Me), 17.91 (s, $^3J_{\text{C-Pt}} = 32$ Hz, PCH₂CH₂), 26.27 (d, $^1J_{\text{C-Pt}} = 37$ Hz, $^2J_{\text{C-Pt}} = 46$ Hz, PCH₂), 27.39 (d, $^2J_{\text{C-Pt}} = 5$ Hz, $^1J_{\text{C-Pt}} = 746$ Hz, C_n), 32.58 (s, $^3J_{\text{C-Pt}} = 59$ Hz, H_m), 34.07 (s, $^3J_{\text{C-Pt}} = 33$ Hz, H_m), 38.49 (s, C_{allyl}), 49.72 (s, $^2J_{\text{C-Pt}} = 23.5$ Hz, C_l), 112.08 (d, $^2J_{\text{C-F}} = \text{Hz}$, C_b), 115.96 (d, $^2J_{\text{C-F}} = \text{Hz}$, C_d), 116.37 (s, C_{allyl}), 119.62 (s, $^3J_{\text{C-Pt}} = 25$ Hz, C_j), 125.67 (d, $^4J_{\text{C-Pt}} = 4$ Hz, C_h), 132.71 (d, $^3J_{\text{C-F}} = 9.5$ Hz, C_h), 136.95 (m, C_f), 137.05 (m, C_{i,allyl}), 141.01 (d, $^3J_{\text{C-F}} = 8.5$ Hz, C_e), 159.90 (s, C_g), 163.21 (d, $^1J_{\text{C-F}} = 249$ Hz, C_c), 174.60 (s, C_k) ppm.

$\delta_{\text{F}} = -113.00$ ppm. $\delta_{\text{P}} = -1.2$ ($^1J_{\text{P-Pt}} = 4330$ Hz) ppm. $\delta_{\text{Pt}} = -4214$ (d, $^1J_{\text{Pt-P}} = \sim 4350$ Hz) ppm.

Complex Bn-3

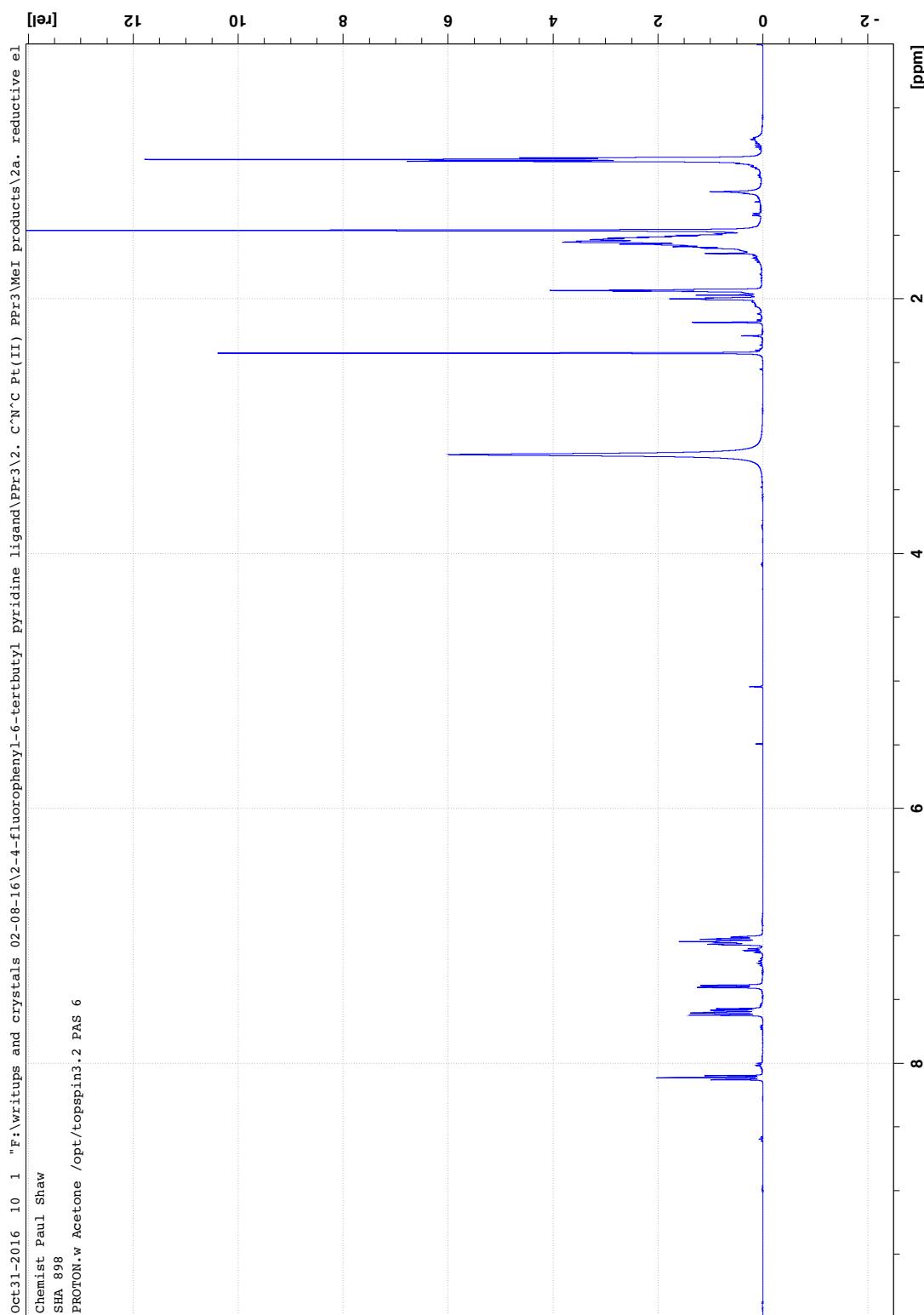


$\delta_H = 7.57$ (2H, m, H_{a,i}), 7.14 (3H, m, H_j, Bn-*m*), 7.08 (1H, m, Bn-*p*), 7.05 (2H, d, ³J_{H-H} = 7.5 Hz, Bn-*o*), 6.99 (2H, d, ³J_{H-H} = 7.5 Hz, H_h), 6.85 (1H, td, ³J_{H-H} = ³J_{H-F} = 8.5 Hz, ⁴J_{H-H} = 2.5 Hz, H_b), 6.70 (1H, td, ³J_{H-F} = 10 Hz, ⁴J_{H-H} = 2.5 Hz, H_b), 4.41 (1H, d, ²J_{H-H} = 16.5 Hz, H_o), 2.41 (1H, d, ²J_{H-H} = 16.5 Hz, Bn-CH₂), 1.93 (1H, d, ²J_{H-H} = 10.5 Hz, ²J_{H-Pt} = 51 Hz, H_n), 1.72 (7H, m, H_n, PCH₂), 1.53 (9H, m, H_m, PCH₂CH₂), 1.43 (3H, s, H_m), 0.96 (9H, t, ³J_{H-H} = 7 Hz, PCH₂CH₂Me) ppm.



Note that at 298K the Bn-CH₂, H_m and H_n resonances have separated into 2 peaks each, separated by 103, 60 and 102 Hz respectively (600 MHz spectrometer). By 328K, multiplicity is lost in all cases due to broadening of the peaks. Using a separation of 102 Hz and a coalescence temperature of 338 K we can calculate a barrier of 68.0 kJmol⁻¹; in line with our previous calculation our estimate of the uncertainty in this is ± 5 kJmol⁻¹.

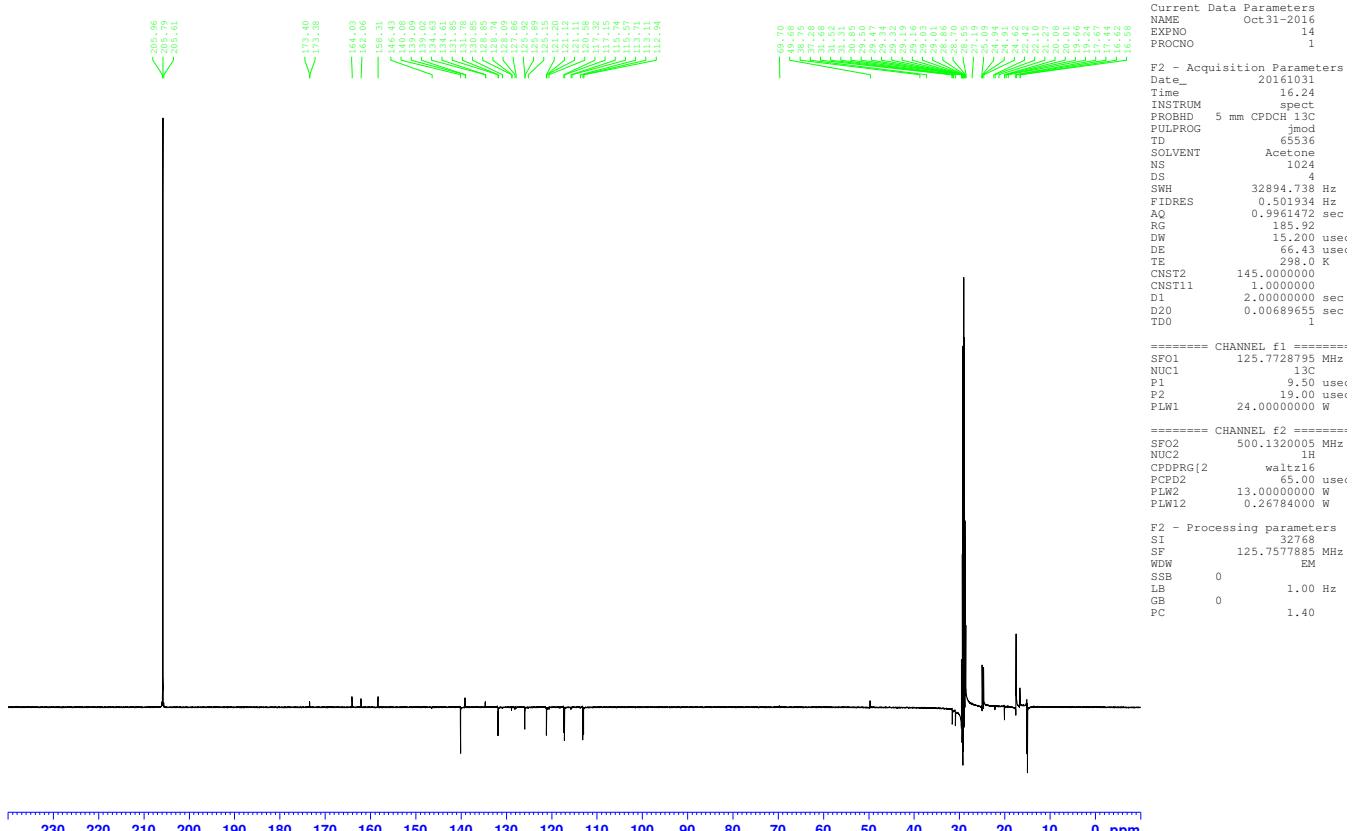
Complex Me-4



δ_H (Acetone-d₆) = 8.11 (1H, t, $^3J_{H-H} = 8$ Hz, H_i), 7.59 (2H, m, H_{j,a}), 7.40 (1H, d, $^3J_{H-H} = 8$ Hz, H_h), 7.04 (2H, m, H_{b,d}), 2.42 (3H, s, Me), 2.00 (2H, m, $^2J_{H-Pt} = \sim 32$ Hz, H_n), 1.54 (12H, m, PCH₂, PCH₂CH₂), 0.91 (9H, t, $^3J_{H-H} = 7$ Hz, PCH₂CH₂Me) ppm.

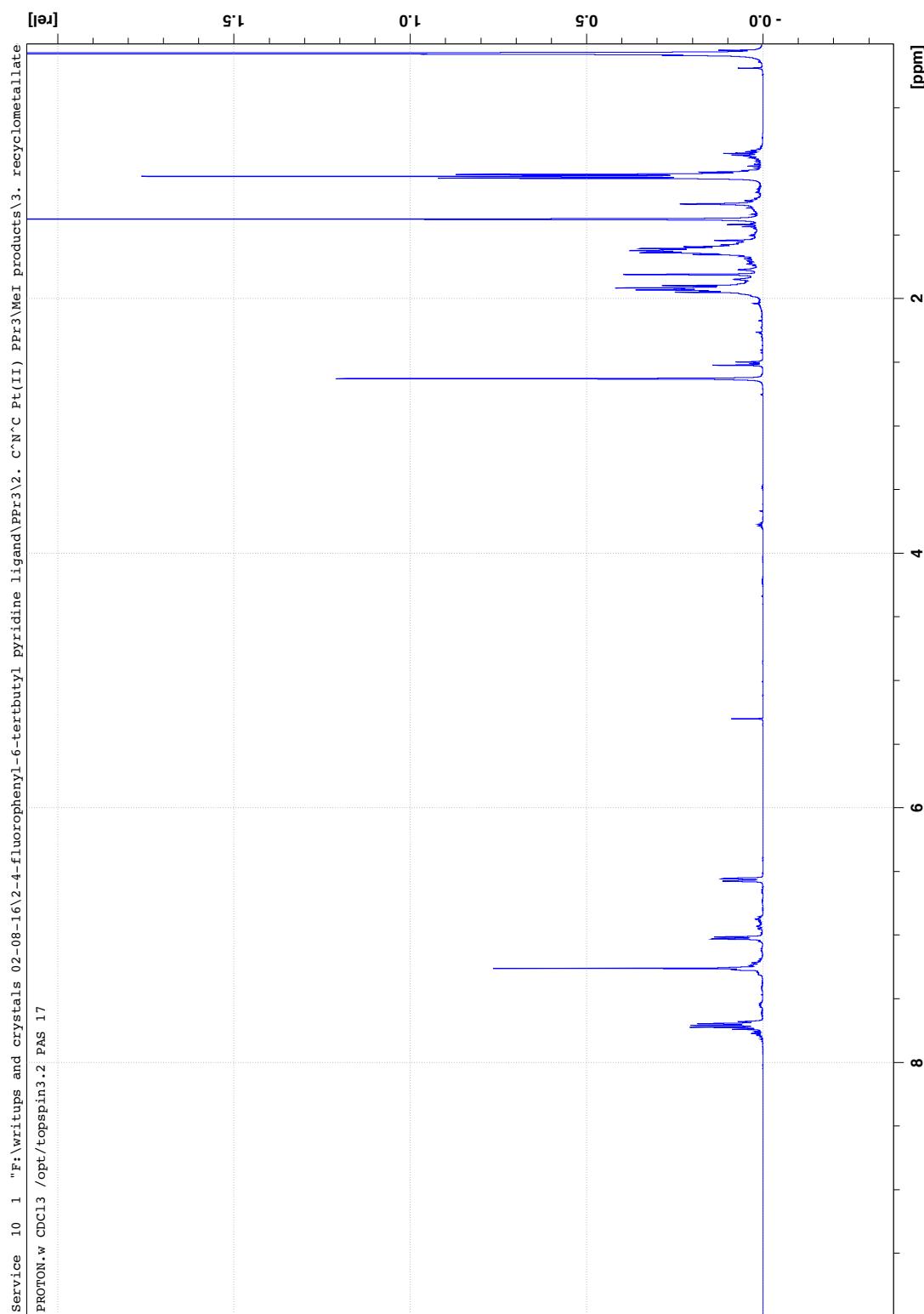
**Chemist Paul Shaw
SHA 898
C13APT.w Acetone /opt/topspin3.2 PAS 6**





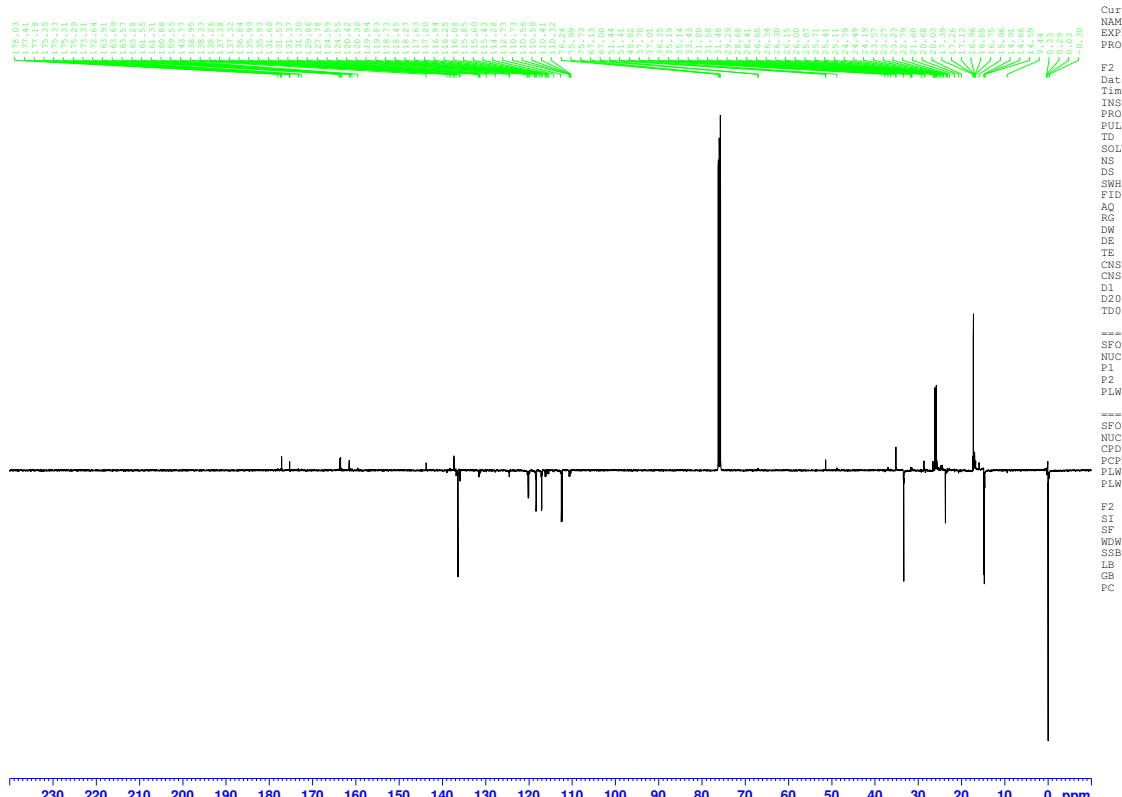
δ_C = 15.01 (d, $^3J_{C-P}$ = 15.5 Hz, PCH₂CH₂Me), 16.61 (d, $^2J_{C-P}$ = 5.5 Hz, $^1J_{C-Pt}$ = 765 Hz, C_n), 17.45 (s, $^3J_{C-Pt}$ = 30 Hz, PCH₂CH₂), 20.03 (s, Me), 24.77 (d, $^1J_{C-P}$ = 38 Hz, $^2J_{C-Pt}$ = 42 Hz, PCH₂), 31.21 (s, C_m), 49.68 (s, C_l), 113.04 (d, $^2J_{C-F}$ = 25 Hz, C_b), 117.25 (d, $^2J_{C-F}$ = 22 Hz, C_d), 121.13 (s, C_j), 125.93 (s, C_h), 131.83 (d, $^3J_{C-F}$ = 8.5 Hz, C_a), 134.63 (s, C_f), 139.07 (d, $^3J_{C-F}$ = 8.5 Hz, C_e), 140.10 (s, C_i), 158.35 (s, C_g), 163.02 (d, $^1J_{C-F}$ = 245 Hz, C_c), 173.43 (s, C_k) ppm. δ_F (Acetone-d₆) = -114.13 ppm. δ_P (Acetone-d₆) = 3.00 ($^1J_{P-Pt}$ = 4272 Hz) ppm. δ_{Pt} (Acetone-d₆) = -4116 (d, $^1J_{Pt-P}$ = ~4400 Hz) ppm.

Complex Me-1

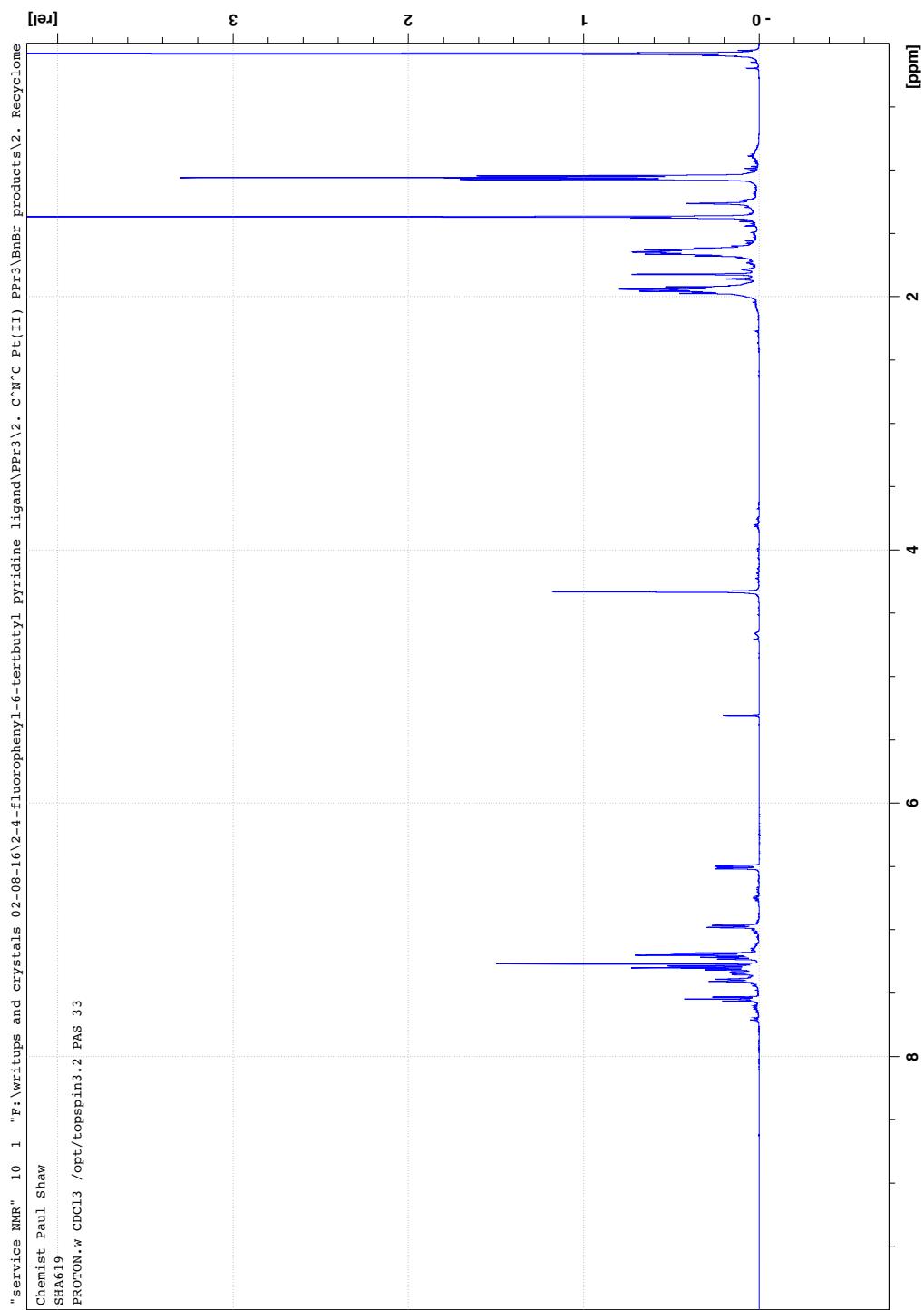


$\delta_H = 7.71$ (2H, m, $H_{h,i}$), 7.25 (1H, m, H_b), 7.02 (1H, d, ${}^3J_{H-H} = 8.5$ Hz, H_j), 6.56 (1H, dd, ${}^3J_{H-H} = 9.5$ Hz, ${}^5J_{H-P} = 2$ Hz, H_d), 2.63 (3H, s, Me), 1.92 (6H, m, PCH_2), 1.81 (2H, s, ${}^2J_{H-Pt} = 37$ Hz, H_n), 1.62 (6H, m, PCH_2CH_2), 1.37 (6H, s, H_m), 1.04 (9H, t, ${}^3J_{H-H} = 7$ Hz, PCH_2CH_2Me) ppm.

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SHA 620
C13APT.w CDCl₃ /opt/topspin3.2 PAS 17

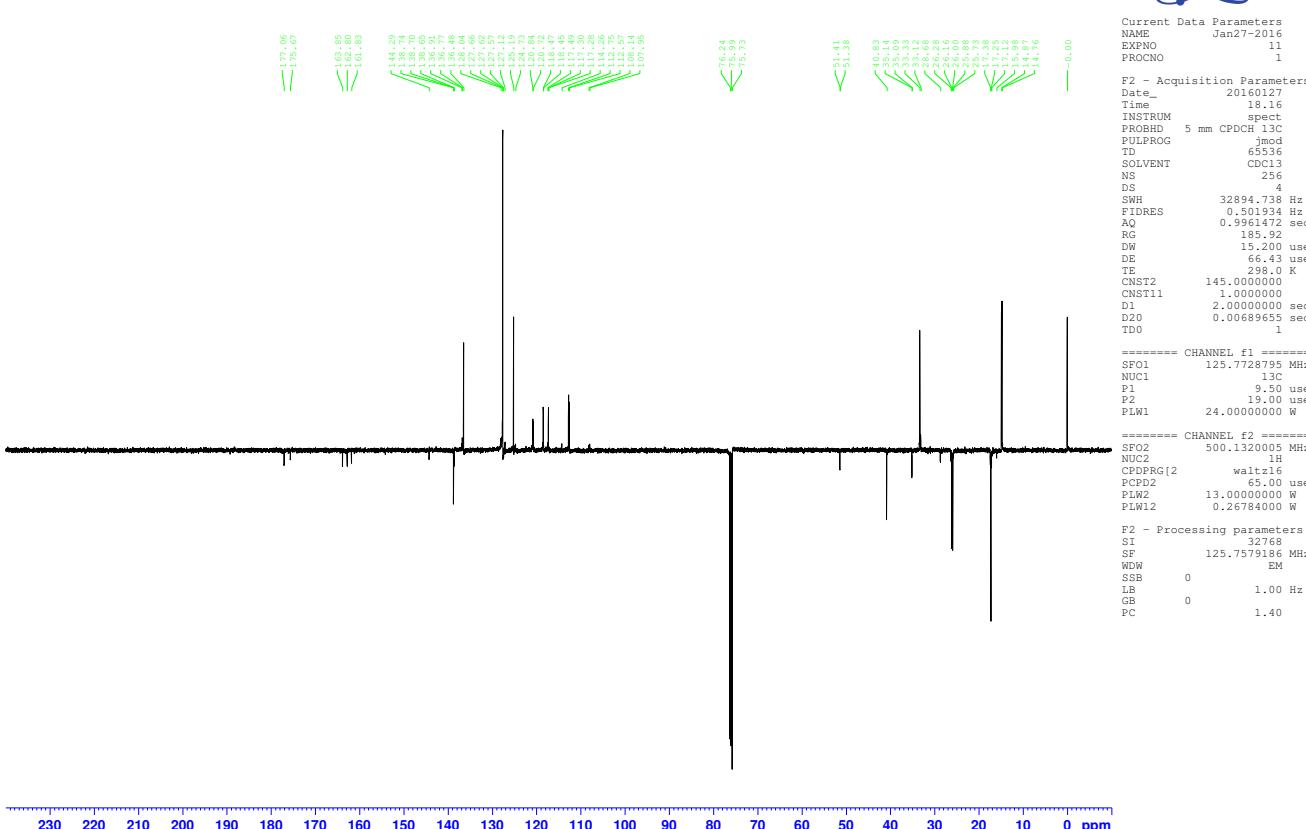


Complex Bn-1



$\delta_H = 7.55$ (1H, t, $^3J_{H-H} = 7.5$ Hz, H_i), 7.41 (1H, d, $^3J_{H-H} = 7.5$ Hz, H_h), 7.35 (d, 1H, $^3J_{H-F} = 8$ Hz, $^4J_{H-H} = 2$ Hz, $^3J_{H-Pt} = \#20$ Hz, H_b), 7.31 (3H, m, Bn-*m,p*), 7.20 (2H, d, $^3J_{H-H} = 8.5$ Hz, Bn-*o*), 6.98 (1H, d, $^3J_{H-H} = 7.5$ Hz, H_j), 6.51 (1H, dd, $^3J_{H-F} = 10$ Hz, $^4J_{H-H} = 2$ Hz, H_d), 4.34 (2H, s, Bn-CH₂), 1.95 (6H, m, PCH₂), 1.83 (2H, s, $^2J_{H-Pt} = 37$ Hz, H_n), 1.66 (6H, m, PCH₂CH₂), 1.38 (6H, s, C_m), 1.07 (9H, t, $^3J_{H-H} = 7$ Hz, PCH₂CH₂Me) ppm.

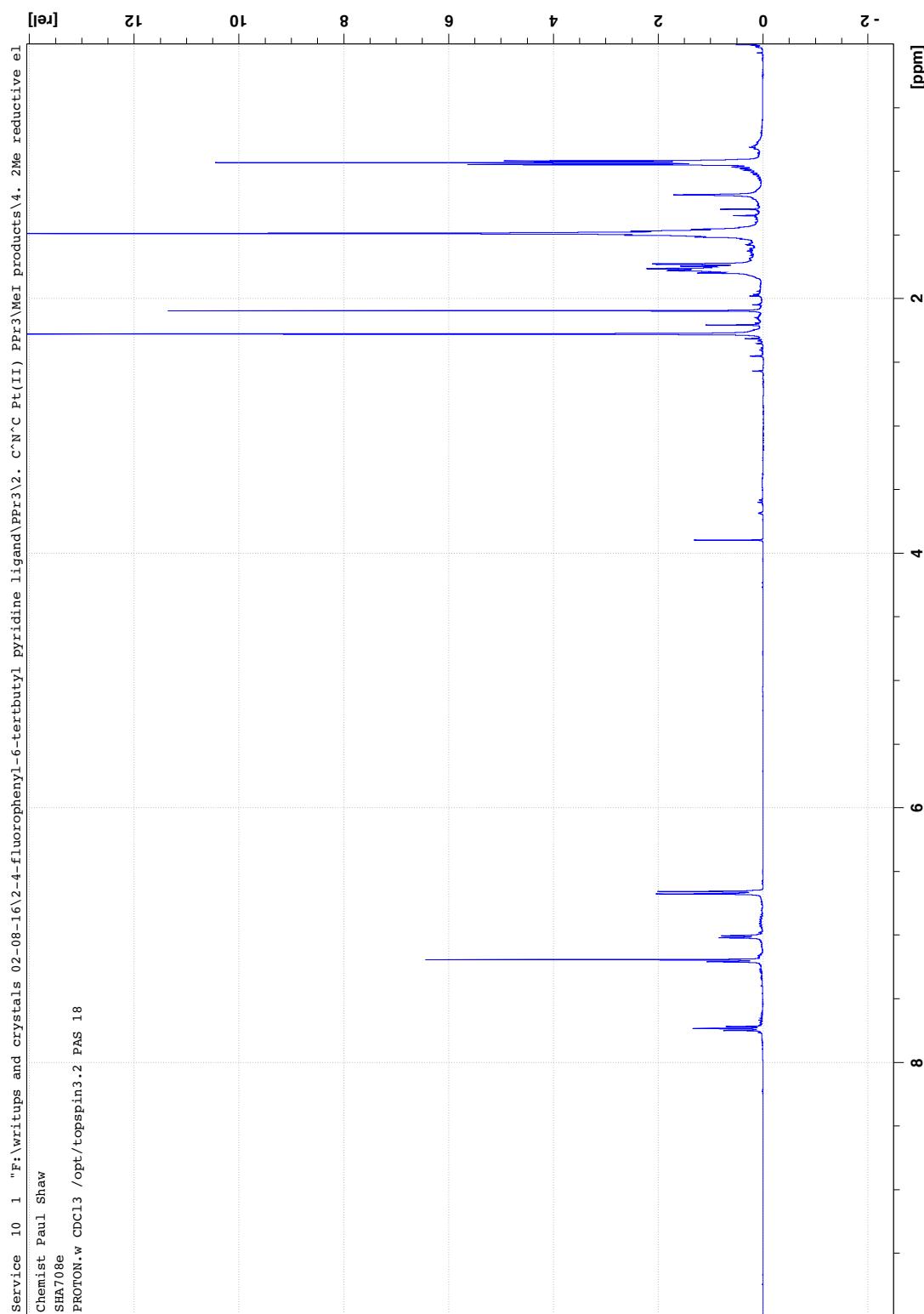
Chemist Paul Shaw
SHA619
C13APT.w CDCl₃ /opt/topspin3.2 PAS 33



$\delta_{\text{C}} = 14.79$ (d, $^3J_{\text{C-P}} = 14$ Hz, PCH₂CH₂Me), 17.26 (s, $^3J_{\text{C-Pt}} = 30.5$ Hz, PCH₂CH₂), 26.00 (d, $^1J_{\text{C-P}} = 36$ Hz, $^2J_{\text{C-Pt}} = 36$ Hz, PCH₂), 33.36 (s, $^3J_{\text{C-Pt}} = 17$ Hz, C_m), 35.17 (s, C_n), 40.86 (s, Bn-CH₂), 51.44 (s, C_i), 112.69 (d, $^2J_{\text{C-F}} = 24$ Hz, C_d), 117.35 (s, C_j), 118.52 (s, C_h), 120.83 (d, $^2J_{\text{C-F}} = 16$ Hz, C_b), 125.29 (s, Bn-*p*), 127.69 (s, Bn-*o,m*), 136.46 (s, C_i), 138.82 (s, Bn-*i*), 144.28 (s, C_f), 162.82 (s, C_g), 162.85 (d, $^1J_{\text{C-F}} = 253$ Hz, C_c), 175.66 (m, C_a), 177.07 (s, C_k) ppm.

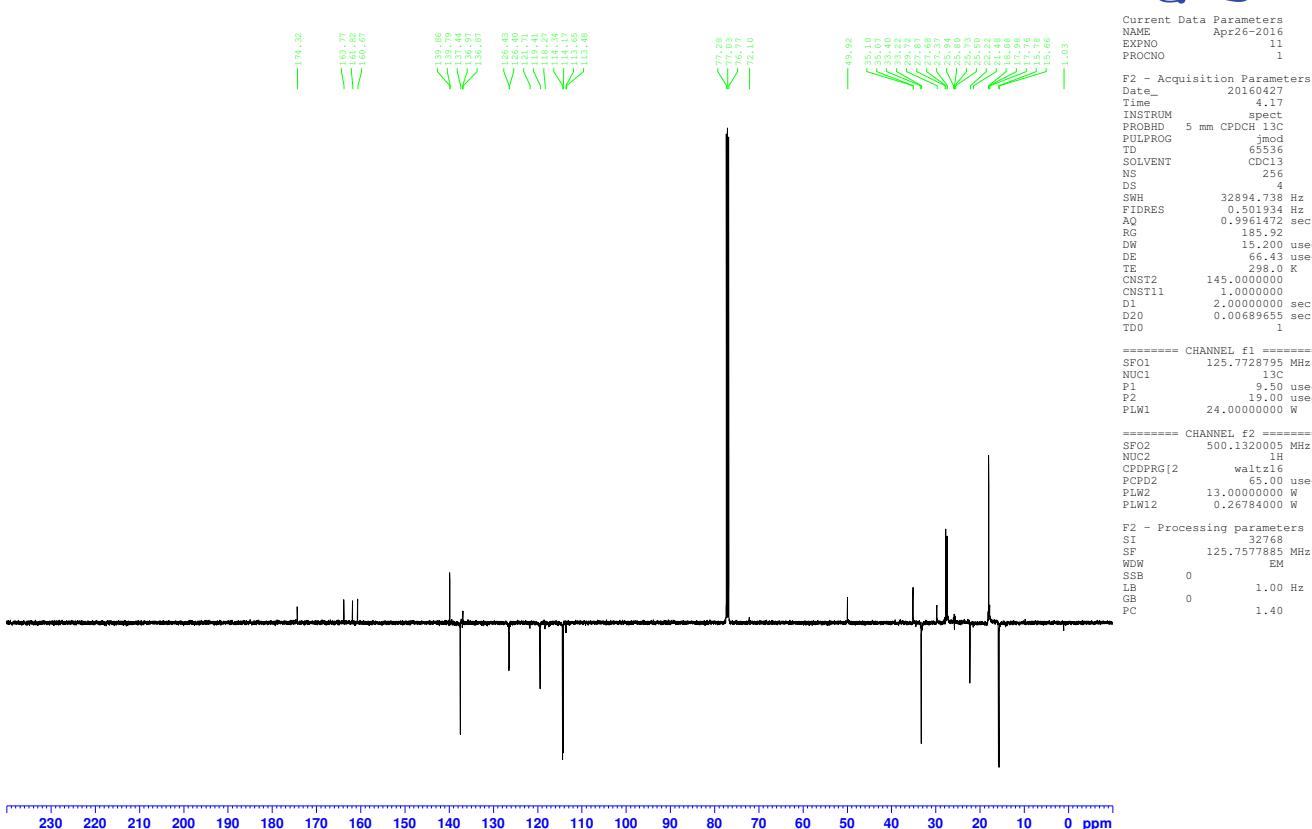
$\delta_{\text{F}} = -113.66$ ($^4J_{\text{F-Pt}} = 26.5$ Hz) ppm. $\delta_{\text{P}} = 0.86$ ($^1J_{\text{P-Pt}} = 3813$ Hz) ppm. $\delta_{\text{Pt}} = -3987$ (d, $^1J_{\text{Pt-P}} = \sim 3850$ Hz) ppm.

Complex Me₂-3



$\delta_H = 7.80$ (1H, t, ${}^3J_{H-H} = 8$ Hz, H_g), 7.27 (1H, d, ${}^3J_{H-H} = 8$ Hz, H_h), 7.08 (1H, d, ${}^3J_{H-H} = 8$ Hz, H_f), 6.73 (2H, d, ${}^3J_{H-F} = 9.5$ Hz, H_b), 2.35 (6H, s, Me), 1.83 (8H, m, H_l, PCH₂), 1.56 (12H, m, H_k, PCH₂CH₂), 0.99 (9H, t, ${}^3J_{H-H} = 7$ Hz, PCH₂CH₂Me) ppm.

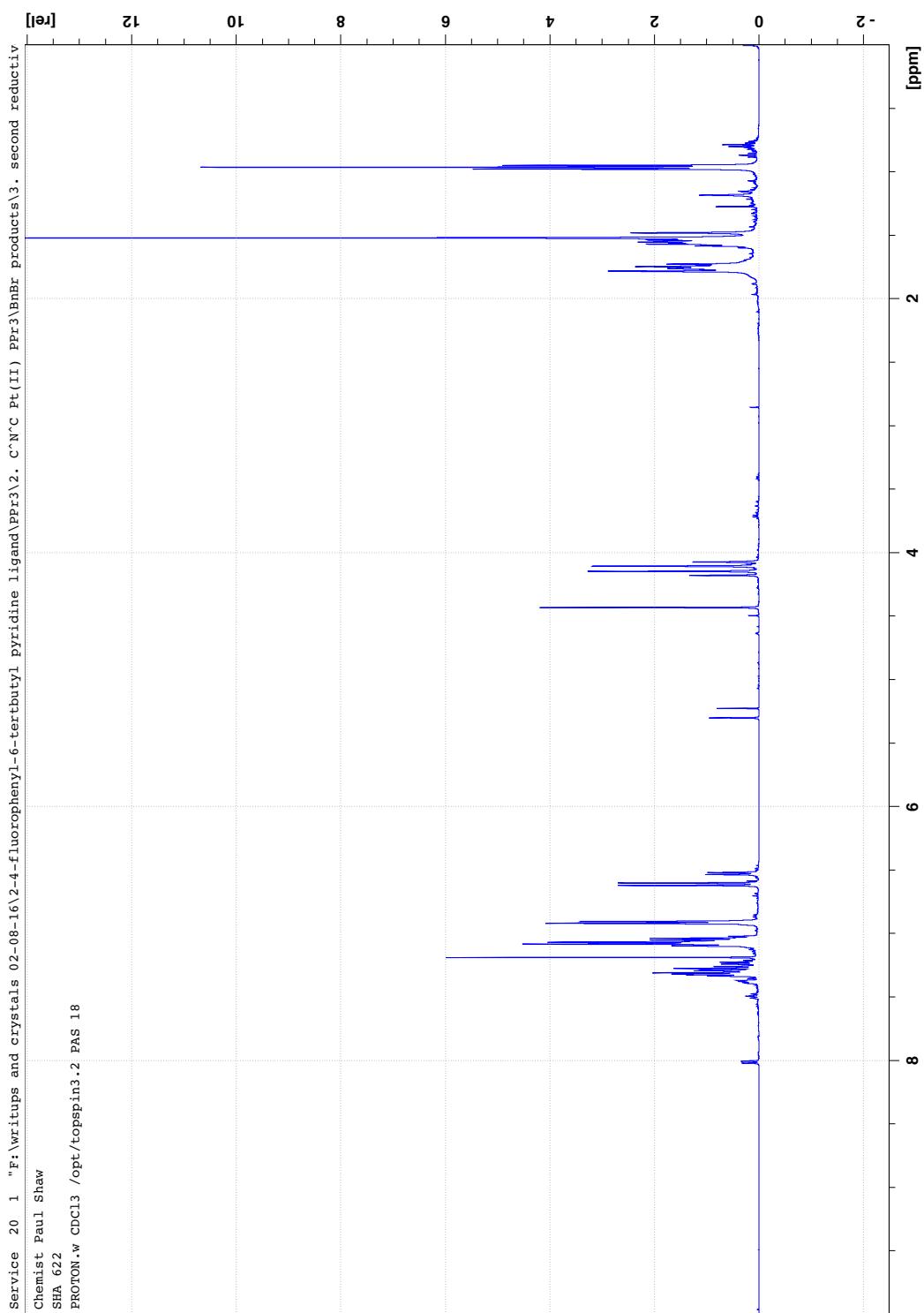
**Chemist Paul Shaw
SHA708e
C13APT.w CDCl₃ /opt/topspin3.2 PAS 18**



$\delta_{\text{C}} = 15.76$ (d, $^3J_{\text{C-P}} = 18$ Hz, $\text{PCH}_2\text{CH}_2\text{Me}$), 18.00 (s, $^3J_{\text{C-Pt}} = 30$ Hz, PCH_2CH_2) 22.24 (s, Me), 27.57 (d, $^1J_{\text{C-P}} = 35$ Hz, $^2J_{\text{C-Pt}} = 48$ Hz, PCH_2), 33.23 (s, $^3J_{\text{C-Pt}} = 44$ Hz, C_k), 35.16 (s, $^1J_{\text{C-Pt}} = 727$ Hz, C_l), 49.98 (s, C_j), 114.22 (d, $^2J_{\text{C-F}} = 23$ Hz, C_b), 119.42 (s, C_h), 126.44 (s, C_f), 136.86 (s, C_d), 137.44 (s, C_g), 139.83 (d, $^3J_{\text{C-F}} = 10$ Hz, C_c), 160.69 (s, C_e), 162.80 (d, $^1J_{\text{C-F}} = 254$ Hz, C_a), 174.32 (s, C_i) ppm.

$\delta_F = -114.84$ ppm, $\delta_P = -1.80$ ($^1J_{P-Pt} = 4292$ Hz) ppm, $\delta_{Pt} = -4382$ (d, $^1J_{Pt-P} = \sim 4300$ Hz) ppm.

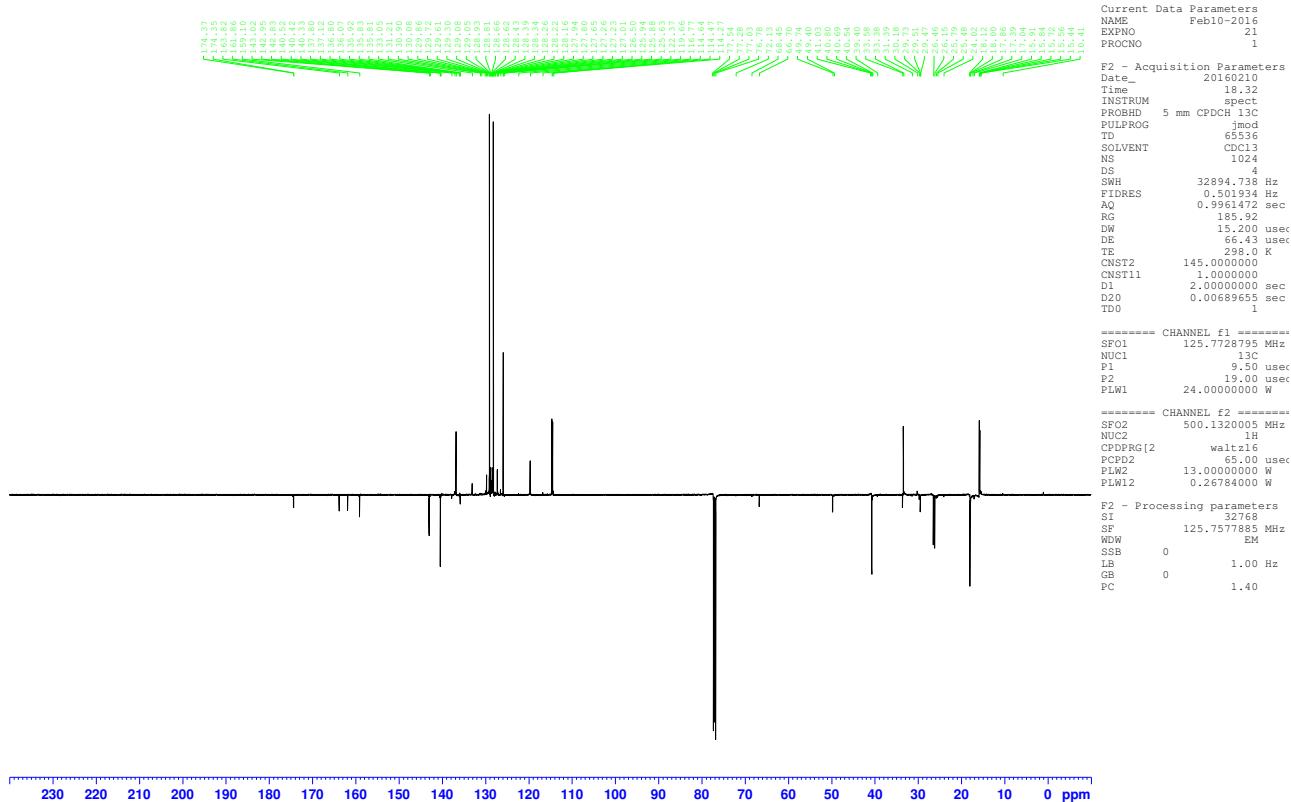
Complex Bn₂-3



δ_H = 7.32 (1H, t, ${}^3J_{H-H}$ = 8.5 Hz, H_g), 7.07 (7H, m, H_h, Bn-*m,p*), 6.91 (4H, d, ${}^3J_{H-H}$ = 8 Hz, Bn-*o*), 6.61 (2H, d, ${}^3J_{H-F}$ = 10 Hz, H_b), 6.53 (2H, d, ${}^3J_{H-H}$ = 8.5 Hz, H_f), 4.16 (2H, d, ${}^2J_{H-H}$ = 16 Hz, Bn-CH₂), 4.09 (2H, d, ${}^2J_{H-H}$ = 16 Hz, Bn-CH₂), 1.75 (8H, m, H_l, PCH₂), 1.55 (6H, m, PCH₂CH₂), 1.52 (6H, m, H_k), 0.96 (9H, t, ${}^3J_{H-H}$ = 7 Hz, PCH₂CH₂Me) ppm.

**Chemist Paul Shaw
SHA 622
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$\delta_{\text{C}} = 15.80$ (d, ${}^3J_{\text{C-P}} = 16$ Hz, $\text{PCH}_2\text{CH}_2\text{Me}$), 18.02 (s, ${}^3J_{\text{C-Pt}} = 27$ Hz, PCH_2CH_2), 26.35 (d, ${}^1J_{\text{C-P}} = 37$ Hz, ${}^2J_{\text{C-Pt}} = 34$ Hz, PCH_2), 29.50 (d, ${}^2J_{\text{C-P}} = 4$ Hz, C_l), 33.40 (s, C_k), 40.69 (s, C_m), 49.60 (s, C_j), 114.56 (d, ${}^2J_{\text{C-F}} = 22$ Hz, C_b), 119.66 (s, C_f), 125.88 (s, $\text{Bn-}p$), 127.26 (s, C_h), 128.18 (s, $\text{Bn-}m$), 129.09 (s, $\text{Bn-}o$), 135.83 (s, C_d), 136.81 (s, C_g), 140.44 (s, C_n), 143.01 (d, ${}^3J_{\text{C-F}} = 8$ Hz, C_c), 159.11 (s, C_e), 162.85 (d, ${}^1J_{\text{C-F}} = 247$ Hz, C_a), 174.50 (s, C_i) ppm.

$\delta_F = -113.65$ ppm. $\delta_P = -0.93$ ($^1J_{P,Pt} = 4298$ Hz) ppm. $\delta_{Pt} = -4250$ (d , $^1J_{Pt-P} \approx 4300$ Hz) ppm.