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Supporting Information for

Isomeric Structure of Pentacoordinate Chiral Spirophosphoranes in Solution by combined use of NMR Experiments and GIAO DFT calculations of NMR parameters

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Experimental Section

NMR Spectroscopy. All NMR experiments were performed with a 600, 500 and 400 MHz (600.1, 500.1 and 400.1 MHz for ¹H NMR; 150.9, 125.8 and 100.6 MHz for ¹³C NMR; 242.9, 202.5 and 162.0 MHz for ³¹P NMR, 60.81 MHz for ¹⁵N NMR, 376.5 for ¹⁹F respectively) spectrometers equipped with a 5 mm diameter probehead and a pulsed gradient unit capable of producing magnetic field pulse gradients in the z-direction of 53.5 G·cm⁻¹. For ¹H-¹³C correlations HSQC experiment optimized for J = 145 Hz. For ¹H-¹³C long range correlations HMBC experiment optimized for J = 8 Hz. For ¹H-¹³C long range correlations HSQC experiment optimized for J = 8 Hz. For ¹H-¹⁵N correlations HMBC experiment optimized for J = 6 Hz. For ¹H-¹⁵N correlations HSQC experiment optimized for J = 6 Hz. For ¹H-¹⁵N correlations HETCOR experiment optimized for J = 3 Hz. DOSY experiments were performed with ledbpgp2s,¹ using a stimulated echo sequence and two spoil gradients. NOE experiments were performed with 1D DPFGNOE techniques.² CS's (δ in ppm) were referenced to the solvent CDCl₃ ($\delta = 7.27$ ppm for ¹H and 77.0 ppm for ¹³C NMR) and to external H₃PO₄ (0.0 ppm) for ³¹P NMR spectra, to external C₆F₆ (-164.9 ppm) for ¹⁹F NMR spectra.

¹W. S. Price, Concepts Magn. Reson., 1997, 9, 299; W. S. Price, Concepts Magn. Reson., 1998, 10, 197; C. S. Johnson, Prog. Nucl. Magn. Reson. Spectrosc., 1999, 34, 203.

² J. Stonehouse, P. Adell, J. Keeler and A.J. Shaka, J. Am. Chem. Soc., 1994, 116, 6037.



Figure S1. ¹H (a), ³¹P{¹H} (b), ³¹P (c), ¹³C{¹H} (d) and ¹³C DEPT (e) spectra of **4a** in CDCl₃ at T = 303 K.



Figure S2. ¹H-¹H COSY spectra of **4a** in CDCl₃ at T = 303 K.



Figure S3. ^{1}H - ^{13}C HSQC spectra of **4a** in CDCl₃ at T = 303 K.



Figure S4. ¹H-¹³C HMBC spectra of 4a in CDCl₃ at T = 303 K.



Figure S5. ¹H-³¹P HMBC spectra of **4a** in CDCl₃ at T = 303 K.



Figure S6. ^{1}H - ^{15}N HSQC spectra of **4a** in CDCl₃ at T = 303 K.



Figure S7. ¹H-¹⁵N HMBC spectra of **4a** in CDCl₃ at T = 303 K.



Figure S8. ¹H (a) and 1D NOESY (b-h) spectra of 4a in CDCl₃ at T = 303 K.



Figure S9. ¹H (a), ³¹P{¹H} (b) spectra of **4a** after 24h in CDCl₃ at T = 303 K (**4a** 80 %, **4c** 20 %).



Figure S10. ¹H-³¹P HMBC spectra of 4a after 24h in CDCl₃ at T = 303 K (4a 80 %, 4c 20 %).



Figure S11. ¹H-¹⁵N HSQC spectra of **4a** after 24h in CDCl₃ at T = 303 K (**4a** 80 %, **4c** 20 %).



Figure S12. ¹H (a), ³¹P{¹H} (b), ³¹P (c), ¹³C{¹H} (d) and ¹³C DEPT (e) spectra of **4b** in CDCl₃ at T = 303 K.



Figure S13. ¹H-¹H COSY spectra of **4b** in CDCl₃ at T = 303 K.



Figure S14. ¹H-¹³C HSQC spectra of **4b** in CDCl₃ at T = 303 K.



Figure S15. ¹H-¹³C HMBC spectra of **4b** in CDCl₃ at T = 303 K.



Figure S16. ¹H-³¹P HMBC spectra of **4b** in CDCl₃ at T = 303 K.



Figure S17. ¹H-¹⁵N HSQC spectra of **4b** in CDCl₃ at T = 303 K.



Figure S18. ¹H (a) and 1D NOESY (b-d) spectra of **4b** in CDCl₃ at T = 303 K.



Figure S19. ¹H (a), ³¹P{¹H} (b), ³¹P (c), ¹³C{¹H} (d) and ¹³C DEPT (e) spectra of **5a** in CDCl₃ at T = 303 K.



Figure S20. ¹H-¹H COSY spectra of **5a** in CDCl₃ at T = 303 K.



Figure S21. ¹H-¹³C HSQC spectra of 5a in CDCl₃ at T = 303 K.



Figure S22. ¹H-¹³C HMBC spectra of 5a in CDCl₃ at T = 303 K.



Figure S23. ¹H-³¹P HMBC spectra of **5a** in CDCl₃ at T = 303 K.

Figure S24. ¹H-¹⁵N HSQC spectra of **5a** in CDCl₃ at T = 303 K.

Figure S25. ¹H-¹⁵N HMBC spectra of **5a** in CDCl₃ at T = 303 K.

Figure S26. ¹H (a) and 1D NOESY (b-f) spectra of 5a in CDCl₃ at T = 303 K.

Figure S27. ¹H (a), ¹H{¹⁹F} (b) and ¹⁹F{¹H} (c) spectra of 5a in CDCl₃ at T = 303 K.

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Figure S29. ${}^{1}H$ (a), ${}^{31}P{}^{1}H$ (b), ${}^{13}C{}^{1}H$ (c) and ${}^{13}C$ DEPT (d) spectra of 5a after 5 days in CDCl₃ at T = 303 K (5a 87 %, 5c 13 %).

Figure S30. ¹H-¹H COSY spectra of **5a** after 5 days in CDCl₃ at T = 303 K (**5a** 87 %, **5c** 13 %).

Figure S31. ¹H-¹³C HSQC spectra of 5a after 5 days in CDCl₃ at T = 303 K (5a 87 %, 5c 13 %).

Figure S32. ¹H-¹³C HMBC spectra of **5a** after 5 days in CDCl₃ at T = 303 K (**5a** 87 %, **5c** 13 %).

Figure S33. ¹H-³¹P HMBC spectra of **5a** after 5 days in CDCl₃ at T = 303 K (**5a** 87 %, **5c** 13 %).

Figure S34. ¹H-¹⁵N HSQC spectra of 5a after 5 days in CDCl₃ at T = 303 K (5a 87 %, 5c 13 %).

Figure S35. 2D DOSY spectra of 5a after 5 days in CDCl₃ at T = 303 K (5a 87 %, 5c 13 %).

Figure S36. ¹H (a), ³¹P{¹H} (b), ³¹P (c), ¹³C{¹H} (d) and ¹³C DEPT (e) spectra of **5b** in CDCl₃ at T = 303 K.

Figure S37. ¹H-¹H COSY spectra of **5b** in CDCl₃ at T = 303 K.

Figure S38. ¹H-¹³C HSQC spectra of **5b** in CDCl₃ at T = 303 K.

Figure S39. ¹H-¹³C HMBC spectra of **5b** in CDCl₃ at T = 303 K.

Figure S40. ¹H-³¹P HMBC spectra of **5b** in CDCl₃ at T = 303 K.

Figure S41. ¹H-¹⁵N HSQC spectra of **5b** in CDCl₃ at T = 303 K.

Figure S42. ¹H (a) and 1D NOESY (b-h) spectra of **5b** in CDCl₃ at T = 303 K.

Figure S43. ¹H (a), ¹H $\{^{19}F\}$ (b) and ¹⁹F $\{^{1}H\}$ (c) spectra of **5b** in CDCl₃ at T = 303 K.

Figure S44. ¹⁹F-¹H HETCOR spectra of **5b** in CDCl₃ at T = 303 K.

Figure S45. ¹H (a), ³¹P{¹H} (b), ³¹P (c), ¹³C{¹H} (d) and ¹³C DEPT (e) spectra of **6a** in CDCl₃ at T = 303 K.

Figure S46. ¹H-¹H COSY spectra of **6a** in CDCl₃ at T = 303 K.

Figure S47. ¹H-¹³C HSQC spectra of **6a** in CDCl₃ at T = 303 K.

Figure S48. ¹H-¹³C HMBC spectra of **6a** in CDCl₃ at T = 303 K.

Figure S49. ¹H-³¹P HMBC spectra of **6a** in CDCl₃ at T = 303 K.

Figure S50. ¹H-¹⁵N HSQC spectra of **6a** in CDCl₃ at T = 303 K.

Figure S51. ¹H-¹⁵N HMBC spectra of **6a** in CDCl₃ at T = 303 K.

Figure S52. ¹H (a), ³¹P{¹H} (b) spectra of **6a** after 24h in CDCl₃ at T = 303 K (**6a** 86 %, **6c** 14 %).

Figure S53. ^{1}H - ^{31}P HMBC spectra of 6a in CDCl₃ at T = 303 K after 24h in CDCl₃ at T = 303 K (6a 86 %, 6c 14 %).

Figure S54. ¹H (a), ³¹P{¹H} (b), ¹³C{¹H} (c) and ¹³C DEPT (d) spectra of **6b** in CDCl₃ at T = 303 K.

Figure S55. ¹H-¹H COSY spectra of **6b** in CDCl₃ at T = 303 K.

Figure S56. ¹H-¹³C HSQC spectra of **6b** in CDCl₃ at T = 303 K.

Figure S57. ¹H-¹³C HMBC spectra of **6b** in CDCl₃ at T = 303 K.

Figure S58. ¹H-³¹P HMBC spectra of **6b** in CDCl₃ at T = 303 K.

Figure S59. ¹H-¹⁵N HSQC spectra of **6b** in CDCl₃ at T = 303 K.

Figure S60. ¹H (a) and 1D NOESY (b-g) spectra of **6b** in CDCl₃ at T = 303 K.

Calculations

The quantum chemical calculations were performed using Gaussian 03. Full geometry optimizations have been carried out within the framework of DFT (PBE1PBE) method using 6-31+G(d) basis sets. Chemical shifts (CSs) and SSC were calculated by the GIAO method at the PBE1PBE/6-311G(2d,2p) level of theory. ³¹P CSs were referred to H₃PO₄ and linear scaling procedure was applied ($\delta_{scaled} = (\delta_{unscaled} - intercept)/slope$, where intepcept = -14.4 ppm, slope = 1.073).³ ¹H CSs were referred to TMS.

³ Sh. K. Latypov, F. M. Polyancev, D. G. Yakhvarov and O. G. Sinyashin, Phys. Chem. Chem. Phys., 2015, 17, 6976.

 $^{3}J_{\mathrm{PH4}}$ $\delta H_{17,19}$ Isomer ΔΕ $\delta^{31}P$ δH_6 $\delta H_{16,20}$ δH_4 δH_5 $^{1}J_{\mathrm{PH}}$ -76.0 7.64 7.59 7.42 5.95 3.78 716.4 36.2 0 Ι 7.43, 7.87 7.23, 7.59 Π 0.4 -77.4 7.76 5.80, 5.84 3.49, 3.84 739.2 27.8, 34.9 2.4 -77.5 7.91 7.71 7.53 5.78 3.57 759.3 25.5 III

Table S1. Calculated NMR parameters for compound 4 with inclusion of solvent effects in frame of PCM^a.

^a CS's calculated in frame of PCM (chloroform), chemical shifts in ppm, spin-spin couplings in Hz, relative energy, in kcal/mol.