

Supporting Information

Facile hydrothermal synthesis of carbon dots modified g-C₃N₄ for enhanced photocatalytic H₂-evolution performance

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Table S1 The element components of various samples based on the EDS results.

Samples	C	N	O	Molar ratio of C/N	g _{CDs} /g _{catalyst} (wt%)
g-C ₃ N ₄	39.46	56.66	3.88	0.70	
CDs/g-C ₃ N ₄ (1 wt%)	39.79	56.34	3.87	0.71	0.53
CDs/g-C ₃ N ₄ (10 wt%)	40.66	56.18	3.16	0.72	1.06
CDs/g-C ₃ N ₄ (50 wt%)	41.96	51.82	6.22	0.81	5.56
CDs/g-C ₃ N ₄ (100 wt%)	44.88	48.87	6.25	0.92	10.54

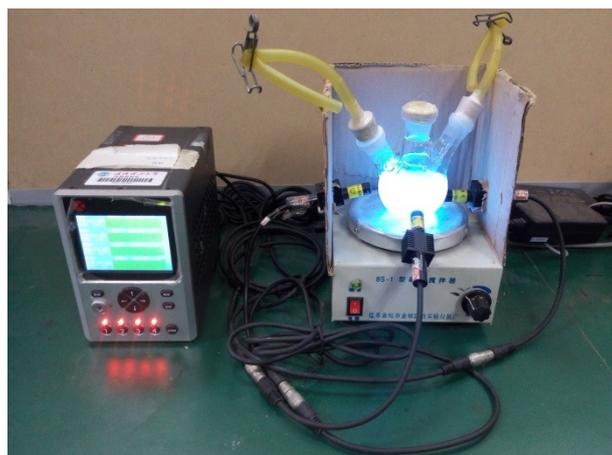


Fig. S1. The set-up diagram for the photocatalytic H₂ production.

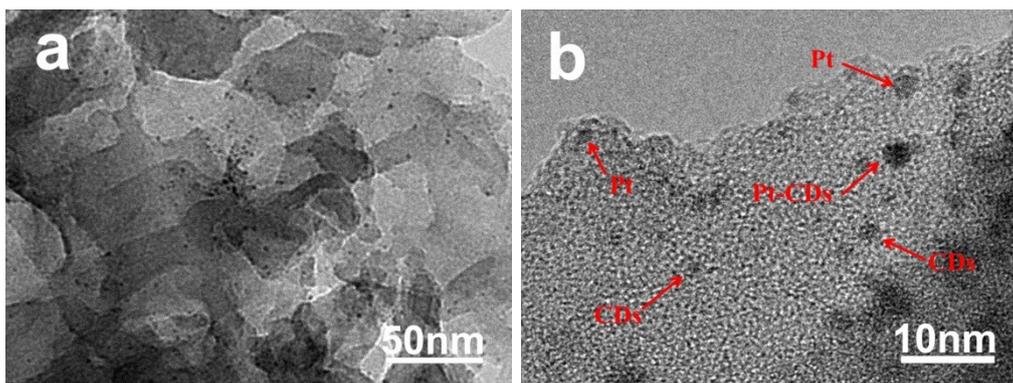


Fig. S2. (a)TEM and (b) HRTEM images of Pt-CDs/g-C₃N₄(100 wt%).

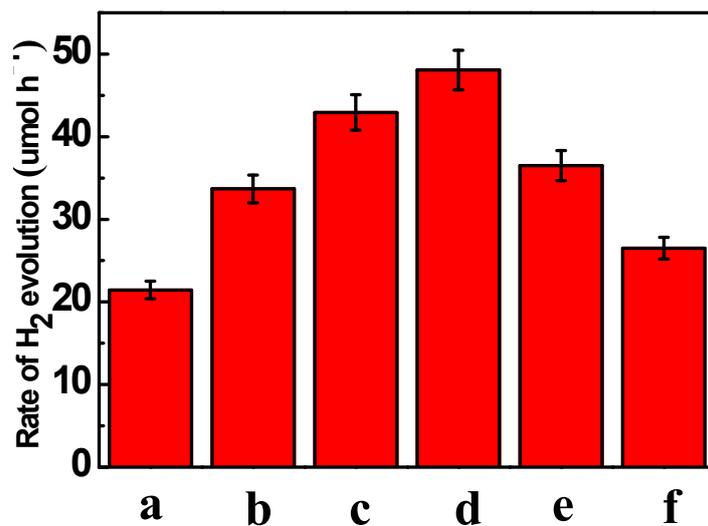


Fig. S3 Photocatalytic H₂-evolution rate of (a) g-C₃N₄, (b) CDs/g-C₃N₄(1 wt%), (c) CDs/g-C₃N₄(5 wt%), (d) CDs/g-C₃N₄(10 wt%), (e) CDs/g-C₃N₄(50 wt%) and (f) CDs/g-C₃N₄(100 wt%) under visible-light irradiation ($\lambda = 420$ nm) with lactic acid as the sacrificial agent and additional Pt cocatalyst.