

ARTICLE

Supporting Information

Organic-inorganic hybrid Three-Dimensional Metal Sulfite-Oxalate with honeycomblike structures

Ranjay K. Tiwari and J. N. Behera*

School of Chemical Sciences, National Institute of Science Education and Research (NISER), Bhubaneswar 751005, India

*e-mail:jnbehera@niser.ac.in

ARTICLE

The bond valence sum (BVS) calculated from the formula as given below

 $S = exp[(R_o-R)/B]$

$$V = \Sigma S$$

Where V is the oxidation state of cation and S is bond valence

For [Na₂Mn₂(C₂O₄)(SO₃)₂(H₂O)₂], 1

Where B = 0.37, R_0 for Mn-O = 1.790 Å and R is Mn-O bond length

 $S1 = \exp[(1.790 - 2.218)/0.37] = 0.3145$

 $S2 = \exp[(1.790 - 2.1797)/0.37] = 0.3216$

S3 = exp[(1.790-2.1535)/0.37] = 0.3743

 $S4 = \exp[(1.790 - 2.2129)/0.37] = 0.3188$

 $S4 = \exp[(1.790 - 2.2299)/0.37] = 0.3045$

 $S4 = \exp[(1.790 - 2.2183)/0.37] = 0.3142$

And $V = \Sigma S$

V = S1 + S2 + S3 + S4 + S5 + S6 = 1.9479

For [NaEu(C₂O₄)(SO₃)(H₂O)₃], **2**

Where B = 0.37, R_o for Eu-O = 2.074 Å and R is Eu-O bond length

 $S1 = \exp[(2.074 - 2.4269)/0.37] = 0.3852$

 $S2 = \exp[(2.074 - 2.5210)/0.37] = 0.2987$

 $S3 = \exp[(2.074 - 2.3655)/0.37] = 0.4548$

 $S4 = \exp[(2.074 - 2.3559)/0.37] = 0.4667$

S5 = exp[(2.074 - 2.4361)/0.37] = 0.3758

S6= exp[(2.074-2.4662/0.37] = 0.3464

 $S7 = \exp[(2.074 - 2.4980)/0.37] = 0.3179$

S8= exp[(2.074-2.4879)/0.37] = 0.3267

 $S9 = \exp[(2.074 - 2.6458 / 0.37)] = 0.2132$

And $V = \Sigma S$

V = S1 + S2 + S3 + S4 + S5 + S6 + S7 + S8 + S9 = 3.1854

(1) I. D. Brown and D. Altermatt, Acta Crystallographica Section B, 1985, 41, 244-247.



Fig. S1 Na oxalate chain along c-axis in $[Na_2Mn_2(C_2O_4)(SO_3)_2(H_2O)_2]$, 1



Fig. S2 Distorted octahedral geometry of Na atom in $[NaEu(C_2O_4)(SO_3)(H_2O)_3]$, 2



Fig. S3 PXRD pattern for (a) [Na₂Mn₂(C₂O₄)(SO₃)₂(H₂O)₂], 1 and (b) [NaEu(C₂O₄)(SO₃)(H₂O)₃], 2



Fig. S4 Post-Calcinated PXRD pattern of [Na₂Mn₂(C₂O₄)(SO₃)₂(H₂O)₂], 1

Address Of 2010; 2014 - 4the Harr 197 Headress Demonstrate Control of Address D Order Address Ref Prove Address Ref Provide A

Fig. S5 Post-Calcinated PXRD pattern of [NaEu(C₂O₄)(SO₃)(H₂O)₃], 2



Fig. S6 Excitation spectrum of [NaEu(C₂O₄)(SO₃)(H₂O)₃], 2

ARTICLE

Journal Name

Table S1. Hydrogen bonding table for complexes 1-2.#

D–H···A	Symmetry of A	D-H	Н…А	D-A	∠D–H…A				
[Na ₂ Mn ₂ (C ₂ O ₄)(SO ₃) ₂ (H ₂ O) ₂] (1)									
01W–H1W1…03	-x,y,1/2-z	0.82(4)	2.02(4)	2.829(2)	168(3)				
01W–H2W1…O2	-x,2-y,-z	0.81(2)	2.15(3)	2.856(2)	146(3)				
[NaEu(C ₂ O ₄)(SO ₃)(H ₂ O)₃] (2)								
01W–H1W1…07	1-x,1-y,1-z	0.81(3)	1.96(3)	2.738(2)	162(3)				
02W–H1W2…01		0.81(3)	2.07(3)	2.813(2)	153(3)				
O3W-H1W3…O1W	1-x,1-y,-z	0.82(3)	2.01(3)	2.795(2)	158(3)				
01W–H2W1…O3		0.82(4)	2.07(4)	2.856(3)	159(3)				
O2W–H2W2…O5	x,1+y,z	0.82(2)	1.980(18)	2.741(2)	155(3)				
O3W–H2W3…O1	-1+x,y,z	0.81(3)	2.30(3)	3.107(3)	170(4)				

"Where 'D' is donor and 'A' is acceptor, the bond lengths are in (Å) and angles are in (°).

Table S2. Complete list of bond lengths [Å] and bond angles [$^{\circ}$] for complex 1-2 [#]									
[Na2Mn2(C2O4)(SC	03)2(H2O)2] (1)		,						
Mn1-01	2.2181(13)	Na1-O2 ^e	2.3589(15)	S1-01	1.5471(12)				
Mn1-O1 ^b	2.1797(12)	Na1-O3 ^e	2.5146(15)	S1-02	1.5225(12)				
Mn1-O2 ^a	2.1535(12)	Na1-04	2.3410(16)	S1-O3	1.5315(13)				
Mn1-03 ^c	2.2129(12)	Na1-O5 ^g	2.9900(19)	O4-C1	1.2500(2)				
Mn1-04	2.2299(13)	Na1-O1W	2.3171(18)	05-C1	1.2560(2)				
Mn1-05 ^d	2.2183(14)	Na1-O1W ^f	2.507(2)	C1-C1 ^d	1.5440(3)				
01 ^b -Mn1-01	74.93(5)	02 ^e -Na1-O5 ^g	111.24(5)	O2-S1-Na1 ^h	48.83(5)				
02ª-Mn1-01	104.82(5)	O1W ^f -Na1-O5 ^g	158.62(6)	O3-S1-Na1 ^h	54.80(5)				
02 ^{<i>a</i>} -Mn1-O1 ^{<i>b</i>}	94.19(5)	03 ^e -Na1-05 ^g	65.26(4)	O1-S1-Na1 ^h	110.90(5)				
03 ^c -Mn1-01	93.78(5)	O1W-Na1-S1 ^e	129.22(5)	S1-O1-Mn1 ^b	134.29(7)				
02 ^{<i>a</i>} -Mn1-03 ^{<i>c</i>}	84.48(5)	O4-Na1-S1 ^e	131.41(5)	S1-01-Mn1	117.27(7)				
01 ^b -Mn1-03 ^c	167.95(5)	O2 ^e -Na1-S1 ^e	29.07(3)	Mn1 ^b -O1-Mn1	105.07(5)				
02 ^a -Mn1-05 ^d	94.14(5)	O1W ^f -Na1-S1 ^e	112.17(5)	S1-O2-Mn1 ⁱ	124.38(7)				
01 ^b -Mn1-05 ^d	107.00(5)	O3 ^e -Na1-S1 ^e	29.85(3)	S1-O2-Na1 ^h	102.10(7)				
03 ^c -Mn1-05 ^d	85.05(5)	O5 ^g -Na1-S1 ^e	89.12(4)	Mn1 ⁱ -O2-Na1 ^h	112.36(6)				
01-Mn1-05 ^d	160.82(5)	O1W-Na1-C1	78.07(6)	S1-O3-Mn1 ^c	114.89(7)				
02ª-Mn1-04	167.36(5)	O4-Na1-C1	20.70(4)	\$1-03-Na1 ^h	95.36(6)				
01 ^b -Mn1-04	93.45(5)	02 ^e -Na1-C1	132,11(6)	Mn1 ^c -O3-Na1 ^h	107.46(5)				
03 ^c -Mn1-04	90.10(5)	O1W ^f -Na1-C1	75.37(5)	C1-O4-Mn1	115.67(11)				
01-Mn1-04	86.89(5)	03 ^e -Na1-C1	148.95(6)	C1-04-Na1	117.87(11)				
05 ^d -Mn1-04	73 97(5)	05 ^g -Na1-C1	84 87(5)	Mn1-04-Na1	120.93(6)				
01W-Na1-04	98.46(6)	S1 ^e -Na1-C1	152 10(5)	C1-O5-Mn1 ^d	116.05(11)				
01W-Na1-02 ^e	142 78(7)	01W-Na1-Na1	132.10(5)	C1-O5-Na19	103 98(12)				
04-Na1-02	11/ 38(6)	04-Na1-Na1	90.46(5)	Mn1 ^d -05-Na1 ^g	93 02(5)				
01W-Na1-01W	78 37(7)	02º_Na1-Na1	118 34(6)	Na1_01W_Na1	101 63(6)				
	78.37(7) 82.87(6)		118.34(0)		126 26(16)				
	82.87(0)		122 42(5)	04-C1-O3	120.20(10) 117.01(10)				
	109 62(6)		132.42(3)		117.01(19) 116.72(10)				
	108.02(0)		120.44(3)	03-C1-C1 04 C1 No1	110.72(19)				
04-Na1-03	130.07(0)		130.13(4)		41.44(0)				
$O2^{\circ}$ -INd $1-O3^{\circ}$	30.00(4) 125 26(6)		102 EE(7)		09.50(10) 146.00(12)				
	155.50(0)	02-31-03	103.33(7)	CICI-Nai	140.99(12)				
04-Na1-05 ^g	89.79(0) 81.34(5)	03-\$1-01	104.77(7)						
[NaEu(C ₂ O ₄)(SO ₃)	(H₂O)₃] (2)								
Eu1-01	2.4269(16)	Eu1-07'''	2.4879(16)	\$1-02	1.5549(16)				
Eu1-02	2.5210(15)	Na1-O2W	2.395(2)	\$1-03	1.5176(17)				
Eu1-02*	2.3655(15)	Na1-O2W"	2.323(2)	04-C1	1.257(3)				
Eu1-03/	2.3559(16)	Na1-O3W	2.478(2)	05-C1	1.251(3)				
Eu1-04	2.4361(16)	Na1-O3W ^o	2.3743(19)	06-C2	1.251(3)				
Eu1-05'	2.4662(17)	Na1-O4	2.3933(17)	07-C2	1.259(3)				
Eu1-06	2.4980(15)	Na1-06	2.3065(18)	C1-C1'	1.538(4)				
Eu1-O1W	2.6458(16)	S1-01	1.5349(16)	C2-C2 ^m	1.541(4)				
O3 ^{<i>j</i>} -Eu1-O2 ^{<i>k</i>}	81.26(5)	O5′-Eu1-C1	46.44(5)	O3W-Na1-Na1 ⁿ	148.86(7)				
03 ^j -Eu1-O1	147.93(5)	07 ^m -Eu1-C1	148.70(5)	C1-Na1-Na1 ⁿ	89.53(5)				
02 ^k -Eu1-O1	122.36(5)	O6-Eu1-C1	84.26(5)	O6-Na1-Na1°	106.08(6)				
03 [/] -Eu1-O4	102.07(6)	O2-Eu1-C1	131.19(5)	O2W ⁿ -Na1-Na1 ^o	96.16(6)				
02 ^k -Eu1-O4	139.44(5)	O1W-Eu1-C1	67.98(5)	O3W°-Na1-Na1°	45.74(5)				
01-Eu1-O4	74.69(5)	S1-Eu1-C1	111.88(4)	O4-Na1-Na1 ^o	121.62(6)				
O3 ^j -Eu1-O5 ^j	69.19(6)	O3 ^j -Eu1-Na1	86.79(4)	O2W-Na1-Na1°	149.63(7)				
O2 ^k -Eu1-O5 [/]	78.47(5)	O2 ^k -Eu1-Na1	165.71(4)	O3W-Na1-Na1°	43.33(4)				
01-Eu1-05'	132.21(6)	O1-Eu1-Na1	71.73(4)	C1-Na1-Na1°	103.06(6)				
O4-Eu1-O5'	65.74(5)	O4-Eu1-Na1	36.44(4)	Na1 ⁿ -Na1-Na1 ^o	133.23(6)				
O3 ^j -Eu1-O7 ^m	78.95(6)	O5 [/] -Eu1-Na1	90.00(4)	O6-Na1-Eu1	38.14(4)				
O2 ^k -Eu1-O7 ^m	87.10(5)	O7 ^m -Eu1-Na1	98.32(4)	O2W ⁿ -Na1-Eu1	119.21(5)				
01-Eu1-07 ^m	80.89(6)	O6-Eu1-Na1	34.76(4)	O3W ^o -Na1-Eu1	157.27(6)				

ARTICLE

Journal Name

04-Eu1-07 ^m	133.42(5)	O2-Eu1-Na1	128.35(4)	O4-Na1-Eu1	37.20(4)
05 [/] -Eu1-07 ^m	146.55(5)	O1W-Eu1-Na1	106.81(4)	O2W-Na1-Eu1	82.35(5)
03 ^j -Eu1-06	73.66(5)	S1-Eu1-Na1	99.751(17)	O3W-Na1-Eu1	79.49(5)
02 ^k -Eu1-06	144.84(5)	C1-Eu1-Na1	50.76(4)	C1-Na1-Eu1	54.52(4)
01-Eu1-06	75.21(5)	03-S1-01	104.37(9)	Na1 ⁿ -Na1-Eu1	103.95(4)
O4-Eu1-O6	71.20(5)	03-S1-O2	103.73(9)	Na1º-Na1-Eu1	120.16(4)
05′-Eu1-O6	113.66(5)	01-S1-O2	100.00(9)	S1-O1-Eu1	103.69(8)
07 ^m -Eu1-O6	64.44(5)	O3-S1-Eu1	109.76(7)	S1-O2-Eu1 ^k	144.96(9)
03 ^{<i>j</i>} -Eu1-O2	134.17(5)	01-S1-Eu1	48.19(6)	S1-O2-Eu1	99.07(7)
02 ^k -Eu1-O2	65.95(6)	O2-S1-Eu1	51.90(6)	Eu1 ^k -O2-Eu1	114.05(6)
01-Eu1-02	57.12(5)	06-Na1-O2W ⁿ	154.74(7)	S1-O3-Eu1 ^p	135.10(10)
O4-Eu1-O2	123.71(5)	O6-Na1-O3W°	121.52(7)	C1-O4-Na1	112.73(13)
05 [/] -Eu1-O2	128.85(5)	O2W ⁿ -Na1-O3W ^o	82.69(7)	C1-O4-Eu1	120.80(14)
07 ^m -Eu1-O2	68.72(5)	O6-Na1-O4	75.33(6)	Na1-O4-Eu1	106.37(6)
06-Eu1-02	116.65(5)	02W ⁿ -Na1-O4	82.94(6)	C1-O5-Eu1 [/]	120.18(14)
03 ^j -Eu1-O1W	136.11(6)	O3Wº-Na1-O4	159.26(8)	C2-O6-Na1	128.22(14)
02 ^k -Eu1-O1W	77.18(5)	06-Na1-O2W	78.44(6)	C2-O6-Eu1	120.80(13)
01-Eu1-01W	74.40(5)	O2W ⁿ -Na1-O2W	88.48(7)	Na1-O6-Eu1	107.10(6)
04-Eu1-01W	72.90(5)	O3Wº-Na1-O2W	105.75(7)	C2-O7-Eu1 ^m	121.32(14)
05 [/] -Eu1-O1W	69.30(5)	O4-Na1-O2W	88.71(6)	Na1 ⁿ -O2W-Na1	91.52(7)
07 ^m -Eu1-O1W	136.56(5)	O6-Na1-O3W	83.44(7)	Na1º-O3W-Na1	90.93(7)
06-Eu1-01W	137.71(5)	O2W ⁿ -Na1-O3W	105.77(7)	05-C1-O4	126.8(2)
02-Eu1-01W	67.88(5)	O3Wº-Na1-O3W	89.07(7)	O5-C1-C1 [/]	116.4(2)
O3 ^j -Eu1-S1	152.73(4)	O4-Na1-O3W	80.56(6)	O4-C1-C1 [/]	116.8(2)
O2 ^k -Eu1-S1	94.45(4)	O2W-Na1-O3W	160.81(7)	O5-C1-Na1	95.21(13)
O1-Eu1-S1	28.12(4)	O6-Na1-C1	91.23(6)	O4-C1-Na1	45.33(10)
O4-Eu1-S1	98.59(4)	O2W ⁿ -Na1-C1	71.85(6)	C1'-C1-Na1	132.60(19)
O5′-Eu1-S1	136.61(4)	O3Wº-Na1-C1	137.54(7)	O5-C1-Eu1	166.60(16)
07 ^m -Eu1-S1	73.93(4)	O4-Na1-C1	21.94(6)	O4-C1-Eu1	39.88(10)
O6-Eu1-S1	96.73(4)	O2W-Na1-C1	106.89(6)	C1'-C1-Eu1	76.93(16)
O2-Eu1-S1	29.04(4)	O3W-Na1-C1	66.96(6)	Na1-C1-Eu1	74.72(5)
O1W-Eu1-S1	67.39(4)	O6-Na1-Na1 ⁿ	118.65(6)	O6-C2-O7	127.2(2)
O3 ^j -Eu1-C1	92.83(6)	O2W ⁿ -Na1-Na1 ⁿ	45.09(5)	O6-C2-C2 ^m	116.8(2)
O2 ^k -Eu1-C1	121.77(5)	O3W ^o -Na1-Na1 ⁿ	96.02(6)	07-C2-C2 ^m	116.0(2)
O1-Eu1-C1	91.58(5)	O4-Na1-Na1 ⁿ	84.24(5)		
O4-Eu1-C1	19.33(5)	O2W-Na1-Na1 ⁿ	43.39(5)		

#Symmetry transformations used to generate equivalent atoms: (a) x,-y,z-1/2; (b) -x+1/2,-y+1/2,-z+1; (c) -x+1/2,-y-1/2,-z+1; (d) -x,y,-z+1/2; (e) -x+1/2,y+1/2,-z+3/2; (f) - x,-y+1,-z+1; (g) -x,-y,-z+1; (h) -x+1/2,y-1/2,-z+3/2; (i) x,-y,z+1/2; (j) x-1,y,z; (k) -x+1,-y+1,-z+1; (l) -x+1,-y+2,-z+1; (n) -x+1,-y+2,-z; (o) -x,-y+2,-z; (p) x+1,y,z (p) x+1,y,z (p) x+1,y,z (p) x+1,y,z (p) x+1,y,z (p) x+1,y,z (p) x+1,-y+1,-z+1; (l) -x+1,-y+1,-z; (p) -x,-y+2,-z+1; (n) -x+1,-y+2,-z; (p) x+1,y,z (p) x+1,y,z (p) x+1,y,z (p) x+1,-y+1,-z+1; (p) -x+1,-y+1,-z; (p) x+1,-y+1,-z; (p) x+1,-z; (p)

ARTICLE

Table S3. Atomic coordinates (×10⁴) and equivalent isotropic displacement parameters (Å×10³) for **1-2** U(eq) is defined as one third of the trace of the orthogonalized U_{ij} tensor.

	х	v	Z	U(eq)		х	Y	Z	U(eq)	
[Na ₂ Mn ₂ (C ₂ O ₄)(SO ₃) ₂ (H ₂ O) ₂] (1)										
Mn1	1862(1)	1035(1)	3633(1)	12(1)	03	3368(1)	-2580(2)	6635(1)	16(1)	
Na1	875(1)	3110(2)	5984(1)	31(1)	04	739(1)	1351(2)	4115(1)	19(1)	
S1	2567(1)	-1256(1)	6413(1)	11(1)	05	-705(1)	1322(3)	3233(1)	24(1)	
01	2679(1)	759(2)	5733(1)	16(1)	01W	-577(1)	3681(3)	5426(2)	26(1)	
02	2733(1)	-519(2)	7760(1)	15(1)	C1	11(1)	1327(3)	3178(2)	15(1)	
[NaEu	[NaEu(C₂O₄)(SO₃)(H₂O)₃] (2)									
Eu1	3687(1)	6287(1)	3406(1)	5(1)	06	1111(3)	9213(2)	3127(2)	8(1)	
S1	7304(1)	7326(1)	4541(1)	7(1)	07	-1231(3)	11846(2)	4228(2)	9(1)	
Na1	2415(2)	9638(1)	650(1)	11(1)	01W	8196(3)	4194(2)	2222(2)	10(1)	
01	6115(3)	7978(2)	3338(2)	10(1)	O2W	4737(3)	10719(2)	1676(2)	13(1)	
02	6107(3)	6032(2)	5155(2)	8(1)	O3W	-94(3)	8054(2)	365(2)	14(1)	
03	9869(3)	6218(2)	3672(2)	11(1)	C1	5304(4)	5787(3)	-228(2)	7(1)	
04	5001(3)	6831(2)	811(2)	9(1)	C2	-30(4)	10302(3)	4231(2)	7(1)	
05	6023(3)	5902(2)	-1596(2)	9(1)						

Table S4. Anisotropic displacement parameters ($Å^2 \times 10^3$) for complex **1-2**. The anisotropic displacement factor exponent takes the form: -2 π^2 [$h^2 a^{*2} U_{11} + ... + 2 h k a^* b^* U_{12}$]

	U11	U22	U33	U23	U13	U12		U11	U22	U33	U23	U13	U12
[Na₂N	[Na ₂ Mn ₂ (C ₂ O ₄)(SO ₃) ₂ (H ₂ O) ₂](1)												
Mn1	13(1)	12(1)	11(1)	-1(1)	5(1)	0(1)	03	20(1)	12(1)	19(1)	1(1)	12(1)	3(1)
Na1	22(1)	50(1)	19(1)	-7(1)	6(1)	-2(1)	04	14(1)	28(1)	13(1)	-3(1)	4(1)	0(1)
S1	14(1)	10(1)	10(1)	0(1)	5(1)	-1(1)	05	14(1)	44(1)	16(1)	-6(1)	8(1)	-3(1)
01	23(1)	11(1)	11(1)	2(1)	6(1)	-2(1)	01W	24(1)	33(1)	27(1)	-7(1)	15(1)	-8(1)
02	20(1)	16(1)	12(1)	-1(1)	10(1)	0(1)	C1	15(1)	16(1)	13(1)	-2(1)	7(1)	0(1)
[NaEu	ı(C₂O₄)(S	D₃)(H₂O);	₃](2)										
Eu1	5(1)	5(1)	3(1)	0(1)	-1(1)	-2(1)	06	10(1)	9(1)	5(1)	0(1)	-1(1)	-3(1)
S1	7(1)	7(1)	6(1)	1(1)	-2(1)	-3(1)	07	10(1)	8(1)	7(1)	1(1)	-2(1)	-1(1)
Na1	13(1)	12(1)	7(1)	2(1)	-2(1)	-4(1)	01W	12(1)	9(1)	9(1)	1(1)	-3(1)	-4(1)
01	10(1)	10(1)	11(1)	4(1)	-5(1)	-4(1)	O2W	19(1)	10(1)	9(1)	2(1)	-2(1)	-6(1)
02	8(1)	10(1)	8(1)	4(1)	-3(1)	-6(1)	O3W	16(1)	14(1)	10(1)	0(1)	-3(1)	-5(1)
03	7(1)	11(1)	12(1)	0(1)	-1(1)	-3(1)	C1	5(1)	7(1)	7(1)	1(1)	-1(1)	-1(1)
04	11(1)	8(1)	6(1)	0(1)	-2(1)	-3(1)	C2	6(1)	7(1)	8(1)	0(1)	-1(1)	-3(1)
05	14(1)	10(1)	4(1)	-1(1)	-1(1)	-6(1)							