## Interconnected hierarchical NiCo<sub>2</sub>O<sub>4</sub> microspheres as high performance

## electrode material for supercapacitor

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**Fig. S1** Typical SEM images of the nickel-cobalt glycolate precursor synthesized with 1 mL deionized water and different amount of IPA and EG: (a) 40 mL IPA and 20 mL EG, (b) 35 mL IPA and 25 mL EG, (c) 30 mL IPA and 20 mL EG, (d) 25 mL IPA and 35 mL EG.



Fig. S2 Typical SEM images of the nickel-cobalt glycolate precursor synthesized with 25 mL IPA and 35 mL EG and different amount of deionized water: (a) 0 mL, (b) 0.5 mL, (c) 1 mL, (d) 1.5 mL.



Fig. S3 TGA curves of the NiCo<sub>2</sub>O<sub>4</sub> precursor.



Fig. S4 SEM images of the obtained  $NiCo_2O_4$  prouducts after calcined at in air at 250 °C (a,b) and 300 °C (c,d).



Fig. S5 XRD pattern (a) and EDS spectrum (b) of the IH-NiCo<sub>2</sub>O<sub>4</sub> sample.



**Fig. S6** (a) XRD pattern of the IH-Co<sub>3</sub>O<sub>4</sub> sample. High resolution XPS spectra for the Co 2p (b) and O 1s (c) of the IH-Co<sub>3</sub>O<sub>4</sub> sample. (d) Typical N<sub>2</sub> adsorption and desorption isotherms and the pore-size-distribution curves (inset) of the IH-Co<sub>3</sub>O<sub>4</sub>.



**Fig. S7** TEM image (a) and corresponding SAED pattern (b) of an individual IH-Co<sub>3</sub>O<sub>4</sub> microsphere. (c) TEM image of the local characteristic region of the IH-Co<sub>3</sub>O<sub>4</sub> microsphere. (d) HRTEM lattice image of the IH-Co<sub>3</sub>O<sub>4</sub> microsphere.



Fig. S8 EIS plots of the  $IH-NiCo_2O_4$  and  $IH-Co_3O_4$  and the equivalent circuit model for the EIS spectra.



Fig. S9 SEM images of the  $IH-NiCo_2O_4$  elecrode before (a) and after (b) 7000 cycles at a high current density of 10 g A<sup>-1</sup>.



**Fig. S10** CV curves at different scan rates (a) and charge-discharge curves at different current densities (b) of the graphene electrode. (c) The corresponding specific capacitances at different current densities of the graphene electrode. (d) Compared CV curves of the IH-NiCo<sub>2</sub>O<sub>4</sub> and graphene at a scan rate of 10 mV s<sup>-1</sup>.

Material	Morphology	Mass loading (mg cm <sup>-2</sup> )	Potential Window (V)	Specific capacitance@curre nt density	Capacity retention (%)
Rambutan-like NiCo <sub>2</sub> O4 <sup>1</sup>	200nm	2.0	0-0.4	798 F g <sup>−1</sup> @0.5 A g <sup>−1</sup>	91.7(1500 cycles@2 A g <sup>-1</sup> )
Urchin like NiCo <sub>2</sub> O4 <sup>2</sup>	B-I Source	>1	0-0.45	658 F g <sup>-1</sup> @1 A g <sup>-1</sup>	98.4(1000 cycles@10 A g⁻¹)
Hollow urchin-like NiCo <sub>2</sub> O <sub>4</sub> microspheres <sup>3</sup>	(a)	2.5-3.5	0-0.5	950 F g <sup>-1</sup> @1 A g <sup>-1</sup>	93.6(1500 cycles@6 A g <sup>-1</sup> )
Hollow NiCo <sub>2</sub> O <sub>4</sub> sub-microspheres <sup>4</sup>	200 nm	8	0-0.4	678 F g <sup>-1</sup> @1 A g <sup>-1</sup>	87(3500 cycles@10 A g <sup>-1</sup> )
Flower-like NiCo <sub>2</sub> O <sub>4</sub> hierarchitectures <sup>5</sup>	d 10 <u>0</u> nm		0-0.55	1191.2 F g <sup>-1</sup> @1 A g <sup>-</sup> 1	78(1200 cycles@1 A g <sup>-1</sup> )
Spinel NiCo <sub>2</sub> O <sub>4</sub> nanostructure <sup>6</sup>	100 pm	1	0-0.4	1362 F g <sup>-1</sup> @1 A g <sup>-1</sup>	41(1500 cycles@4 A g <sup>-1</sup> )
NiCo <sub>2</sub> O <sub>4</sub> nanoflake composites <sup>7</sup>		2	0-0.4	1468 F g <sup>-1</sup> @4 A g <sup>-1</sup>	85.5(5000 cycles@4 A g <sup>-1</sup> )

**Table S1** A comparison with previously reported  $NiCo_2O_4$  nanomaterials.

Nanowires buliding NiCo <sub>2</sub> O <sub>4</sub> architectures <sup>8</sup>	b 50 <u>0 nm</u>	1.3	0-0.5	1080 F g <sup>-1</sup> @2 A g <sup>-1</sup>	86.5(1500 cycles@4 A g <sup>-1</sup> )
Nanosheets buliding NiCo <sub>2</sub> O <sub>4</sub> architectures <sup>8</sup>		1.3	0-0.5	1400 F g <sup>-1</sup> @5 A g <sup>-1</sup>	94(1500 cycles@4 A g <sup>-1</sup> )
Nanorod- assembled NiCo <sub>2</sub> O <sub>4</sub> hollow microspheres <sup>9</sup>	(с) <u>- 1µт</u>		0-0.5	764 F g <sup>-1</sup> @2 A g <sup>-1</sup>	101.7(1500 cycles@2 A g <sup>-1</sup> )
NiCo <sub>2</sub> O <sub>4</sub> nanosheets <sup>10</sup>			0-0.45	876 F g <sup>-1</sup> @1 A g <sup>-1</sup>	88(1000 cycles@1 A g <sup>-1</sup> )
Hierarchical mesoporous NiCo <sub>2</sub> O <sub>4</sub> hollow nanocubes <sup>11</sup>	IOnm	1	0-0.45	795.6 F g⁻¹@1 A g⁻¹	97.5(2000 cycles@1 A g <sup>-1</sup> ) 96.1(2000 cycles@2 A g <sup>-1</sup> )
Urchin like NiCo <sub>2</sub> O4 <sup>12</sup>	(i) 1 µm	2.1	0-0.55	436.1 F g⁻¹@1 A g⁻¹	No decay(1000 cycles@10 A g <sup>-1</sup> )
Mesoporous NiCo <sub>2</sub> O <sub>4</sub> nanospheres <sup>13</sup>	(b) 100 nm	2	0-0.4	842 F g <sup>-1</sup> @2 A g <sup>-1</sup>	107(1000 cycles@30 mV s <sup>-</sup> <sup>1</sup> )
Multiple hierarchical NiCo <sub>2</sub> O <sub>4</sub> <sup>14</sup>		5.0-6.0	0-0.45	1393 F g <sup>-1</sup> @ 0.5 A g <sup>-1</sup>	

Hierarchical NiCo <sub>2</sub> O <sub>4</sub> tetragonal microtubes <sup>15</sup>		1.0	0-0.53	1387.9 F g⁻¹@2 A g⁻ 1	89.4(12000 cycles@10 A g <sup>−1</sup> )
Hollow NiCo <sub>2</sub> O <sub>4</sub> submicrospheres <sup>16</sup>			0-0.5	987 F g <sup>-1</sup> @1 A g <sup>-1</sup>	No decay(5000 cycles@5 A g⁻¹)
NiCo₂O₄ double- shell hollow spheres <sup>17</sup>	ŝ	3.76	0-0.4	781 F g <sup>-1</sup> @1 A g <sup>-1</sup>	85.8(2000 cycles@2 A g <sup>-1</sup> )
Mesoporous NiCo <sub>2</sub> O <sub>4</sub> nanosheets <sup>18</sup>	A Contraction of the second se	1	0-0.35	292 F g <sup>-1</sup> @1 A g <sup>-1</sup>	90(2000 cycles@8 A g <sup>-1</sup> )
3D network-like mesoporous NiCo <sub>2</sub> O4 <sup>19</sup>	(d) 	1.0	0-0.5	931 F g <sup>-1</sup> @3 A g <sup>-1</sup>	125.5(1000 cycles@3 A g <sup>-1</sup> )
IH-NiCo₂O₄ (This work)	e 5 <u>00 nm</u>	2.1	0-0.55	1822.3 F g⁻¹ @2 A g⁻¹	87.6(7000 cycles@10 A g <sup>−1</sup> )

## References

- 1. Y. Shang, Y. Gai, L. Wang, L. Hao, H. Lv, F. Dong and L. Gong, *Eur. J. Inorg.c Chem.*, 2017, **2017**, 2340.
- 2. J. Xiao and S. Yang, *RSC Adv.*, 2011, **1**, 588.
- 3. Y. Lei, Y. Wang, W. Yang, H. Yuan and D. Xiao, *RSC Adv.*, 2015, **5**, 7575.
- 4. C. Yuan, J. Li, L. Hou, J. Lin, G. Pang, L. Zhang, L. Lian and X. Zhang, *RSC Adv.*, 2013, **3**, 18573.
- 5. C. An, Y. Wang, Y. Huang, Y. Xu, C. Xu, L. Jiao and H. Yuan, *CrystEngComm*, 2014, **16**, 385.
- 6. S. Jiang, Y. Sun, H. Dai, P. Ni, W. Lu, Y. Wang, Z. Li and Z. Li, *Electrochim. Acta*, 2016, **191**, 364.
- 7. Y. Sun, X. Xiao, P. Ni, Y. Shi, H. Dai, J. Hu, Y. Wang, Z. Li and Z. Li, *Electrochim. Acta*, 2014, **121**,

270.

- 8. L. Yu, H. Wu, T. Wu and C. Yuan, *RSC Adv.*, 2013, **3**, 23709.
- 9. Y. Zhu, X. Ji, R. Yin, Z. Hu, X. Qiu, Z. Wu and Y. Liu, *RSC Adv.*, 2017, **7**, 11123.
- 10. L. Zhang, W. Zheng, H. Jiu, C. Ni, J. Chang and G. Qi, *Electrochim. Acta.*, 2016, **215**, 212.
- 11. C. Zheng, C. Cao, R. Chang, J. Hou and H. Zhai, *Phys. Chem. Chem. Phys.*, 2016, **18**, 6268.
- 12. J. Wang, Y. Zhang, J. Ye, H. Wei, J. Hao, J. Mu, S. Zhao and S. Hussain, *RSC Adv.*, 2016, **6**, 70077.
- 13. M. J. Pang, S. Jiang, G. H. Long, Y. Ji, W. Han, B. Wang, X. L. Liu, Y. L. Xi, F. Z. Xu and G. D. Wei, *RSC Adv.*, 2016, **6**, 67839.
- 14. S. Wang, S. Sun, S. Li, F. Gong, Y. Li, Q. Wu, P. Song, S. Fang and P. Wang, *Dalton T.*, 2016, **45**, 7469.
- 15. F.-X. Ma, L. Yu, C.-Y. Xu and X. W. Lou, *Energy Environ. Sci.*, 2016, **9**, 862.
- 16. Y. Zhu, J. Wang, Z. Wu, M. Jing, H. Hou, X. Jia and X. Ji, *J. Power Sources*, 2015, **287**, 307.
- 17. X. Li, L. Jiang, C. Zhou, J. Liu and H. Zeng, *NPG Asia Mater.*, 2015, **7**, e165.
- S. Khalid, C. Cao, A. Ahmad, L. Wang, M. Tanveer, I. Aslam, M. Tahir, F. Idrees and Y. Zhu, *RSC Adv.*, 2015, 5, 33146.
- Y. Zhu, Z. Wu, M. Jing, W. Song, H. Hou, X. Yang, Q. Chen and X. Ji, *Electrochim. Acta*, 2014, 149, 144.