Supporting Information

Syntheses of asymmetric zinc porphyrins bearing different pseudo-pyridine substituents and their photosensitization for visible-light-driven H_2 production activity

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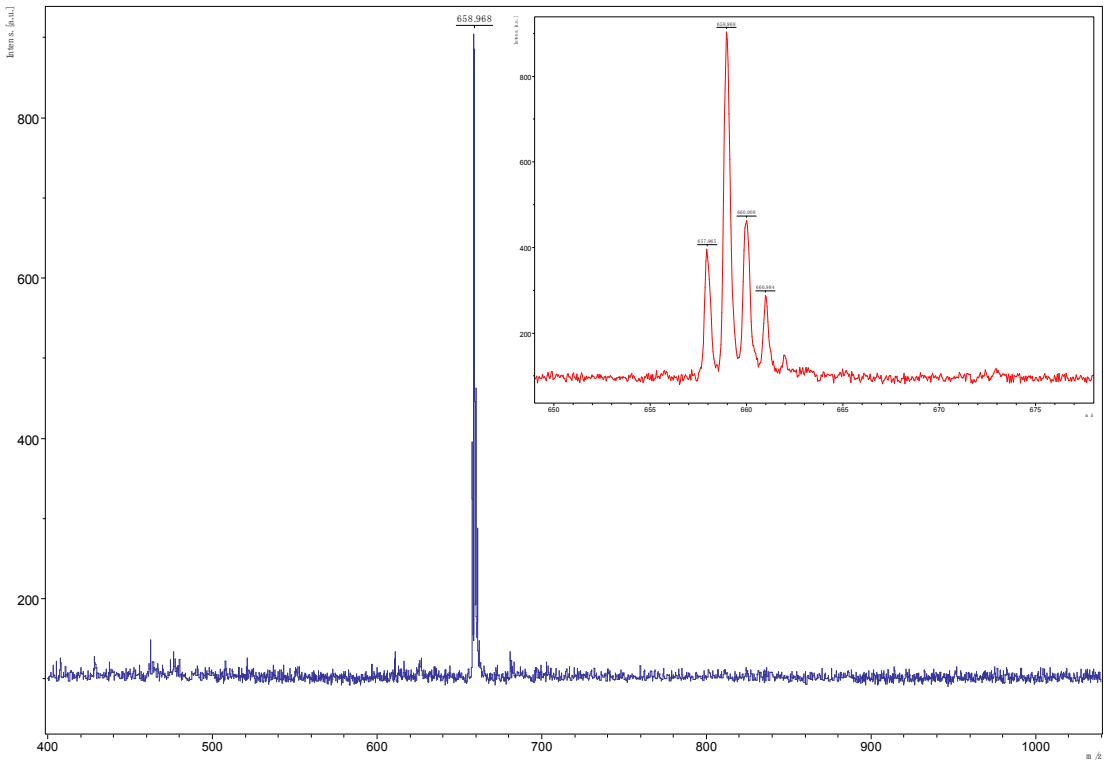

Figure S1. MALDI-TOF mass spectrum of H_2Py-1.

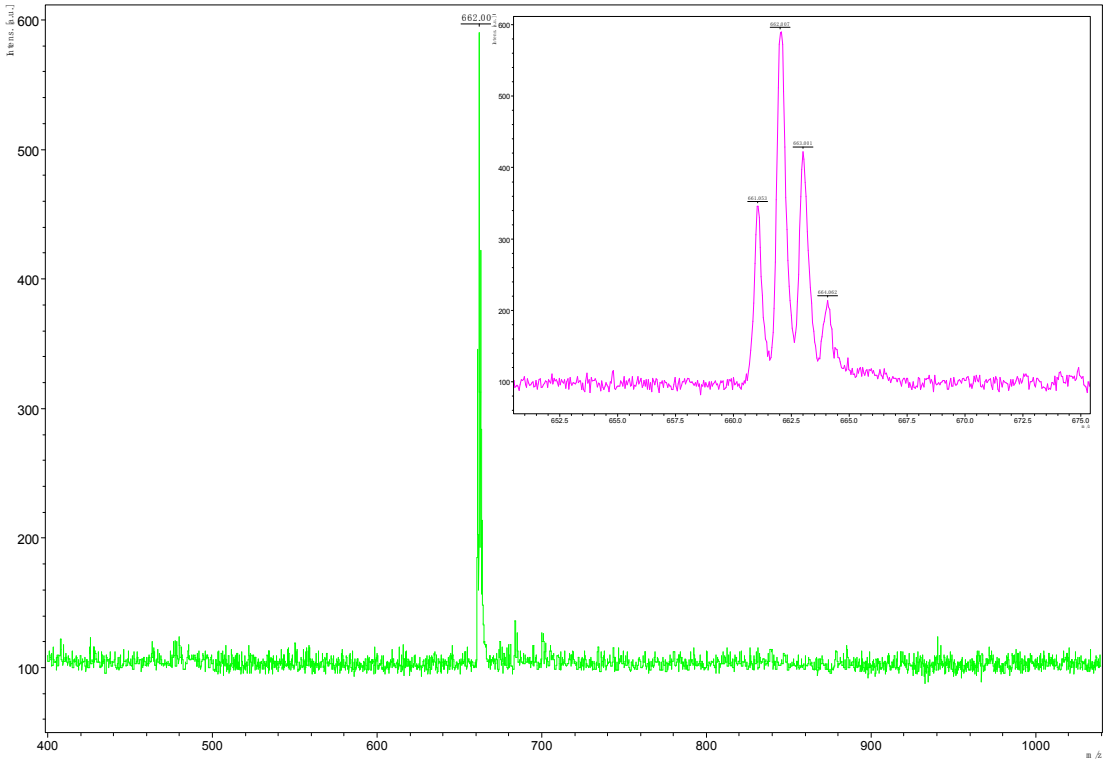

Figure S2. MALDI-TOF mass spectrum of H_2Py-2.

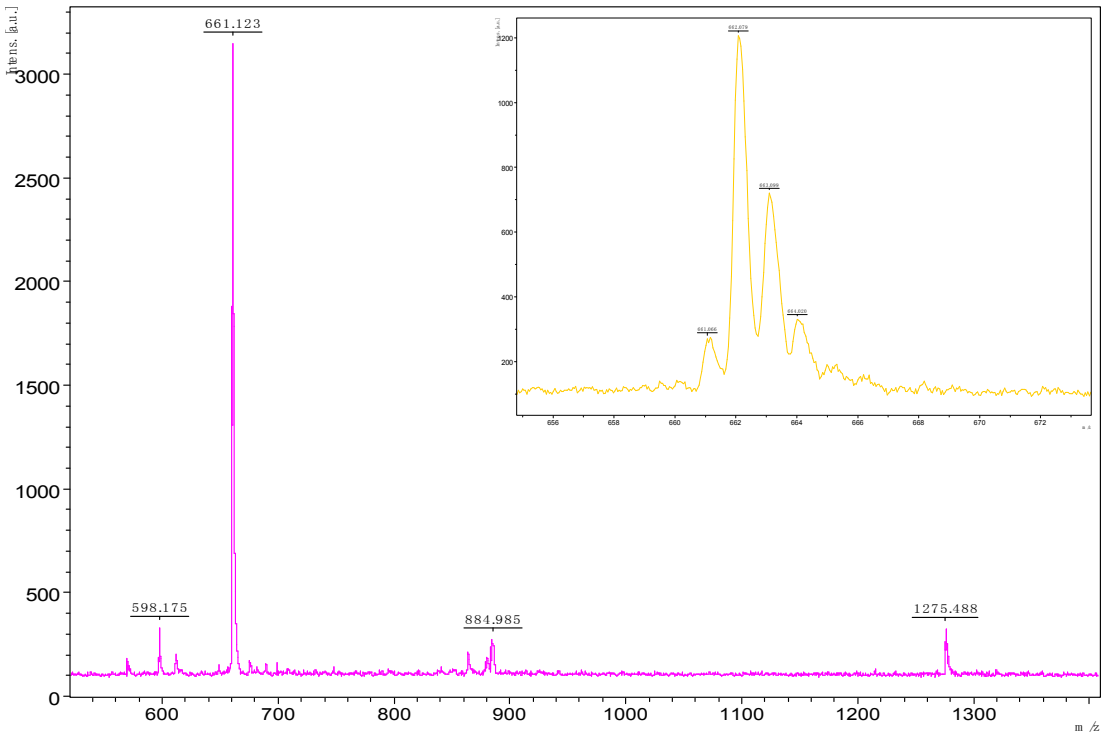

Figure S3. MALDI-TOF mass spectrum of H_2Py-3.

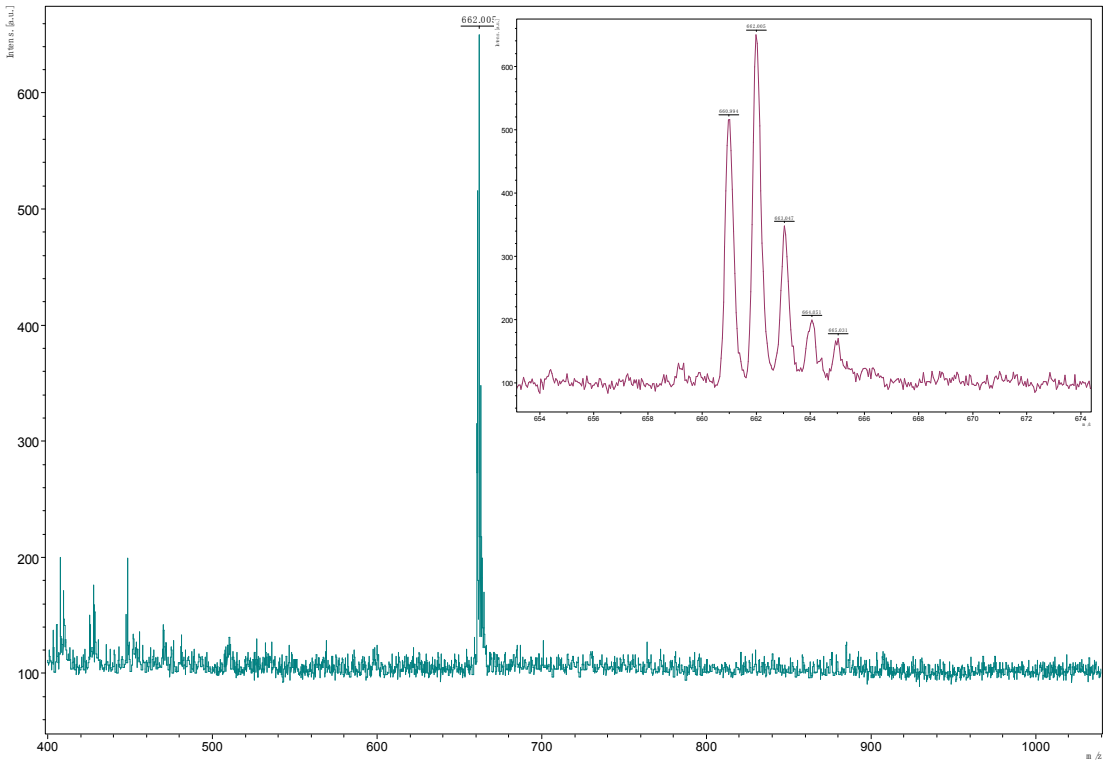

Figure S4. MALDI-TOF mass spectrum of H_2Py-4.

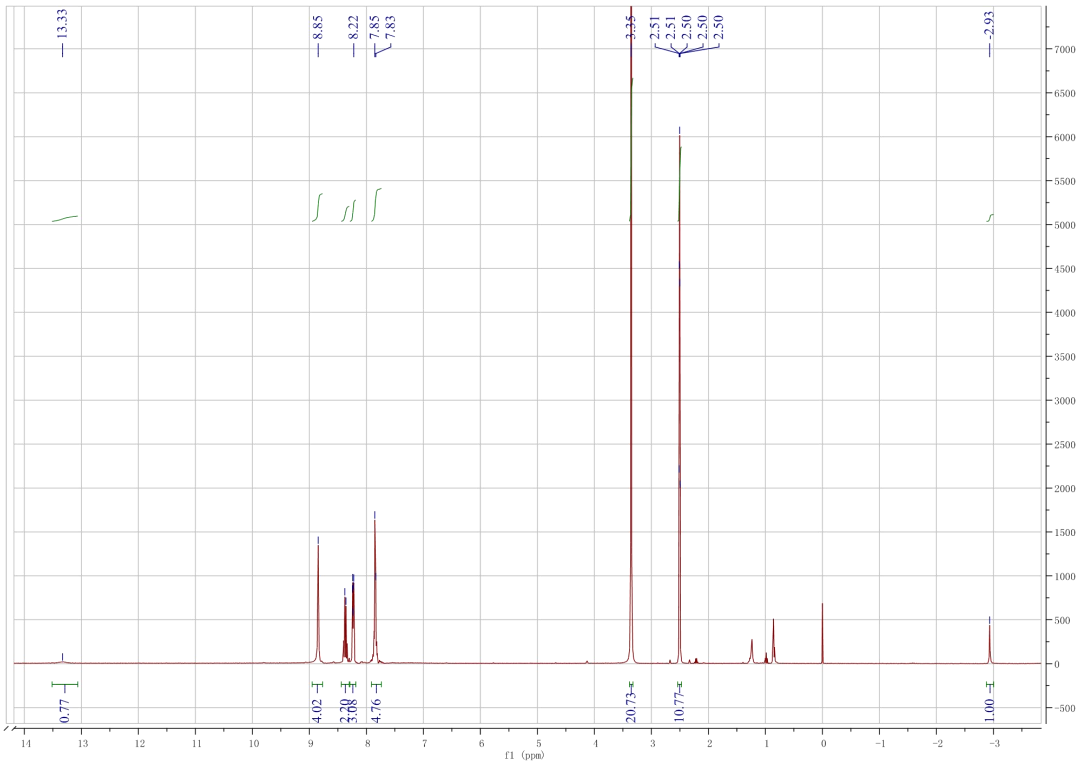

Figure S5. ^{1}H MNR (in DMSO-d_6) spectrum of the molecular ion of H_2Py-1.

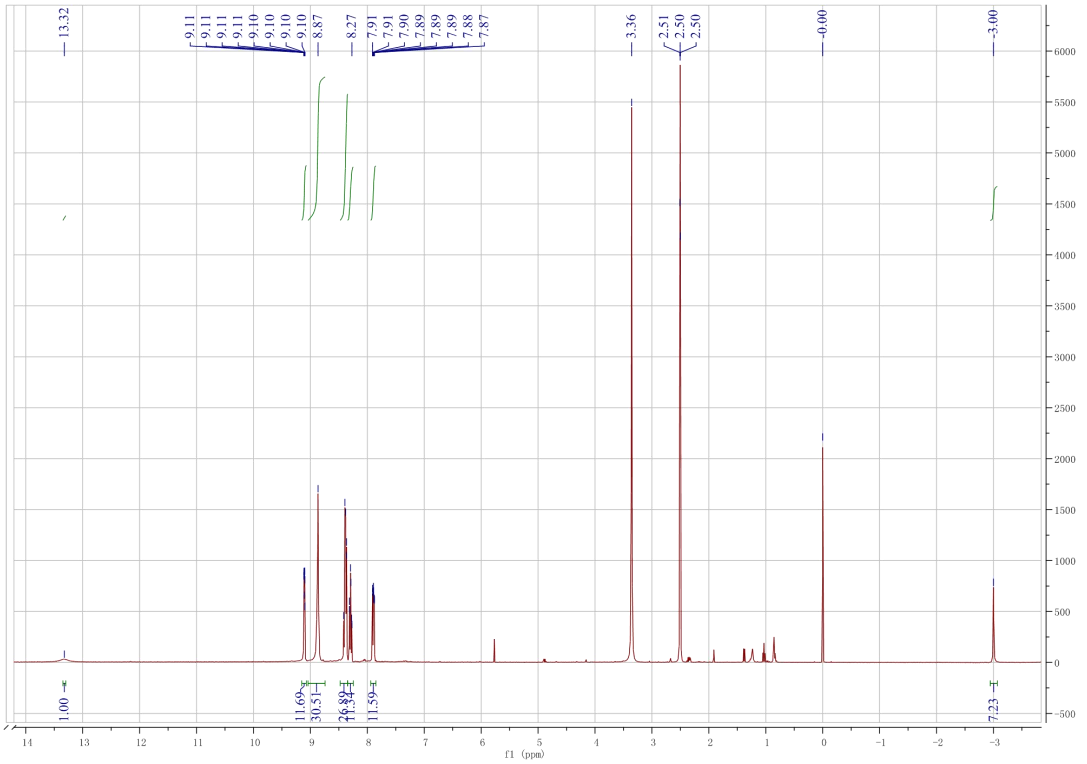

Figure S6. ^{1}H MNR (in DMSO-d_6) spectrum of the molecular ion of H_2Py-2.

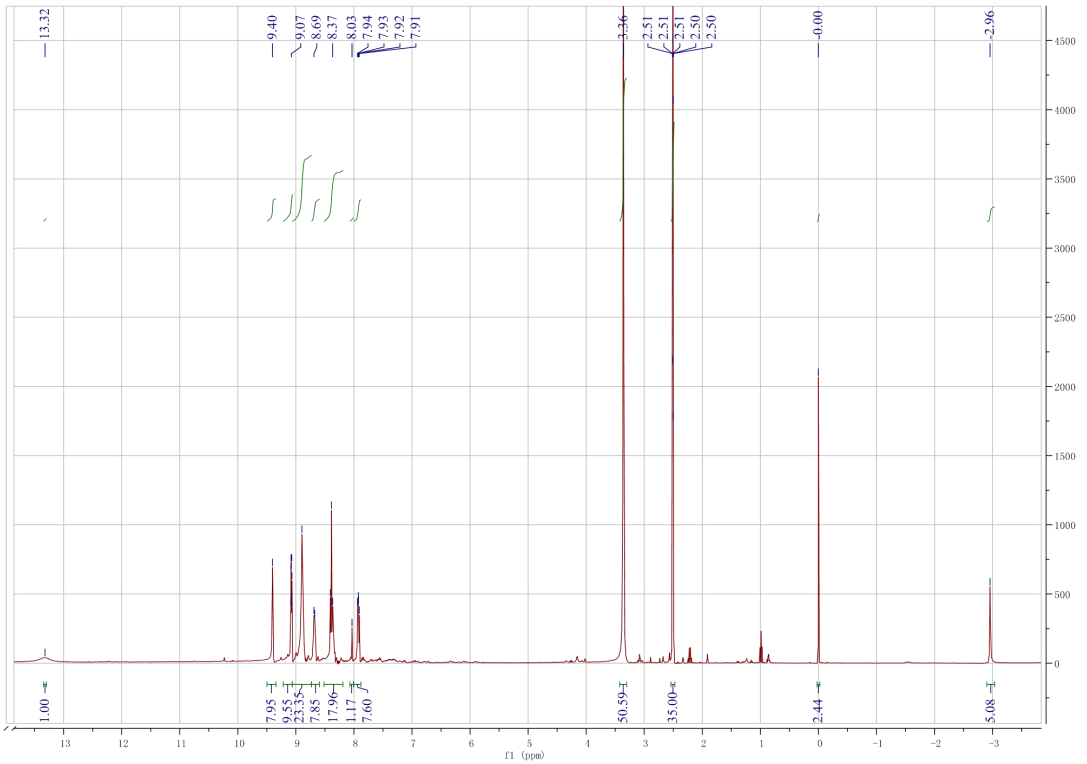

Figure S7. ^{1}H MNR (in DMSO-d$_6$) spectrum of the molecular ion of H_2Py-3.

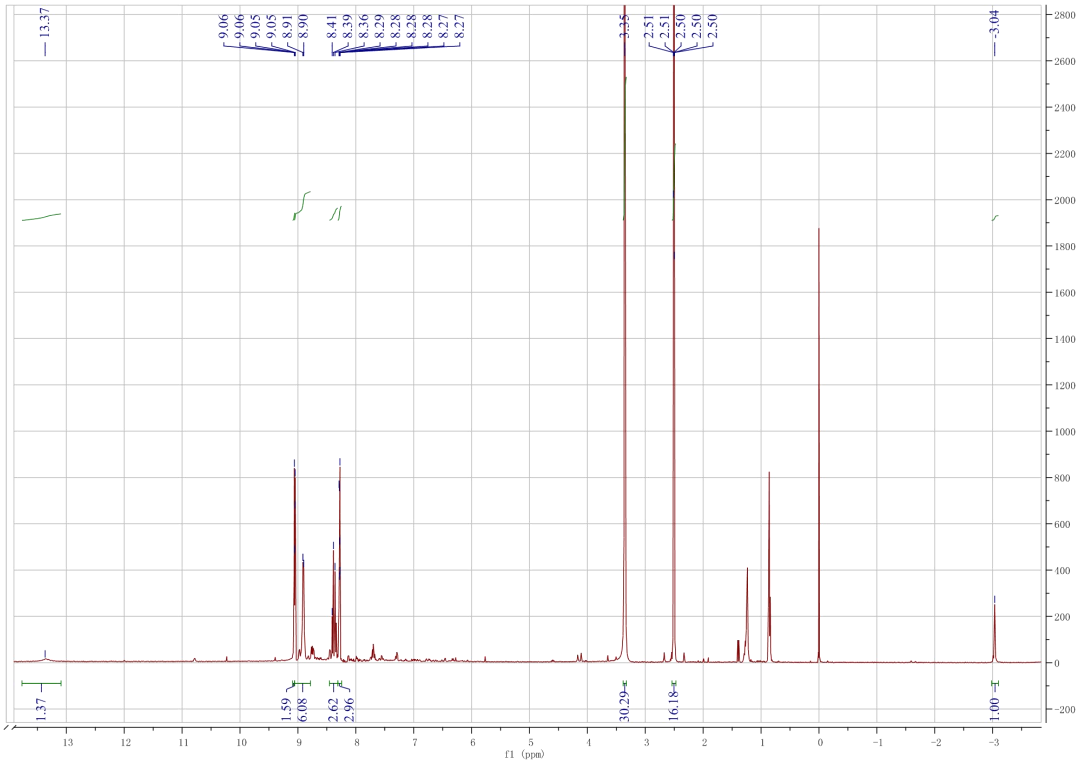

Figure S8. ^{1}H MNR (in DMSO-d$_6$) spectrum of the molecular ion of H_2Py-4.

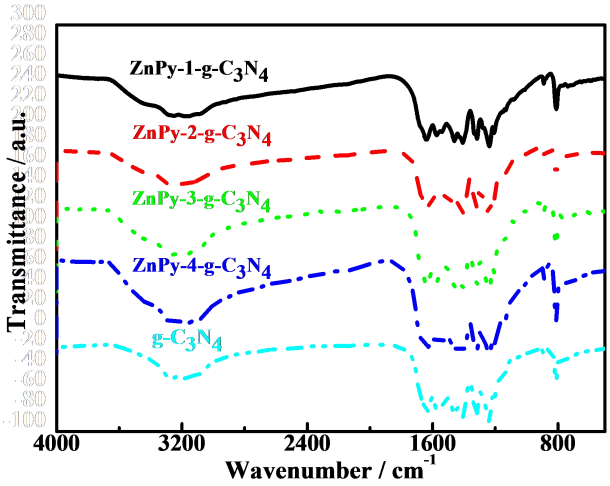

Figure S9. Comparison of the FTIR spectra of g-C_3N_4 and its ZnPy-sensitized products.

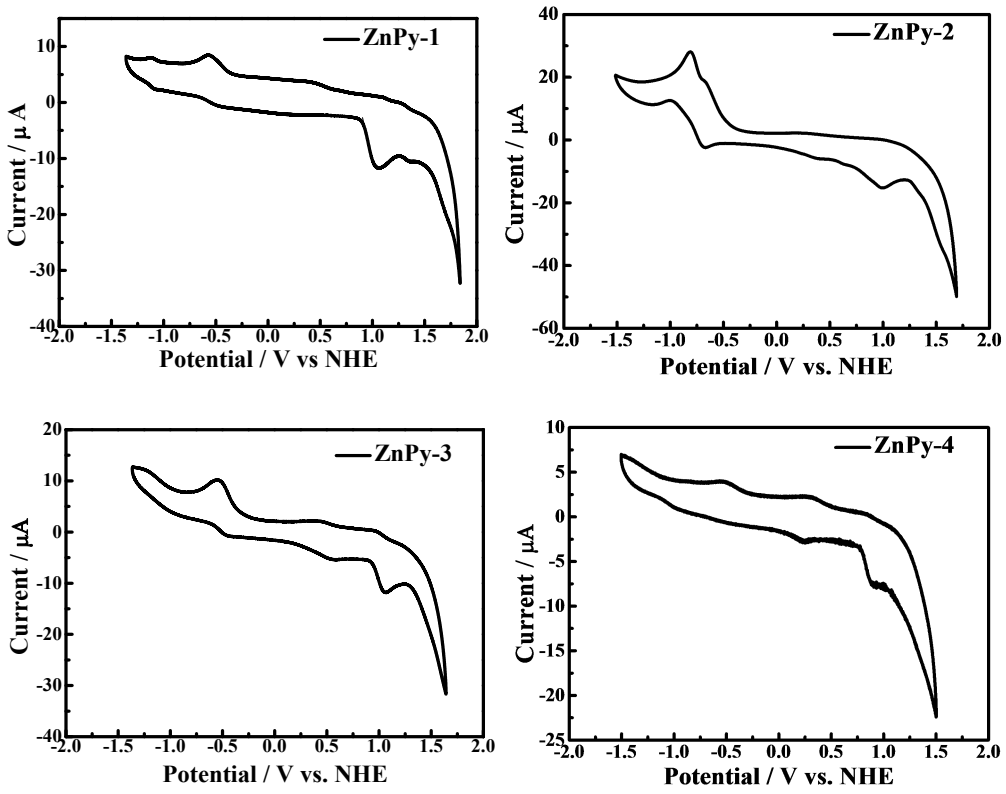

Figure S10. The typical cyclic voltammogram of ZnPys in DMF containing 0.1 M [NBu_4][ClO_4] at a scan rate of 100 $mV \cdot S^{-1}$.

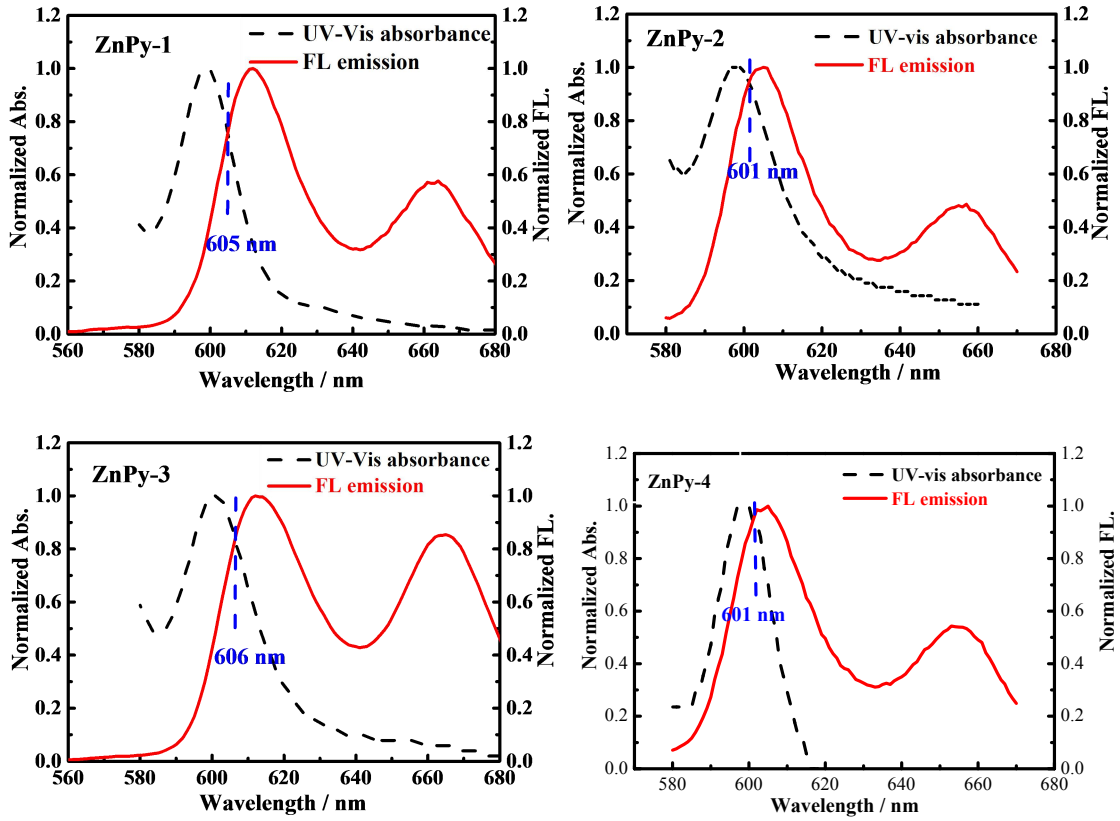

Figure S11. UV-vis absorption and fluorescence emission spectra of ZnPys in DMF solution with excitation fixed at 426 nm.

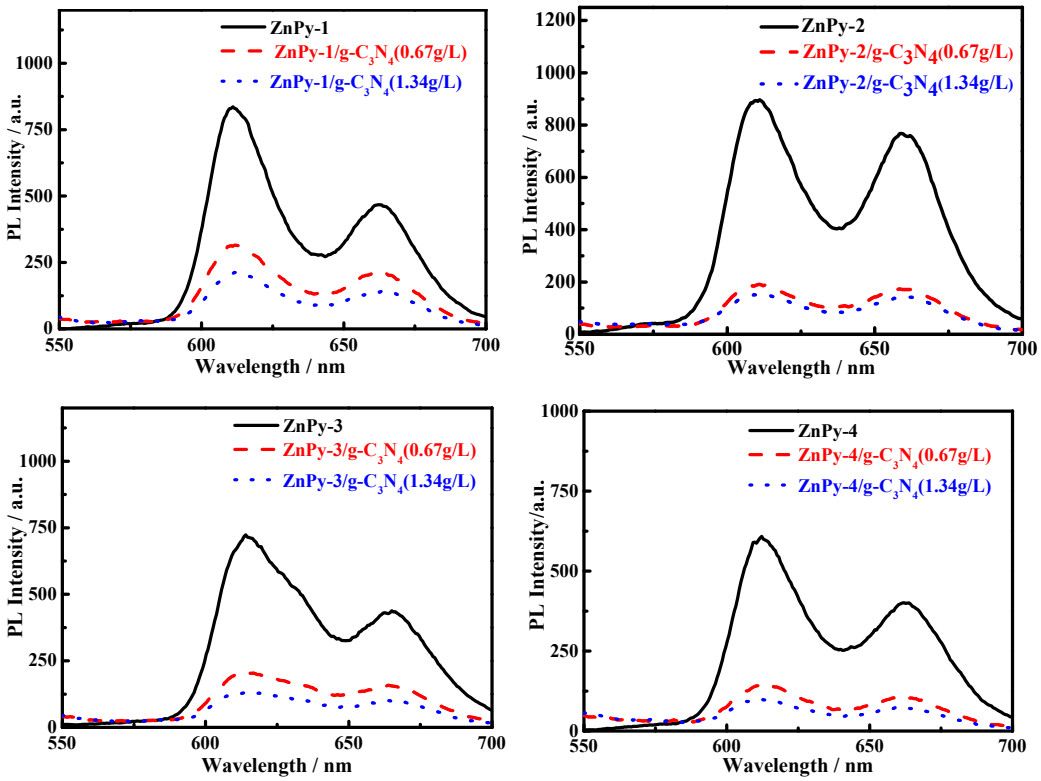

Figure S12. Photoluminescence spectra (excited at 426 nm) of ZnPys and g-C_3N_4/ZnPys dispersion (0.67 g L^{-1} and 1.34 g L^{-1}), DMF was used as the solvent.

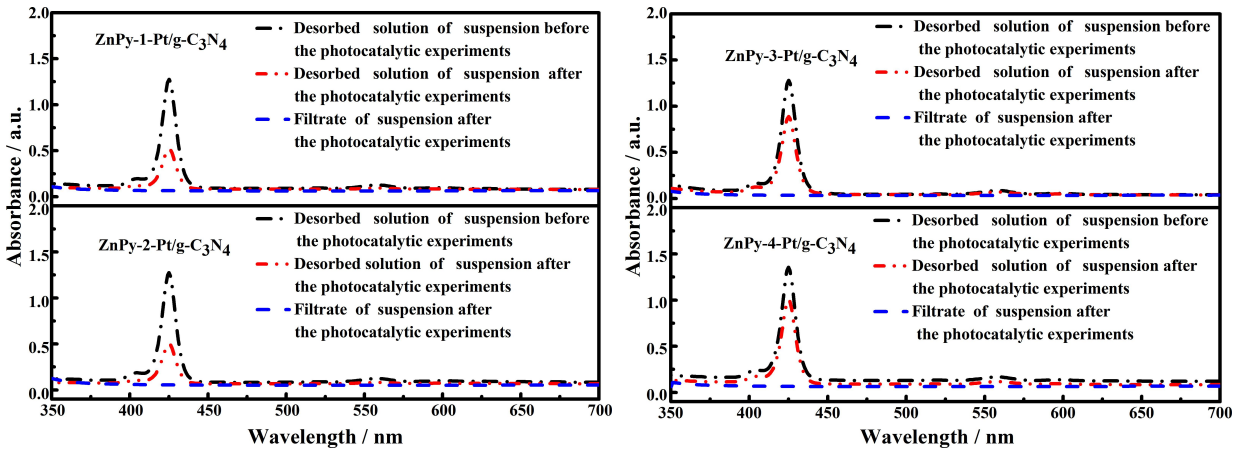

Figure S13. The UV-vis absorption spectra of ZnPys-Pt/g-C_3N_4 before and after the photocatalytic experiments.