Supporting Information

Ligand sensitized strong luminescence from Eu³⁺-doped LiYF₄ nanocrystals: A photon downshifting strategy to increase solar to current conversion efficiency

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Fig. S1 XRD patterns of bare Eu^{3+} doped $LiYF_4$ nanocrystals along with standard pattern.



Fig. S2. Overlap between the emission spectrum of the TPB and the excitation spectrum of the Eu^{3+} -doped LiYF₄ NCs.



Fig. S3. The lifetime of TPB capped Eu³⁺-doped LiYF₄ NCs, by exciting the nanocrystals at $\lambda_{ex} = 394$ nm and $\lambda_{ex} = 365$ nm and monitoring the emission at $\lambda_{em} = 615$ nm.



Fig. S4. The lifetime of (A) bare Eu³⁺-doped LiYF₄ NCs and (B) Eu³⁺-TPB complex by exciting the nanocrystals at $\lambda_{ex} = 394$ nm and monitoring the emission at $\lambda_{em} = 615$ nm.



Fig. S5. The absorbance spectra of (A) TPB capped Eu³⁺-doped LiYF₄ NCs in water (0.1 wt%). (B) TPB ligands (1 X 10⁻⁵ M).



Fig. S6. AFM image of the EVA film of TPB capped Eu^{3+} -doped $LiYF_4$ nanocrystals (A) and thickness of the film (B)

Quantum Yield Calculation

The quantum yield was determined by comparing the luminescence with quinine-sulphate. The quantum yield of TPB capped Eu³⁺-doped LiYF₄NCs and bare Eu³⁺-doped LiYF₄ NCs were calculated from the following equation- $Q_{sample} = Q_{ref} (A/A_{ref}) (I_{ref}/I) (n^2 /n^2_{ref})$ where, Q_{sample} and Q_{ref} are the quantum yields of the nanocrystals and quinine-sulphate respectively, A is the absorbance, I is the integrated area of photoluminescence spectra, and n is the refractive index of the solution. The quantum yield of Quinine sulphate as the reference is 0.546. The quantum yield ofEu³⁺ doped TPB capped and bare NCs were estimated by comparing the integrated emission spectra of the aqueous dispersion with that of Quinine sulphate solution. The sample and the reference have the identical optical density at the excitation wavelength. The calculated quantum yield was about 5% for Eu³⁺-doped LiYF₄NCs and 31 % TPB capped Eu³⁺-doped LiYF₄NCs.